

# **Integrated socio-hydrogeological approach to tackle nitrate contamination in groundwater resources. The case of Grombalia Basin (Tunisia).**

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## **Abstract**

Nitrate contamination still remains one of the main groundwater quality issues in several aquifers worldwide, despite the perduring efforts of the international scientific community to effectively tackle this problem. The classical hydrogeological and isotopic investigations are obviously of paramount importance for the characterization of contaminant sources, but are clearly not sufficient for the correct and long-term protection of groundwater resources. This paper aims at demonstrating the effectiveness of the socio-hydrogeological approach as the best tool to tackle groundwater quality issues, while contributing bridging the gap between science and society. An integrated survey, including land use, hydrochemical (physicochemical parameters and major ions) and isotopic ( $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$ ) analyses, coupled to capacity building and participatory activities was carried out to correctly attribute the nitrate origin in groundwater from the Grombalia Basin (North Tunisia), a region where only synthetic fertilizers have been generally identified as the main source of such pollution. Results demonstrates that the basin is characterized by high nitrate concentrations, often exceeding the statutory limits for drinking water, in both the shallow and deep aquifers, whereas sources are associated to both agricultural and urban activities.

The public participation of local actors proved to be a fundamental element for the development of the hydrogeological investigation, as it permitted to obtain relevant information to support data interpretation, and eventually guaranteed the correct assessment of contaminant sources in the studied area. In addition, such activity, if adequately transferred to regulators, will ensure the effective adoption of management practices based on the research outcomes and tailored on the real needs of the local population, proving the added value to include it in any integrated investigation.

**Keywords:** Socio-hydrogeology, aquifer pollution, isotopes, public engagement, stakeholder analysis

## 1 **1. Introduction**

2 A stronger integration between science and society could contribute solving nitrate contamination issues  
3 affecting rural regions worldwide. In these areas groundwater often represents the main freshwater source for  
4 both domestic and agricultural uses, providing farms and local households with -generally- free supplies in  
5 close proximity to the users and commonly without the need for complex treatment (Morris et al., 2003).  
6 Accordingly, worldwide 49% of the rural population depends on groundwater for domestic supply mostly  
7 extracted from private boreholes and/or hand dug wells (UNICEF and WHO, 2012). In addition,  
8 approximately 38% of global irrigated areas rely on groundwater resources (Siebert et al., 2013), which has  
9 contributed to a ten-fold increase of groundwater abstraction for agricultural irrigation over the last 50 years  
10 (WWAP, 2016).

11 Undoubtedly one of the main consequences of this high aquifer dependency is that any contamination of  
12 these waters can have serious repercussions on local population, either directly, when groundwater is used  
13 for drinking purposes or indirectly (e.g. affecting food security).

14 The intensification of agriculture to sustain human needs, occurred since the second half of the twentieth  
15 century, positively contributed to improve the wellbeing of many developing countries, where agriculture is a  
16 fundamental part of the economy (Hazell and Wood, 2008). On the other hand, however, the need to respond  
17 to rapid population growth and the shift towards more water-dependent economies resulted in severe aquifer  
18 exploitation and contamination, the latter mainly associated to high fertilizers use rates (Foster and Chilton,  
19 2003), contributing to increase the adverse impacts of agriculture on underlying groundwater resources. As a  
20 result, rural population is primarily affected by groundwater pollution, shown by an overall enhanced  
21 mineralization generally associated to nitrate contamination (FAO, 1996).

22 In fact, N-compounds are among the principal nutrients provided by synthetic fertilizers and manure spread  
23 on soils to improve crop growth and development, eventually increasing the grain/seed yield (Bose and  
24 Srivastava, 2011). However, mismanaged fertilizers use and irrigation practices can lead to an accumulation  
25 of nitrates in the subsoil that can leach into the aquifer, contribute to groundwater quality degradation,  
26 especially in shallow aquifers (Singh et al., 1995), and trigger complex water-rock interaction processes,  
27 ultimately enhancing salinization, particularly in coastal aquifers (Re and Sacchi, 2017, and references  
28 therein). In fact, when too large amounts of fertilizers are used (i.e. abundantly exceeding their use efficiency,  
29 as the percent recovery of fertilizer-N by a crop), a significant fraction of N can remain unutilized in the soil  
30 and create the potential for aquifer contamination. This is the case of many rural areas, especially in  
31 developing countries, where agricultural-led nitrate contamination is often associated to other anthropogenic  
32 sources (e.g. animal manure, sewage effluents, and untreated wastewaters) jointly increasing N  
33 concentration in groundwater bodies (Keeney, 1989). For example, in the sub-urban and rural areas of the  
34 Cap Vert Peninsula in the Dakar region, Senegal the interaction of agricultural activities, wastewater and  
35 septic effluent infiltration cause the occurrence of high nitrate concentrations in water for irrigation and human  
36 consumption, largely exceeding the statutory limits for drinking water (50 mg/L; WHO, 2011) and reaching  
37 concentrations up to 800 mg/L (Deme et al., 2006; Re et al., 2011; Diédhiou et al., 2012). High groundwater

38 nitrate concentrations, both industrial and agricultural origin, have also been recorded in many rural areas in  
39 China, with concentrations up to 560 mg/L found in high-yielding areas of northern China (Liu et al., 2005; Yu  
40 et al., 2015, and references therein). In other regions the agricultural impact can be considered negligible, if  
41 compared with the on-site sanitation contribution. This is the case for example of different urban/peri-urban  
42 areas and high-density rural settlements in South Africa, Namibia and Botswana, where this anthropogenic  
43 source causes the occurrence of nitrate concentrations exceeding 800 mg/L (Tredoux et al., 2009).

44 In these situations, nitrate contamination of groundwater goes beyond the issue of water resources  
45 conservation *per se*, but also becomes a strong economic and social concern, given the severe impacts that  
46 excess of nitrates in groundwater can have on local populations, potentially arising food security and health  
47 (i.e gastric cancer and methemoglobinemia in infants; Fan and Steinberg, 1996) issues. In addition, nitrate  
48 pollution, according to its origin, it's often associated with other contaminants, as pesticides or coliforms,  
49 therefore evidencing the aquifer vulnerability. This is why over the years the international scientific community  
50 widely analysed these effects and worked towards finding methods to identify the sources of contamination,  
51 such as the application of environmental isotopes (e.g.  $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$ ; Aravena et al., 1993; Panno et  
52 al., 2001; Baily et al., 2011).

53 So why nitrate pollution continues being such a great issue worldwide? Why farmers continue over applying  
54 fertilizers and mismanage farm manure? Why local authorities underestimate the effects of the lack of proper  
55 sewage collection and treatment facilities on the subsoil and groundwater quality? This can be partially  
56 attributed to the "hidden nature" of both groundwater resources and nitrate contamination (i.e. odourless,  
57 absence of unpleasant taste, colour or turbidity), that makes more difficult its identification without specific  
58 analysis. But, leaving aside limitations in governance and central-local relations, which are out of the scope of  
59 this research, it is clear that a gap between the scientific community and the water end-  
60 users/managers/polluters still hampers adequate capacity building and knowledge transfer in this regard. The  
61 main effect of this fault is therefore the lack of awareness not only on the strong connections between human  
62 activities and the quality of the natural environment, but also on the fact that groundwater protection can lead  
63 to long-term benefits from the socio-economic point of view. This is why it is nowadays unrealistic to obtain  
64 effective solutions to water (and more generally environmental) issues keeping hydrogeology (and science)  
65 separate from the socio-economic and political domain. It is clearly important that the risks to groundwater  
66 quality (and consequently to human health) by the widespread use of fertilizers and unconstrained water use  
67 are assessed, so that the necessary control measures can be introduced. Nevertheless, effective durable  
68 solutions can not be implemented without coupling sound scientific assessments with adequate public  
69 engagement and capacity building on the importance of water resources protection. In this regards it is  
70 crucial to strengthen the interactions between scientists and both farmers and water users, in order to make  
71 the most effective use of the outcomes of hydrogeological and environmental investigations tailored to water  
72 security. This implies bridging the gap between science policy and society, based on the assumption that a  
73 common understanding of the implication of scientific research would help solving the issues arising from  
74 opposing perceptions of environmental matters (Busche, 2015). Thus, improving information sharing and  
75 public engagement when different interests are at stake would be an asset in any research willing to lead to

76 effective governance measures also in the groundwater sector. In this framework a new approach was  
77 proposed by Re (2015), named socio-hydrogeology, as a way of incorporating the social dimension into  
78 hydrogeological investigations and calling for a stronger engagement of hydrogeologists as advocates for  
79 public participation in water management and governance. In line with socio-hydrology (Sivapalan et al.,  
80 2012), aimed at studying the dynamic interactions and feedbacks between water and people, socio-  
81 hydrogeology focuses on the understanding of the mutual relations between people and groundwater (i.e.,  
82 the impact of human activities on aquifer quality and the impact of groundwater on human wellbeing and life),  
83 by fostering the inclusion of the social dimension in hydrogeological investigations.

84 This paper aims at promoting the implementation of socio-hydrogeological approach as the best tool to tackle  
85 groundwater quality assessment, while contributing bridging the gap between science and society, by  
86 presenting the results of a research undertaken in the Grombalia Basin (North Tunisia). In particular, the  
87 effectiveness of performing specific groundwater quality assessments together with capacity building and  
88 participatory activities is tested. The public participation of local actors is a fundamental element for the  
89 development of the hydrogeological investigation, as it may ensure the effective adoption of management  
90 practices based on the research outcomes and tailored on the real needs of the local population. In this  
91 regard, direct engagement and confrontation with well owners and farmers allows hydrogeologists to tackle  
92 the investigation more productively, to retrieve reliable information on water and land use, and to create a  
93 relationship of mutual trust with local stakeholders. Hence the manuscript examines whether incorporating  
94 structured questionnaire administration to local farmers while performing hydrogeological samplings can  
95 value the effort, although it may seem time and money consuming. Indeed, these activities can provide  
96 precious information, useful for data interpretation, and at the same time favour dissemination and capacity  
97 building to support the implementation of new management practices based on the results of the scientific  
98 investigation.

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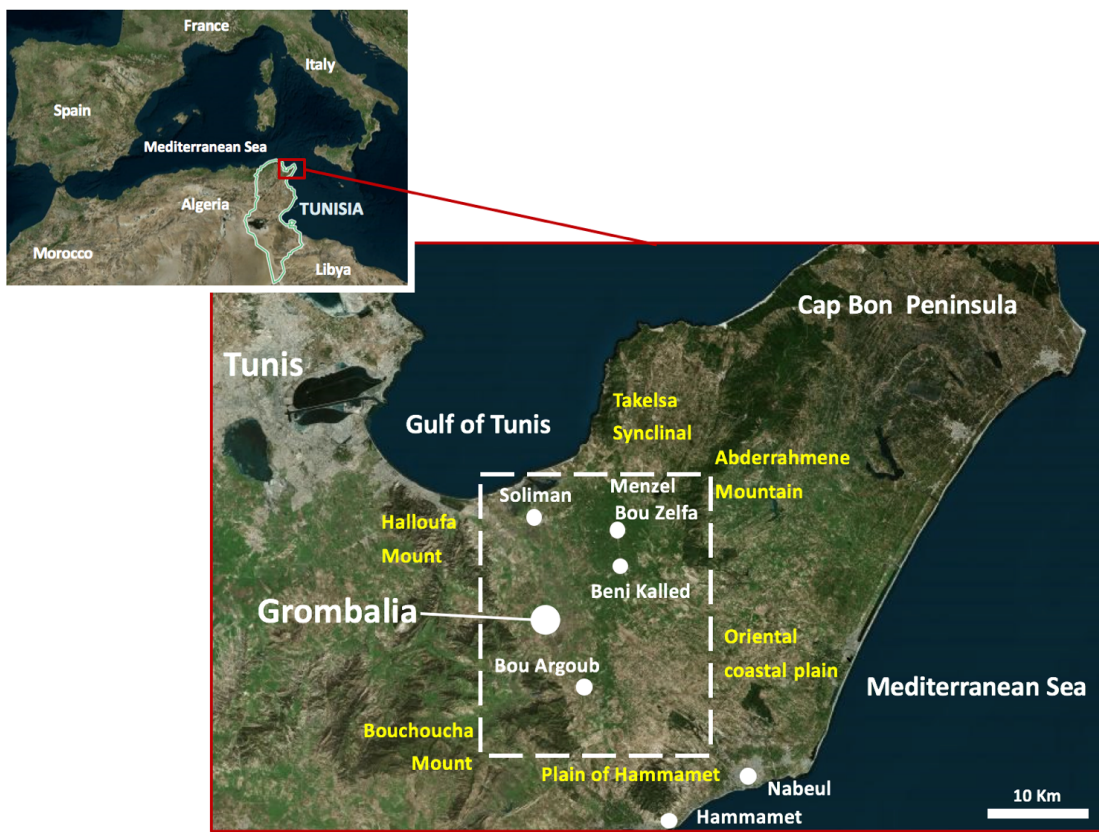
## 100 **2. Site description**

101 The Grombalia region is located in south-western part of the Cap Bon Peninsula (Tunisia) and covers a  
102 surface of about 719 km<sup>2</sup>. The basin is bordered by the Gulf of Tunis (N), the Takelsa Syncline (N-E), the  
103 anticlinal of the Abderrahmene Mountain and the oriental coastal plain (E), the plain of Hammamet (S) and  
104 the Bouchoucha and Halloufa reliefs (W; Figure 1).

105 The climate of the region is semi-arid to Mediterranean sub-humid, with average precipitations of about 500  
106 mm/y (1954-2006; DGRE, 2006), with maximum precipitations between October and January, and mean  
107 annual temperature of around 18 °C (max 28.9 °C in July, min 8.6 °C in January; 2003-2013, INM, 2014).  
108 The average monthly potential evapotranspiration is 76.8 mm, with lowest value in January (40.7 mm) and  
109 the maximum in July (134.5) (2003-2012; INM, 2014). In the area several ephemeral rivers (*wadis*) are  
110 present, collecting surface runoff from the surrounding highlands toward the Gulf of Tunis.

111 The Grombalia Basin is situated astride the African–Eurasian plate boundary (Elmejdoub and Jedoui 2009).  
112 Geologically, it is described as a graben delimited by two normal faults developed during the Middle Miocene

113 (Hadj Sassi et al. 2006), namely the Borj Cedria NNW–SSE normal fault and the Hammamet NE–SW normal  
 114 fault (Ben Ayed 1993; Ben Salem 1995; Chihi 1995) and filled by 500 m of Quaternary sediments. These  
 115 mainly consist of fine to coarse grained sands, clayey sands, sandstone, silt and abundant evaporate  
 116 deposits (Schoeller, 1939; Colleuil, 1976; Ben Salem, 1995; Ben Moussa et al., 2010).  
 117 From a hydrogeologic point of view, the Grombalia aquifer is a multi-layer system constituted by a shallow  
 118 phreatic aquifer, with an average thickness of about 50 m, hosted in the Quaternary continental sand, clayey  
 119 sand and sandstones deposits, and different confined aquifers with average thickness of about 100 m each  
 120 separated by marl layers but communicating through discontinuities (Castany, 1948; Ennabli, 1980). The  
 121 recharge in the shallow unconfined aquifer mainly occurs in the pediments of the surrounding mountains and  
 122 converges to the central part of the basin. There, a general southeast–northwest flow carries groundwater to  
 123 the Gulf of Tunis discharge areas (Ben Moussa 2007; Gaaloul et al., 2014).



124  
 125 **Figure 1. Location of the Grombalia Basin. Background satellite image from Microsoft® Bing™ Maps. The white**  
 126 **rectangle highlights the area represented in Figure 2.**

127  
 128 The Gombalia region is particularly devoted to arboriculture (mainly citrus -representing 82% of the national  
 129 production-, grapes -80% of Tunisian vineyards- and olives) and horticulture (mainly tomatoes, strawberries  
 130 and legumes). Most of the agricultural production, if not used for personal consumption, is sold on both the  
 131 national and international markets (Gafsi and Ben Hadj, 2007), and the agro-industrial sector is also rapidly

132 expanding with more than 1250 factories, including food processing plants located in surrounding areas of  
133 Nabeul, Grombalia and Soliman.

134 In the region, groundwater represents the main source of water supply for both agricultural and industrial use,  
135 and consequently, the aquifer is increasingly exposed to external pressure. More than 11,000 wells tap the  
136 shallow aquifer, with a current abstraction around 250 Mm<sup>3</sup>/y, causing an estimated decrease of the water  
137 table depth of about 0.3 m/y (DGRE 2010). Aquifer overexploitation had led to a severe piezometric level  
138 decrease over the years ( about 10 m in the last 50 years; Charfi et al. 2013a; Gaaloul et al. 2014), that is  
139 especially evident during the dry months, when abstraction rates exceed natural aquifer recharge from rainfall  
140 infiltration. In addition to scarcity, groundwater quality depletion is increasingly harming natural water  
141 resources. The latter is generally associated to aquifer salinization, salt water intrusion near to the sea shore  
142 and nitrate pollution due to anthropogenic activities (Ben Moussa et al., 2010; Ben Moussa and Zouari,  
143 2011). As multiple contamination sources are present in the region of Grombalia, it is fundamental not only to  
144 clearly identify them, but also to effectively raise awareness among all the concerned stakeholders in order to  
145 promote shared strategies for contamination reduction and remediation, based both on sound scientific  
146 results and participative processes.

147

### 148 **3. Methods**

149 A socio-hydrogeological investigation, combining both classical hydrogeological analysis with socio-economic  
150 assessment (Re, 2015) was performed in the Grombalia plain (N-E Tunisia) targeted to a complete nitrate  
151 vulnerability assessment of the region. The sampling campaign combined the typical activities of a  
152 groundwater quality monitoring, together with a structured social analysis performed by the research team  
153 while conducting the field works.

154

#### 155 **3.1. Hydrogeochemical investigation**

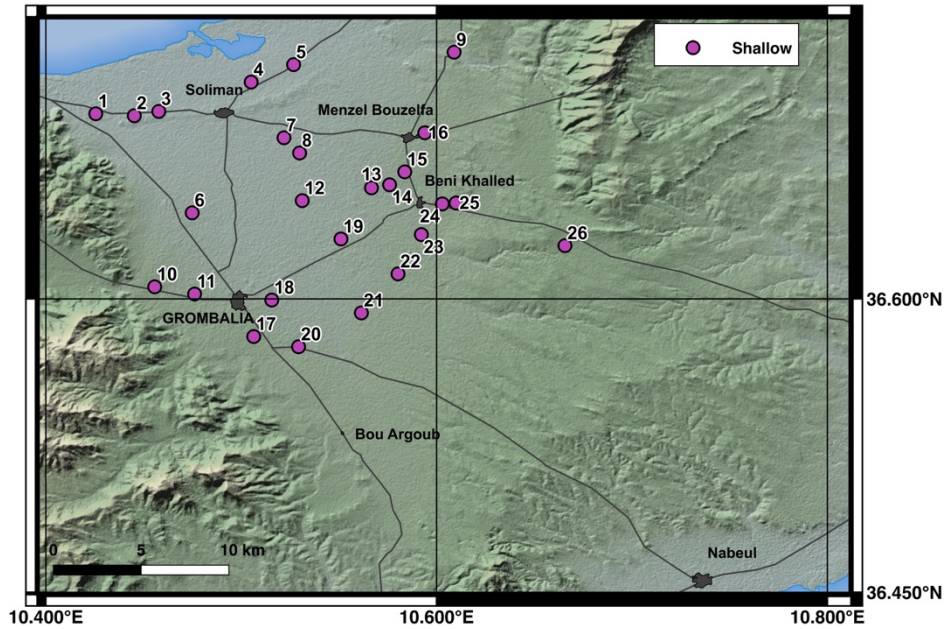
156 Between February and March 2014 (i.e. at the end of the rainy season/winter) a total of 51 groundwater  
157 samples were collected in both the shallow (depth generally < 50 m) and deep Grombalia (depth generally >  
158 50 m) aquifers (26 and 25 respectively). Samples were collected from private hand-dug wells, mostly  
159 equipped with electrical pump, and from both private and public boreholes.

160 The sampling strategy took into account the historical sampling network and the previous studies in the  
161 region (Ben Moussa et al., 2010; Ben Moussa and Zouari, 2011; Ben Moussa et al., 2012; Charfi et al.,  
162 2013). This favoured the design of the sampling network presented in Figure 2, covering the whole Grombalia  
163 Basin and including the sites potentially more susceptible to nitrate contamination.

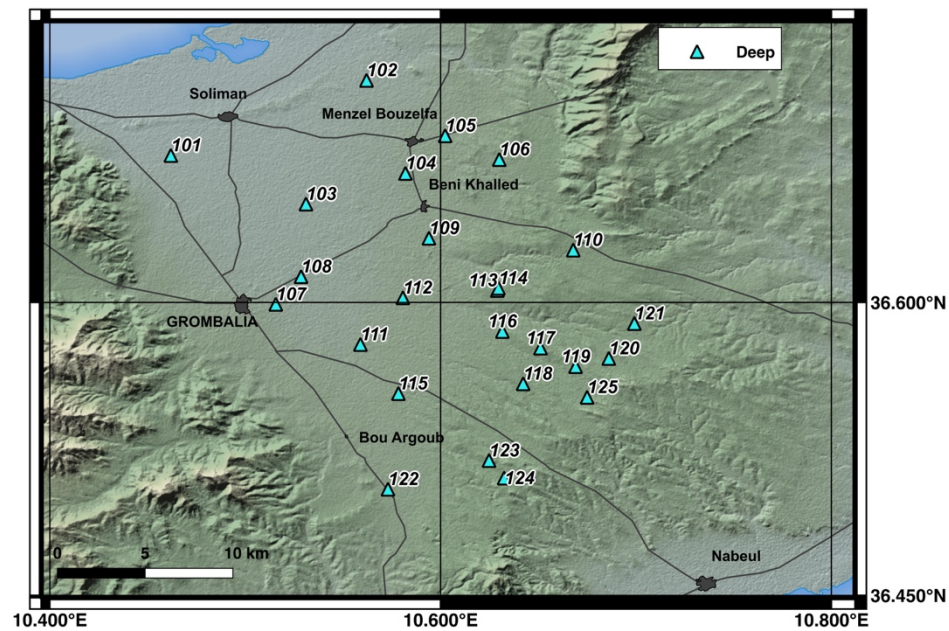
164 *In situ* measurements of electrical conductivity, pH and water temperature (Table X1 – Supplementary  
165 Materials) were performed, using a WTW 340i multimeter. Samples for major ion analysis were filtered  
166 through 0.45 µm cellulose membrane and stored in high density polyethylene bottles. Chemical and isotopic  
167 analyses of the water samples were performed at the Laboratory of Radio-Analyses and Environment (LRAE)  
168 of the National School of Engineers of Sfax (Tunisia). Major elements were analysed using a Dionex DX 100



169 ion chromatograph equipped with a CS12 and an AS14A-SC Ion Pac columns and an AS-40 auto- sampler.  
170 The total alkalinity (as  $\text{HCO}_3^-$ ) was determined by titration with standard hydrochloric acid (0.1N) using methyl  
171 orange and phenolphthalein as indicators. The error, based on the charge balance, was calculated to be  
172 <5%. The isotopes of dissolved nitrate ( $\delta^{15}\text{N}_{\text{NO}_3}$  and  $\delta^{18}\text{O}_{\text{NO}_3}$ ) were prepared and analysed at the ISO4 private  
173 laboratory (Turin, Italy) using a Finningan™ MAT 250 Mass Spectrometer, following the procedures  
174 described by Silva et al. (2000). Results are expressed in ‰ and refer to AIR and V-SMOW (Gonfiantini et  
175 al., 1995) with uncertainties ( $2\sigma$ ) of  $\pm 0.5\text{‰}$  and  $\pm 1\text{‰}$  respectively.



A



B

176 **Figure 2. Location of the sampling sites: (A) shallow aquifer; (B) deep aquifer.**

### 177 **3.2. Socio-hydrogeological approach**

178 The socio-hydrogeological approach proposed by Re (2015) combines the groundwater quality assessment  
179 with capacity building and participatory activities, and it is centered on the role of hydrogeologists as  
180 advocates for public engagement in water management and governance.

181 Therefore, during the previously described field work performed, all the farmers and well owners of the 51  
182 sampled sites were asked to respond to structured interviews on water use and agricultural practices (Tringali  
183 et al., 2017). The main goal of this activity was to create a momentum for dialogue on local groundwater  
184 protection and capacity building, while also collecting relevant information on groundwater use and pollution  
185 issues. The interviews were administered directly by the research team and were focused on the collection of  
186 information related to sampled wells' features, groundwater uses and perceived anthropogenic impacts on  
187 water resources, crop production, irrigation and fertilizer use (Re, 2015; Table 1).

188

189 **Table 1. Summary of the structure and information retrieved with the structured questionnaires proposed by Re (2015).**

<b>Part</b>	<b>Goal</b>	<b>Information</b>
<b>Personal information</b>	Obtain information (to be treated anonymously) on the rural population features	Gender, age, education, occupation, contacts
<b>Water use</b>	Retrieve information on regional and local characteristics to support data interpretation	Well features (age, depth, main characteristics), groundwater withdrawal rates, groundwater use trends, perceived or ascertained groundwater quality issues
<b>Purposes of groundwater uses</b>	Obtain information on local activities and priorities to support data interpretation	Groundwater use, kinds of crops cultivated, seasonal production, kinds and quantity of fertilizers used, irrigation type
<b>Awareness of water issues</b>	Know farmers and well holders perception about water issues	Perception of: water scarcity, climate change, integrated water resources management and groundwater pollution
<b>Potential for Participation</b>	Evaluation of the potential for the implementation of participatory monitoring and management initiatives	Farmers' role in groundwater protection, perceived groundwater issues in the region, perception of scientists and policy makers regarding local groundwater management, willingness to be included in the groundwater monitoring network

190

## 191 **4. Results and discussion**

192

### 193 **4.1. Hydrogeochemical characteristics of the Grombalia aquifer**

194 The high salinity recorded in the shallow aquifer has been pointed out by different authors, highlighting that  
195 mineralization processes in the region are relevant (Charfi et al., 2013) and concern areas where farming and  
196 agricultural activities are more intensive. The abundance of dissolved salts, especially of nitrates, chlorides  
197 and sulphates, indicates an alteration of physical-chemical properties due to anthropogenic activities (Ben  
198 Moussa and Zouari, 2011; Ben Moussa et al., 2014), thus constituting a serious threat for public health and  
199 crop production. Indeed, in their study on the unconfined aquifer, Tlili-Zrelli et al. (2013) indicate that all  
200 groundwater samples exceeded the drinking water limits for Na, Cl and SO<sub>4</sub>, whereas 70% of the samples



201 exceeded also that for Ca and Mg, imparting strong limitations on the use of groundwater for irrigation  
 202 purposes.

203 Although numerous authors have investigated groundwater quality from the shallow aquifer, little is known  
 204 about the composition of the deep groundwater and the factors regulating its chemistry. Our work therefore  
 205 focused on the comparison between these two aquifer layers.

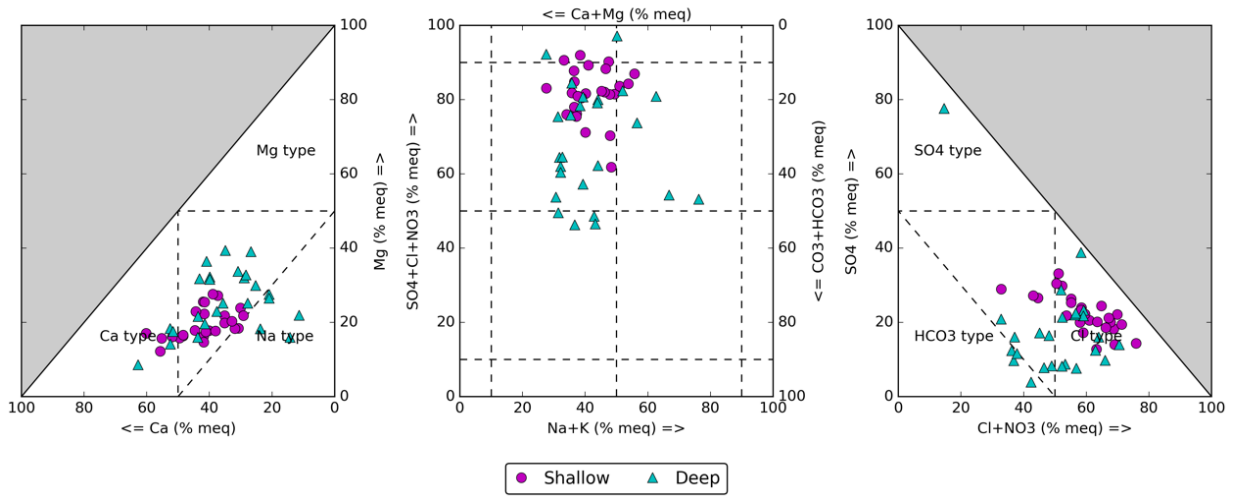
206 Samples collected in our study show Electrical conductivity ranges from 1.04 to 9.18 mS/cm (mean 3.87  
 207 mS/cm) in the shallow aquifer and from 1.04 to 7.13 mS/cm (mean 2.37 mS/cm) in the deep one (Table X1 –  
 208 Supplementary Materials and Table 2). Chloride concentrations range from 112.4 mg/L (well 18), to 2932.4  
 209 mg/L (well 1), with average of 838.8 mg/L, in the phreatic aquifer, and from 107.3 mg/L (well 122) to 3436  
 210 mg/L (well 108), with average of 473.3 mg/L in the deep one. This confirms the high mineralization of the  
 211 studied system for both the shallow and deep aquifer.

212  
 213 **Table 2. Descriptive statistics and Mann–Whitney U test results. Concentrations are expressed in mg/L, while isotopic values as**  
 214 **permil. Underlined values correspond to not-statistically significant parameters. Values in bold indicate the group with the**  
 215 **highest mean ranks relative to each statistically significant parameter.**

		EC	pH	HCO <sub>3</sub> <sup>-</sup>	CO <sub>3</sub> <sup>-</sup>	SO <sub>4</sub> <sup>-</sup>	Cl <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	Ca <sup>++</sup>	Mg <sup>++</sup>	Na <sup>+</sup>	K <sup>+</sup>	δ <sup>18</sup> O	δ <sup>2</sup> H	δ <sup>15</sup> N	δ <sup>18</sup> O <sub>Nos</sub>
Shallow	Min.	1.0	6.9	164.7	0.0	145.3	112.4	0.0	69.0	26.1	111.1	0.0	-5.4	-37.1	4.6	7.8
	Max.	9.2	7.8	481.9	0.0	723.1	1450.0	514.7	677.4	186.6	734.7	40.3	-3.5	-20.8	25.4	22.2
	Average	3.9	7.2	332.5	0.0	434.1	838.8	148.0	349.5	99.9	397.5	16.8	-4.6	-28.4	10.8	12.8
	Std. Dev	1.5	0.2	75.8	0.0	165.0	534.1	140.9	156.8	54.2	220.0	10.7	0.5	3.2	5.3	4.4
	Mean M-W Rank	<b>33.7</b>	23.5	27.8	23.5	<b>34.2</b>	<b>34.4</b>	31.6	<b>34.8</b>	<b>31.7</b>	<b>32.7</b>	<b>28.6</b>	<b>34.9</b>	<b>33.9</b>	11.6	10.6
Deep	Min.	1.0	7.0	128.1	0.0	23.8	107.3	2.0	31.6	12.7	72.8	4.2	-6.1	-34.8	4.9	7.9
	Max.	7.1	7.7	688.0	42.0	3105.0	3436.0	230.6	696.5	557.3	1888.4	63.6	-4.5	-28.7	11.6	12.7
	Average	2.4	7.3	321.3	4.8	383.0	473.3	55.7	181.8	89.7	275.9	16.1	-5.4	-31.8	8.1	10.7
	Std. Dev	1.4	0.2	99.9	11.1	777.1	650.6	62.1	157.6	114.0	359.0	15.0	0.4	1.7	2.0	1.6
	Mean M-W Rank	18.0	28.6	24.1	<b>28.6</b>	17.4	17.3	20.2	16.9	20.0	19.0	23.3	15.4	16.4	7.8	9.3
Mann-Whitney U	125.0	<u>260.5</u>	<u>277.5</u>	<u>260.0</u>	111.0	108.0	179.0	97.0	176.0	150.0	<u>258.0</u>	69.0	93.0	<u>26.0</u>	<u>38.0</u>	
p value	0.0	<u>0.2</u>	<u>0.4</u>	0.0	0.0	0.0	0.0	0.0	0.0	0.0	<u>0.2</u>	0.0	0.0	<u>0.1</u>	<u>0.6</u>	

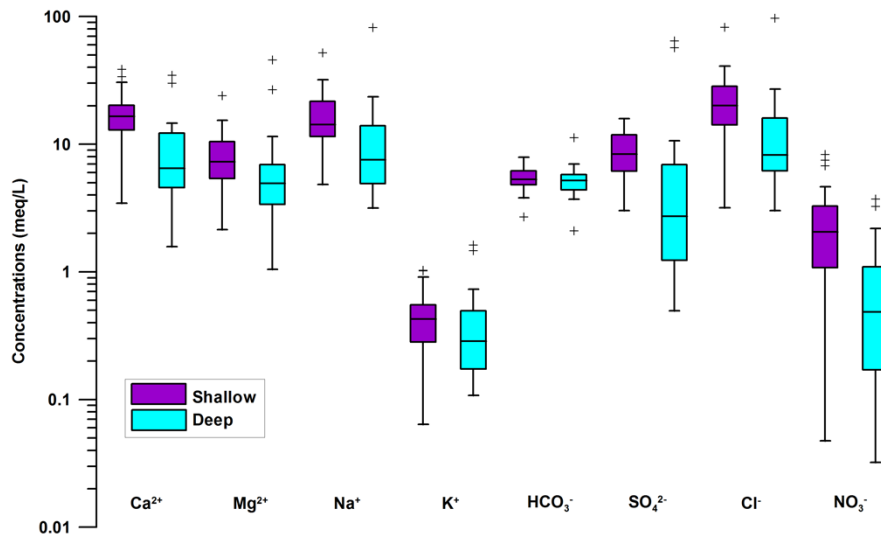
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 217  
 218 A Piper diagram, plotted in rectangular coordinates (Ray and Mukherjee, 2008) was used to highlight the  
 219 different groundwater *facies* in the studied area. Nitrate concentrations were taken into account for the plot,  
 220 due to its abundance in both the shallow and deep aquifers (Table X1 – Supplementary Materials). Grombalia  
 221 groundwater can be generally classified as Ca(Mg)-SO<sub>4</sub>(Cl+NO<sub>3</sub>) water type (Figure 3). Only some samples  
 222 from the deep aquifer (102, 103, 107, 112) display a Na-SO<sub>4</sub>(Cl+NO<sub>3</sub>) *facies*, while for others (110,113,119  
 223 and 122) bicarbonate is the dominant anion. From this diagram, no clear differences in water types between

224 the shallow and the deep aquifer appear; only the composition of deep groundwater seems slightly more  
 225 variable than that of the shallow.



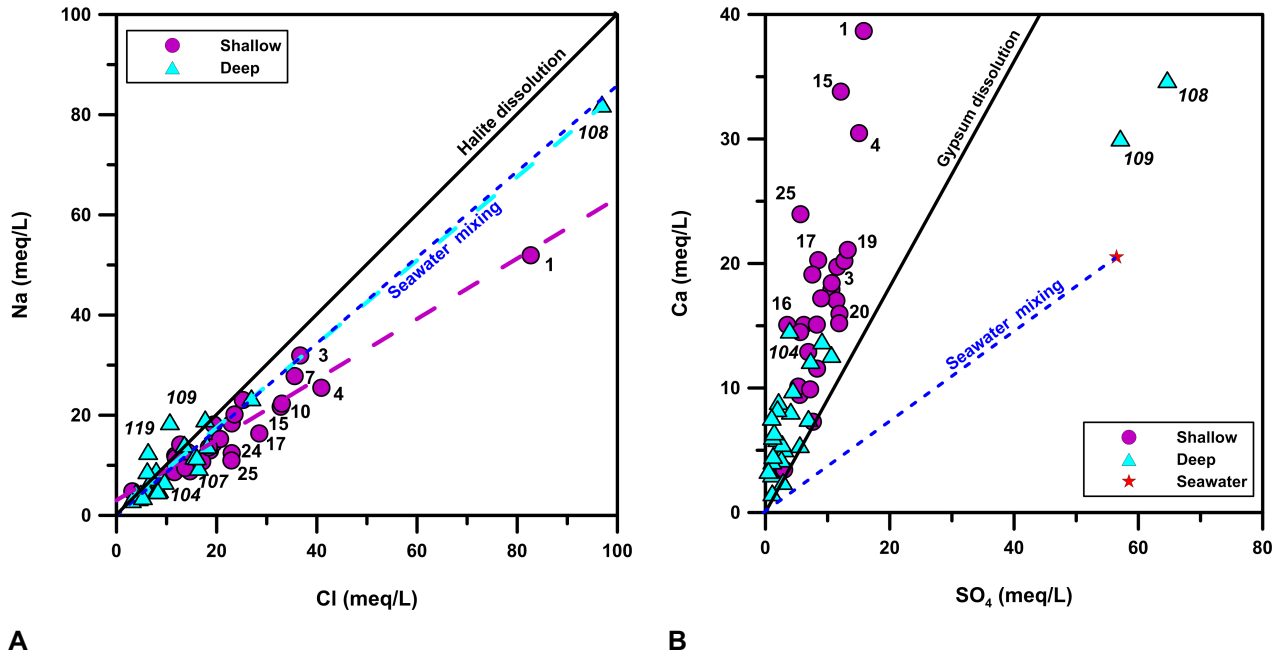
226  
 227 **Figure 3. Piper diagram in rectangular coordinates (Modif. After Ray and Mukherjee, 2008) for the samples**  
 228 **collected between February and March 2014 in the shallow and deep Grombalia aquifers.**

229  
 230 Box plots comparing the major ion contents in groundwater from the two aquifers are reported in Figure 4.  
 231 Accordingly, the shallow aquifer displays higher contents in dissolved cations and anions, with the highest  
 232 differences observed for nitrate, chloride, sulphate and calcium contents, while the distribution of the other  
 233 ions is rather similar. This evidence is also supported by the Mann – Whitney U test (Table 2) highlighting that  
 234 only for pH, HCO<sub>3</sub>, K, δ<sup>15</sup>N<sub>NO<sub>3</sub></sub> and δ<sup>18</sup>O<sub>NO<sub>3</sub></sub> the two aquifers do not show statistically significant differences (p  
 235 value > 0.05).



237  
 238 **Figure 4. Box plots of the major ion contents for the shallow and deep aquifers in the Grombalia basin.**

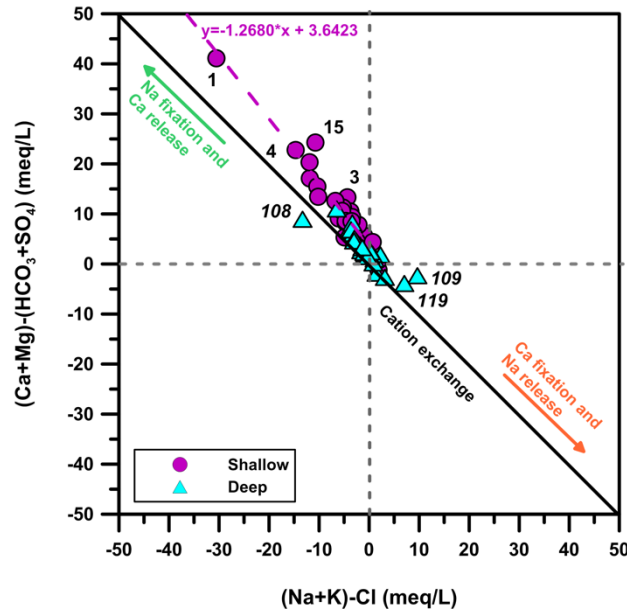
240 Previous studies (e.g. Ben Moussa et al., 2010), based on ion ratios and saturation indexes, indicated that  
 241 the origin of mineralization is mostly from the dissolution of evaporites from the aquifer matrix, namely halite  
 242 and gypsum, whereas the high Ca excess, not balanced by sulphate ions, is due to cation exchange  
 243 processes (aquifer salinization).  
 244



245 **Figure 5. (A) Plot of Na vs Cl and (B) Ca vs SO<sub>4</sub>. Dashed blue line: Seawater mixing trend; Purple and cyan dashed lines correspond**  
 246 **to the regressions for the shallow and deep aquifers respectively ( $Na = 0.6036 \cdot Cl + 3.0061$ ,  $R^2=0.903$  and  $Na= 0.8342 \cdot Cl + 0.8607$ ,**  
 247  **$R^2= 0.962$ ).**

248  
 249 In the Na versus Cl plot (Figure 5A), samples from the shallow aquifer align following the equation  $Na = 0.6036 \cdot Cl + 3.0061$  ( $R^2=0.903$ ) which does not correspond exactly to the 1:1 line expected for halite  
 250 dissolution but shows an excess in Cl (or a deficiency in Na). Another source of Na and Cl could also be  
 251 considered, i.e. sea water, but even in this case several samples (e.g. 7, 4, 10, 15, 17, 24, 25) plot below the  
 252 seawater mixing line. Samples from the deep aquifer have a different behavior, and tend to align more clearly  
 253 on both the halite dissolution and seawater mixing lines ( $Na= 0.8342 \cdot Cl + 0.8607$ ;  $R^2= 0.962$ ), with the more  
 254 saline sample (108) plotting directly on the latter, and two samples (109 and 119) with an excess in Na,  
 255 plotting above the two previously mentioned lines. Also in the Ca vs SO<sub>4</sub> plot (Figure 5B) samples do not  
 256 align on the 1:1 line, indicative of gypsum dissolution, but show an excess in Ca for both the shallow and the  
 257 deep aquifers. Two deep samples instead (108 and 109) plot close to the seawater composition. This excess  
 258 in Ca, according to previous interpretations, is due to cation exchanges, which sequester Na and release Ca  
 259 in solution. This process can be evidenced in the plot of (Na+K)-Cl versus (Ca+Mg)-(HCO<sub>3</sub>+SO<sub>4</sub>) (Figure 6)  
 260 where samples should align on a -1 slope line (McLean et al., 2000). Indeed, for our case study, samples  
 261 from the shallow aquifer are aligned and fall in the field of aquifer salinization (i.e. Na uptake and Ca release)  
 262 but with a higher slope (-1.27), suggesting that other processes should be considered to account for  
 263

264 mineralization. Samples from the deep aquifer are more in agreement with the aquifer salinization  
 265 interpretation (i.e. fall on a -1 slope), although some samples (108, 109 and 119) deviate from this trend.  
 266



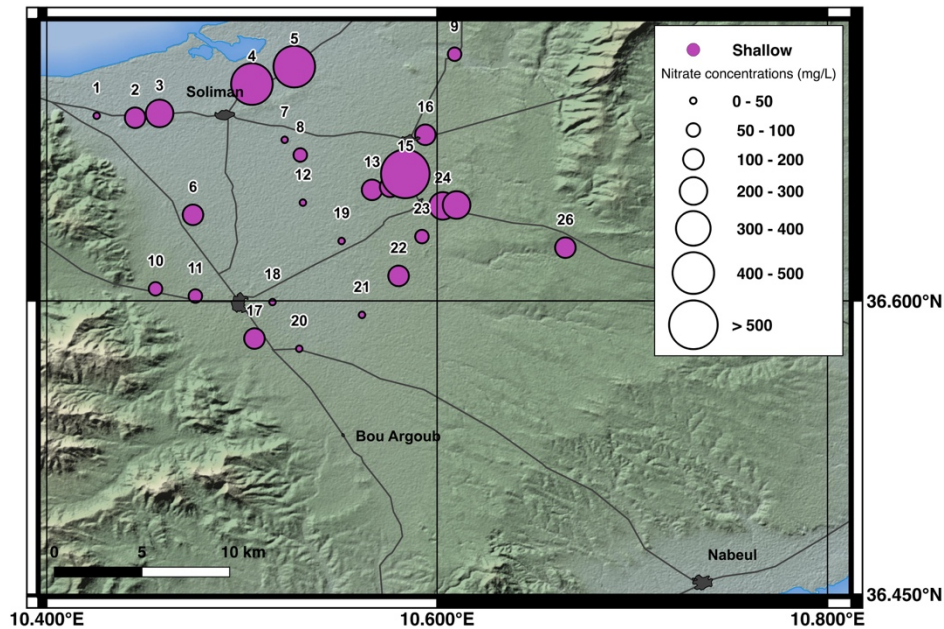
267  
 268 **Figure 6. (Na+K)-Cl versus (Ca+Mg)-(HCO<sub>3</sub>+SO<sub>4</sub>). Black line: cation exchange (-1: 1) line.**

269  
 270 In conclusion, the processes regulating groundwater chemistry in the shallow and in the deep aquifers show  
 271 similarities (salinity and hydrochemical facies) but also differences (some lower ionic content for the deep  
 272 aquifer). A full understanding of water-rock interaction for these waters would require geochemical modelling  
 273 and is out of the scope of this paper. Nevertheless, it should be noted that the most mineralized waters form  
 274 the shallow aquifer are located along the coast in the industrial area around Soliman: here groundwater  
 275 overexploitation could have induced sea water intrusion and associated aquifer salinization processes, which  
 276 modify the Ca/Na ratio. On the other hand, the highest mineralization in the deep aquifer is observed for wells  
 277 108 and 109, and is mostly due to the abundance of sulphate ions in solution. These samples are located at  
 278 more than 15 km inland in an area between Beni Khalled and Grombalia, corresponding to an area where  
 279 Tlili-Zrelli et al. (2013) already evidenced elevated sulphate contents not associated to Ca in groundwater.  
 280 Hence, another source of salinity should be considered, of more continental origin and located at depth (e.g.  
 281 evaporites containing MgSO<sub>4</sub>).

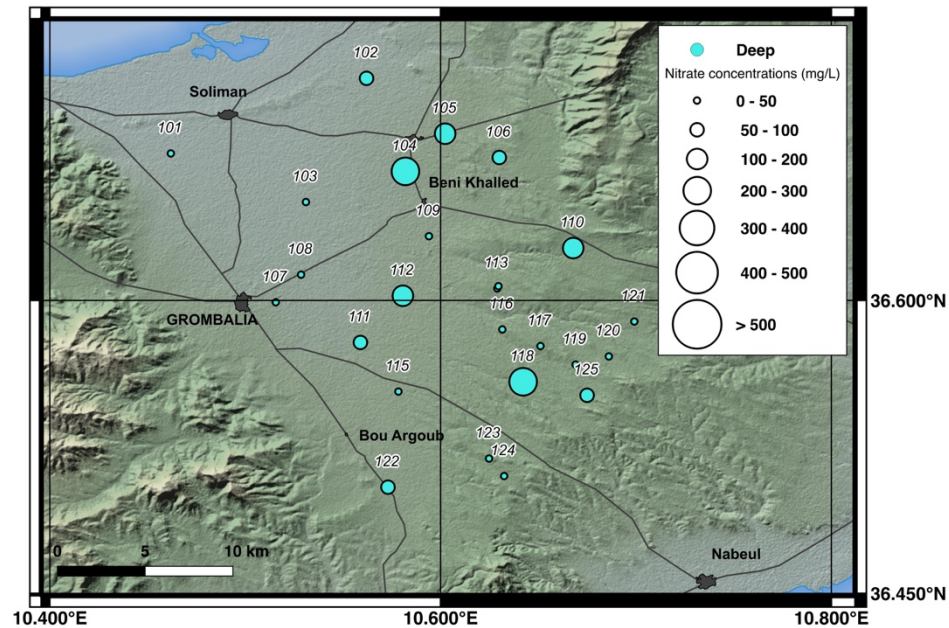
282  
 283 **4.2. Nitrate contamination**

284  
 285 High nitrate concentrations in the shallow aquifer have been remarked over the years by different authors  
 286 (e.g. Ben Moussa and Zouari, 2011; Charfi et al., 2013, Kammoun et al., submitted), and are confirmed by  
 287 our survey. In the NO<sub>3</sub><sup>-</sup> distribution map (Figure 7), only few shallow wells (1, 7, 12, 18, 19, 20 and 21) show  
 288 concentrations below the drinking water statutory limits of 50 mg/L (WHO, 2011), while most of the samples

289 are not suitable for human consumption, with concentrations ranging from 67 mg/l (well 23) to 515 mg/L (well  
 290 15). Two pollution “hotspots” can be identified. The first is located in the central part of the plain (well 15 and  
 291 surroundings), an area identified by Chenini et al. (2015) as high to very high groundwater contamination risk,  
 292 due to the dominance of irrigated areas. This evidence would support the proposed strong contribution of  
 293 agricultural activities to nitrate contamination, enhanced by long-term flood irrigation practices.  
 294



A



B

295 **Figure 7. Nitrate concentration distribution map in the Grombalia Basin: (A) shallow aquifer, (B) deep aquifer.**

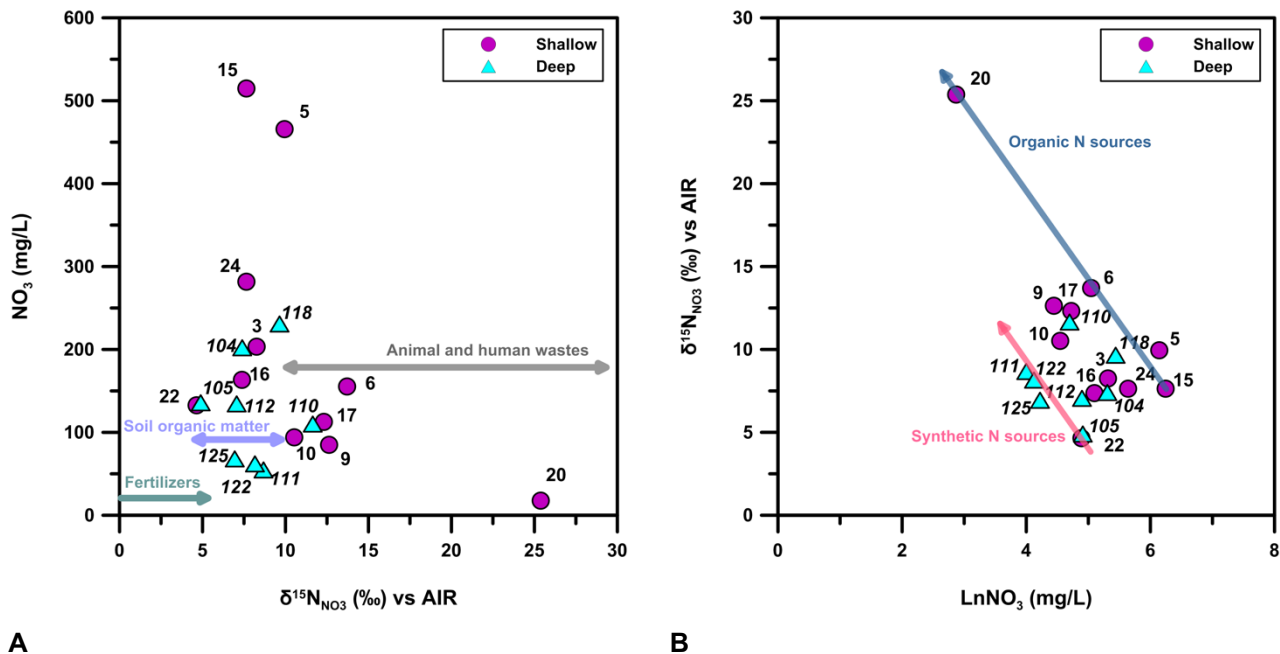
296 On the other hand, the second pollution hotspot is located along the coast (e.g. wells 4 and 5), where urban  
 297 settlements and industrial activities dominate the land use (Chenini et al., 2015), testifying for the presence of  
 298 multiple nitrate sources in the investigated area. Concerning the deep aquifers samples, 15 out of 25 can be  
 299 considered suitable for drinking purposes, whereas in the others  $\text{NO}_3^-$  concentrations range from 55 mg/L  
 300 (well 111) to 231 (well 118): these values, although significantly lower than those found in the phreatic  
 301 aquifer, indicate that the deep aquifer may be severely impacted by nitrate pollution as well. Also in this case,  
 302 the highest concentrations are recorded in the cultivated area (sample 104), but also closer to the recharge  
 303 area to both aquifers (sample 118).

304 Another common feature to both aquifers is the presence of an area where nitrate contents are relatively low,  
 305 located between the cities of Grombalia and Beni Kalled, the first identified pollution hotspot. In order to  
 306 explain such low concentrations, Charfi et al. (2013) suggested the possible presence of denitrification  
 307 processes.

308 The isotopic investigation was undertaken to identify the nitrate pollution sources and the processes affecting  
 309 their concentration. The isotopic compositions of dissolved nitrates range between +4.65 and +25.38‰ vs  
 310 AIR in  $\delta^{15}\text{N}_{\text{NO}_3}$  and between +7.8 and +19.4‰ vs SMOW in  $\delta^{18}\text{O}_{\text{NO}_3}$  in the shallow aquifer, and between  
 311 +4.90 and +11.65‰ vs AIR in  $\delta^{15}\text{N}_{\text{NO}_3}$  and between +7.9 and +12.7‰ vs SMOW in  $\delta^{18}\text{O}_{\text{NO}_3}$  in the deep  
 312 aquifer.

313 In order to identify the dominant sources, samples were plotted in a diagram displaying the  $\text{NO}_3^-$   
 314 concentration vs  $\delta^{15}\text{N}_{\text{NO}_3}$  (Figure 8A) together with the normal ranges reported in the literature for each  
 315 source. Even in this case, no significant differences can be observed for the shallow and the deep aquifer.

316



317 **Figure 8. (A) Nitrate concentration vs  $\delta^{15}\text{N}$ . The reported ranges for nitrogen pollution sources are based on Clark**  
 318 **and Fritz, 1997). (B)  $\delta^{15}\text{N}$  vs  $\text{LnNO}_3^-$ , with the possible denitrification trends identified for synthetic (red arrow) and**  
 319 **organic (blue arrow) nitrogen sources.**



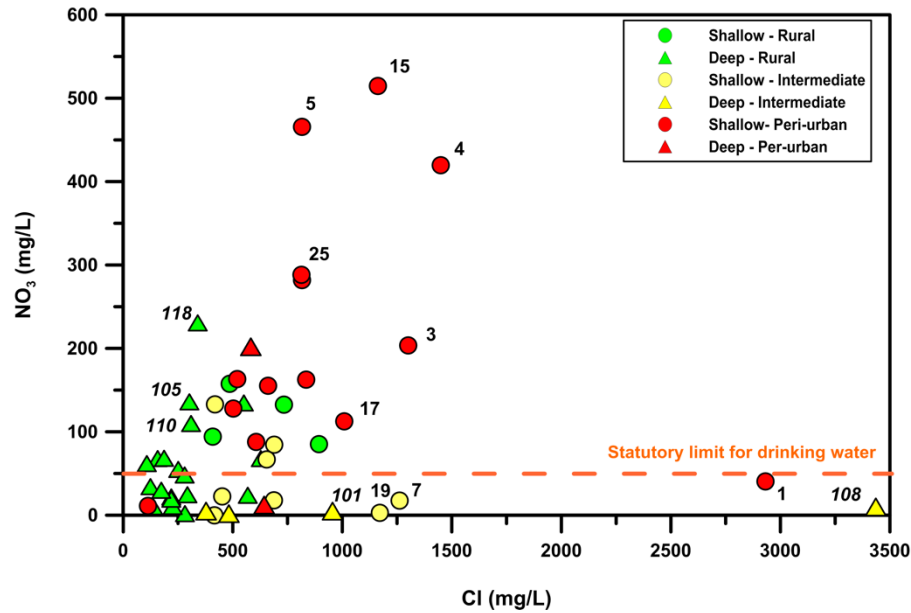
320 Some groundwater samples (22, 105) fall close to the compositional range of synthetic fertilizers, although  
321 their nitrogen isotopic composition is rather enriched, since fertilizers generally range between -4‰ and +4‰  
322 in  $\delta^{15}\text{N}_{\text{NO}_3}$  (Kendall et al., 2007). Nevertheless, the original isotopic composition of fertilizers can be enriched  
323 by other processes occurring in the soil prior to be leached to groundwater, namely volatilization, which is  
324 enhanced in alkaline soils and arid climates (Kendall et al., 2007). Similar isotopic compositions for  
325 groundwater nitrates are reported for the Bou Areg plain in Morocco by Re and Sacchi (2017, and references  
326 therein), and attributed by the authors to synthetic fertilizers enriched by different degrees of ammonia  
327 volatilization.

328 Most of the samples display an  $\delta^{15}\text{N}_{\text{NO}_3}$  in the compositional range of soil organic matter. Nevertheless, since  
329 nitrate concentrations largely exceed the expected natural background level, (10-12 mg/L; Shand and  
330 Edmunds, 2008; Sacchi et al., 2013) these samples likely record a mixed contamination from both synthetic  
331 fertilizers and anthropogenic organic matter (animal or human waste). The latter source is characterized by  
332 enriched  $\delta^{15}\text{N}_{\text{NO}_3}$  values, exceeding +10‰; nevertheless, we considered this organic matter contribution  
333 dominant for samples showing a  $\delta^{15}\text{N}_{\text{NO}_3}$  greater than +8.6‰ (Re and Sacchi, 2017). It should be noted that  
334 the most enriched sample in  $\delta^{15}\text{N}_{\text{NO}_3}$  also displays a low nitrate concentration, which could be due to  
335 denitrification processes.

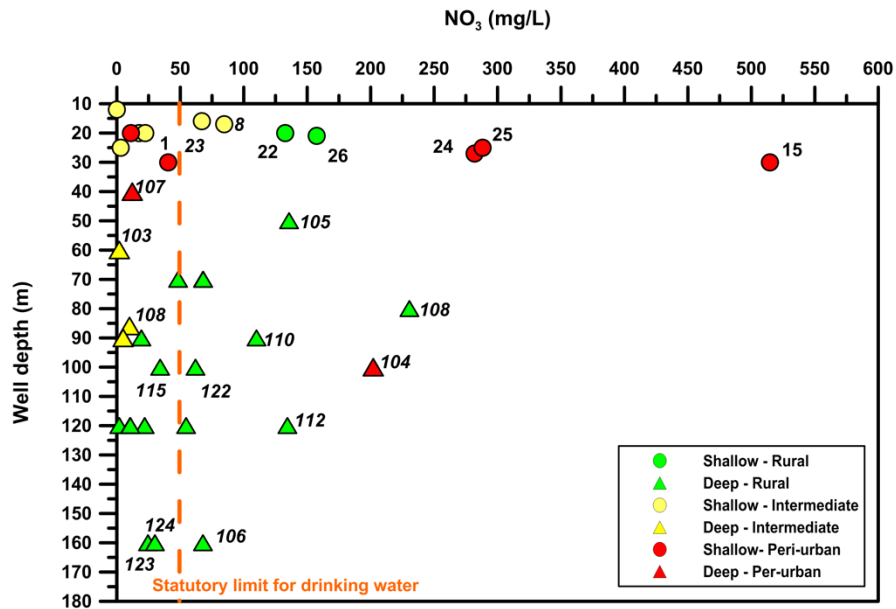
336 Indeed, when plotting  $\delta^{15}\text{N}_{\text{NO}_3}$  vs  $\ln\text{NO}_3^-$  (Figure 8B), samples affected by denitrification should plot on a  
337 straight line. Two of such trends can be identified in the plot: one, mostly displayed by samples from the  
338 shallow aquifer (15, 5, 6, 20), originates from rather enriched isotopic compositions and therefore would  
339 indicate the denitrification path followed by samples contaminated by organic matter sources. The second  
340 trend with the same slope, mostly displayed by samples from the deep aquifer (105, 125, 122, 111) originates  
341 from isotopic compositions in the range of synthetic fertilizers. Samples in between the two trends would  
342 represent a mixture of these two dominant sources with variable degrees of enrichment due to denitrification.  
343 The slope of the two identified trends corresponds to that indicated for a fractionation factor  $\epsilon$  of about 10‰  
344 (Kendall et al., 2007). In this case, the isotopic enrichment would be due to the denitrification of about 30-  
345 45% of the original nitrate content, reaching up to 80% in the case of sample 20. This sample is located in the  
346 previously mentioned area of low groundwater nitrate concentrations, despite the presence of a very shallow  
347 water table depth. Here, denitrification processes could be favored by the presence of clay rich vertisols  
348 (Chenini et al., 2015).

349 To verify the abovementioned hypothesis, the  $\text{NO}_3^-$  vs  $\text{Cl}^-$  diagram was plotted taking into account the  
350 localization of the samples in the Grombalia plain (Table X1 – Supplementary Materials). In this graph (Figure  
351 9) it is possible to notice a remarkable enrichment trend of nitrate and chloride concentrations, especially  
352 characterizing the wells located in the peri-urban areas, confirming the relevance of anthropogenic pollution  
353 associated to domestic activities. Only two wells (e.g. 7, 1), with low or absent nitrate concentrations, show a  
354 significant enrichment in chloride concentration that can be associate to the occurrence of saline water  
355 intrusion in the coastal area near Soliman.

356 On the other hand, most of the wells located in the rural zones or in areas with no dominant land use (hence  
 357 classified as intermediate) relatively lower  $\text{NO}_3^-$  concentrations, corresponding to the previously identified  
 358 trend of synthetic fertilizers pollution origin. Based on this observation, rural areas seem to be relatively less  
 359 contaminated than peri-urban ones, although, the presence of most of the deep wells in these zones  
 360 highlights the high vulnerability to agricultural pollution also in the deeper parts of the aquifer (Figure 10).  
 361 To better understand the factors controlling nitrate contamination origin and evolution, the hydrogeochemical  
 362 results have been analysed taking into account the territorial reality at regional level, based on the outcomes  
 363 of the results of structured interviews performed during *in situ* measurements.



364 Figure 9. Nitrate versus chloride concentrations for the samples collected in the Grombalia Basin in the  
 365 February-March 2014 campaign. Orange dashed line: WHO (2011) statutory limit for drinking water (50 mg/L).



366 Figure 10. Nitrate concentrations versus well depth.  
 367

368 **4.3. Socio-hydrogeological approach to identify nitrate pollution origin**

369

370 Field surveys were completed for 85% of the sampled sites (78% of the shallow wells and 92% of the deep  
371 ones; Table 3).

372

373 **Table 3. Summary of the information retrieved with the structured interviews administered during the *in situ***  
374 **measurements. Percentages indicated as “out of total” take into account that for some categories responses are**  
375 **higher to the number of respondents (i.e. corresponding to possibility to have multiple groundwater use and**  
376 **crop production in the same site).**

	Shallow	Deep
<b>Administered interviews</b>	21	23
Well owners	6	10
Tenants	15	13
<b>Well type</b>		
Hand dug well	23%	-
Pumped well	77%	32%
Borehole	-	68%
<b>Irrigated area</b>		
Average	2 hectares	12 hectares
Min.	1 hectare	1 hectare
Max.	4 hectares	75 hectares
<b>Groundwater use (% out of total)</b>		
Irrigation	63%	90%
Domestic	47%	81%
Drinking	32%	81%
Animal husbandry	5%	38%
<b>Water type used for irrigation</b>		
Groundwater	75%	86%
Groundwater and Irrigation channel water	25%	14%
<b>Irrigation type</b>		
Flood	33%	-
Drip	54%	89%
Spray	-	-
Mix (Flood-Drip/Spray-Drip)	13%	11%
<b>Crops (% out of total)</b>		
Horticulture	28%	52%
Arboriculture	89%	90%
<b>Fertilizers use</b>		
Manure	16%	-
Binary synthetic fertilizers (DAP)	-	5%
Three-component synthetic fertilizers (NPK)		
Manure and NPK	10%	43%
Manure, DAP and NPK	37%	19%
DAP and NPK	32%	28%
	5%	5%

377

378 Overall, well owners and farmers have shown high level of cooperation and interest in the outcomes of the  
379 hydrogeological assessment, although few declined to respond to the interview but gave the permission to  
380 collect groundwater samples in their property. The interviews confirmed that local people are generally aware  
381 of the existence of groundwater issues in the region, and that they consider the most crucial problems to be:  
382 (i) the aquifer salinity increase, (ii) the decrease in the piezometric level due to groundwater overexploitation,  
383 and (iii) a clearly perceived degradation of water quality with respect to the previous years (Tringali et al.,  
384 2017). As a result of this high awareness, most of them expressed the willingness to be part of a collaborative  
385 water management process to improve groundwater quality in the region.

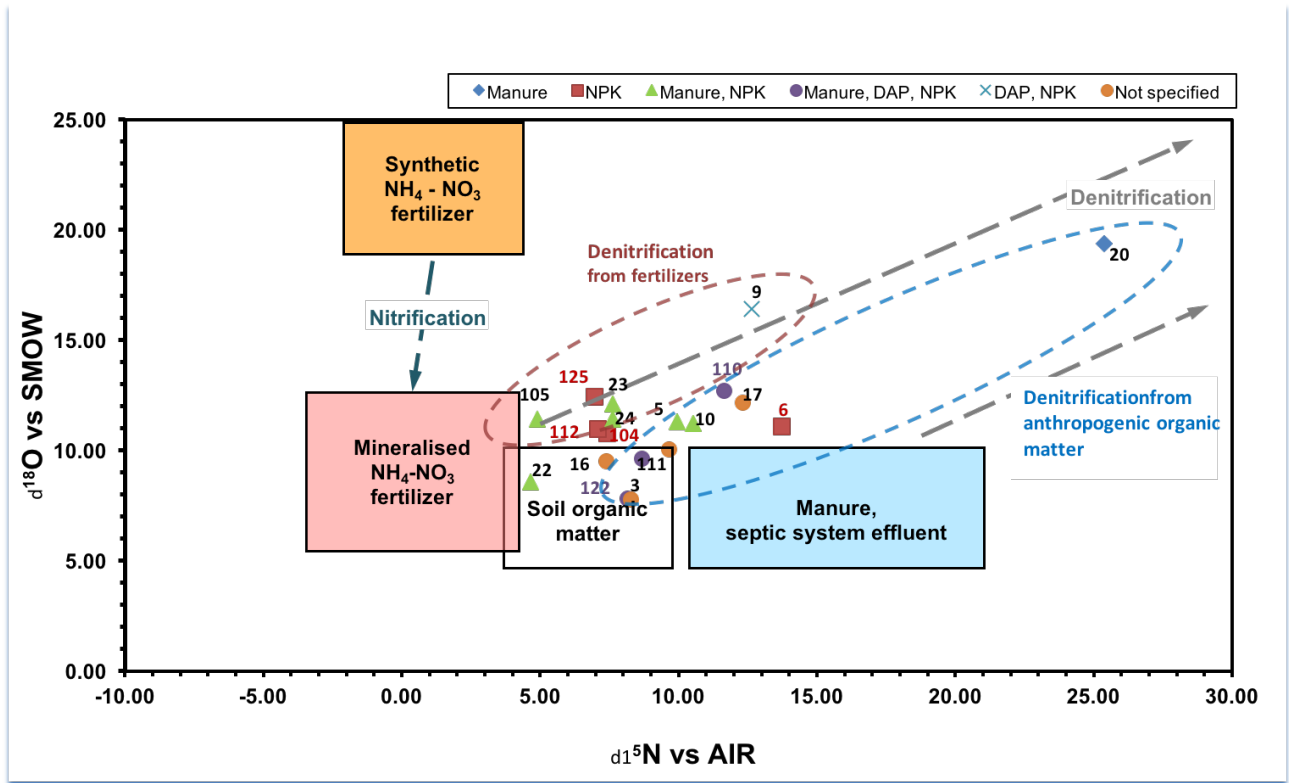
386 Considering all the sampled sites, deep wells are used to irrigate larger agricultural parcels (up to 75  
387 hectares; Table 3), probably due to the prevailing use of boreholes and resulting water availability. In fact,  
388 almost all the wells sampled in the deep aquifer are used for irrigation, often with a multiple utilization that  
389 includes drinking and domestic (i.e. household cleaning). The same groundwater use pattern can be  
390 observed for the shallow wells, mainly providing water for irrigation, but in some cases also used for  
391 household consumption, even though the high salinity makes people less inclined to its consumption for  
392 drinking water purposes except when other sources are not available. As it emerged during the interviews,  
393 local farmers are generally aware of groundwater salinity issues but have a scarce knowledge of the other  
394 potential contamination sources. As a result, besides the environmental and crops' productivity implications of  
395 using contaminated groundwater, many people are unaware of drinking water that is not so adequate for  
396 human consumption, with potential severe consequences on their health in the long run. This is why, as part  
397 of the proposed socio-hydrogeological approach, during interviews administration some time was also  
398 dedicated to knowledge transfer to farmers and wells' owners, and particularly targeted to information sharing  
399 on the general status of the studied aquifers and to raising awareness on the implication of groundwater  
400 misuse and pollution (Tringali et al., 2017).

401 As previously mentioned (Section 4.2), most of the authors have attributed the high  $\text{NO}_3^-$  concentration in the  
402 shallow aquifer to the impact agricultural return flow due to the fertilizers application rates and long-term flood  
403 irrigation. However, of all the interviewees only 33% of the shallow well's holders (i.e. 16% of the total) still  
404 uses flood irrigation in their fields, while the majority is opting for drip irrigation (78% of the total), generally  
405 considered the most effective irrigation practice in arid and semi-arid regions. Apparently, this shift towards  
406 more sustainable irrigation practices occurred in late 90s has had little influence on the nitrate contamination,  
407 which is still increasing (Kammoun et al., submitted), suggesting long recovery times for the aquifers.

408 Finally, it is interesting to note that the farmers' majority indicated the use of manure, alone or in combination  
409 with other types of synthetic fertilizers to increase crop production.

410 In order to identify the different sources of nitrate in the investigated area, the isotopic composition of nitrogen  
411 and oxygen ( $\delta^{15}\text{N}_{\text{NO}_3}$  versus  $\delta^{18}\text{O}_{\text{NO}_3}$ ) was compared to the information on fertilizer's use retrieved during  
412 semi-structured interviews. In Figure 11 the isotopic compositions of the samples do not show a dominance  
413 of mineralized synthetic fertilizers as contaminants (as also evidenced in Figure 8a), coherently with the  
414 information provided by interviewed wells owners. Denitrification processes are observed in both aquifers,  
415 following two trends: one originated from fertilizers (dashed brown oval in Figure 11) and one (dashed blue

416 oval in Figure 11) from anthropogenic organic matter, confirming the previously described dual origin of  
 417 nitrate contamination in the region. This supports the theory that agricultural activities are not the unique  
 418 cause of aquifer pollution and that significant contribution also come from domestic activities, and specifically  
 419 from the lack of adequate sanitation facilities in the rural and peri-urban zones For example sample number  
 420 6, which shows a composition fitting in the field of anthropogenic organic matter (i.e. manure or septic system  
 421 effluents in Figure 11), belongs a site where the farmer declared to use only synthetic fertilizers, and  
 422 consequently this contamination should have a civil origin.



423  
 424  
 425 **Figure 11. Stable isotope composition of dissolved nitrates in groundwater from the Grombalia Basin, with ranges**  
 426 **for groundwater with  $\delta^{18}\text{O}_{\text{H}_2\text{O}} \sim -4$  VSMOW. Modified after Clark and Fritz (1997) and Kendall et al. (2007). Dashed**  
 427 **brown and blue oval corresponds to the denitrification trend originated from fertilizers and anthropogenic**  
 428 **organic matter respectively.**

429  
 430 **5. Science-based management implications**

431  
 432 Results of the socio-hydrogeological investigation performed in the Grombalia aquifer have important  
 433 implication for the local water management.

434 The clear identification of pollution origin crucial for drawing new science-based reductions measures. To this  
 435 end both the hydrogeochemical and social data have proved to be fundamental for the contamination source  
 436 characterization, supporting the need for integrated investigations: socio-hydrogeology can hence be used to

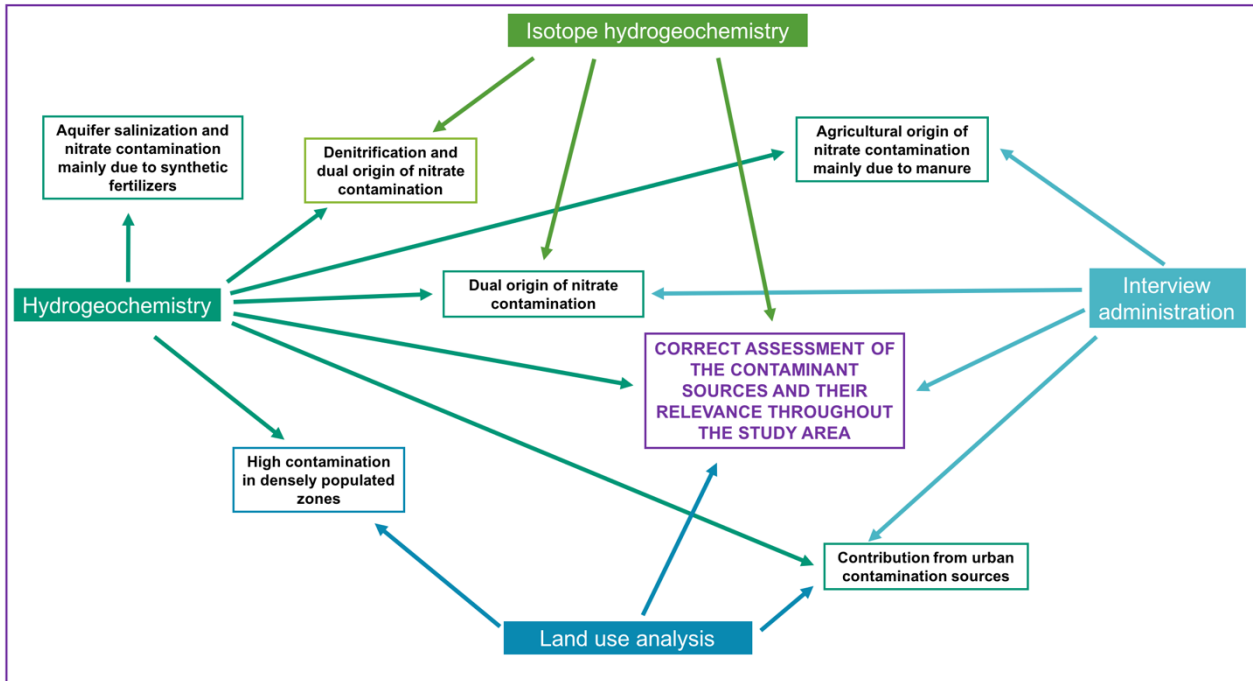
437 integrate vulnerability assessment, especially in areas with intense anthropization, and might be considered  
438 as an asset in any hydrogeological investigation.

439 In particular, the study demonstrates the added value of socio-hydrogeology compared to classical  
440 hydrogeological assessment (Figure 12). The integrated analysis permitted to better constraint the nitrate  
441 contamination sources in the studied region. Previous studies, where only hydrogeochemical analyses were  
442 performed, generally attributed nitrate contamination to agricultural activities, and particularly synthetic  
443 fertilizers, given the dominant land use in the region. However, the information retrieved with the interviews to  
444 local farmers and well owners revealed that farmers use manure in combination to synthetic fertilizers. The  
445 isotopic investigation also confirmed the presence of manure-induced contamination and highlighted the dual  
446 origin (urban/domestic and agricultural) of nitrate. In addition, a more detailed land-use analysis, also coupled  
447 with the interview administration, demonstrated that higher concentrations are found in more densely  
448 populated areas, evidencing the relevance of domestic pollution especially in the shallow aquifer. Indeed,  
449 only with the integrated approach the correct attribution of nitrate contamination would be possible, avoiding  
450 the implementation of improper management actions or penalizing farmers. The study also highlights the  
451 need to foster better sanitation and waste management at local and regional level. Therefore, results will be  
452 adequately shared with competent authorities to support new actions plans in the region.

453 As concerns the implementation of new management practices, it will be of paramount importance to take  
454 into account both the needs of the local stakeholders (including managers and water-end users) and the  
455 outcomes of the integrated socio-hydrogeological assessment. To this end, results will be shared with the  
456 key-stakeholders identified at the early stages of the project by means of a Social Network Analysis: the so  
457 called Groups of Agricultural Development (GDAs; composed by landowners, farmers and water users  
458 sharing water resources in each irrigated area, and coordinated by a board of democratically elected local  
459 members), the Regional Commissariat for Agricultural Development (CRDA, i.e. the institution responsible for  
460 water resource management and control in the Grombalia region), and representatives of local farmers  
461 (Tringali et al., 2017). For the identification of a new and shared strategy for long-term groundwater protection  
462 priority will be given to the identification of new actions for groundwater protection that will not compromise  
463 the farmer's wellbeing and productivity and that will take into account the contribution of domestic and urban  
464 contamination sources. In this process the role of scientist and local mediators will also be fundamental to  
465 ensure adequate information sharing to the general public and civil society.



Socio-hydrogeological approach



466  
 467 **Figure 12. Highlights of the information provided by integrating the different components of the socio-hydrogeological approach.**  
 468 **The filled boxes indicate the activity, the empty boxes the conclusions that may be reached using the information provided by**  
 469 **the activities. Partial or incorrect conclusions can be obtained if using only 2 or 3 information activities.**

470  
 471 **6. Conclusions**

472  
 473 The Grombalia basin is characterized by a general high nitrate concentration, in both the shallow and deep  
 474 aquifers. Results of an integrated investigation performed between February and March 2014 coupling  
 475 hydrogeochemistry and isotope geochemistry highlighted the presence of a dual origin of such contamination,  
 476 associated to both agricultural and urban activities, in a region where only synthetic fertilizers were generally  
 477 identified as the main source of nitrate pollution. The presence of high nitrate contents also in the deep  
 478 aquifer demonstrate its vulnerability to anthropogenic contamination and points to a hydraulic connection with  
 479 the shallow aquifer for either natural (e.g. presence of discontinuities in the clay layer separating the two) or  
 480 anthropogenic reasons (e.g. multilayer wells). This evidence would require further investigation as it is a  
 481 crucial issue for the sound groundwater management in the area.

482 Interview administration provided useful information supporting the hydrogeochemical analysis, and, as in the  
 483 case of fertilizers use, in agreement with the findings of the isotopic assessment. Indeed, when budget  
 484 limitations do not permit a full isotopic assessment, public engagement activities could represent a useful tool  
 485 to provide insight on possible contamination sources. Coherently, public engagement and capacity building  
 486 are fundamental to inform farmers and households on the impact of agricultural practices and domestic  
 487 activities (also with regard to the long-term health and food security implications) as well as to assess their  
 488 needs and perceptions of environmental issues.

489 Overall, results of the investigation performed in the Grombalia basin supported the necessity to perform  
490 integrated investigations to correctly assess contaminant sources and their relevance throughout the study  
491 area. Multidisciplinary approaches, that also include socio-economic analysis can permit fostering connection  
492 between providers and users of water science, also including decision makers. In fact, for a correct long-term  
493 management of groundwater resources not only it is fundamental to know the hydrogeological characteristics  
494 of the studied region and the causes of aquifer contamination, but also to understand the socio-economic  
495 drivers that lead to such contamination. For example, only through public engagement it will be possible to  
496 understand if and why farmers over apply nutrients, or whether lacking sanitation facilities are still dominant.  
497 Additionally, this information could guide new policies prescriptions targeted to contamination reduction that  
498 also take into account the needs and issues of water end-users. Indeed, through socio-hydrogeology  
499 (ground)water scientists can act as advocates for good governance, and effectively contribute solving nitrate  
500 contamination issues.

501

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510

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697

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