



Article

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Article Influence of Resins on the Structure and Dynamics of SBR Compounds: A Solid-State NMR Study

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Abstract: The tackifying effect of resins used in the tire industry highly depends on the compatibility and interaction strength with the rubber matrix. Here, uncured and cured styrene/butadiene rubber compounds, either in the presence or absence of a hydrocarbon aromatic tackifying resin, were studied by means of high-resolution and time-domain solid-state NMR (SSNMR) techniques to investigate resin/polymer interactions and the effect of the resin on the dynamics of polymer chains. ¹³C direct excitation and cross-polarization spectra, combined with low-field measurements of ¹H T_1 and analysis of ¹H on-resonance free-induction decay, provided information on the dynamic heterogeneity of the samples and the degree of mixing between the resin and the rubber matrix. Moreover, ¹H T_1 and $T_{1\rho}$ relaxation times at variable temperatures were used to investigate the effect of resin on both segmental dynamics activated at the glass transition and collective polymer dynamics. SSNMR findings were discussed in relation to crosslink density and T_g data. The obtained results show that the resin is intimately mixed with the polymer, while maintaining its rigid character. A slowdown of segmental dynamics, related to an increase in T_g , was found as a consequence of resin addition, while no effect was evidenced on fragility and collective polymer dynamics.

Keywords: time-domain NMR; MAS NMR; field-cycling NMR; spin–spin nuclear relaxation; spin– lattice nuclear relaxation; segmental dynamics; polymer dynamics; styrene/butadiene rubber; α -methylstyrene/styrene resin; vulcanization

1. Introduction

Elastomer-based compounds are widely employed in the tire industry thanks to the possibility to tune their applicative properties by modifying their formulation. Usually, they are composed of one or more polymers, such as natural rubber or synthetic polymers (isoprene rubber, butadiene rubber, styrene/butadiene rubber), which are subjected to vulcanization in the presence of a vulcanizing agent and other ingredients, including fillers, stabilizers, processing oils and resins, with the purpose of providing defined mechanical proprieties to the final product [1]. In particular, resins are employed in the tire industry for the following three different functions: as tackifiers; as reinforcing agents; and as curing agents [2]. Specifically, tackifying resins are added to modify the rheological behavior of the rubber compound, with the final aim of favoring the processability of the uncured compound, promoting filler dispersion and improving mechanical properties of tires, such as rolling resistance and wet traction [1-7]. The achievement of precise requirements in the final product strongly depends on the polymer–resin miscibility, which in turn is related to physico-chemical characteristics of the resin itself, such as the softening point, glass transition temperature (T_g), viscosity, chemical structure and molecular weight [6–14]. Indeed, the presence of resin alters the dynamics of the polymer chains, resulting in a modification



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Copyright: © 2023 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). of the viscoelastic behavior and final mechanical properties of the vulcanizate [14–16]. In this frame, the comprehension of the relationship between macroscopic proprieties of technological interest and structural and dynamic features arising from polymer–resin interactions at the molecular level plays a key role in rationalizing the design of new materials with improved performances. A common way to evaluate polymer–resin miscibility and to predict the effect of resin on the mechanical behavior of the final product relies on monitoring the T_g of the vulcanizate upon resin addition. This is often conducted either by calorimetric measurements [12,17,18] or by dynamic mechanical analyses [6–11,19]; in the latter case, the tan δ curve as a function of temperature is measured, whose profile is directly influenced by the change in T_g and provides a way to predict the behavior of the vulcanizate the effect of resins on the dynamic [13–15], viscoelastic [16], and mechanical [17,18,20,21] properties of rubbers.

Solid-state NMR spectroscopy (SSNMR) represents a valuable technique to gain insight into the structure and dynamics of polymeric materials [22–26]. In particular, ¹H timedomain (TD) SSNMR experiments can be exploited to measure ¹H spin–lattice relaxation times in the laboratory frame (T_1) and in the rotating frame (T_{10}), as well as ¹H spin–spin relaxation times (T_2), which depend on the modulation of ¹H-¹H dipolar couplings by molecular motions, thus providing information on the dynamics of polymer chains [27–31]. In addition, in the case of multicomponent systems with domain sizes lower than tens of nm (or nm), the T_1 's (or $T_{1\rho}$'s) of protons in different dynamic environments tend to be averaged to a single value due to the spin diffusion phenomenon, thus allowing an assessment of the structural homogeneity of a sample. Moreover, one can also benefit from magic angle spinning (MAS), high-power decoupling, and cross-polarization (CP) techniques to obtain high-resolution SSNMR ¹³C spectra, which yield both structural and dynamic information on each component of the composite [22,24]. Thanks to the wealth of information that can be obtained, SSNMR has been successfully employed to investigate the dynamic and structural properties of elastomer compounds and vulcanizates. However, to the best of our knowledge, it has not previously been employed to characterize the effect of resins on elastomer structure and dynamics.

In this work, TD and high-resolution SSNMR techniques were applied to investigate uncured and cured styrene/butadiene rubber compounds, either in the presence or absence of the tackifying resin KristalexTM 5140, a low molecular weight ($M_n = 1690$ g/mol) α -methylstyrene/styrene copolymer characterized by excellent thermal stability, a high softening point (413 K) and a glass transition temperature of 363 K [32]. ¹³C high-resolution SSNMR spectra and analyses of on-resonance ¹H free-induction decays (FIDs) were used to investigate the structural properties and dynamic and structural heterogeneity of the samples. Information on the heterogeneity of the samples was also obtained from ${}^{1}HT_{1}$ and $T_{1\rho}$. Moreover, ¹H T_1 and $T_{1\rho}$ measured at different temperatures and at different magnetic fields were exploited to characterize the dynamics of the polymeric chains on a wide range of motion timescales. In particular, we took advantage of field-cycling (FC) NMR relaxometry to measure 1 H T_{1} on a wide range of Larmor frequencies (0.01–35 MHz). The NMR results were discussed in comparison with data on crosslink density and glass transition temperature, two macroscopic properties of interest for rubber applications that are usually affected by the addition of resins. This comparison allowed the effect of the resin on the structure and dynamics of the polymer chains in the rubber compound and the corresponding vulcanizate to be correlated with these applicative properties. Moreover, information could be obtained on the degree of mixing between polymer and resin.

2. Materials and Methods

2.1. Samples

All samples were provided by Pirelli Tyre SpA (Milano, Italy). The sample indicated as SBR consists of a blend of styrene/butadiene rubber (39.5% styrene, vinyl content on the dienic portion 38.5%, M_n = 530,000 g/mol, M_w = 750,000 g/mol), treated distillated aromatic

extract (TDAE) as plasticizer oil, carbon black (N100 series, surface area: 158 m²/g) as a reinforcing filler and a vulcanization package containing sulfur, N-cyclohexyl-2-benzothiazole sulfenamide (CBS), zinc oxide and stearic acid. Sample SBR_k has the same formulation as SBR, but it also contains KristalexTM 5140 resin (poly(α -methylstyrene-co-styrene), Synthomer, $M_n = 1690$ g/mol, $M_w = 4750$ g/mol). vSBR and vSBR_k indicate samples obtained by the vulcanization of SBR and SBR_k, respectively. All compounds were mixed in a 1.5 L internal mixer (Harburg Freudenberger, Hamburg, Germany) in a two-step mixing process. In the first step, all ingredients, except the vulcanization system, were mixed for 200 s, reaching a dumping temperature of approximately 413 K with a rotor speed of 75 rpm. In the second step, the vulcanization system was added, and the compound was finalized by mixing for 120 s at 313 K with a rotor speed of 50 rpm, and the maximum dumping temperature was set at 383 K. vSBR and vSBR_k were vulcanized at 443 K for 10 min.

The composition of the samples is reported in detail in Table 1.

Component	SBR and vSBR	SBR_k and vSBR_k		
Styrene/butadiene rubber	100	100		
TDAE	37.5	37.5		
Carbon black	45	45		
Sulfur	2	2		
CBS	4	4		
Zinc oxide	3.5	3.5		
Stearic acid	2	2		
Kristalex TM 5140	0	15		

Table 1. Sample composition in phr (parts per hundred rubber).

2.2. Differential Scanning Calorimetry and Equilibrium Swelling Experiments

Differential scanning calorimetry (DSC) measurements were carried out with a DSC Mettler-Toledo 820 instrument. Thermal cycles between 183 and 473 K were performed. The cooling/heating rate was 10 K/min. For all samples, T_g was determined as the intersection point of the two tangents to the DSC curve at the endothermic step. The obtained values are reported in Table 2.

Equilibrium swelling experiments were performed in duplicate to determine the crosslink density ($1/M_c$, where M_c is the average molar mass between two adjacent crosslinks) of vSBR_k and vSBR. The samples were weighed and then soaked in toluene for 72 h in the dark. Thereafter, they were dried with absorbent paper and quickly weighed. After being dried overnight in an oven at 343 K under vacuum, the samples were weighed again in order to determine the amount of adsorbed solvent. $1/M_c$ values were then calculated using the Flory–Rehner equation [33]. The obtained $1/M_c$ values are reported in Table 2.

Table 2. T_g values obtained by DSC experiments and crosslink density (1/ M_c) values determined by equilibrium swelling experiments.

Sample	<i>Т</i> _g (К)	$1/M_{ m c}$ (10 $^{-5}$ mol/g)	
SBR	246	-	
SBR_k	249	-	
vSBR	253	1.80	
vSBR_k	257	1.38	

2.3. SSNMR Experiments

¹H on-resonance FIDs were recorded at 303 K on a Niumag permanent magnet interfaced with a Stelar PC-NMR console, working at a ¹H Larmor frequency of 20.8 MHz and equipped with a single-channel static 5 mm probe. Specifically, the mixed Magic Sandwich Echo (MSE) pulse sequence was applied [23] using a total echo duration $\tau_{MSE} = 6 (4\tau_{\varphi} + 2\tau_{90})$, with $\tau_{\varphi} = 1.5 \ \mu s$ and $\tau_{90} = 3.3 \ \mu s$. For compounds and vulcanizates, a dwell time of 1 μs was used and 3k data points were acquired. For KristalexTM 5140, the dwell time was 0.1 μs and 2k data points were acquired. A total of 200 scans were accumulated using a recycle delay of 0.5 s for the compounds and vulcanizates, and one of 1 s for KristalexTM 5140. The experimental MSE ¹H FIDs were analyzed by a discrete approach using a non-linear least-square fitting procedure implemented in the Mathematica[®] environment [34]. For all samples, ¹H longitudinal relaxation time, T_1 , values were also measured at 303 K using the inversion recovery pulse sequence coupled with a solid echo pulse scheme (IRSE), with delay times ranging from 1 ms to 0.5 s in the case of the vulcanized samples and from 1 ms to 1 s for KristalexTM 5140. All IRSE experiments were acquired using 4–16 scans and recycle delays of 0.5 and 1 s for vulcanized samples and KristalexTM 5140, respectively.

 13 C high-resolution SSNMR experiments and 1 H $T_{1\rho}$ measurements were performed on a Bruker Avance Neo spectrometer working at Larmor frequencies of 500.13 and 125.77 MHz for ¹H and ¹³C nuclei, respectively, using a double-resonance 4 mm CP-MAS probe. 90° pulses with a duration of 4.3 μ s and 4.1 μ s were employed for ¹H and ¹³C excitation, respectively. ¹³C direct excitation (DE) spectra under MAS were recorded on vSBR and vSBR_k using a recycle delay of 20 s, optimized to obtain quantitative measurements and accumulating 1600 scans. ¹H-¹³C CP/MAS spectra were recorded on vulcanized samples and on Kristalex[™] 5140 by applying a linear ramp on the ¹³C channel during the contact time, using a constant ¹H spin-lock field of 69 kHz. To investigate CP dynamics, contact times ranging from 0.05 to 4 ms were employed for vSBR and vSBR_k, while for Kristalex™ 5140, contact times ranged from 0.05 to 10 ms; 1000 transients were accumulated, using a recycle delay of 4 s. All ¹³C experiments were recorded at 303 K, applying the SPINAL-64 scheme during acquisition for high-power proton decoupling. An MAS frequency of 5 kHz was used to avoid spinning instability due to the elastic character of the samples. ${}^{1}H T_{10}$ measurements were performed at different temperatures between 303 and 343 K and under static conditions by applying a spin-lock field of 46 kHz for durations ranging from 0.4 to 20 ms.

¹H longitudinal relaxation rates, $R_1 = 1/T_1$, were measured in a 0.01–35 MHz Larmor frequency range at different temperatures by means of a Stelar Spin Master FFC-2000 FC NMR relaxometer. The switching time was 3 ms, while the 90° pulse duration was 10.9 µs, and a single scan was acquired. Prepolarized and non-prepolarized pulse sequences were used below and above 12 MHz, respectively. The polarizing and detection frequencies were set at 25 and 16.3 MHz, respectively, while all other parameters were optimized for each experiment. All the ¹H magnetization curves vs. time were reproduced using a monoexponential function, with errors on ¹H R_1 lower than 3%. Samples were cut into small pieces and loaded in a 10 mm NMR glass tube. Measurements were performed every 10 K from 303 K to 393 K. The sample temperature was controlled within ±0.1 K by a Stelar VTC90 variable temperature unit. Considering the adopted instrumental conditions, R_1 values higher than 1000 s⁻¹ were disregarded.

3. Results and Discussion

3.1. Structural Characterization and Degree of Mixing

The presence of domains with different molecular mobility in the samples was investigated through the analysis of on-resonance ¹H FIDs acquired using the MSE pulse sequence [35]. Indeed, this analysis allowed protons in different dynamic environments to be distinguished on the basis of spin–spin relaxation times, ¹H T_2 , whose values monotonically increase as the degree of mobility increases [35–40].

Figure 1 shows the ¹H FIDs obtained for all samples at 303 K. The curves were fitted to a linear combination of three functions (Equation (1)), each characterized by a decay time $T_{2,i}$ and multiplied by a fractional weight W_i (i = g, e1, e2) [38,41,42].

$$I(t) = I(0) \left(W_g e^{-\left(\frac{t}{T_{2,g}}\right)^2} + W_{e1} e^{-\frac{t}{T_{2,e1}}} + W_{e2} e^{-\frac{t}{T_{2,e2}}} \right)$$
(1)

The first function is a Gaussian function (g), representative of the fraction of protons in rigid environments. The other two are exponential functions (e1, e2), ascribable to protons in regions of intermediate and fast mobility. An example of FID fitting is shown in the inset of Figure 1, while the best fit values of W_i and $T_{2,i}$ are reported in Table 3.



Figure 1. Expansion of the first 0.5 ms of the experimental ¹H FIDs of the investigated samples at 303 K. The inset shows the experimental curve (blue) and the total fitting function (magenta) obtained for vSBR, together with the single contributions of the Gaussian (cyan), intermediate- T_2 (orange), and long- T_2 (brown) exponential functions.

Table 3. Weight percentages (W_i) and $T_{2,i}$ values obtained as best-fitting parameters from the analysis of the ¹H FIDs of the investigated samples through Equation (1).

Sample	Wg (%)	W _{e1} (%)	W _{e2} (%)	$T_{2,g}$ (µs)	$T_{2,e1}$ (µs)	$T_{2,e2}$ (µs)
SBR	8	65	27	22	296	773
SBR_k	13	56	31	25	216	607
vSBR	11	59	30	27	183	615
vSBR_k	19	57	24	26	139	535

For all the analyzed samples, a small proton fraction associated to the Gaussian function was detected, with a very short T_2 on the order of 20–30 µs, ascribable to molecular fragments with highly restricted mobility, including polymer segments involved in physical and chemical constraints, such as entanglements and crosslinks, and in interactions with filler particles. The two exponential components are instead characterized by longer T_2 values ranging from ~100 to ~800 µs, and they account for most of the protons in the polymer chain segments and for protons in liquid-like components, including TDAE and dangling chains.

By passing from the uncured to the vulcanized samples, an increase in W_g and a concomitant decrease in the T_2 values of both exponential functions were observed. This effect can be ascribed to the formation of chemical crosslinks between the polymer chains, causing a reduction in molecular mobility. As the resin was added, an increase in the rigid fraction on the order of 5–8% was observed, accompanied by a decrease in both $T_{2,e1}$ and $T_{2,e2}$ values. Because the ¹H FID of pure resin can be nicely reproduced by a Gaussian function characterized by a T_2 value of 20 µs (Figure S1), the higher W_g values obtained for SBR_k and vSBR_k can reasonably be ascribed to a contribution of the resin protons to the Gaussian component. Indeed, it was estimated that the resin protons account for 10% of the total proton content in both samples. Furthermore, the decrease in $T_{2,e1}$ and $T_{2,e2}$ values reflects a slowdown of the dynamics of all the other components of the samples, mainly the polymer, induced by the presence of resin.

In order to ascribe the regions at different mobilities to different sample components, ¹³C high-resolution SSNMR experiments were carried out on vSBR and vSBR_k. In particular, ¹³C DE/MAS spectra were recorded to obtain quantitative spectra, while ¹H-¹³C CP/MAS spectra selectively highlighted the more rigid components of the samples. In fact, the CP process relies on the magnetization transfer from abundant (¹H) to rare (¹³C) nuclei mediated by the heteronuclear dipolar interaction, enhancing the detection of ¹³C nuclei in close proximity to protons in rigid environments. Furthermore, the CP dynamics is strongly influenced by molecular motions, as they directly affect the extent of dipolar interaction, as well as ¹H $T_{1\rho}$, and can be studied by analyzing CP spectra recorded at different contact times.

The ¹³C DE/MAS spectra acquired for vSBR and vSBR_k are reported in Figure 2a,b. Both spectra presents broad signals, mainly due to the distribution of chemical shifts typical of amorphous materials. Specifically, a group of signals at low chemical shifts (0–50 ppm) is present, ascribable to the aliphatic carbon nuclei of the polymer and TDAE, and to naphthenic carbons in TDAE. In the spectrum of vSBR_k, an additional contribution from CH_2 carbons of KristalexTM 5140 can be observed between 30 and 50 ppm. At higher chemical shifts (100–160 ppm), signals from aromatic and alkene carbons were detected. In the spectrum of vSBR_k, it is also possible to observe low intensity signals at about 167 and 87 ppm due to the first-order spinning sidebands of the tertiary aromatic carbons of the resin. In the ¹³C CP/MAS spectra, recorded with a short contact time of 0.2 ms (Figure 2c,d), signals arising from rigid domains are enhanced, showing broader lines with respect to the DE spectra. In particular, the signals of resin carbons are favored compared to those of SBR, as can be seen from the higher intensity of the spinning sidebands. The isotropic signal and the corresponding sidebands of aromatic tertiary carbons of KristalexTM 5140 were assigned on the basis of the ¹H-¹³C CP spectrum recorded for the resin under the same experimental conditions (Figure 2e). Interestingly, the observation of spinning sidebands for KristalexTM 5140 in the spectra of vSBR_k indicates that the resin maintains a rigid character even when blended with the polymer, in agreement with the results from ¹H FID analysis.

(b)

(a)

250





Figure 2. ¹³C DE/MAS spectra of (**a**) vSBR and (**b**) vSBR_k recorded using a recycle delay of 20 s. ¹H-¹³C CP/MAS spectra of (**c**) vSBR, (**d**) vSBR_k and (**e**) KristalexTM 5140 recorded using a contact time of 0.2 ms and a recycle delay of 4 s. Spinning sidebands are marked with an asterisk; impurities are marked with a filled triangle. All spectra were recorded using an MAS frequency of 5 kHz.

The CP dynamics was then studied for the tertiary aromatic carbons in pure KristalexTM 5140 and in vSBR_k and compared with that of aromatic/alkene carbons in vSBR and vSBR_k (see Supplementary Materials). Interestingly, the ¹H $T_{1\rho}$ of the aromatic protons of the resin in pure KristalexTM 5140 (~9 ms) is significantly longer than that measured in both vSBR_k and vSBR (0.6–1 ms) (Table S1). This seems to be ascribable to the proton spin diffusion between resin and polymer domains occurring in vSBR_k and suggests that in this sample the resin is intimately mixed with the polymer. To this regard, further information could be obtained by the low-resolution measurement of ¹H T_1 : mono-exponential trends for the relaxation curves were obtained for all the samples and ¹H T_1 values of 114, 66 and 74 ms were measured for pure Kristalex, vSBR and vSBR_k, respectively. Although these values are quite similar, these results seem to confirm an intimate mixing (on a tens of nm scale) between resin and polymer in vSBR_k.

3.2. Characterization of Dynamics

The dynamic properties of all samples were investigated by means of FC NMR experiments for the measurement of ¹H spin–lattice relaxation rates ($R_1 = 1/T_1$) at variable Larmor frequencies, from 0.01 to 35 MHz. In fact, ¹H spin–lattice relaxation is driven by the modulation of the ¹H-¹H dipolar couplings by molecular motions. The dependence of ¹H R_1 on frequency (v or $\omega = 2\pi v$), called nuclear magnetic relaxation dispersion (NMRD), can be described by a linear combination of spectral densities, $J(\omega)$, which are the Fourier transform of the autocorrelation functions of motion [43]. For polymers far above T_g , R_1 dispersions are mainly governed by segmental dynamics, that is local motions within the Kuhn segment, activated at the glass transition. Segmental dynamics, also referred to as "glassy" dynamics, is responsible for the so-called α -relaxation. Additional contributions to R_1 dispersions arise from collective motions involving longer and longer chain portions, indicated as polymer dynamics. The contribution to $R_1(\omega)$ arising from segmental dynamics can be well represented using the Cole-Davidson spectral density, while contributions from polymer dynamics result in power law dependences of the type $R_1(\omega) \propto \omega^{-\gamma}$, with different values of the γ exponent depending on the motional regime as defined by the mostly accepted tube–reptation theory [44–46]. Therefore, FC NMR has proven to be an effective technique to characterize the dynamics of polymers far above

glass transition over a broad range of characteristic motion times [47,48]. In particular, it has been successfully applied to obtain information on segmental and polymer dynamics in polymer melts and vulcanized rubbers [49–54]. For this aim, $R_1(\omega)$ data are usually transformed to the susceptibility representation, $\chi''(\omega) = \omega R_1(\omega)$, and $\chi''(\omega\tau_s)$ master curves are built from $\chi''(\omega)$ curves acquired at different temperatures by exploiting the frequency–temperature superposition (FTS) principle [55,56] to determine the values of the correlation time for segmental dynamics, τ_s . At low temperatures, $\chi''(\omega)$ curves show a maximum, and the high-frequency branch of the curves, which mainly arises from the contribution of segmental dynamics ($\chi''_{seg}(\omega)$), can be fitted to Equation (2):

$$\chi''_{seg}(\omega) = \omega K_{CD}[J_{CD}(\omega) + 4J_{CD}(2\omega)]$$
⁽²⁾

using the Cole–Davidson spectral density function:

$$J_{CD}(\omega) = \frac{2\sin[\beta_{CD}\arctan(\omega\tau_{CD})]}{\omega[1 + \omega\tau_{CD}]^{\beta_{CD}/2}}$$
(3)

with $0 < \beta_{CD} \le 1$. Thus, τ_s values can be obtained from τ_{CD} using the relation $\tau_s = \beta_{CD}\tau_{CD}$. At higher temperatures, the $\chi''(\omega)$ curves do not show a maximum, and the frequency axis is scaled until they overlap the curves obtained at lower temperatures to build the master curve. In this case, τ_s can be determined as the frequency scaling factor.

Here, ¹H NMRD curves were acquired for all the investigated samples at different temperatures in the 303–393 K range. In Figure 3a, the curves recorded at 303, 333 and 363 K are reported as examples. In Figure S3, the corresponding $\chi''(\omega)$ curves are also shown.



Figure 3. (a) ¹H NMRD curves for the investigated samples at the indicated temperatures. (b) Susceptibility master curves of the indicated samples; the black line represents the contribution of the segmental dynamics to the $\chi''(\omega\tau_s)$ master curve of SBR_k.

For all samples, at low temperatures, R_1 shows a power law dependence on Larmor frequency with $\gamma \cong 1$, suggesting that R_1 relaxation is dominated by segmental dynamics (Regime 0). Correspondingly, the $\chi''(\omega)$ curves display a maximum associated with the

condition $\omega \tau_s \cong 1$, which shifts towards lower frequencies passing from uncured to vulcanized samples. Moreover, SBR_k and vSBR_k show $\chi''(\omega)$ maxima at lower frequencies than the corresponding samples without KristalexTM 5140. These results indicate a slowdown of segmental dynamics as a consequence of both crosslinking and resin addition, in agreement with the higher T_g values measured by DSC (Table 2). At $T \ge 323$ K, the NMRD curves show two regions with different power law dependences. A region with γ values in the range 0.7–0.8, with γ decreasing by increasing the temperature, was detected at higher frequencies, ascribable to the overlap of Regime 0 with the Rouse regime (Regime I of the tube–reptation model). At lower frequencies, a region with γ of 0.25–0.28 was observed, which is attributable to the Rouse regime. For all samples, R_1 increases with increasing temperature at high frequencies, while it decreases at lower frequencies, with a shift in the crossover point between Regime 0 and Regime I toward lower frequencies, due to the corresponding increase in τ_s .

For each sample, the NMR susceptibility $\chi''(\omega)$ curves obtained at different temperatures were combined together to build $\chi''(\omega\tau_s)$ master curves, under the assumption that the FTS principle holds true, to obtain τ_s values. $\chi''(\omega)$ curves at 303 K were fitted to Equation (2) in order to obtain τ_s . β_{CD} was found to be 0.12, in agreement with the value determined for SBR samples in a previous work [52]. At the remaining temperatures, τ_s values were determined as the frequency scaling factors used to build the master curves. The $\chi''(\omega\tau_s)$ master curves obtained for all samples are shown in Figure 3b, while the values of τ_s as a function of 1000/T are reported in Figure 4a. The obtained $\chi''(\omega\tau_s)$ master curves cover a frequency range of six decades and display characteristic shapes that well reflect the different dynamic regimes described above: at a high frequency, the contribution of segmental dynamics is dominant, while polymer dynamics assumes a significant importance for $\omega\tau_s < 0.1$.



Figure 4. (a) Correlation times for segmental dynamics, τ_s , vs. 1000/*T* and (b) Log τ_s vs. (*T*/*T*_g - 1) for the indicated samples. *T*_g values were determined by DSC measurements (Table 2). Errors on τ_s are lower than 5%.

As expected from the trend of the maxima of the $\chi''(\omega)$ curves, τ_s values are longer in the vulcanized samples (vSBR, vSBR_k) with respect to the uncured ones (SBR, SBR_k), and in samples containing the resin (SBR_k, vSBR_k) with respect to those without it (SBR, vSBR). The increase in τ_s upon vulcanization is ascribable to the formation of crosslinks, which restrict segmental dynamics, also resulting in increased T_g (Table 2). The addition of the resin, although lowering the efficiency of vulcanization, as it was observed when comparing the crosslink density of vSBR and vSBR_k (Table 2), yielded a slowdown of segmental dynamics, paralleled by an increase in T_g . Similar trends in τ_s upon resin addition to rubber matrices were found in investigations of α -relaxation by broadband dielectric spectroscopy [13–15]

For all samples, τ_s values show, as expected, decreasing trends by increasing the temperature (Figure 4a). However, quantitative analyses of these trends in terms of the commonly used Vogel–Fulcher–Tammann (VFT) function to obtain information on polymer fragility is hampered by the limited temperature range investigated. On the other hand, hints on the variation of fragility upon vulcanization or resin addition can be achieved by plotting Log τ_s as a function of $(T/T_g - 1)$ according to Equation (4):

$$Log\left(\frac{\tau_s(T)}{\tau_0}\right) = \frac{Log^2\left(\frac{\tau_s(T_g)}{\tau_0}\right)}{m\left(\frac{T}{T_g} - 1\right) + Log\left(\frac{\tau_s(T_g)}{\tau_0}\right)}$$
(4)

Equation (4) is obtained by recasting the VFT equation [57]. In Equation (4), *m* is the fragility index, τ_0 is the pre-exponential factor of the VFT equation and the value of τ_s at the glass transition ($\tau_s(T_g)$) is, by definition, a constant (100 s) [58]. If τ_0 and *m* do not change significantly, all samples are expected to exhibit the same behavior. Indeed, in our case, using T_g values determined by DSC in ($T/T_g - 1$), a good superposition was observed for the Log τ_s curves of all the samples (Figure 4b). These results indicate that both crosslinking by vulcanization and resin addition do not significantly affect the polymer fragility index, which yields a measurement of the rapidity with which segmental dynamics slow down upon approaching the glass transition by decreasing the temperature. Unchanged polymer fragility upon vulcanization was also reported for rubbers with low crosslinking degrees [52]. On the other hand, the addition of resin was previously found to increase the fragility index at high resin concentrations [13], suggesting that in our case the amount of resin was too low to show such an effect.

For the vulcanized samples, the dynamics at low frequency was further investigated by analyzing ¹H spin–lattice relaxation in the rotating frame ($T_{1\rho}$). Although an estimate of ¹H $T_{1\rho}$ was already obtained by analyzing the CP dynamics, ¹H $T_{1\rho}$ data were more accurately measured at a spinlock field $(\omega_1/2\pi)$ of 46 kHz by means of a dedicated experiment with direct 1 H detection. At all the investigated temperatures (from 303 to 343 K), the relaxation curves can be well reproduced by the combination of two exponential functions $I(t) = I(0) \left(W_a e^{-\frac{t}{T_{1\rho,a}}} + W_b e^{-\frac{t}{T_{1\rho,b}}} \right), \text{ with } W_a \cong 40-90\%, T_{1\rho,a} \cong 0.6-2.3 \text{ ms}, W_b \cong 10-60\%$ and $T_{1\rho,b} \cong 5$ –11 ms, which is very similar for vSBR and vSBR_k, as shown in Table S2. The effect of spin diffusion, which tends to average out differences between intrinsic relaxation times, makes a physical interpretation of the individual $T_{1\rho}$ components very difficult. Nevertheless, dynamic information could be obtained by looking at the population weighted rate average $R_{1\rho}^{PWRA} = \frac{W_a}{T_{1\rho,a}} + \frac{W_b}{T_{1\rho,b}}$, which is not affected by spin diffusion [29] (Table S3). As shown in Figure 5, ¹H $R_{1\rho}^{PWRA}$ decreases by increasing the temperature for both vSBR and vSBR_k, indicating the presence of dynamic processes in the fast motion regime. In agreement with R_1 measurements, higher $R_{1\rho}^{PWRA}$ values were measured for vSBR_k, which shows a higher T_g , indicating a slowdown of dynamics induced by the presence of the resin. In order to better understand the origin of $R_{1\rho}^{PWRA}$ relaxation, the

contribution of segmental dynamics ($R_{1\rho}^{seg}$) at different temperatures was calculated from Equation (5):

$$R_{1\rho}^{seg}(\omega,\omega_1) = \frac{K_{CD}}{2} [5J_{CD}(\omega) + 2J_{CD}(\omega) + 3J_{CD}(2\omega_1)]$$
(5)

The values of τ_s obtained from the analysis of ¹H NMRD curves and reported in Figure 4 were used for the calculation. As we can see, $R_{1\rho}^{seg}$ shows a decreasing trend with increasing temperature, but it accounts for only a fraction, ranging from about 52% at 303 K to 14% at 343 K, of the experimental ¹H $R_{1\rho}^{PWRA}$. This confirms the increasing contribution of collective polymer dynamics in the kHz frequency range as the temperature increases. The contribution of polymer dynamics was estimated as the difference $R_{1\rho}^{pol} = R_{1\rho}^{PWRA} - R_{1\rho}^{seg}$ (Figure 5, Table S3), under the assumption of statistical independence and time-scale separation between segmental and polymer dynamics. $R_{1\rho}^{pol}$ increases by decreasing the temperature, approaching a maximum at 303–313 K that corresponds to the condition $\omega_1\tau_{pol} \cong 1$. These results suggest the presence of Rouse motions with characteristic correlation times (τ_{pol}) on the order of tens of μ s in the investigated temperature range.



Figure 5. ¹H $R_{1\rho}^{PWRA}$ (experimental), $R_{1\rho}^{seg}$ (calculated) and $R_{1\rho}^{pol}$ (calculated) vs. temperature for the indicated samples. The lines are plotted to guide the eye.

4. Conclusions

In this work, the effects of the tackifying resin KristalexTM 5140 on the structural and dynamic properties of SBR compounds of interest to the tire industry were investigated through a combination of high-resolution and TD SSNMR techniques. To this end, uncured and cured SBR compounds, either in the absence or presence of 15 phr of resin, as well as a pure KristalexTM 5140 sample, were studied, and the obtained results were discussed in comparison with T_g and crosslink density data.

From the "macroscopic" point of view, an increase in T_g and a slight decrease in crosslink density were observed as a consequence of resin addition. From the "microscopic" point of view, the resin was found to be intimately mixed with the polymer in the rubber matrix, while maintaining its rigid character. Moreover, the resin induced a slowdown of segmental dynamics of the polymer in both uncured and cured samples, while it did

not affect the polymer fragility and the spectrum of collective polymer motions in the Rouse regime.

In conclusion, the obtained results showed for the first time that SSNMR can provide very useful information on the degree of mixing and interaction between resin and polymer in rubber compounds. This information could help the comprehension of the molecular origin of the observed macroscopic and mechanical properties, which is fundamental to drive research towards formulations with improved and optimized performances.

Supplementary Materials: The following supporting information can be downloaded at: https: //www.mdpi.com/article/10.3390/app13031939/s1, Figure S1: ¹H FID (black points) and Gaussian fitting function (green line) obtained for Kristalex; Figure S2: Integral intensities of the ¹³C CP/MAS signals (points) and fitting functions (black lines) obtained for vSBR (**a**), vSBR_k (**b**) and Kristalex (**c**) as a function of the contact time τ_{CP} ; Figure S3: ¹H NMR susceptibility curves of the investigated samples at the indicated temperatures. Table S1: T_{CH} and $T_{1\rho}$ values obtained from the analysis of the CP curves for the investigated samples; Table S2: ¹H $T_{1\rho,i}$ values and relative weight percentages (W_i) obtained for vSBR and vSBR_k at the investigated temperatures; Table S3: ¹H $R_{1\rho}^{PWRA}$ (experimental), $R_{1\rho}^{seg}$ (calculated) and $R_{1\rho}^{pol}$ (calculated) obtained for vSBR and vSBR_k at the investigated temperatures.

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