

 Analytical pyrolysis with *in situ* silylation - gas chromatography coupled with mass spectrometry (Py(HMDS)-GC/MS) was used to distinguish the various types of Asian 26 lacquers and identify additional original or restoration materials, by using a decisional 27 scheme based on the presence of characteristic pyrolytic profiles and molecular markers. Scanning electron microscopy with energy-dispersive X-ray spectrometry (SEM-EDS) was used on cross sections to study the stratigraphy of the objects and understand the artistic techniques. The results revealed a multifaceted picture, as some of the objects showed interesting mixtures of lacquers, such as *urushi* and *thitsi*, or unexpected compositions, which questioned their initial attributions. Complex or unexpectedly simple preparations and restoration treatments of the decorative surfaces were highlighted too. The data are here presented in a systematic way, showcasing the interest of these uncommon analytical scenarios and the interpretational challenges provided by such complex museum objects.

Keywords: Asian lacquers; analytical pyrolysis; *in situ* silylation; GC/MS; SEM-EDS;

provenance

1. INTRODUCTION

 Asian lacquers have been used since ancient times in East Asia as coatings for all kinds of surfaces, and lacquered objects have always been much-appreciated because of the beauty, brightness, toughness, durability and waterproofing of their coatings [1, 2]. These objects and artworks are widespread in Oriental art, but they have also become 47 extremely popular in Europe since the $16th$ century [3], and have been collected since then.

 As far as museums are concerned, lacquered objects often enter the collections from very diverse geographical regions and sometimes after multiple stops along the way. Provenance is often attributed based on art historical knowledge, and, even when it is officially recorded, debates and questions easily arise [4]. Given the popularity of lacquered objects, European craftsmen started imitating and paraphrasing them, using locally available materials and techniques to closely reproduce the glossy luxury of their Oriental equivalents [5]. It is therefore extremely important to determine the chronology, manufacture origin or even authenticity of such objects to avoid misleading attributions [4]. The preparation of the lacquer coating is a complex process, in which mixtures of several materials are used in multiple layers [6-8]. The complexity is enhanced by the possible presence of additional substances, such as plant oils and pigments, used to

produce the lacquered decoration. The identification of all these materials is the first

- crucial step to understand the overall composition of an object, solve problems of
- conservation and authenticity, differentiate between the different Asian lacquers, point
- towards provenance and distinguish European from Asian productions [9-11]. If the

 exact preparation process and materials used are identified, hypotheses on specific production centres are also possible [7].

 The raw material of Oriental lacquerware is produced from the sap of three lacquer trees: *Rhus vernicifera* (China, Japan and Korea), *Rhus succedanea* (Vietnam and Taiwan), and *Melanorrhoea usitata* (Laos, Myanmar, Cambodia and Thailand) [12, 13]. Although the geographical distribution of these trees is a more complex topic than what 71 appears in the literature [14], the use of their sap is mostly reported in agreement with the above-mentioned regions. The chemical composition of all saps is a complex mixture of catechol and phenol derivatives (60–65%), proteins (glycoproteins (2%) and a laccase enzyme (1%)), polysaccharides (7%) and water (30%) [15]. The laccase enzyme is the responsible trigger for the polymerisation reaction that leads to the formation of the hardened lacquer film [16]. The catechol/phenol mixture composition is different in the three trees and such difference is maintained in the final polymer, thus enabling the three types of lacquers to be chemically distinguished [17, 18]. The main components of the product extracted from *Rhus vernicifera* are referred to as urushiol; laccol is obtained from *Rhus succedanea*, and thitsiol from *Melanorrhoea usitata* [19]. Urushiol 81 and laccol main compounds are catechol derivatives with C15 and C17 alkenyl chain, respectively. Thitsiol contains additional catechol derivatives with an ω-phenylalkyl chain of C10 and C12 [15, 20]. Relatively abundant research has been undertaken to characterise Asian lacquers, by focusing on finding molecular markers to identify the three different saps [21, 22], 86 studying the interactions between plant oils (used as additives in lacquer formulations) and lacquers [23] and investigating the pathways of lacquer degradation [24, 25]. Due

88 to the high polymerisation degree and complex chemical composition of oriental

89 lacquers, pyrolysis-gas chromatography coupled with mass spectrometry (Py-GC/MS) is the most suitable analytical approach for their chemical characterisation and for their identification in samples of unknown composition [26-28]. Due to the high number of pyrolysis products obtained and the difficulty of interpretation, a remarkable effort has been made to systematise the identification process by using the AMDIS software in combination with a complex Excel spreadsheet [27]. The polar nature of most pyrolysis products derived from lacquer makes a derivatisation step highly recommended. Both methylation (TMAH) [8, 9, 27] and silylation (HMDS) [15, 20] have been proposed in the literature with good results. Advantages and disadvantages of these derivatisation methods are mostly related to the identification of materials other than lacquer possibly present in a sample, with TMAH being particularly suitable for lipids, and HMDS being more suitable for proteins and polysaccharides [29]. Regardless of the derivatisation method, the strategy to differentiate among lacquers is based on the study of specific pyrolytic profiles and the identification of molecular markers [9, 12, 13, 15, 17, 20, 22, 26, 27, 30-33]. The study presented here was undertaken in order to delve into some significant objects from the Asian art collection at the Museum of Zaragoza, Spain ("Federico

Torralba" collection) [34], with the aim to characterise the complex mixtures of

materials possibly present, assess the provenance of the objects and describe the

analytical difficulties arising from the study of these types of objects. To support and

guide the data interpretation, a simple decisional scheme was implemented, which was

- then applied to the data obtained from the museum objects. Because many of the
- lacquerwares were coloured or decorated, scanning electron microscopy (SEM) coupled

to energy-dispersive X-ray spectrometry (EDS) was used as complementary technique to

- Py(HMDS)-GC/MS, in order to determine the elemental composition of the coatings and 114 study the artistic technique.
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2. MATERIAL AND METHODS

2.1. Objects and samples

- Eight lacquered objects were selected from the Asian art collection at the Museum of
- Zaragoza (Spain). The objects include three food containers, three boxes, one plate and
- one Buddha sculpture. They were selected in order to cover a chronological period from
- 121 the $16th$ to the $20th$ century and represent different possible provenances (China, Japan,
- Myanmar and Thailand), according to stylistic interpretation. Table 1 includes an image
- and a description of the objects under investigation. Some of these lacquerwares show
- a European influence, as they were probably intended for the export market (49009,
- 125 49189 and 54562) and are referred to as *Nanban* art objects ¹ (49189 and 54562).

¹ This style was developed in Japan between the $16th$ and $17th$ centuries, and the word "*Nanban*" means "Southern barbarians", because it was linked to the contacts and trade with European people, especially from Portugal.

Museum	Object	Description	Decoration	Chronology	Provenance	Size
code						(l, w, h: cm)
49005		Htamin-gyaing	Black (exterior) and	First half of the 20 th	Myanmar	23 (\emptyset), 48.5
		(food container	red (interior)	century		
		made of bamboo)	lacquer; kyaukka			
			decoration			
49006		Htamin-gyaing	Red and black	First half of the 20 th	Myanmar	21.8 (\varnothing), 45
		(food container	lacquer, with yellow,	century		
		made of bamboo)	red and black			
			decoration; yun			
			(incised) technique			
49188		Food container	Black lacquer with	First half of the 20 th	Thailand or	29.5 (\emptyset) , 39
		(basketry)	red and gold	century	Myanmar	
			decoration			
49189		Nanban yōhitsu	Black lacquer with	End of the	Japan	45.2, 24.9,
		(small chest)	maki-e and raden	Momoyama period		29.5
			decoration	or beginning of the		
				Edo era, 17 th century		

126 Table 1. Summary of the lacquered objects from the Asian art collection at the Museum of Zaragoza included in this research.

 Two sets of micro-samples (with total diameter less than 1 mm) were taken from all the 128 objects by using a scalpel and then stored in glass vials. To minimise sampling of these precious objects, one set of micro-samples was used for SEM-EDS examination and the other one for Py(HMDS)-GC/MS. Monochrome and non-decorated micro-samples were mainly used for Py(HMDS)-GC/MS analyses, and coloured micro-samples were selected for SEM-EDS.

2.2. Py(HMDS)-GC/MS

 Analytical pyrolysis was performed using 1,1,1,3,3,3-hexamethyldisilazane (HMDS, chemical purity 99.9%, SigmaAldrich Inc., USA) as a silylating agent for the *in situ* derivatisation of pyrolysis products. The instrumentation consisted of a micro-furnace Multi-Shot Pyrolyzer EGA/Py-3030D (Frontier Lab) coupled to a gas chromatograph 6890 (Agilent Technologies, Palo Alto, CA, USA), equipped with an HP-5MS fused silica 140 capillary column (stationary phase 5% diphenyl and 95% dimethyl-polysiloxane, 30 m \times 141 0.25 mm i.d., Hewlett Packard, USA) and with a deactivated silica pre-column (2 m \times 0.32 mm i.d., Agilent J&W, USA). The GC instrument was coupled to an Agilent 5973 Mass Selective Detector operating in electron impact mode (EI) at 70 eV. The micro-144 sample (*ca*. 100 µg) was admixed with 3 µL HMDS into a stainless steel cup and inserted into the micro-furnace. The pyrolysis temperature was 550°C and interface temperature was 280°C. The split/splitless injector was used with a 1:20 split ratio and kept at 280°C. Chromatographic conditions were as follows: initial temperature 36°C, 10 min 148 isothermal; 10° C min⁻¹ up to 280°C, 2 min isothermal; 20° C min⁻¹ up to 310°C, 50 min isothermal. Helium (purity 99.995%) was used as carrier gas with constant flow 1.0 mL min^{-1} .

2.3. SEM-EDS The observations and analyses were obtained by Field-Emission Scanning Electron Microscopy (FESEM) with Energy-Dispersive X-ray Spectrometry (EDS); they were performed with a Merlin™ FESEM microscope equipped with a Gemini column (both from Carl Zeiss Nano Technology Systems, Germany), and coupled with an X-Max X-ray microanalyzer (Oxford Instruments, UK). Cross-sections of the micro-samples were prepared by embedding them in epoxy resin, polished with diamond polishing suspensions, coated with carbon and fixed to the FESEM holder. The FESEM observations and EDS analyses were performed at 15.0 kV and 300 pA. Polished cross-sections were previously observed by Optical Microscopy (OM) using a Zeiss Axio Imager Reflected-light Microscope (Carl Zeiss, Jena, Germany) with polarised light. 3. RESULTS AND DISCUSSION *Data analysis* The pyrolysis data were interrogated with the aim to highlight pyrolytic profiles and molecular markers based on what is reported in the literature [9, 12, 13, 15, 17, 20, 22, 26, 27, 30-33]. A summary of these profiles and markers is reported in Table 2, together with the mass-to-charge (*m/z*) values used to extract chromatographic profiles of non-derivatised compounds, as well as methylated and silylated derivatives.

172 Table 2. Summary of pyrolytic profiles and molecular markers commonly used in the literature to distinguish Asian lacquers [9, 12, 13, 15, 17, 20, 173 22, 26, 27, 30-33]. The *m/z* values of the ions used to extract them are reported together with additional description of the profiles obtained. The 174 classes of compounds are indicated with abbreviations: C – aliphatic hydrocarbons; B – alkylbenzenes; Ph – alkylphenols; Ct – alkylcatechols; ωPh 175 – ω-alkylphenylphenols; ω Ct – ω-alkylphenylcatechols; K – alkylphenylketones; Ct_{ox} – acid alkylcatechols; Ph-A – alkylphenyl carboxylic acids; Ox-176 A – alkyl-oxo-phenylcarboxylic acids. The carbon atoms in the alkyl chains and the unsaturations are indicated with numbers: Cn:m – n=number

177 of carbon atoms; m=number of unsaturations.

178 * methylated - **trimethylsilylated

 Based on previous published research, as well as on the experience of the authors, a decisional scheme was implemented and is reported in Figure 1. The scheme shows the *m/z* values for silylated compounds, but it can be used in the case of methylation or non-derivatised compounds by using the *m/z* values reported in Table 2.

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185 Figure 1. Proposed decisional scheme for the differentiation of *urushi*, laccol and *thitsi*

186 using Py(HMDS)-GC/MS. Further confirmation of identification is then obtained by

187 extracting the profiles of alkylphenols (Ph, *m/z* 180) and bis-TMS alkylcatechols (*m/z*

188 253) in the same way proposed for mono-TMS alkylcatechols. The corresponding *m/z*

189 values for these additional profiles are reported in Table 2.

190

191 FESEM-EDS results were used to distinguish decoration layers and microstructure, and

192 to identify the pigments, in order to improve the knowledge of the lacquer artistic

193 techniques. In the following paragraphs, the results are presented according to the area of

provenance of the objects.

Objects 49005, 49006 and 49188 – food containers, Myanmar and Thailand, early 20th

century

- The total ion chromatograms (TICs) of the samples taken from objects 49005 and 49006
- are similar (Figure 2a), and show the whole molecular pattern generally reported for
- *thitsi* [24], including the distinctive profile of alkylbenzenes (B, *m/z* 91, Figure 2b) and
- alkylphenylketones (K, *m/z* 105, not shown).
- The profile of alkylcatechols (Ct, *m/z* 179, Figure 2c) is more complex but still contains
- all the features expected for *thitsi*, including the molecular markers, such as 3-(10-
- phenyldecyl)catechol (ωCt10) and 3-(12-phenyldodecyl)catechol (ωCt12), despite their
- 206 low relative abundance. A more successful visualisation of the molecular markers was
- generally obtained by directly extracting the *m/z* values corresponding to their
- molecular ions (Table 2). A cleaner profile was obtained from the alkylphenols (Ph, *m/z*
- 180, not shown), showing the typical Gaussian profile and the molecular markers 2- and
- 3-(10-phenyldecyl)phenols (ωPh10), and 2- and 3-(12-phenyldodecyl)phenols (ωPh12).
- The first part of the pyrolytic profile of aliphatic hydrocarbons (C, *m/z* 57, Figure 2d)
- shows saturated and unsaturated compounds up to C17, in accordance with the
- composition of *thitsi*, which naturally contains alkylcatechols with up to 17 carbon
- atoms. All these features satisfy the steps described in the decisional scheme reported
- in Figure 1 for the straightforward identification of *thitsi*. However, at higher retention
- times, longer-chain aliphatic hydrocarbons (C23-C35) are present. The Gaussian shape
- 217 of the profile and the distribution of the hydrocarbons is in good agreement with the

- 221 Other materials are commonly found mixed with lacquer, including oils, proteins and
- sugars [27, 36, 37]. No molecular markers for these materials were identified in sample
- 49006, whereas some markers for lipids were detected in sample 49005, in particular
- nonandioic acid (azelaic acid), thus suggesting the presence of a drying or semi-drying
- 225 oil, which is commonly added to Asian lacquer formulations [7, 24, 27].
- 226 These results confirm the Southeast Asian provenance of both objects, as relatively pure
- *thitsi* appeared to be applied on their surfaces, in agreement with traditional practice in
- Myanmar.
-

231 Figure 2. Chromatographic profiles obtained by Py(HMDS)–GC/MS of sample 49006. a)

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233 propylbenzene, B4) butylbenzene, B5) pentylbenzene, B6) hexylbenzene, B7)

234 heptylbenzene, B8) octylbenzene, B9:1) nonenylbenzene, B10) decylbenzene, B11:1)

- 235 undecenylbenzene, B12) dodecylbenzene, c) extracted ion chromatogram (*m/z* 179),
- 236 showing Ct4) 3-butylcatechol (TMS), Ct5) 3-pentylcatechol (TMS), Ct6) 3-hexylcatechol
- 237 (TMS), Ct7) 3-heptylcatechol (TMS), Ct8) 3-octylcatechol (TMS), Ph15) 3-
- 238 pentadecylphenol (TMS), Ct15) 3-pentadecylcatechol (TMS), ωPh10) 3-(10-

 phenyldecyl)phenol, ωPh12) 3-(12-phenyldodecyl)phenol; d) extracted ion chromatogram (*m/z* 57), showing C10) decane, C11) undecane, C12) dodecane, C13) tridecane, C14:1) tetradecene, C14) tetradecane, C15) pentadecane, C16) hexadecane, C17) heptadecane, C23) tricosane, C24) tetracosane, C25) pentacosane, C26) hexacosane, C27) heptacosane, C28) octacosane, C29) nonacosane, C30) triacontane, C31) hentriacontane, C32) dotriacontane, C33) tritriacontane, C34) tetratriacontane, C35) pentatriacontane.

 Another food container (object 49188), supposedly from Thailand, was present among 248 the objects under investigation. The TIC does not present the typical pyrolytic profile of *thitsi*, which would be expected in the case of a Thai lacquered object. A series of compounds not related to the presence of an Asian lacquer were identified (Figure S1a, Supplementary Information), including, short-chain aliphatic carboxylic acids, glycerol, phthalic anhydride, several phthalates, ethylene and diethylene glycols and polysaccharide pyrolysis products. These last compounds (mostly levoglucosan derivatives) might indicate the presence of a cellulose- or starch-based material, but contamination from the wooden support cannot be excluded. Short-chain aliphatic carboxylic acids, as well as small amounts of palmitic and stearic acids, suggest the presence of a lipid material. Phthalic anhydride, phthalates and glycerol were also detected in the same sample. These are known additives in alkyd resin formulations [38], which are modified drying oils [39]. The lack of a pyrolytic profile of carboxylic 260 acids similar to the ones normally obtained for drying oils (presence of azelaic acid and high relative abundance of palmitic and stearic acids [40]) makes the identification of 262 the lipid material uncertain. Nonetheless, it has to be noted that the derivatisation of

263 the sample was not completely successful. Additionally, the presence of a broad tailed chromatographic peak at 21.35 min, whose mass spectrum showed the isotopic pattern 265 of Hg, was highlighted. This is reported in the literature as indicative of the presence of vermillion (HgS) [41]. As the derivatisation reaction is generally affected by the presence 267 of inorganic compounds, it is possible that the carboxylic acid profile was consequently affected, not allowing the straightforward identification of the alkyd resin.

269 The TIC also shows long-chain aliphatic hydrocarbons indicative of the presence of a mineral wax, but the distribution of the hydrocarbons is different compared to the 271 paraffin wax identified in the two samples previously discussed. However, the profile is still consistent with reported paraffin waxes recovered from various petroleum products [35].

274 Despite the predominance of peaks not related to lacquer in the TIC, the use of extract ion chromatograms (EICs), as specified in the scheme of Figure 1, enabled the presence of *thitsi* to be assessed. In particular, the pyrolytic profile of aliphatic hydrocarbons (C, *m/z* 57, Figure S1d, Supplementary Information) is in good agreement with the general profile of *thitsi* and similar to those obtained for objects 49005 and 49006. The pyrolytic profile of alkylphenols (Ph, *m/z* 180, Figure S1c, Supplementary Information) also shows several compounds, with Ph7 as the most abundant one and with ωPh10 and ωPh12 clearly detectable in the profile. Surprisingly, the pyrolytic profile of alkylbenzenes (B, *m/z* 91, Figure S1b, Supplementary Information) did not show the typical Gaussian profile of *thitsi*, which is considered one of the most reliable features to identify the lacquer. Ageing, or possibly interactions with other materials, might have affected the lacquer molecular composition, or pyrolytic behaviour. This result is important because 286 it shows how complex the interpretation of the results can be when additional materials 287 from restoration are present in the same sample. Nevertheless, the identification of *thitsi* based on the molecular markers is solid and the provenance of the object is thus confirmed.

 The cross sections of samples from objects 49006 and 49188 were studied by FESEM- EDS (Figure 3). The decoration of lacquered objects is a complex process, which requires 292 the application of several layers [1]. In the case of these objects, the wooden support was coated with a clayey powder, a process referred to as *shitachi* (surface treatment in Japanese) or, more appropriately *thayo* ground, for Burmese objects. This clayey layer was sealed with a layer of black lacquer about 50-80 µm thick (Figures 3a and 3c), which usually constitutes the finished ground for further decoration. The coloured layer was in 297 fact applied on the top of the black lacquer layer.

298 In the case of sample 49006, the red colour was obtained by vermilion (HgS) mixed with 299 lacquer, resulting in a layer about 10 µm thick (Figures 3a and 3b). Most of the HgS particles are less than 1 µm long. An additional top layer is also clearly present (Figure 301 3b, marked with an arrow), about 3-4 μ m thick, whitish in colour in OM (Figure 3b). The inorganic elemental composition, as determined by EDS, showed the presence of Si, Al and some other minor elements, which suggest an environmental contamination. The layer, mostly organic, might therefore correspond to an original finishing layer or a protective layer applied during a conservation treatment. Nonetheless, the stratigraphic analysis matched the visual observations and decoration design, thus confirming the object to be traditionally decorated using the *yun* (incised) technique [7]. In the case of sample 49188, the red layer (Figures 3c and 3d), whose thickness is around 15 µm, was prepared using vermilion (HgS), with the addition of gypsum (calcium sulphate, CaSO4). Ca, S and O were detected by EDS and gypsum particles are marked with an arrow in

- 311 Figure 3d. The HgS particles are averagely about 2 μ m long (slightly bigger than for
- 312 object 49006) and traces of iron oxides were also detected.

313

315 Figure 3. Images obtained by FESEM of the cross-sections of samples: a) 49006,

- 316 secondary electron (SE) image showing the whole stratigraphy; b) 49006, SE image of
- 317 the pigment at the surface (white particles in the image) and optical photomicrograph
- 318 of the top layer; c) 49188, SE image showing the whole stratigraphy and optical
- 319 photomicrograph of the top layer; **d**) 49188, SE image of the red pigment (white
- 320 particles in the image; arrow marks a calcium sulphate inclusion, in grey).
- 321

Objects 49189 and 54562 – boxes, Japan, 16th -17th 322 *century – and object 48990 - Buddha sculpture, Japan, early 19th* 323 *century*

 The two boxes are considered the oldest examples among the items selected for this 326 study and they are supposed to be made in Japan, possibly for the export market. In the case of object 49189, the TIC shows the presence of aliphatic carboxylic acids among the most abundant pyrolysis products (Figure S2a, Supplementary Information). It has been proved in a previous study that, when a drying oil is mixed with *urushi*, the curing of the two materials is mutually affected [23]. In particular, the pyrolytic profile of the oil shows short-chain aliphatic carboxylic acids (C5-C9) with high relative abundance and azelaic acid (diC9) is present with lower relative abundance compared 333 to a drying oil curing on its own. The results from this sample were in perfect agreement with these observations, thus suggesting the presence of a drying oil mixed with an Oriental lacquer. Additionally, a series of sesquiterpenes including cedrene, cuparene, cadalene and mayurone were identified. These compounds are all present in essential oils extracted from trees of the *Cuprassaceae* family and are reported as possible additives during the preparation of urushi lacquer [27]. The presence of the lacquer was assessed by using the EICs, following the steps described for *urushi* in Figure 1. In particular, the alkylbenzenes (B, *m/z* 91, Figure S2b, Supplementary Information) show homologues up to nonylbenzene (B9); the alkylcathecols (Ct, *m/z* 179, Figure S2c, Supplementary Information) show that the most abundant peak is the one attributed to 3-heptylcatechol (Ct7), and the aliphatic hydrocarbons (C, *m/z* 57, Figure S2d, Supplementary Information) show tetradecene (C14:1) and pentadecane (C15) as the most abundant homologues. 3- pentadecylcatechol (Ct15) and 3-pentadecylphenol (Ph15) are present with very low relative abundance, but still detectable, especially visible extracting the *m/z* values corresponding to their molecular weights (Table 2). These results enabled *urushi* to be

 identified as the Asian lacquer used to decorate this object. Paraffin wax was also present in the sample, as previously observed in other objects.

 A cross-section of sample 49189 was observed by FESEM-EDS. This micro-fragment corresponds to a decorated surface in black with gold powder (*maki-e* decoration). A layered structure is observed (Figure 4a), but this is more complex compared to the Burmese samples. The ground layer is a clayey coating, as identified by elemental analysis by EDS, with some inclusions composed of Ca, S and O, and sealed by the lacquer (Figures 4a and 4b). Two different sub-layers of lacquer can be distinguished (Figure 4a, marked with dashed arrows). The first one shows an irregular thickness of 5- 15 μ m and the lacquer is mixed with a finely ground powder, following the typical Japanese *shitachi* process. The second sub-layer is about 20 µm thick. This layer was probably smoothed and polished. The outer coating (about 5 µm thick) corresponds to 361 the decoration and is composed of golden particles in a red layer. The pigment used is an extremely fine iron oxide powder (Figure 4b, plain arrow) and a lacquer/oil mixture was probably used as adhesive. The chemical composition of the metal particles is gold, 364 with traces of silver, and they are less than 10 μ m long and extremely thin (see Figure 4b). On the top of the golden design, along the cross-section surface, a final layer is observed, composed of C, O, Ca, and others (Figure 4b, dashed arrow). According to the *shitachi* process, after the final lacquer polishing, a last coating with at least three fine lacquer layers is applied, although some sophisticated lacquerwares could be painted with dozens of layers of lacquer coatings [1]. All these observations were consistent with the *maki-e* decoration technique, which is a very traditional and widespread type of sprinkled-gold design commonly seen on Japanese lacquer objects [42].

374 Figure 4. Images obtained by FESEM of the cross-sections of samples: a) 49189, 375 backscattering electron (BSE) image showing the whole stratigraphy and an optical 376 photomicrograph of the top layer; b) 49189, secondary electron (SE) image of the 377 surface decoration (plain arrow indicates iron oxide particles and dashed arrow marks 378 the top upper layer); c) 48990, SE image and optical photomicrograph showing a 379 decoration made of bone or ivory on the top (Ca, P and O detected by EDS) and with 380 arrow pointing to the gold foil on the lacquer; **d**) 49009, BSE image showing the whole 381 stratigraphy and optical photomicrograph of the red layer. 382 383 In the case of object 54562, the TIC (Figure 5a) shows the presence of aliphatic 384 carboxylic acids with a similar distribution to that observed for object 49819, therefore

385 indicating the use of a drying oil. Camphor was identified, which is another known

 additive of *urushi* formulations [27]. A paraffin wax was also present, as suggested by the presence and distribution of long-chain aliphatic hydrocarbons, as previously discussed.

 The EIC of the alkylbenzenes (Figure 5b) is in agreement with the profile reported for *urushi*, but the EIC of the alkylcatechols (Figure 5c) clearly shows the presence of all the *thitsi* markers, thus placing the sample in none of the scenarios shown in the decisional scheme of Figure 1. In the case of this sample, the profile of hydrocarbons (Figure 5d) was not useful to distinguish between *urushi* and *thitsi*, as these compounds overlap with the ones deriving from the paraffin wax. Overall, the lack of the alkylbenzenes typical of *thitsi* together with the presence of *thitsi* molecular markers makes the interpretation of the results challenging. A mixture of *urushi* and *thitsi* is the most likely hypothesis, but it is difficult to ascertain whether this mixture was originally used to 398 decorate the object or is the result of a later addition. The design of the box definitely places it as a Japanese object, but it is suspected to be made for the export market. The use of *thitsi* imported to Japan and used in mixture with *urushi* for Japanese export 401 lacquer is reported since the $17th$ century [35]. This object could therefore be one of the few scientifically-confirmed examples of this practice. In the hypothesis of an original mixture of *urushi* and *thitsi*, it is reasonable to think that the two lacquers mixed together would polymerise differently than when they polymerise separately. In particular, *urushi* forms C-C aromatic nucleus-side chain coupling bonds, C-O phenolic oxygen-side chain coupling bonds and C-C bonds between side chains [43], whereas in *thitsi* nuclear-nuclear C-C couplings are predominant [44]. A co-polymerisation would 408 therefore affect the distribution of dimeric bonds and consequently the pyrolytic

415 chromatogram (*m/z* 179); d) extracted ion chromatogram (*m/z* 57). Refer to Figure 2

These two objects are believed to be produced in China, therefore the use of *urushi*

lacquer (*qi* lacquer in Chinese) was expected.

438 In the case of object 49009, the pyrogram is dominated by the presence of aliphatic

carboxylic acids, including dicarboxylic acids (sebacic, azelaic and suberic acids) with

high relative abundance and a characteristic distribution indicative of the use of a drying

oil, possibly *tung* oil [23, 27, 45]. The pyrogram also enabled alpha-cedrene, beta-

- cedrene and cedrol to be identified. These are the main components of the essential oil
- from cedar, another known additive in lacquer formulations [27].
- 444 The extracted ion chromatograms of alkylbenzenes and alkylcatechols are in perfect
- agreement with the chromatographic features of *urushi* (Figure 1, Table 2), therefore
- pointing towards the authenticity of the object as being produced in China with
- traditional materials used in lacquer preparation. Additionally, a series of compounds
- referred to as alkylphenylcarboxylic acids was identified and these compounds form
- when the drying oil is mixed with the lacquer [23, 24].
- A very similar composition was found for object 48165, which was therefore also
- confirmed as a traditional Chinese lacquered object.
- 452 In the case of object 49009, the surface was not shiny and this led us to consider its
- 453 study by FESEM-EDS. Observations of a micro-sample from this plate highlighted a
- different way of decorating the surface compared to all the other samples. In fact, the
- object is coated with a first layer of lacquer probably containing calcium carbonate (Ca,
- C and O detected by EDS analysis). However, instead of a second sealing coating with
- lacquer, a red pigmented lacquer layer is directly applied onto the ground (Figure 4d).
- The decorative layer is 10-15 µm thick and is composed of vermilion with pigment
- particles with 2-4 µm of total diameter. Finally, a thicker transparent finishing layer (15-
- 30 μ m thick) is applied on the top of the coloured layer. The irregular thickness of the
- layers and the lack of some expected layers seem to be the result of a decoration
- process carried out with poor polishing and for an object of poor quality or value.
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4. CONCLUSIONS

 This study presents the results of an investigation carried out to assess the provenance of a series of lacquered objects from the Museum of Zaragoza (Spain). The lacquer 467 identification was carried out based on the pyrolytic profiles and molecular markers reported in the literature and following a decisional scheme which was implemented to help the data analysis in order to distinguish the three Asian lacquers - *urushi*, laccol and *thitsi*. In addition, some of the reported cases represent analytical situations, which are often neglected in the literature. In particular, a possible mixture of *urushi* and *thitsi* is discussed, as well as examples of *urushi* and *thitsi* identified after careful examination 473 despite the predominant presence of restoration materials. The chemical information obtained by Py(HMDS)-GC/MS was coupled with the study of 475 the decoration techniques by FESEM-EDS. The study of the cross-sections added 476 significant information about the nature of the pigments (mercury sulphide, iron oxides, 477 gold powder, etc.) used to prepare the coloured layers and highlighted different ways of applying the coating layers. Traditional techniques (*yun*, *maki-e*, *shitachi*, *shwe-zawa*, *etc*.) were recognised and some chemical considerations were supported. The overall results allowed us to confirm the geographical provenance of six out of the eight objects investigated, whereas for two of them this information must be re- evaluated. This confirms the robustness of Py(HMDS)-GC/MS as analytical technique to characterise and distinguish Asian lacquers, as well as the importance of undertaking scientific analysis to support historical and art historical information on museum objects.

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