1	Challenges in the characterisation of complex formulations of Asian lacquers from
2	museum objects by pyrolysis gas chromatography/mass spectrometry
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17	Abstract
18	A group of lacquered objects belonging to the Asian art collection of the Museum of
19	Zaragoza (Spain) were chemically studied with the aim to investigate the composition of
20	their coatings and possibly confirm their authenticity and geographical provenance.
21	Based on stylistic and art historical information, the objects are dated from the 16 <sup>th</sup> to
22	the 20 <sup>th</sup> century and are hypothetically produced in Japan, China, Myanmar and
23	Thailand.

24 Analytical pyrolysis with in situ silylation - gas chromatography coupled with mass 25 spectrometry (Py(HMDS)-GC/MS) was used to distinguish the various types of Asian 26 lacquers and identify additional original or restoration materials, by using a decisional scheme based on the presence of characteristic pyrolytic profiles and molecular 27 markers. Scanning electron microscopy with energy-dispersive X-ray spectrometry 28 29 (SEM-EDS) was used on cross sections to study the stratigraphy of the objects and understand the artistic techniques. 30 31 The results revealed a multifaceted picture, as some of the objects showed interesting mixtures of lacquers, such as *urushi* and *thitsi*, or unexpected compositions, which 32 questioned their initial attributions. Complex or unexpectedly simple preparations and 33 34 restoration treatments of the decorative surfaces were highlighted too. The data are

36 analytical scenarios and the interpretational challenges provided by such complex

here presented in a systematic way, showcasing the interest of these uncommon

37 museum objects.

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39 *Keywords*: Asian lacquers; analytical pyrolysis; *in situ* silylation; GC/MS; SEM-EDS;

40 provenance

### 42 1. INTRODUCTION

Asian lacquers have been used since ancient times in East Asia as coatings for all kinds
of surfaces, and lacquered objects have always been much-appreciated because of the
beauty, brightness, toughness, durability and waterproofing of their coatings [1, 2].
These objects and artworks are widespread in Oriental art, but they have also become
extremely popular in Europe since the 16<sup>th</sup> century [3], and have been collected since
then.

As far as museums are concerned, lacquered objects often enter the collections from 49 very diverse geographical regions and sometimes after multiple stops along the way. 50 51 Provenance is often attributed based on art historical knowledge, and, even when it is 52 officially recorded, debates and questions easily arise [4]. Given the popularity of 53 lacquered objects, European craftsmen started imitating and paraphrasing them, using 54 locally available materials and techniques to closely reproduce the glossy luxury of their 55 Oriental equivalents [5]. It is therefore extremely important to determine the chronology, manufacture origin or even authenticity of such objects to avoid misleading 56 57 attributions [4]. 58 The preparation of the lacquer coating is a complex process, in which mixtures of

59 several materials are used in multiple layers [6-8]. The complexity is enhanced by the 59 possible presence of additional substances, such as plant oils and pigments, used to 61 produce the lacquered decoration. The identification of all these materials is the first 62 crucial step to understand the overall composition of an object, solve problems of 63 conservation and authenticity, differentiate between the different Asian lacquers, point 64 towards provenance and distinguish European from Asian productions [9-11]. If the exact preparation process and materials used are identified, hypotheses on specificproduction centres are also possible [7].

The raw material of Oriental lacquerware is produced from the sap of three lacquer 67 68 trees: Rhus vernicifera (China, Japan and Korea), Rhus succedanea (Vietnam and 69 Taiwan), and Melanorrhoea usitata (Laos, Myanmar, Cambodia and Thailand) [12, 13]. 70 Although the geographical distribution of these trees is a more complex topic than what 71 appears in the literature [14], the use of their sap is mostly reported in agreement with 72 the above-mentioned regions. The chemical composition of all saps is a complex 73 mixture of catechol and phenol derivatives (60–65%), proteins (glycoproteins (2%) and a 74 laccase enzyme (1%)), polysaccharides (7%) and water (30%) [15]. The laccase enzyme is 75 the responsible trigger for the polymerisation reaction that leads to the formation of 76 the hardened lacquer film [16]. The catechol/phenol mixture composition is different in 77 the three trees and such difference is maintained in the final polymer, thus enabling the three types of lacquers to be chemically distinguished [17, 18]. The main components of 78 79 the product extracted from *Rhus vernicifera* are referred to as urushiol; laccol is 80 obtained from Rhus succedanea, and thitsiol from Melanorrhoea usitata [19]. Urushiol 81 and laccol main compounds are catechol derivatives with C15 and C17 alkenyl chain, respectively. Thitsiol contains additional catechol derivatives with an  $\omega$ -phenylalkyl 82 83 chain of C10 and C12 [15, 20]. Relatively abundant research has been undertaken to characterise Asian lacquers, by 84 85 focusing on finding molecular markers to identify the three different saps [21, 22], 86 studying the interactions between plant oils (used as additives in lacquer formulations)

and lacquers [23] and investigating the pathways of lacquer degradation [24, 25]. Due

to the high polymerisation degree and complex chemical composition of oriental

89 lacquers, pyrolysis-gas chromatography coupled with mass spectrometry (Py-GC/MS) is 90 the most suitable analytical approach for their chemical characterisation and for their identification in samples of unknown composition [26-28]. Due to the high number of 91 pyrolysis products obtained and the difficulty of interpretation, a remarkable effort has 92 93 been made to systematise the identification process by using the AMDIS software in 94 combination with a complex Excel spreadsheet [27]. The polar nature of most pyrolysis 95 products derived from lacquer makes a derivatisation step highly recommended. Both 96 methylation (TMAH) [8, 9, 27] and silvlation (HMDS) [15, 20] have been proposed in the 97 literature with good results. Advantages and disadvantages of these derivatisation 98 methods are mostly related to the identification of materials other than lacquer 99 possibly present in a sample, with TMAH being particularly suitable for lipids, and HMDS 100 being more suitable for proteins and polysaccharides [29]. Regardless of the 101 derivatisation method, the strategy to differentiate among lacquers is based on the 102 study of specific pyrolytic profiles and the identification of molecular markers [9, 12, 13, 103 15, 17, 20, 22, 26, 27, 30-33]. 104 The study presented here was undertaken in order to delve into some significant

105 objects from the Asian art collection at the Museum of Zaragoza, Spain ("Federico 106 Torralba" collection) [34], with the aim to characterise the complex mixtures of 107 materials possibly present, assess the provenance of the objects and describe the 108 analytical difficulties arising from the study of these types of objects. To support and 109 guide the data interpretation, a simple decisional scheme was implemented, which was 110 then applied to the data obtained from the museum objects. Because many of the 111 lacquerwares were coloured or decorated, scanning electron microscopy (SEM) coupled 112 to energy-dispersive X-ray spectrometry (EDS) was used as complementary technique to

- Py(HMDS)-GC/MS, in order to determine the elemental composition of the coatings andstudy the artistic technique.
- 115

## 116 2. MATERIAL AND METHODS

## 117 2.1. Objects and samples

118 Eight lacquered objects were selected from the Asian art collection at the Museum of

119 Zaragoza (Spain). The objects include three food containers, three boxes, one plate and

- 120 one Buddha sculpture. They were selected in order to cover a chronological period from
- 121 the 16<sup>th</sup> to the 20<sup>th</sup> century and represent different possible provenances (China, Japan,
- 122 Myanmar and Thailand), according to stylistic interpretation. Table 1 includes an image
- and a description of the objects under investigation. Some of these lacquerwares show
- a European influence, as they were probably intended for the export market (49009,
- 125 49189 and 54562) and are referred to as *Nanban* art objects <sup>1</sup> (49189 and 54562).

<sup>&</sup>lt;sup>1</sup> This style was developed in Japan between the 16<sup>th</sup> and 17<sup>th</sup> centuries, and the word "*Nanban*" means "Southern barbarians", because it was linked to the contacts and trade with European people, especially from Portugal.

Museum	Object	Description	Decoration	Chronology	Provenance	Size
code						(l, w, h: cm)
49005	Por	Htamin-gyaing	Black (exterior) and	First half of the 20 <sup>th</sup>	Myanmar	23 (Ø), 48.5
		(food container	red (interior)	century		
		made of bamboo)	lacquer; <i>kyaukka</i>			
			decoration			
49006		Htamin-gyaing	Red and black	First half of the 20 <sup>th</sup>	Myanmar	21.8 (Ø), 45
		(food container	lacquer, with yellow,	century		
		made of bamboo)	red and black			
			decoration; yun			
			(incised) technique			
49188		Food container	Black lacquer with	First half of the 20 <sup>th</sup>	Thailand or	29.5 (Ø), 39
		(basketry)	red and gold	century	Myanmar	
			decoration			
	And an and a second sec					
49189	Ani part anima	Nanban <i>yōhitsu</i>	Black lacquer with	End of the	Japan	45.2, 24.9,
		(small chest)	maki-e and raden	Momoyama period		29.5
			decoration	or beginning of the		
	and the second se			Edo era, 17 <sup>th</sup> century		

126	Table 1. Summary	y of the lacquered ol	pjects from the Asian a	art collection at the Muse	um of Zaragoza inclu	ded in this research.
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54562	Nanban <i>yōdansu</i> (writing box)	Black lacquer with red <i>, maki-e</i> and <i>raden</i> decoration	Momoyama period, end of the 16 <sup>th</sup> century or beginning of the 17 <sup>th</sup> century	Japan	25.4, 25.5, 23.3
48990	Amida-Buddha	Gilded decoration	Edo period, 19 <sup>th</sup> century	Japan	23.5, 12.5, 56.5
48165	Sewing box	Brown lacquer with golden decoration	First half of the 19 <sup>th</sup> century	China	63.7, 43.7, 71
49009	Plate	Non-shinning red lacquer with golden and black decoration	Qing period, 18 <sup>th</sup> century or beginning of the 19 <sup>th</sup> c.	China	25.2 (Ø), 4.8

Two sets of micro-samples (with total diameter less than 1 mm) were taken from all the objects by using a scalpel and then stored in glass vials. To minimise sampling of these precious objects, one set of micro-samples was used for SEM-EDS examination and the other one for Py(HMDS)-GC/MS. Monochrome and non-decorated micro-samples were mainly used for Py(HMDS)-GC/MS analyses, and coloured micro-samples were selected for SEM-EDS.

133

### 134 2.2. Py(HMDS)-GC/MS

Analytical pyrolysis was performed using 1,1,1,3,3,3-hexamethyldisilazane (HMDS, 135 chemical purity 99.9%, SigmaAldrich Inc., USA) as a silylating agent for the *in situ* 136 137 derivatisation of pyrolysis products. The instrumentation consisted of a micro-furnace 138 Multi-Shot Pyrolyzer EGA/Py-3030D (Frontier Lab) coupled to a gas chromatograph 6890 (Agilent Technologies, Palo Alto, CA, USA), equipped with an HP-5MS fused silica 139 capillary column (stationary phase 5% diphenyl and 95% dimethyl-polysiloxane, 30 m × 140 141 0.25 mm i.d., Hewlett Packard, USA) and with a deactivated silica pre-column (2 m imes142 0.32 mm i.d., Agilent J&W, USA). The GC instrument was coupled to an Agilent 5973 143 Mass Selective Detector operating in electron impact mode (EI) at 70 eV. The micro-144 sample (ca. 100 µg) was admixed with 3 µL HMDS into a stainless steel cup and inserted 145 into the micro-furnace. The pyrolysis temperature was 550°C and interface temperature 146 was 280°C. The split/splitless injector was used with a 1:20 split ratio and kept at 280°C. 147 Chromatographic conditions were as follows: initial temperature 36°C, 10 min 148 isothermal; 10°C min<sup>-1</sup> up to 280°C, 2 min isothermal; 20°C min<sup>-1</sup> up to 310°C, 50 min 149 isothermal. Helium (purity 99.995%) was used as carrier gas with constant flow 1.0 mL 150 min⁻¹.

151

# 152 2.3. SEM-EDS 153 The observations and analyses were obtained by Field-Emission Scanning Electron Microscopy (FESEM) with Energy-Dispersive X-ray Spectrometry (EDS); they were 154 performed with a Merlin<sup>™</sup> FESEM microscope equipped with a Gemini column (both 155 156 from Carl Zeiss Nano Technology Systems, Germany), and coupled with an X-Max X-ray 157 microanalyzer (Oxford Instruments, UK). Cross-sections of the micro-samples were 158 prepared by embedding them in epoxy resin, polished with diamond polishing 159 suspensions, coated with carbon and fixed to the FESEM holder. The FESEM 160 observations and EDS analyses were performed at 15.0 kV and 300 pA. 161 Polished cross-sections were previously observed by Optical Microscopy (OM) using a 162 Zeiss Axio Imager Reflected-light Microscope (Carl Zeiss, Jena, Germany) with polarised 163 light. 164 165 3. RESULTS AND DISCUSSION 166 Data analysis 167 The pyrolysis data were interrogated with the aim to highlight pyrolytic profiles and 168 molecular markers based on what is reported in the literature [9, 12, 13, 15, 17, 20, 22, 169 26, 27, 30-33]. A summary of these profiles and markers is reported in Table 2, together 170 with the mass-to-charge (m/z) values used to extract chromatographic profiles of non-171 derivatised compounds, as well as methylated and silylated derivatives.

**Table 2.** Summary of pyrolytic profiles and molecular markers commonly used in the literature to distinguish Asian lacquers [9, 12, 13, 15, 17, 20, 22, 26, 27, 30-33]. The *m/z* values of the ions used to extract them are reported together with additional description of the profiles obtained. The classes of compounds are indicated with abbreviations: C – aliphatic hydrocarbons; B – alkylbenzenes; Ph – alkylphenols; Ct – alkylcatechols;  $\omega$ Ph -  $\omega$ -alkylphenylphenols;  $\omega$ Ct –  $\omega$ -alkylphenylcatechols; K – alkylphenylketones; Ct<sub>ox</sub> – acid alkylcatechols; Ph-A – alkylphenyl carboxylic acids; Ox-A – alkyl-oxo-phenylcarboxylic acids. The carbon atoms in the alkyl chains and the unsaturations are indicated with numbers: Cn:m – n=number

177 of carbon atoms; m=number of unsaturations.

	m/z	Urushi	Laccol	Thitsi	Comment
Pyrolytic profiles				ł	
Aliphatic	55 (alkenes)	Present from	Present from C9:1	Present from C9:1 to	Thitsi is difficult to identify based only
hydrocarbons (C)		C9:1 to C15:1,	to C17:1, peaking	C17:1, with C14:1 as	on C profiles
		peaking at	at C16:1	most abundant	
		C14:1			
	57 (alkanes)	Present from	Present from C9	Present from C9 to	
		C9 to C15,	to C17, peaking at	C17, peaking at C15	
		peaking at C15	C17		
Alkylbenzenes (B)	91	Decreasing	Decreasing profile	Gaussian profile from	The B profile of <i>thitsi</i> is highly
		profile from	from B1 to B11-	B1 to B12, peaking at	distinctive
		B1 to B9-B10	B12	B11:1	
Alkylphenols (Ph)	108	Gaussian	Gaussian profile	Complex profile from	All profiles are highly distinctive
	122 (Me)*	profile from	from Ph1 to Ph17,	Ph1 to Ph17 with	

	180 (TMS)**	Ph1 to Ph15, peaking at Ph7	peaking at Ph9	additional presence of ωPh10 and ωPh12	
Alkylcatechols (Ct)	123 151 (Me) 179 (TMS) 253-267 (2TMS)	Gaussian profile from Ct1 to Ct15, peaking at Ct7	Gaussian profile from Ct1 to Ct17, peaking at Ct9	Complex profile from Ct1 to Ct17 with additional presence of ωCt10 and ωCt12	The profile of disilylated catechols is sometimes more complicated, as it is dependent on steric hindrance issues during the derivatisation process
Molecular markers	·	-	·		
Ph15:1	302	Present	Present	Present	
	316 (Me)				
	374 (TMS)				
Ph15	304	Present	Present	Present	
	318 (Me)				
	376 (TMS)				
Ct15:1	318	Present	Present	Present	
	346 (Me)				
	390 (TMS)				
	462 (2TMS)				
Ct15	320	Present	Present	Present	
	348 (Me)	_			
	392 (TMS)	_			
	464 (2TMS)				
Ph17:1	330	Absent	Present	Present	
	344 (Me)	_			
	402 (TMS)				
Ph17	332	Absent	Present	Present	
	346 (Me)				

	404 (TMS)				
Ct17:1	346	Absent	Present	Present	
	374 (Me)				
	418 (TMS)				
	490 (2TMS)				
Ct17	348	Absent	Present	Present	
	376 (Me)				
	420 (TMS)				
	492 (2TMS)				
ωPh10	310	Absent	Absent	Present	
	324 (Me)				
	382 (TMS)				
ωCt10	326	Absent	Absent	Present	
	354 (Me)				
	398 (TMS)				
	470 (2TMS)				
ωPh12	338	Absent	Absent	Present	
	352 (Me)				
	410 (TMS)				
ωCt12	354	Absent	Absent	Present	
	382 (Me)				
	426 (TMS)				
	498 (2TMS)				
Degradation produc	ts				
Acid alkylcatechols	123	Present from	Present from	Ct <sub>ox</sub> 8 generally present	
(Ct <sub>ox</sub> )	151 (Me)	Ct <sub>ox</sub> 6 to Ct <sub>ox</sub> 9,	Ct <sub>ox</sub> 7 to Ct <sub>ox</sub> 12,	as most abundant	
	179 (TMS)	peaking at	peaking at Ct <sub>ox</sub> 10		
		Ct <sub>ox</sub> 8			

Alkylphenylketones	105	Not present	Not present	Gaussian profile from	They are considered degradation
(K)				K1 to K12, peaking at	products, but they appear in relative
				K11:1	fresh <i>thitsi</i> as well
Alkylphenyl	91	Not present	Not present	Gaussian profile from	
carboxylic acids				Ph-A10 to Ph-A14,	
(Ph-A)				peaking at Ph-A12 and	
				Ph-A13	
Alkyl-oxo-	105	Not present	Not present	Gaussian profile from	
phenylcarboxylic				Ox-A10 to Ox-A14,	
acids (Ox-A)				peaking at Ox-A12 and	
				Ox-A13	

178 \* methylated - \*\*trimethylsilylated

Based on previous published research, as well as on the experience of the authors, a decisional scheme was implemented and is reported in Figure 1. The scheme shows the m/z values for silylated compounds, but it can be used in the case of methylation or non-derivatised compounds by using the m/z values reported in Table 2.

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184

**Figure 1.** Proposed decisional scheme for the differentiation of *urushi*, laccol and *thitsi* 

using Py(HMDS)-GC/MS. Further confirmation of identification is then obtained by

extracting the profiles of alkylphenols (Ph, m/z 180) and bis-TMS alkylcatechols (m/z

188 253) in the same way proposed for mono-TMS alkylcatechols. The corresponding m/z

values for these additional profiles are reported in Table 2.

190

**191** FESEM-EDS results were used to distinguish decoration layers and microstructure, and

to identify the pigments, in order to improve the knowledge of the lacquer artistic

193 techniques.

194 In the following paragraphs, the results are presented according to the area of

195 provenance of the objects.

196

# 197 Objects 49005, 49006 and 49188 – food containers, Myanmar and Thailand, early 20<sup>th</sup>

- 198 *century*
- **199** The total ion chromatograms (TICs) of the samples taken from objects 49005 and 49006
- are similar (Figure 2a), and show the whole molecular pattern generally reported for
- thitsi [24], including the distinctive profile of alkylbenzenes (B, m/z 91, Figure 2b) and
- alkylphenylketones (K, *m/z* 105, not shown).
- 203 The profile of alkylcatechols (Ct, *m/z* 179, Figure 2c) is more complex but still contains
- all the features expected for *thitsi*, including the molecular markers, such as 3-(10-
- 205 phenyldecyl)catechol ( $\omega$ Ct10) and 3-(12-phenyldodecyl)catechol ( $\omega$ Ct12), despite their
- 206 low relative abundance. A more successful visualisation of the molecular markers was
- 207 generally obtained by directly extracting the m/z values corresponding to their
- 208 molecular ions (Table 2). A cleaner profile was obtained from the alkylphenols (Ph, *m/z*
- 209 180, not shown), showing the typical Gaussian profile and the molecular markers 2- and
- 210  $3-(10-\text{phenyldecyl})\text{phenols} (\omega \text{Ph10})$ , and 2- and  $3-(12-\text{phenyldodecyl})\text{phenols} (\omega \text{Ph12})$ .
- 211 The first part of the pyrolytic profile of aliphatic hydrocarbons (C, *m*/z 57, Figure 2d)
- shows saturated and unsaturated compounds up to C17, in accordance with the
- 213 composition of *thitsi*, which naturally contains alkylcatechols with up to 17 carbon
- atoms. All these features satisfy the steps described in the decisional scheme reported
- in Figure 1 for the straightforward identification of *thitsi*. However, at higher retention
- times, longer-chain aliphatic hydrocarbons (C23-C35) are present. The Gaussian shape
- of the profile and the distribution of the hydrocarbons is in good agreement with the

218	presence of paraffin wax [35]. There is no record of paraffin wax used in the traditional
219	preparation of <i>thitsi</i> [7], suggesting that this material most likely belongs to a past
220	conservation treatment.

- 221 Other materials are commonly found mixed with lacquer, including oils, proteins and
- sugars [27, 36, 37]. No molecular markers for these materials were identified in sample
- 49006, whereas some markers for lipids were detected in sample 49005, in particular
- nonandioic acid (azelaic acid), thus suggesting the presence of a drying or semi-drying
- oil, which is commonly added to Asian lacquer formulations [7, 24, 27].
- 226 These results confirm the Southeast Asian provenance of both objects, as relatively pure
- 227 *thitsi* appeared to be applied on their surfaces, in agreement with traditional practice in
- 228 Myanmar.
- 229



propylbenzene, B4) butylbenzene, B5) pentylbenzene, B6) hexylbenzene, B7) 233

230

231

234 heptylbenzene, B8) octylbenzene, B9:1) nonenylbenzene, B10) decylbenzene, B11:1)

235 undecenylbenzene, B12) dodecylbenzene, **c)** extracted ion chromatogram (m/z 179),

showing Ct4) 3-butylcatechol (TMS), Ct5) 3-pentylcatechol (TMS), Ct6) 3-hexylcatechol 236

(TMS), Ct7) 3-heptylcatechol (TMS), Ct8) 3-octylcatechol (TMS), Ph15) 3-237

pentadecylphenol (TMS), Ct15) 3-pentadecylcatechol (TMS), ωPh10) 3-(10-238

239 phenyldecyl)phenol,  $\omega$ Ph12) 3-(12-phenyldodecyl)phenol; **d)** extracted ion

chromatogram (*m*/*z* 57), showing C10) decane, C11) undecane, C12) dodecane, C13)

tridecane, C14:1) tetradecene, C14) tetradecane, C15) pentadecane, C16) hexadecane,

242 C17) heptadecane, C23) tricosane, C24) tetracosane, C25) pentacosane, C26)

hexacosane, C27) heptacosane, C28) octacosane, C29) nonacosane, C30) triacontane,

244 C31) hentriacontane, C32) dotriacontane, C33) tritriacontane, C34) tetratriacontane,

245 C35) pentatriacontane.

246

247 Another food container (object 49188), supposedly from Thailand, was present among 248 the objects under investigation. The TIC does not present the typical pyrolytic profile of 249 thitsi, which would be expected in the case of a Thai lacquered object. A series of 250 compounds not related to the presence of an Asian lacquer were identified (Figure S1a, 251 Supplementary Information), including, short-chain aliphatic carboxylic acids, glycerol, 252 phthalic anhydride, several phthalates, ethylene and diethylene glycols and 253 polysaccharide pyrolysis products. These last compounds (mostly levoglucosan 254 derivatives) might indicate the presence of a cellulose- or starch-based material, but 255 contamination from the wooden support cannot be excluded. Short-chain aliphatic 256 carboxylic acids, as well as small amounts of palmitic and stearic acids, suggest the 257 presence of a lipid material. Phthalic anhydride, phthalates and glycerol were also detected in the same sample. These are known additives in alkyd resin formulations 258 [38], which are modified drying oils [39]. The lack of a pyrolytic profile of carboxylic 259 260 acids similar to the ones normally obtained for drying oils (presence of azelaic acid and 261 high relative abundance of palmitic and stearic acids [40]) makes the identification of 262 the lipid material uncertain. Nonetheless, it has to be noted that the derivatisation of

the sample was not completely successful. Additionally, the presence of a broad tailed
chromatographic peak at 21.35 min, whose mass spectrum showed the isotopic pattern
of Hg, was highlighted. This is reported in the literature as indicative of the presence of
vermillion (HgS) [41]. As the derivatisation reaction is generally affected by the presence
of inorganic compounds, it is possible that the carboxylic acid profile was consequently
affected, not allowing the straightforward identification of the alkyd resin.

mineral wax, but the distribution of the hydrocarbons is different compared to the
paraffin wax identified in the two samples previously discussed. However, the profile is
still consistent with reported paraffin waxes recovered from various petroleum
products [35].

269

The TIC also shows long-chain aliphatic hydrocarbons indicative of the presence of a

Despite the predominance of peaks not related to lacquer in the TIC, the use of extract 274 275 ion chromatograms (EICs), as specified in the scheme of Figure 1, enabled the presence 276 of thitsi to be assessed. In particular, the pyrolytic profile of aliphatic hydrocarbons (C, 277 m/z 57, Figure S1d, Supplementary Information) is in good agreement with the general profile of *thitsi* and similar to those obtained for objects 49005 and 49006. The pyrolytic 278 279 profile of alkylphenols (Ph, m/z 180, Figure S1c, Supplementary Information) also shows 280 several compounds, with Ph7 as the most abundant one and with  $\omega$ Ph10 and  $\omega$ Ph12 281 clearly detectable in the profile. Surprisingly, the pyrolytic profile of alkylbenzenes (B, *m*/*z* 91, Figure S1b, Supplementary Information) did not show the typical Gaussian 282 profile of *thitsi*, which is considered one of the most reliable features to identify the 283 284 lacquer. Ageing, or possibly interactions with other materials, might have affected the 285 lacquer molecular composition, or pyrolytic behaviour. This result is important because 286 it shows how complex the interpretation of the results can be when additional materials from restoration are present in the same sample. Nevertheless, the identification of *thitsi* based on the molecular markers is solid and the provenance of the object is thus
confirmed.

290 The cross sections of samples from objects 49006 and 49188 were studied by FESEM-291 EDS (Figure 3). The decoration of lacquered objects is a complex process, which requires 292 the application of several layers [1]. In the case of these objects, the wooden support was coated with a clayey powder, a process referred to as *shitachi* (surface treatment in 293 Japanese) or, more appropriately *thayo* ground, for Burmese objects. This clayey layer 294 295 was sealed with a layer of black lacquer about 50-80 µm thick (Figures 3a and 3c), which 296 usually constitutes the finished ground for further decoration. The coloured layer was in 297 fact applied on the top of the black lacquer layer.

298 In the case of sample 49006, the red colour was obtained by vermilion (HgS) mixed with 299 lacquer, resulting in a layer about 10 μm thick (Figures 3a and 3b). Most of the HgS 300 particles are less than 1 µm long. An additional top layer is also clearly present (Figure 301 3b, marked with an arrow), about 3-4  $\mu$ m thick, whitish in colour in OM (Figure 3b). The 302 inorganic elemental composition, as determined by EDS, showed the presence of Si, Al 303 and some other minor elements, which suggest an environmental contamination. The 304 layer, mostly organic, might therefore correspond to an original finishing layer or a 305 protective layer applied during a conservation treatment. Nonetheless, the stratigraphic 306 analysis matched the visual observations and decoration design, thus confirming the 307 object to be traditionally decorated using the yun (incised) technique [7]. In the case of 308 sample 49188, the red layer (Figures 3c and 3d), whose thickness is around 15  $\mu$ m, was 309 prepared using vermilion (HgS), with the addition of gypsum (calcium sulphate,  $CaSO_4$ ). 310 Ca, S and O were detected by EDS and gypsum particles are marked with an arrow in

- Figure 3d. The HgS particles are averagely about 2 μm long (slightly bigger than for
- object 49006) and traces of iron oxides were also detected.

## 313



Figure 3. Images obtained by FESEM of the cross-sections of samples: a) 49006,

secondary electron (SE) image showing the whole stratigraphy; b) 49006, SE image of

the pigment at the surface (white particles in the image) and optical photomicrograph

- of the top layer; c) 49188, SE image showing the whole stratigraphy and optical
- 319 photomicrograph of the top layer; d) 49188, SE image of the red pigment (white
- 320 particles in the image; arrow marks a calcium sulphate inclusion, in grey).
- 321

322 *Objects 49189 and 54562 – boxes, Japan, 16<sup>th</sup>-17<sup>th</sup> century – and object 48990 - Buddha* 323 *sculpture, Japan, early 19<sup>th</sup> century* 

325 The two boxes are considered the oldest examples among the items selected for this study and they are supposed to be made in Japan, possibly for the export market. 326 327 In the case of object 49189, the TIC shows the presence of aliphatic carboxylic acids 328 among the most abundant pyrolysis products (Figure S2a, Supplementary Information). 329 It has been proved in a previous study that, when a drying oil is mixed with *urushi*, the 330 curing of the two materials is mutually affected [23]. In particular, the pyrolytic profile of the oil shows short-chain aliphatic carboxylic acids (C5-C9) with high relative 331 332 abundance and azelaic acid (diC9) is present with lower relative abundance compared 333 to a drying oil curing on its own. The results from this sample were in perfect agreement 334 with these observations, thus suggesting the presence of a drying oil mixed with an 335 Oriental lacquer. Additionally, a series of sesquiterpenes including cedrene, cuparene, 336 cadalene and mayurone were identified. These compounds are all present in essential 337 oils extracted from trees of the Cuprassaceae family and are reported as possible 338 additives during the preparation of urushi lacquer [27]. 339 The presence of the lacquer was assessed by using the EICs, following the steps described for *urushi* in Figure 1. In particular, the alkylbenzenes (B, *m/z* 91, Figure S2b, 340 341 Supplementary Information) show homologues up to nonylbenzene (B9); the 342 alkylcathecols (Ct, *m/z* 179, Figure S2c, Supplementary Information) show that the most 343 abundant peak is the one attributed to 3-heptylcatechol (Ct7), and the aliphatic hydrocarbons (C, *m/z* 57, Figure S2d, Supplementary Information) show tetradecene 344 (C14:1) and pentadecane (C15) as the most abundant homologues. 3-345 346 pentadecylcatechol (Ct15) and 3-pentadecylphenol (Ph15) are present with very low relative abundance, but still detectable, especially visible extracting the *m*/z values 347 348 corresponding to their molecular weights (Table 2). These results enabled *urushi* to be

identified as the Asian lacquer used to decorate this object. Paraffin wax was alsopresent in the sample, as previously observed in other objects.

351 A cross-section of sample 49189 was observed by FESEM-EDS. This micro-fragment 352 corresponds to a decorated surface in black with gold powder (maki-e decoration). A 353 layered structure is observed (Figure 4a), but this is more complex compared to the 354 Burmese samples. The ground layer is a clayey coating, as identified by elemental analysis by EDS, with some inclusions composed of Ca, S and O, and sealed by the 355 lacquer (Figures 4a and 4b). Two different sub-layers of lacquer can be distinguished 356 357 (Figure 4a, marked with dashed arrows). The first one shows an irregular thickness of 5-358 15  $\mu$ m and the lacquer is mixed with a finely ground powder, following the typical 359 Japanese *shitachi* process. The second sub-layer is about 20 µm thick. This layer was 360 probably smoothed and polished. The outer coating (about 5  $\mu$ m thick) corresponds to 361 the decoration and is composed of golden particles in a red layer. The pigment used is 362 an extremely fine iron oxide powder (Figure 4b, plain arrow) and a lacquer/oil mixture 363 was probably used as adhesive. The chemical composition of the metal particles is gold, with traces of silver, and they are less than 10 µm long and extremely thin (see Figure 364 365 4b). On the top of the golden design, along the cross-section surface, a final layer is 366 observed, composed of C, O, Ca, and others (Figure 4b, dashed arrow). According to the 367 shitachi process, after the final lacquer polishing, a last coating with at least three fine lacquer layers is applied, although some sophisticated lacquerwares could be painted 368 369 with dozens of layers of lacquer coatings [1]. All these observations were consistent 370 with the maki-e decoration technique, which is a very traditional and widespread type 371 of sprinkled-gold design commonly seen on Japanese lacquer objects [42].



Figure 4. Images obtained by FESEM of the cross-sections of samples: a) 49189, 374 375 backscattering electron (BSE) image showing the whole stratigraphy and an optical 376 photomicrograph of the top layer; b) 49189, secondary electron (SE) image of the surface decoration (plain arrow indicates iron oxide particles and dashed arrow marks 377 378 the top upper layer); c) 48990, SE image and optical photomicrograph showing a decoration made of bone or ivory on the top (Ca, P and O detected by EDS) and with 379 380 arrow pointing to the gold foil on the lacquer; d) 49009, BSE image showing the whole 381 stratigraphy and optical photomicrograph of the red layer. 382 383 In the case of object 54562, the TIC (Figure 5a) shows the presence of aliphatic

- carboxylic acids with a similar distribution to that observed for object 49819, therefore
- indicating the use of a drying oil. Camphor was identified, which is another known

additive of *urushi* formulations [27]. A paraffin wax was also present, as suggested by
the presence and distribution of long-chain aliphatic hydrocarbons, as previously
discussed.

The EIC of the alkylbenzenes (Figure 5b) is in agreement with the profile reported for 389 390 urushi, but the EIC of the alkylcatechols (Figure 5c) clearly shows the presence of all the 391 thitsi markers, thus placing the sample in none of the scenarios shown in the decisional scheme of Figure 1. In the case of this sample, the profile of hydrocarbons (Figure 5d) 392 393 was not useful to distinguish between *urushi* and *thitsi*, as these compounds overlap 394 with the ones deriving from the paraffin wax. Overall, the lack of the alkylbenzenes 395 typical of *thitsi* together with the presence of *thitsi* molecular markers makes the interpretation of the results challenging. A mixture of *urushi* and *thitsi* is the most likely 396 397 hypothesis, but it is difficult to ascertain whether this mixture was originally used to decorate the object or is the result of a later addition. The design of the box definitely 398 399 places it as a Japanese object, but it is suspected to be made for the export market. The 400 use of *thitsi* imported to Japan and used in mixture with *urushi* for Japanese export lacquer is reported since the 17<sup>th</sup> century [35]. This object could therefore be one of the 401 402 few scientifically-confirmed examples of this practice. In the hypothesis of an original 403 mixture of *urushi* and *thitsi*, it is reasonable to think that the two lacquers mixed 404 together would polymerise differently than when they polymerise separately. In particular, urushi forms C-C aromatic nucleus-side chain coupling bonds, C-O phenolic 405 oxygen-side chain coupling bonds and C-C bonds between side chains [43], whereas in 406 407 thitsi nuclear-nuclear C-C couplings are predominant [44]. A co-polymerisation would 408 therefore affect the distribution of dimeric bonds and consequently the pyrolytic

# 409 profiles, possibly justifying the profile of alkylbenzenes obtained. This would be an







417

418	The Buddha sculpture (object 48990) is described as produced in Japan in the early 19 <sup>th</sup>
419	century. However, a mixture of drying oil and thitsi was clearly identified. All the
420	extracted ion chromatograms matched with the profiles expected for this lacquer, in
421	agreement with the scheme in Figure 1, therefore strongly questioning the Japanese
422	origin of this object. In the case of a real Japanese production, this object would have
423	probably been made for the export market in a similar way to object 54562, but using
424	pure thitsi, which was less expensive than urushi. However, furniture and boxes were
425	dominating the export market in the 19 <sup>th</sup> century, whereas a Buddha sculpture would
426	have been less attractive for European customers, strengthening a possible Southeast
427	Asian origin of the object.
428	The results of sample 48990 by FESEM-EDS may also support a possible Burmese or Thai
429	manufacture, because a decoration with a last coating of extremely thin gold foil is
430	observed. This finishing is typical of lacquerware from Myanmar (Figure 4c) and is
431	referred to as <i>shwe zawa</i> [24]. The object is also decorated with bone or ivory, and this
432	material was confirmed by the detection of Ca, P and O by EDS analysis in the layer
433	above the gold foil (Figure 4c, top).
434	
435	Objects 49009 (plate) and 48165 (sewing box) – China, 19 <sup>th</sup> century
436	These two objects are believed to be produced in China, therefore the use of <i>urushi</i>

**437** lacquer (*qi* lacquer in Chinese) was expected.

438 In the case of object 49009, the pyrogram is dominated by the presence of aliphatic

439 carboxylic acids, including dicarboxylic acids (sebacic, azelaic and suberic acids) with

440 high relative abundance and a characteristic distribution indicative of the use of a drying

oil, possibly *tung* oil [23, 27, 45]. The pyrogram also enabled alpha-cedrene, beta-

442 cedrene and cedrol to be identified. These are the main components of the essential oil

443 from cedar, another known additive in lacquer formulations [27].

444 The extracted ion chromatograms of alkylbenzenes and alkylcatechols are in perfect

agreement with the chromatographic features of *urushi* (Figure 1, Table 2), therefore

446 pointing towards the authenticity of the object as being produced in China with

447 traditional materials used in lacquer preparation. Additionally, a series of compounds

448 referred to as alkylphenylcarboxylic acids was identified and these compounds form

449 when the drying oil is mixed with the lacquer [23, 24].

450 A very similar composition was found for object 48165, which was therefore also

451 confirmed as a traditional Chinese lacquered object.

452 In the case of object 49009, the surface was not shiny and this led us to consider its

453 study by FESEM-EDS. Observations of a micro-sample from this plate highlighted a

454 different way of decorating the surface compared to all the other samples. In fact, the

455 object is coated with a first layer of lacquer probably containing calcium carbonate (Ca,

456 C and O detected by EDS analysis). However, instead of a second sealing coating with

457 lacquer, a red pigmented lacquer layer is directly applied onto the ground (Figure 4d).

458 The decorative layer is 10-15  $\mu$ m thick and is composed of vermilion with pigment

459 particles with 2-4  $\mu$ m of total diameter. Finally, a thicker transparent finishing layer (15-

460 30 μm thick) is applied on the top of the coloured layer. The irregular thickness of the

461 layers and the lack of some expected layers seem to be the result of a decoration

462 process carried out with poor polishing and for an object of poor quality or value.

463

464 4. CONCLUSIONS

465 This study presents the results of an investigation carried out to assess the provenance 466 of a series of lacquered objects from the Museum of Zaragoza (Spain). The lacquer 467 identification was carried out based on the pyrolytic profiles and molecular markers reported in the literature and following a decisional scheme which was implemented to 468 469 help the data analysis in order to distinguish the three Asian lacquers - urushi, laccol and 470 thitsi. In addition, some of the reported cases represent analytical situations, which are often neglected in the literature. In particular, a possible mixture of *urushi* and *thitsi* is 471 discussed, as well as examples of urushi and thitsi identified after careful examination 472 473 despite the predominant presence of restoration materials. 474 The chemical information obtained by Py(HMDS)-GC/MS was coupled with the study of 475 the decoration techniques by FESEM-EDS. The study of the cross-sections added 476 significant information about the nature of the pigments (mercury sulphide, iron oxides, gold powder, etc.) used to prepare the coloured layers and highlighted different ways of 477 478 applying the coating layers. Traditional techniques (yun, maki-e, shitachi, shwe-zawa, 479 etc.) were recognised and some chemical considerations were supported. The overall results allowed us to confirm the geographical provenance of six out of the 480 481 eight objects investigated, whereas for two of them this information must be re-482 evaluated. This confirms the robustness of Py(HMDS)-GC/MS as analytical technique to 483 characterise and distinguish Asian lacquers, as well as the importance of undertaking scientific analysis to support historical and art historical information on museum 484 objects. 485

486

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- 495

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