

1 **Water quality decline in coastal aquifers under anthropic pressure: the case of a**
2 **suburban area of Dakar (Senegal)**

3
4 V. Re⁺, S. Cissé Faye, A. Faye, S. Faye, C.B. Gaye, E. Sacchi, G.M. Zuppi

5
6 Re Viviana
7 Dipartimento di Scienze Ambientali, Università Cà Foscari, Venezia, Italy
8 Calle larga Santa Marta- Dorsoduro 2137
9 30123 Venezia-Italy
10 Tel: +39 041 2348666, fax +39 041 2348584, e-mail re@unive.it

11
12 S. Cissé Faye
13 Département de Géologie, Faculté des Sciences et Techniques, Université Cheich Anta Diop, Dakar, Senegal
14 B.P. 5005 Dakar-Fann, Senegal

15
16 A.Faye
17 Département de Géologie, Faculté des Sciences et Techniques, Université Cheich Anta Diop, Dakar, Senegal
18 B.P. 5005 Dakar-Fann, Senegal

19
20 S.Faye
21 Département de Géologie, Faculté des Sciences et Techniques, Université Cheich Anta Diop, Dakar, Senegal
22 B.P. 5005 Dakar-Fann, Senegal

23
24 C.B. Gaye
25 Département de Géologie, Faculté des Sciences et Techniques, Université Cheich Anta Diop, Dakar, Senegal
26 B.P. 5005 Dakar-Fann, Senegal

27
28 E.Sacchi
29 Dipartimento di Scienze della Terra, Università di Pavia, and CNR-IGG, Sezione di Pavia, Italy
30 Via Ferrata 1
31 27100, Pavia-Italy

32
33 G.M.Zuppi
34 Istituto di Geologia Ambientale e Geoingegneria CNR, Area della Ricerca di Roma 1, Italy
35 Via Salaria Km 29,300
36 C.P. 10 - 00015 Monterotondo- Italy

37
38
39
40
41
42
43

44 **Abstract.** In recent years, the unregulated increase of the population in coastal areas of developing countries has
45 become source of concern for both water supply and quality control.

46 In the region of Dakar (Senegal) approximately 80% of water resources come from groundwater reservoirs, which are
47 increasingly affected by anthropogenic pressures. The identification of the main sources of pollution, and thus the
48 aquifer vulnerability, is essential to provide a sound basis for the implementation of long term geochemically based
49 water management plans in this sub-saharian area.

50 With this aim a hydrochemical and isotopic survey on 26 wells was performed in the so called Peninsula of Cap-Vert.
51 Results show that sea water intrusion represents the main process affecting groundwater chemical characteristics.
52 Nitrates often exceed the WHO drinking water limits: stable isotopes of dissolved nitrate ($\delta^{15}\text{N}$ and $\delta^{18}\text{O}$) indicate urban
53 sewage and fertilizers as a major source of contamination. Results depict a complex situation in which groundwater is
54 affected by direct and indirect infiltration of effluents, mixing with seawater and freshening processes from below.
55 Beside the relevance of the investigation at a regional level, it represents a basis for decision making processes in an
56 integrated water resources management, and in the planning of similar monitoring strategies for other urban coastal
57 regions.

58
59 **Keywords:** Urban and coastal aquifers, groundwater quality, hydrochemistry, environmental isotopes.

60

61 INTRODUCTION

62

63 Water shortages, droughts and groundwater pollution due to human activities are increasingly becoming a worldwide
64 source of concern.

65 Although the world is not yet experiencing a global water shortage, more than 1.2 billion people live under conditions
66 of physical water scarcity, which occurs when more than 75 per cent of the river flows are withdrawn (WHO/UNICEF,
67 2006). Another 1.6 billion people live in areas of economic water scarcity, where human, institutional and financial
68 capital limits access to water. These conditions are prevalent in much of sub-Saharan Africa, and at continental level
69 there is an alarming trend in artificially-created scarcity, even in areas where water is apparently abundant (IWMI,
70 2006). During the past century, water use has increased at double rate if compared to the one of population growth
71 (IWMI, 2006), with consequent impacts in water availability and contamination. Symptoms often include lack of
72 adequate water infrastructure, high vulnerability to short and long-term drought, and difficult access to reliable water
73 supplies, especially for rural people.

74 This is why groundwater resources have been increasingly used to rectify the supply backlogs and many urban areas
75 rely on groundwater as fundamental part of the drinking water supply (Xu and Usher, 2006).

76 As many African urban and rural areas present the aforementioned lack of adequate and accurate information on both
77 quality and quantity of water resources (Showers, 2002), and as the demands on freshwater resources is rapidly
78 growing, there is an urgent need to link research with improved water management. Provision of safe water, access to
79 water and sustainable water use need to be assessed in order to avoid the water crisis, and this should be achieved only
80 applying science based long term management projects.

81 Better monitoring, assessment, and forecasting of water resources will help allocate water more efficiently among
82 competing needs. Without protection there is a serious risk if an irreversible decline of water, with a severe impact on
83 ecosystems and human health.

84 However, under need to rapidly develop new water supplies, there is rarely adequate attention to, and investment in, the
85 maintenance, protection and longer-term sustainability of groundwater.

86 Groundwater use is dominated by three features: depletion due to overexploitation; salinization and water logging due
87 mostly to inadequate drainage and insufficient conjunctive use or to seawater intrusion linked to the costal aquifer
88 overdraft; and pollution due to agricultural, industrial and other human activities.

89 The three aforementioned points summarize three key areas: water scarcity, water quality and water-related disasters,
90 linked reciprocally, which often create serious and irreversible problems to underground resources in many coastal
91 areas (Abramovitz 2001).

92 The region of Dakar (Senegal), collectively called Peninsula of Cap-Vert, is a classical example of these problems. As
93 several other African cities, the Peninsula has seen a rapid population growth increased by movements of refugees from
94 nearby countries in conflict (Deme et al., 2006). This fast population growth has resulted in an increase in water
95 requirements and is putting even more pressure on already insufficient sanitation infrastructures. Moreover as many
96 other coastal African cities (Showers, 2002) Dakar's water supply depends upon groundwater for its daily function, and
97 population use pit latrines and dump waste resources already threatened by seawater intrusion.

98 These peculiarities make the studied area a classical example of several key problems, and can be used as a reference
99 case study representative for groundwater management in urban and coastal areas in both developed and developing
100 countries.

101 At present the urban population of Dakar and its suburb zone accounts for 23% of the total population of Senegal and
102 only 0.3% of the surface area. As a consequence many informal settlements are present in marginal land with lack of
103 both control and sanitation measures. The lack of use and diffusion of sewage drainage infrastructures can be
104 considered as one of the main sources of groundwater pollution in this area, as in most developing countries. These
105 considerations are not limited to Dakar and its outskirts but can be applied to all cities and villages in the peninsula.
106 Thiaroye, Pikine, Yeumbeul, Keur Massar are only few examples of cities in which most houses are equipped with
107 improper septic tanks, seeping from below into the groundwater (Cissé et al., 2004).

108 In addition to this social situation, 90% of the local industries located in this region (including agribusiness, textile and
109 fertilizer production) are recognized as the main drivers of the severe groundwater pollution by nitrate in the region
110 (Collin and Salem 1989, Tandia et al., 1999).

111 In this area, both point and non-point sources of chemical contamination contribute to groundwater quality decline for
112 this reason the apportioning of the relative loads in groundwater, with a focus on anthropogenic inputs of nitrogen and
113 occurrence of saline water intrusion is required.

114 Despite of the rising awareness and attention to such problems (SONES 1986; Diop and Tandia 1997; Cissé et al., 2004;
115 Faye et al., 2005, Deme et al., 2006) the region is still strongly affected by nitrogenous pollution, and the quality of the
116 water has diminished over the last few years with a considerable increase in nitrate content. Concentrations close to 300
117 mg/L have been measured while the WHO recommends 50 mg/L as a limit for the drinking water supply. In fact, the
118 deterioration in groundwater quality can be primarily explained by the proximity of confirmed sources of pollution to
119 the wells (Cissé et al., 2004). In the region of Dakar, the presence of different sources of chemical pollutants is widely
120 recognized. Hence it has become a primary objective to discriminate the contribution of different loads in drainage
121 waters.

122 The continuous overexploitation of coastal aquifers increases the groundwater mineralization and produces a decline in
123 the water quality, with important drawbacks on agricultural development and on the population's health. Groundwater

124 salinization is a common problem for both developing and developed countries (Ghassemi et al., 1995; Foster and
125 Chilton 2003).

126 The use of isotopic tracers has been recognized to be useful in terms of providing new insights into hydrologic
127 processes (Girard and Hillaire-Marcel 1997; Kellman and Hillaire-Marcel 2002; Fukada et al., 2004). In fact, tracers
128 permit to encompass the small-scale variability and provide a description of catchment-scale processes (McDonnell and
129 Kendall 1992). The development of effective management strategies to preserve water quality, and remediation plans
130 for sites with suspected contamination, requires the identification of the pollutant sources and understanding of the
131 effective processes affecting local nitrate concentrations (Kendall 1998). An integrated, holistic approach is therefore
132 required to tackle the big challenge of sustainable water management. In order to implement and integrate equitable
133 policies and projects a sound knowledge of both the natural and human induced phenomena occurring in the aquifer is
134 required.

135 This work tries to deal with the need for the development of an integrated management, concerning both groundwater
136 and wastewater, by providing a sound data basis related to the major processes occurring in shallow aquifers of the
137 Dakar region.

138 Moreover the present study has to be considered not only for its relevance at local level, as a basis for decision making
139 processes in integrated water resources management, but also for the value of the used of approach for broad
140 applications to similar urban coastal areas in Africa.

141 This work can contribute at the provision of geochemical and isotopic data in the framework of the assessment of
142 groundwater quality at regional level and as a basis for constructing geochemical and isotopic based programs for water
143 management and international level.

144

145 **1. SITE DESCRIPTION**

146

147 The region of Dakar is located in the mid-western part of Senegal between the 14° 24' and 15° 5' North and between
148 the 16° 55' and 17° 32' West (Fig. 1). It is characterized by a rocky headland, the peninsula of Cap-Vert, and is linked
149 to the continent by the isthmus of Thiaroye. The morphology of the peninsula is formed by two units: the extreme
150 westward unit, characterised by an uplift of the sedimentary deposits in relation to the Dakar quaternary volcanism, and
151 the eastward depression zone located in the Thiaroye suburban area (Hébrard and Elouard 1976).

152

here Fig.1

153

In this area it is possible to distinguish:

154•

The northern coast, where the dune zone is separated from the Ocean by a continuous littoral strip. The
155 Quaternary sand dunes are oriented SW–NE while the “Niayes” constitute the main surface features along the ocean
156 facade. In this zone, some lakes occur beside the dune deposits; most of them are dry except for the hypersaline Lake
157 Retba (Seck 1988).

158•

The southern coast, where the pre-Quaternary basement, essentially marly, outcrops in Rufisque and Bargny as
159 a plateau relief of 30 m high.

160

161 Dakar climate is controlled by the dual influence of oceanic and continental processes forced by the relative position of
162 the Azores anticyclone, the North African anticyclone, the Saharan thermal low and the Saint Helena anticyclone (Fall
163 et al., 2006). It is governed by the position of the Inter-Tropical Front (ITF), which regulates the rainfall mechanism in
164 this region (Lebel and Le Barbé 1997). As a consequence, in the region of Dakar the dry season (October-June) occurs

165 when the InterTropical Convergence Zone (ITCZ) migrates southward and the area is out of reach from the moist
166 monsoonal flow. The migration of the ITCZ determines the duration of the wet season (June–July), also called
167 *hivernage* (Fontaine 1991; Leroux 2001).

168 The mean annual rainfall rate is 480 mm/year (mean 1961–1990), although it was reduced in 1970-1998 to 342 mm
169 (Dasylyva et al., 2004). The mean annual air temperature (1971-1988) is 24°C, the annual potential evapotranspiration,
170 (Turc’s method) is 1830 mm/yr. The relative humidity varies little during the day or in the course of the year. It stays
171 higher than 70% and reaches 98% in June. The real evapotranspiration is estimated at about 700 mm/yr. The present-
172 day infiltration rate is assumed to vary from year to year between 40 and 60 mm with a mean value of 48 mm per year
173 (Travi et al., 1987; Travi et al., 1991; Tandia 2000).

174 The Geology of Dakar is mainly dominated by Tertiary igneous rocks, and overlying Quaternary sediments. The
175 geology of this region is characterized by a long period of volcanic activity from Tertiary (southern horst) to Quaternary
176 (northern horst) and by cycles of transgression/regression of the Atlantic Ocean. The southern horst forms the Dakar
177 plateau, some 50 m above sea level, which consists mainly of Tertiary volcanic products. The northern horst reaches a
178 height of 105 m and is mainly formed of Quaternary volcanic deposits. From the Mesozoic to the Quaternary, the
179 general evolution of the area is marked by deep faults (Bellion 1980; Bellion 1987) which delimit two horsts, the Dakar
180 and Ndiass horsts, and the Rufisque graben. The effect of this tectonic control is highlighted by the presence of only
181 Tertiary sediments: marls and limestones from the Eocene, karstic limestones from the Paleocene and sandstones and
182 limestones from the Maestrichtian periods. The entire area is covered by eolian sands dating from the Quaternary.

183 Quaternary phreatic aquifer is composed of unconsolidated sands. The sediments consist largely of fine and medium
184 grained sands with a porosity value of about 20%. The aquifer overlies Palaeocene and Maestrichtian limestone aquifers
185 (Martin 1970).

186 Recharge to the phreatic aquifer is yearly provided by direct rainfall infiltration. Outflows are due to evaporation,
187 occurring during the dry season in the Niayes, direct flow into the sea and, in recent years, human exploitation through
188 wells (Audibert 1971).

189 The flows are usually from South-East to North-West, hence from the sea to inner zones (Forkasiewicz 1982; Cissé
190 2001).

191 The presence of two piezometric domes (the first located at North-East, among Mbawane lake and Tanma lake, and the
192 second sited in South-West in proximity of the Niayes of Pikine) allowed the isolation of the studied reservoir as an
193 independent hydrodynamic system.

194 The western zone is marked by the catchment basin of Thiaroye, characterized by a depression generated from its
195 exploitation. East of this basin the hydraulic gradient is generally important, with an order of 1-3%.

196 Aquifers in the Cap Vert peninsula have been over-exploited since 1952, when they first started being used in response
197 to Dakar’s water demand (SONES 1986). This has resulted in some encroachment of salt water (Faye et al., 2005). It
198 appears that the encroachment of the salt water is continuing, albeit slowly, in the Southern area. The chloride contents
199 of some piezometers have increased considerably since 1985 (increasing from 35 to 300 mg/L) (Cissé 2001). Therefore,
200 the groundwater is still slightly over-exploited and maintaining the current level of chloride. However, it should be
201 observed that the quality of the water has degraded over the last few years with a considerable increase in nitrate
202 content up to 300 mg/L (Diop and Tandia 1997; Cissé et al., 2004).

203
204
205

206 **2. Materials and methods**

207

208 In order to discriminate between different sources of contamination in the region of Dakar, groundwater has been
209 sampled in 26 wells (13 drilled wells and 13 dug wells) spread across the Quaternary aquifer. Water sampling was
210 conducted during the dry season (March and April 2006) in an area of approximately 300 km² in the Cap Vert
211 Peninsula, from Dakar Yoff to Kayar (Fig. 1).

212 Different isotopic techniques associated with the hydrochemistry of major and trace elements have been applied in
213 order to investigate the pollution sources affecting groundwater quality.

214 The hydrochemistry of minor elements, namely boron (B), strontium (Sr) and bromide (Br) together with the
215 environmental stable isotopes ($\delta^{18}\text{O}$ and $\delta^2\text{H}$) and the major-ion chemistry (chloride, sodium and calcium) has been
216 used to restrict the sources and the processes of salinization in the Quaternary sand aquifer. Stable isotopes of oxygen-
217 18 and deuterium are commonly used in regional groundwater studies to identify flow regimes mixing and sources of
218 recharge (Fritz and Fontes 1980). In addition, the use of $\delta^{13}\text{C}$ in DIC (Dissolved Organic Carbon) can be important to
219 trace carbon sources in groundwater (Clark and Fritz 1997).

220 The basis for the identification of NO_3^- is the use of a natural abundance of $\delta^{15}\text{N}$. since it allows to distinguish between
221 synthetic and natural fertilizers (Clark and Fritz 1997). The analysis of nitrogen isotope composition permits the
222 identification of the occurrence of contamination by septic effluents apart from agricultural sources and to verify the
223 expected correlation between groundwater pollution and land use. The $\delta^{18}\text{O}$ composition of nitrate provides further
224 information on the processes affecting NO_3^- , namely on the occurrence of nitrification and denitrification phenomena.

225 Sampling was performed by pumping water from wells until electrical conductivity became constant. Pumping was
226 performed using a GRUNDFOS pump, BMP/MPI-230 V, with an output flux of 0.4 L/s.

227 During the sampling phase chemical-physical parameters were measured: water level, temperature, pH, EC and Eh
228 using 340/SET-WTW while dissolved oxygen was measured with an electrode Eh (Sentix ORP).

229 Samples for major ion analysis were collected in polyethylene bottles, filtered in the laboratory through 0.45 μm
230 cellulose nitrate membrane filter papers, stored in 500 ml PE bottles, preserved with HgCl_2 except those for Cl^- analysis.
231 These samples were analysed at the Département de Géologie (UCAD, Dakar) by ion chromatography. Alkalinity was
232 determined by titration immediately after the sampling campaign. All reported values have ionic balance within 5%.

233 Trace elements analyses (B, Li, Sr) were performed at the Dipartimento di Scienze della Terra (Università di Pavia),
234 using an ICP-AES. Br analyses were performed at the IGG-CNR (Pisa) using ionic chromatography (Dionex 100) with
235 an accuracy of 3%.

236 Samples for stable isotope analysis were collected according to the procedures described by Clark and Fritz (1997).

237 Hydrogen isotope composition was measured by water reduction over metallic zinc (Coleman et al., 1982), while $\delta^{18}\text{O}$
238 was analyzed by water- CO_2 equilibration at 25 °C (Epstein and Mayeda 1953); both results are expressed in V-SMOW
239 (Gonfiantini 1978; Gonfiantini et al., 1995). The analytical errors are ± 1 and ± 0.1 ‰ respectively. The $\delta^{13}\text{C}$ of DIC was
240 analysed by direct acidification of the water sample with phosphoric acid (Kroopnick 1974) and the released CO_2
241 recuperated in vacuum line for further analysis through the dual inlet of the mass spectrometer. Results are expressed in
242 ‰ V-PDB (Gonfiantini 1978; Gonfiantini et al., 1995). Analytical errors are ± 0.3 . $\delta^{15}\text{N}_{\text{NO}_3}$ and $\delta^{18}\text{O}_{\text{NO}_3}$ of dissolved
243 nitrate were analysed following the procedures described by Kendall and Grim (1990) and by Silva et al., (2000) and
244 refer to V-AIR (Gonfiantini et al., 1995). The analytical error is ± 0.5 ‰ and ± 1 ‰ respectively. All gases were analysed
245 on a Finnigan MAT 250 Mass Spectrometer at ISO4 s.s., (Pavia, Italy). □

246

247 **3. RESULTS**

248

249 Water temperature varies between 27°C and 31°C, with an average of 28.4 °C. pH has a mean value of 6.9 and a
250 standard deviation of 1.2 (Table 1). Waters from wells number 16 (Darou Mbayakh) and number 17 (Mbayakh) are
251 strongly acidic (4.1 and 4.2, respectively), and are located in the eastern part of the Cap-Vert peninsula; on the contrary,
252 waters from well 11-Malika are very basic (pH of 9.9).

253

here Table 1

254 Eh and dissolved oxygen (Table 2) varies significantly between drilled wells (with a mean of +0.33 V and 5.8 mg/L
255 respectively) and dug wells (+0.44 V and 11 mg/L). Electrical conductivity suggests the presence of exceedingly
256 mineralized waters, with an average of 1,325 µS/cm, and a maximum of 4,150 µS/cm in well 24 (Bambilor).

257 Dominant major ions are sodium-chloride. Mineralization processes are relevant and concern areas where farming and
258 rural or urban life can affect groundwater quality. Data indicate high cation content and low alkalinity values, together
259 with an increase in dissolved nitrate and chloride. The abundance of major ions, especially of nitrate, chloride and
260 sulphate suggest an elevated undesirable evolution of physical-chemical properties of fresh water resources, and thus an
261 increased risk for public health.

262 In particular, the content of nitrate is very high, and, for 50% of wells, the concentrations exceed the WHO regulatory
263 limit of 50 mg/L, with a maximum value of 790 mg/L in dug well 24. Lithium is the least abundant, with an average
264 concentration of 11 µg/L, and only two wells (20 and 22) reach Strontium concentrations exceeding 1 mg/L. Boron
265 concentration, higher than 1 mg/L, is significant only in well number 7, drilled near Retba Lake.

266

here Table 2

267 The $\delta^{15}\text{N}_{\text{-NO}_3}$ values (Table 3) range from approximately +7 to +19 ‰, and are in the range of signatures for manure and
268 septic systems (Kendall 1998). $\delta^{18}\text{O}_{\text{-NO}_3}$ varies between +10 to +16 ‰, while the mean value for $\delta^{13}\text{C}$ of DIC is -12 ‰.
269 Mean $\delta^{18}\text{O}$ and $\delta^2\text{H}$ are -4.8 and -33.5 respectively, corresponding to expected values for meteoric waters in the region
270 (Travi 1988).

271

here Table 3

272

273 **4. DISCUSSION**

274

275 The extreme pH values observed in groundwater could be controlled both by the aquifer mineralogical matrix, and by
276 the dissociation of dissolved carbon dioxide mostly originated in the soil (Tandia 2000). In the case of acidic pH, the
277 hydrosystem might be considered hydrologically phreatic and geochemically open to the atmosphere. On the contrary,
278 waters with higher pH values circulate in a confined aquifer and thus, in a geochemically closed system, where CO₂ is
279 consumed by the active interaction with aquifer matrix. These waters are limited to the basaltic aquifer located
280 dominantly in the western part.

281 In the Eh-pH diagram (Fig. 2) waters define a trend progressively leading from an open hydrosystem (dug wells),
282 characterized by a more oxidising environment, to a closed hydrosystem (drilled wells) where the environment is more
283 reducing.

284

here Fig.2

285 These observations are in agreement with the significant levels of DO relatively more present in waters from shallower
286 wells, either from phreatic or confined aquifers, than in the waters from deeper wells (Fig. 3).

287

here Fig.3

288 On the basis of the anion distribution (Fig. 4) it is possible to recognize two main trends: on the one hand a progressive
289 enrichment in chloride, associated with very low nitrate concentrations, on the other a significant enrichment in nitrate
290 affecting both drilled and dug wells.

291 These two trends could be considered as the major processes controlling groundwater decline, in the region of Dakar:
292 sea water intrusion and anthropogenic pollution.

293 ***here Fig.4***

294 The coexistence of these two sources of contamination adds complexity to the system, in which direct and indirect
295 infiltration of effluent, and saline or fresh water flushing, work together, limiting the detection of representative end
296 members for each phenomenon.

297

298 4.1 Salinization

299

300 The hydrochemistry of trace elements boron (B), strontium (Sr), lithium (Li) and bromide (Br), environmental isotopes
301 of water molecule (^2H and ^{18}O) together with major ion chemistry (mainly Na and Cl) has been used to investigate the
302 occurrence of saline water encroachment.

303 Plots of dissolved species versus chloride can be used to investigate mixing processes between groundwater, whose
304 composition is determined by water-rock interaction, and saline water (Faye et al., 2005). Thus, waters with low salinity
305 (i.e. less than 100 mg/L of chloride) were assumed as representative of fresh groundwater or more diluted salty waters.

306 By plotting Na versus Cl (Fig.5a) it is possible to point out the process of progressive mixing with saline water.

307 ***here Fig. 5***

308 Some wells (7, 15 and 24) show an excess in Na: in particular, well number 7 is characterized by a great enrichment,
309 due to its location in respect to the hypersaline Retba Lake, mainly composed of sodium carbonate. This enrichment in
310 sodium, with respect to calcium, is dominant in this part of the aquifer, as groundwater flows from the Retba Lake
311 south-eastward.

312 Nevertheless, in the Na/Cl ratio versus Cl (Fig. 5b) diagram, evidence of depletion or enrichment of Na appears which
313 seem to result from other processes rather than mixing with saline water.

314 Waters with a Na/Cl weight ratio higher than the reference seawater value (0.858) may be derived by the mixing of
315 fresh waters, whose chemical composition is close the one of deep groundwater, and seawater. Indeed, according to
316 Faye *et al.*, (2005), deep groundwater flowing throughout sandstones shows a Na/Cl ratio of 1.6, due to the presence of
317 Quaternary evaporitic deposits mainly composed of Na_2CO_3 (Garnier 1978) or of glauberite-rich sediments
318 ($\text{Na}_2\text{Ca}(\text{SO}_4)_2$). On the other hand, enrichment in chloride suggests a possible contribution from anthropogenic sources,
319 in association to nitrate (Fig. 4).

320 The detection of traces of dissolved carbonates in wells 7, 15 and 24 is supported by the enrichment of strontium (Fig.
321 5c and 6). In general, the increase in the Sr concentration, parallel to that of Cl, is largely due to the saltwater intrusion
322 processes (Faye et al., 2005).

323 ***here Fig.6***

324 Fig. 6 shows enrichment in Sr indicative of carbonate dissolution. In fact, the incongruent dissolution of carbonates
325 readily increases the Sr concentration of the solution because the large Sr^{2+} ions are rejected from recrystallized calcite
326 (Hem, 1985; Edmunds et al., 1987; Chou et al., 1989; Drever 1997). This typically occurs in the presence of paleo-
327 lacustrine systems in warm regions, where the alternation of clays, carbonate-clays and clays-carbonates is dominant
328 (Mees 1999).

329 The trend of progressive mixing with saline waters can also be pointed out considering boron distribution respect to
330 chloride (Fig. 5d).

331 Apart from sea-water mixing, boron in groundwater can be derived by the leaching of clay minerals (Vengosh et al.,
332 2004; Bouchaou et al., 2008), as possibly in the case of well number 7. The high B concentrations of wells 20 and 21
333 however, may derive from fertilizers and detergents, due to the location of these wells (orchard and barnyard).

334 The analysis of bromine content in groundwater has been used to distinguish between salinization and anthropogenic
335 pollution (Barros et al., 2008). Fig. 7 shows that, although the Br concentration is partly controlled by sea water
336 intrusion, as already evidenced by Faye et al., (2005), an enrichment of chloride can be observed. This confirms an
337 anthropogenic input of chloride.

338 ***here Fig.7***

339 The carbon isotopic composition for the samples range from approximately -7 to -15 ‰ indicating a significant
340 difference in the $\delta^{13}\text{C}$ values of Dissolved Inorganic Carbon (DIC). The most negative $\delta^{13}\text{C}$ values correspond to water
341 in shallow sandy aquifers (open system conditions), and are controlled by the isotopic composition of soil CO_2 and
342 microbial activity affecting OM, which is usually quite negative (-15 to -22 ‰; Clark and Fritz 1997). The shift in $\delta^{13}\text{C}$
343 composition of DIC toward more negative values is indicative of an involvement of an isotopically light carbon into the
344 DIC phase, which may be derived from the oxidation of organic carbon to carbon dioxide.

345 Comparing the Cl/Br ratio with $\delta^{13}\text{C}$ (Fig. 8), a negative correlation between these two parameters appears. Water
346 characterized by high values of Cl/Br (800-600) and by highly negative carbon isotope ratios (-12 - -16‰) may be
347 representative of an organic matter rich environment with an preferential uptake of Br (Gerritse and George 1988). The
348 progressive lowering of Cl/Br ratio and the isotopic enrichment indicates the above mentioned mixing process with
349 saline waters. Two wells (6 and 12) give different results, which could be due to the complete oxidation of the original
350 organic matter content in an open system.

351 ***here Fig. 8***

352 The same results can be found if comparing the Cl/Br ratio with the nitrate content (Fig. 9). By applying the range
353 proposed by Vengosh and Pankratov (1998) for sewage and seawater in particular, and by considering the complete
354 oxidation of organic matter in an oxidising environment, it is possible to use Cl/Br ratio to distinguish between different
355 sources of pollution affecting groundwater of the studied system: domestic waste waters (sewage), seawater intrusion
356 and organic matter degradation. In particular it is important to underline the increase of NO_3^- associated with the first
357 group. The origin and the fate of nitrate pollution is investigated in the following section.

358 ***here Fig.9***

359 Finally, in addition to these minor elements, environmental isotopes (^{18}O and ^2H) were also used for supporting
360 evidence of the mixing processes (Fig. 10).

361 ***here Fig.10***

362 The groundwater isotope composition falls below the LMWL (Travi et al., 1987), which appears to be very close to the
363 Global Meteoric Water Line (Craig 1961; Gonfiantini 1998). Deviations from this line reflect an enrichment of heavy
364 isotope concentration, with a slope of 5.49, confirming once more the mixing process with seawater.

365

366 **4.2 Nitrogen pollution**

367

368 Intensive anthropic activity in this area results in nitrate concentrations in groundwater often exceeding drinking water
369 standards. In order to identify the different sources of nitrate and the processes affecting its concentration in

370 groundwater (nitrification and denitrification), the isotopic composition of nitrogen and oxygen in dissolved nitrates
371 was investigated.

372 Most of the samples lay within the range consistent with an origin from animal wastes, manure and septic system
373 effluents ($\delta^{15}\text{N}_{\text{-NO}_3} = +10 - +15\text{‰}$); this result is consistent with the position and the social situation of the region of
374 Dakar (Fig. 11). In fact, in both rural areas and sub urban areas the main impact on groundwater is represented by
375 organic fertilizers and septic effluents.

376 The second registered trend involves three wells (11, 14 and 23), all located in areas of intensive agriculture (Niayes),
377 and is compatible with the signature of soil organic matter ($\delta^{15}\text{N}_{\text{-NO}_3} = +4 - +9\text{‰}$), therefore indicating that nitrates
378 are derived from soil organic nitrogen by mineralization and nitrification (Clark and Fritz 1997)

379 ***here Fig.11***

380 In other words, on one side nitrates are directly derived from septic effluents (point source), and on the other infiltration
381 of organic matter spread on soils provide an additional non-point source input. In the latter case groundwater takes the
382 same $\delta^{15}\text{N}_{\text{-NO}_3}$ composition of soil organic matter, as a consequence of mixing effects (Kendall 1998). Both direct and
383 indirect inputs cause the processes of oxidation or nitrification in sandy media, as found in the shallow drilled and dug
384 wells (Reynaud and Roger 1981).

385 On the other hand, denitrification is observed in the deepest wells, specifically where they drain water from aquifers
386 located underneath clay layers in closed system conditions. Both the extent and rate of denitrification vary depending on
387 the groundwater flow path. While little or no denitrification occurs in much of the upland portions of the aquifer, a
388 gradual redox gradient is observed as aerobic upland groundwater moves deeper in the aquifer (Fig. 2).

389 A loss of nitrate provides evidence that denitrification occurs as groundwater enters deep layers. Electron and mass
390 balance calculations suggest that organic matter oxidation is the primary source of electrons for denitrification.
391 Therefore the major presence of anthropic contributions to the shallow systems and the presence of iron and aluminium
392 species in a confined aquifer explain these observations (Fall 1986). Indeed, the removal rate of dissolved oxygen is
393 strongly dependent on the presence of electron donors such as dissolved organic matter or Fe^{+2} (Clark and Fritz 1997;
394 Edmunds and Shand 2008)

395 This is also confirmed by the $\delta^{15}\text{N}$ and $\delta^{18}\text{O}$ diagram (Fig. 12).

396 ***here Fig.12***

397 Indeed, the oxygen isotope composition of dissolved nitrates is not governed by the relative proportions of atmospheric
398 and groundwater oxygen (Clark and Fritz 1997), but is related to the electron donors involved in the oxidation processes
399 affecting the organic products. This is more evident in samples collected from sand-point wells, more vulnerable to
400 septic-waste contamination spread at the surface. Results also indicate that groundwater under clayey soil or close to
401 septic drain fields are clearly affected by denitrification processes as indicated by the trend observed in $\delta^{18}\text{O}$ vs $\delta^{15}\text{N}$.

402

403 5. CONCLUSION

404

405 Dakar, as well as other African coastal capitals like Abidjan, Lomè, Lagos or Dar Es Salaam is characterized by an
406 increase of nitrates and other contaminants in groundwater (Abiodun, 1997; Showers, 2002 ; Oga et al., 2007; Akouvi et
407 al., 2008), strongly associated with seawater intrusion processes..

408 This underlines the fact that problems associated with rapid urbanization, inadequate sanitation, lack of alternative
409 water supplies and lack of public awareness and implementation requires urgent addressing. In particular no action is

410 longer possible without a clear understanding on the physical and geochemical processes occurring in natural
411 groundwater.

412 Anthropogenic inputs of nitrogen from human activities are a central concern in urban watershed management.
413 However, identifying the locations of these inputs, in particular if represented by improperly functioning or ill-
414 maintained septic tanks or from leaking sewer lines is difficult (Steffy and Kilham 2004).

415 The hydrochemistry of major and trace elements, coupled with environmental isotopes shed new light on the
416 geochemical processes affecting groundwater quality in the suburban area of Dakar. They have shown that groundwater
417 from drilled and dug wells in near-coastal aquifers are characterized by relatively high chloride content, due to the
418 seawater intrusion related to the overexploitation of the aquifer. The high nitrate concentration is evidence of the
419 presence of several point and non-point pollution sources, principally linked to sewage and septic effluents. This type of
420 pollution provides a supplementary input of chloride.

421 These two phenomena affecting water quality are not separated but representative of a complex situation in which
422 groundwater is affected by both direct and indirect infiltration of contaminants, mixing with seawater and the freshening
423 process from below.

424 All these processes have significant consequences on groundwater quality and pose serious threats to the health
425 conditions of the inhabitants in the region of Dakar (Gaye et al., 1998; Tandia, 1997; Cissé, 2001). In addition to the
426 mentioned pollutants, other substances strictly connected with human activities, i.e. bacteria, parasites and heavy
427 metals, which are liable to deteriorate groundwater quality, should be taken into account and be actively monitored in
428 the future.

429 To better understand the aquifer vulnerability, and to provide a sound basis for the implementation of groundwater and
430 wastewater management plans in this coastal and urban area, it is important to distinguish:

- 431 • Geochemically and hydrologically closed systems, laying underneath clay strata, often affected by
432 denitrification processes controlled by the organic matter degradation.
- 433 • Systems which are geochemically and hydrologically open to deep inputs, affected by groundwater freshening
434 and dissolution of carbonates.
- 435 • Systems which are geochemically and hydrologically open to surface inputs, affected by the direct infiltration
436 of pollutants.

437

438 **Acknowledgements**

439

440 Authors wish to thank Mr. Deme and Mr. Diaw from the Hydrogeology unit of the Université Cheick Anta Diop, for the
441 support during field activities and analytical determinations. We thank Dr. Enrico Allais and ISO4 s.s. for the help in
442 isotope analysis and interpretation. The quality of this paper was greatly improved by the critical review Prof. Mike
443 Edmunds.

444

445

446

447

448

449

450

451 **REFERENCES**

452

453 Abiodun, J. O. (1997). The challenges of growth and development in Metropolitan Lagos. In C. Rakodi (Ed.), *The*
454 *urban challenge in Africa: Growth and management of it's largest cities*. Tokyo: United Nations University Press.

455

456 Abramovitz, J. (2001). *Unnatural Disasters*. Worldwatch Paper 158, Worldwatch Institute, Washington DC.
457 <http://www.worldwatch.org/node/832>.

458

459 Akouvi, A., Dray, M., Violette, S., de Marsily, G., Zuppi, G.M. (2008). The sedimentary coastal basin of Togo:
460 example of a multilayered aquifer still influenced by a palaeo-seawater intrusion. *Hydrogeology Journal*, 16(3), 419-
461 436.

462

463 Audibert, M. (1971). *Etude hydrogéologique de la nappe profonde du Sénégal (Nappe Maestrichtienne)*.
464 *Fonctionnement hydraulique du système*. Rapport BRGM, Orléans, 71 RME 035 : 49–65.

465

466 Barros, G. V., Mas-Pla J., Oliveira Novais T., Sacchi, E., Zuppi, G.M. (2008). Hydrological mixing and geochemical
467 processes characterization in an estuarine mangrove system using environmental tracers in Babitonga Bay (Santa
468 Catarina, Brazil). *Continental Shelf Research*. 28 : 682–695. doi:10.1016/j.csr.2007.12.006.

469

470 Bellion, Y. J. C. (1980). Première contribution à la connaissance de la tectonique cassante de la région du Cap-Vert
471 (Sénégal). *Ann. Fac. Sci. Dakar*. 32: 97–101.

472

473 Bellion, Y. J. C., (1987). *Histoire géodynamique post-paléozoïque de l'Afrique de l'Ouest d'après l'étude de quelque*
474 *bassin sédimentaires (Sénégal, Taoidéni, Tullemden, Tchad)*. Thèse Science, Université d'Avignon et des pays
475 *Vaucluse*.

476

477 Bouchaou, L., Michelot, J. L., Vengosh, A., Hsissou, Y., Qurtobi, M., Gaye, C. B., Bullen, T. D., Zuppi, G. M. (2008).
478 Application of multiple isotopic and geochemical tracers for investigation of recharge, salinization, and residence time
479 of water in the Souss–Massa aquifer, southwest of Morocco. *Journal of Hydrology*. 352, 267-287.

480

481 Chou, L., Garrels, R. M., Wollast, R. (1989). Comparative study of the kinetics and mechanisms of dissolution of
482 carbonate minerals, *Chemical Geology*. 78, 269–282.

483

484 Cisse Faye, S. (2001). *Nappe libre des sables Quaternaires Thiaroye/ Beer Thialane (Dakar, Sénégal)*. Etude sur la
485 contamination par le nitrates sur la base d'un Système d'information Géographique (PC ARC/INFO). Reihe B,Heft 12
486 XXVII, *Munchner Geologische Hefte*,Munich.

487

488 Cisse Faye, S., Faye, S., Wohnlich, S., Gaye, C.B. (2004). An assessment of the risk associated with urban development
489 in the Thiaroye area (Senegal). *Environmental Geology*. 45, 312-322.

490

491 Clark, I., & Fritz, P. (1997). *Environmental Isotopes in Hydrogeology*, Lewis Publishers, New York.

492

493 Coleman, M. L., Sheppard, T. J., Durham, J. J., Rouse, J. E., Moore, G. R. (1982). Reduction of water with zinc for
 494 hydrogen isotope analysis - Analytical Chemistry 54, 993-995.

495

496 Collin, J. J., Salem, G. (1989). Pollution des eaux souterraines par les nitrates dans les banlieues non assainies des pays
 497 en développement. Cas de Pikine (Sénégal). Note technique BRGM SGN/3E, 89/27, 1-11.

498

499 Craig, H. (1961). Isotopic variations in meteoric waters. Science. 133,1702-1703.

500

501 Dasylyva, S., Cosandey, C., Orange, D., Sambou, S., (2004). Rainwater infiltration rate and groundwater sustainable
 502 management in the Dakar region. Agricultural Engineering International, 6, 12-15, The CIGR E-journal of Scientific
 503 Research and Development. Manuscript LW 04 004. Vol. VI.
 504 <http://cigr-e-journal.tamu.edu/submissions/volume6/LW%2004%20004%20Dasylyva%20final%2028Dec2004.pdf>.

505

506 Deme I., Tandia, A. A., Faye, A., Malou, R., Dia, I., Diallo, M. S., Sarr, M. (2006). Management of nitrate pollution of
 507 groundwater in African cities: The case of Dakar, Senegal. In: Y. Xu and B.H. Usher, Editors, Groundwater Pollution in
 508 Africa, Taylor & Francis, 181-192.

509

510 Diop, E. S., Tandia, A. A. (1997). Qualité de l'eau de la nappe phréatique à Yeumbeul, Sénégal. Étude sur le terrain,
 511 CSI info N° 3, UNESCO, Paris. 27,1-35.

512

513 Drever, J. I. (1997). The Geochemistry of Natural Waters, Prentice Hall, New Jersey.

514

515 Edmunds, W. M., Cook, J. M., Darling, W. G., Kinniburgh, D. G., Miles, D. L., Bath, A. H., Morgan-Jones, M.,
 516 Andrews, J. N. (1987). Baseline geochemical conditions in the chalk aquifer, Berkshire, UK: a basis for groundwater
 517 quality management. Applied Geochemistry, 2,251-274.

518

519 Edmunds, W. M., & Shand, P. (2008). Natural Groundwater Quality. Blackwells, Oxford

520

521 Epstein, S., Mayeda, T. K. (1953). Variations of the $^{18}O/^{16}O$ ratio in natural waters. Geochimica Cosmochimica Acta,
 522 4, 213 – 224.

523

524 Fall, M. (1986). Environnements sédimentaires quaternaires et actuels des tourbières des niayes de la grande côte du
 525 Sénégal. Thèse de 3ème cycle UCAD, ORSTOM, 1986, 130 p. Fonds IRD [F A27625],Dakar.

526

527 Fall, M., Azzam, R., Noubactep, C. (2006). A multi-method approach to study the stability of natural slopes and
 528 landslide susceptibility mapping. Engineering Geology. 82, 241-263.

529

530 Faye, S., Maloszewsky, P., Stichler, W., Trimborn, P., CISse Faye, S., Gaye, C. B. (2005). Groundwater salinization in
 531 the Saloum (Senegal) delta aquifer: minor elements and isotopic indicators. Science of The Total Environment. 343,
 532 243-259.

533

534 Fontaine, B. (1991). Variations pluviométriques et connexions climatiques: l'exemple des aires de moussons indienne et
535 ouest-africaine. *Sécheresse*. 4, 259-264

536

537 Forkasiewicz, J. (1982). Aquifère du Maestrichtien du bassin sédimentaire sénégal-mauritanien. *Bull. BRGM* 3: 185–
538 196

539

540 Foster, S. S. D., Chilton P. J. (2003) Groundwater: the processes and global significance of aquifer degradation,
541 *Philosophical Transactions of the Royal Society B: Biological Sciences*. 358: 1957 – 1972. DOI
542 10.1098/rstb.2003.1380.

543

544 Fritz, P., Fontes, J. Ch. (Editors) (1980). *Handbook of environmental isotope geochemistry, vol. 1. The terrestrial
545 environment* : A.. Elsevier Science Publishers B.V., Amsterdam-Oxford-New York-Tokyo.

546

547 Fukada, T., Hiscock, K. M., Dennis, P. F. (2004). A dual-isotope approach to the nitrogen hydrochemistry of an urban
548 aquifer. *Applied Geochemistry*. 19, 709-719.

549

550 Garnier, J. M. (1978). Evolution géochimique d'un milieu confiné : le lac Retba (Cap-Vert), Sénégal. *Rev. Géogr. Phys.
551 Géol. Dyn.* 20, 43-58.

552

553 Gaye, C. B., Faye, A., Gelinat, P. J., Thierrien, P. (1998). Analyse de l'intrusion saline dans les aquifères de la presqu'île
554 du Cap-Vert, analyse des processus de minéralisation et de dégradation de la qualité de l'eau dans les nappes
555 infrabasaltiques et des sables quaternaires. Rapt. GGL-89-CRDI-80 CREGI, Université Laval et UCAD.

556

557 Gerritse, R., George, R. (1988). The role of soil organic matter in the geochemical cycling of chloride and bromide.
558 *Journal of Hydrology*. 101, 83-95.

559

560 Ghassemi, F., Jakeman, A. J., Nix, H. A. (1995). *Salinization of Land and Water Resources: Human Causes, Extent,
561 Management and Case Studies*. New South Wales Press, Sydney, Australia.

562

563 Girard, P., Hillaire-Marcel, C. (1997). Determining the source of nitrate pollution in the Niger discontinuous aquifers
564 using the natural ¹⁵N/¹⁴N ratios. *Journal of Hydrology*. 199, 239-251.

565

566 Gonfiantini, R. (1978). Standards for stable isotope measurements in natural compounds. *Nature*. 271: 534 – 536.

567

568 Gonfiantini, R. (1998). On the isotopic composition of precipitation. In: C. Causse and F. Gasse, Editors, *Hydrologie et
569 Géochimie Isotopique*, ORSTOM, Paris: 3–22.

570

571 Gonfiantini, R., Stichler, W., Rozanski, K. (1995). Standards and Intercomparison Materials Distributed by
572 International Atomic Energy Agency for Stable Isotope Measurements: Reference and intercomparison materials for

573 stable isotopes of light elements. Proceedings of a consultants meeting held in Vienna, 1-3 December, 1993, IAEA-
574 TECDOC 825.

575

576 Hebrard, L., Elouard, P. (1976). Note explicative de la carte géologique de la presqu'île du Cap-Vert. Laboratoire de
577 Géologie, Université Cheikh Anta Diop de Dakar Faculté de Science.

578 HEM, J. D. (1985). Study and interpretation of the chemical characteristics of natural water (3d ed.). U.S. Geological
579 Survey Water-Supply Paper 2254.

580

581 IWMI, (2006). Water for food, water for life Insights from the Comprehensive Assessment of Water Management in
582 Agriculture. Publ. Stockholm World Water Week, 2006.

583

584 Kellman, L. M., Hillaire-Marcel, C. (2002). Evaluation of nitrogen isotopes as indicators of nitrate contamination
585 sources in an agricultural watershed. *Agriculture, Ecosystems and Environment*. 95: 87-102.

586

587 Kendall, C. (1998). Tracing nitrogen sources and cycling in catchments, Chapter 16, in Kendall, C. and McDonnell, J.J.,
588 eds., *Isotope Tracers in Catchment Hydrology*, Elsevier, Amsterdam. 519-576.
589 <http://wwwrcamnl.wr.usgs.gov/isoig/isopubs/itchch16.html>.

590

591 Kendall, C., Grim, E. (1990). Combustion tube method for measurement of nitrogen isotope ratios using calcium oxide
592 for total removal of carbon dioxide and water. *Analytical chemistry*. 62: 526-529.

593

594 Kroopnick, P. (1974). The dissolved O₂ - CO₂ - ¹³C system in the eastern equatorial Pacific. *Deep Sea Res.* 21: 211-
595 227

596

597 Lebel, T., Le Barbe, L. (1997). Rainfall monitoring during HAPEX-Sahel. 2. Point and areal estimation at the event and
598 seasonal scales. *Journal of Hydrology*. 188-189, 97-122.

599

600 Leroux, M. (2001). *The Meteorology and Climate of Tropical Africa*. Springer/Praxis Publishing, Chichester.

601

602 Martin, A. (1970). Les nappes de la Presqu'île du Cap Vert (République du Sénégal). Leur utilisation pur l'alimentation
603 en eau de Dakar. Pub. Bureau de Recherches Géologiques et Minières.

604

605 Mc Donnell, J. J., Kendall, C. (1992). Stable isotopes in catchment hydrology. *EOS, Transactions American*
606 *Geophysical Union*. 73, 260-261.

607

608 Mees, F. (1999). Textural features of Holocene perennial saline lake deposits of the Taoudenni–Agorgott basin,
609 northern Mali. *Sedimentary Geology*, 127 , 65-84

610

611 Oga Yei, M.S., Sacchi, E., Zuppi, G.M. (2007) - Origin and effects of nitrogen pollution in groundwater traced by the
612 case of Abidjan (Ivory Coast). *Int. Symp. on Advances in Isotope Hydrology and its role in Sustainable Water*
613 *Resources Management, IAEA, Vienna 21-25 May 2007, IAEA-CN-151/31, vol. 1, 139-147.*

614
615 Reynaud, P. A., Roger, P. A. (1981). Variations saisonnières de la flore algale et de l'activité fixatrice d'azote dans un
616 sol engorgé de bas de dune. *Rev. Ecol. Biol. Sol.* 18 : 9-27.
617
618 Seck, A. A. (1988). Synthèse hydrogéologiques de la nappe des sables quaternaires. Géométrie et structure du réservoir.
619 Thèse de D.E.A de Géologie appliquée. Université Cheikh Anta Diop de Dakar, Faculté des sciences.
620
621 Showers, K B, (2002). Water Scarcity and Urban Africa: An Overview of Urban–Rural Water Linkages *World*
622 *Development*, 30(4), 621-648.
623
624 Silva, S. R., Kendall, C., Wilkinson, D. H., Ziegler, A. C., Chang, C. C. Y., Avanzino, R. J. (2000). A new method for
625 collection of nitrate from fresh water and the analysis of nitrogen and oxygen isotope ratios *Journal of Hydrology* 228,
626 22-36.
627
628 SONES, (1986). Renforcement de l'approvisionnement en eau de la région de Dakar (1986-1991) phase intérimaire.
629 Rapport définitif, tome I.
630
631 Steffy, L. Y., Kilham, S. S. (2004), Elevated $\delta^{15}\text{N}$ in stream biota in areas with septic tank systems in an urban
632 watershed. *Ecological Applications*. 14, 637–641.
633
634 Tandia, A. A. (2000). Origine, évolution et migration des formes de l'azote minéral dans les aquifères situés sous
635 environnement péri-urbain non assaini : cas de la nappe des sables quaternaires de la région de Dakar (Sénégal). Thèse
636 d'état, Univ.de Dakar.
637
638 Tandia, A. A., Diop, E. S., Gaye, C. B. (1999). Pollution par les nitrates des nappes phréatiques sous environnement
639 semi-urbain non assaini: exemple de la nappe de Yeumbeul, Sénégal. *Journal of African Earth Sciences*. 29, 809-822.
640
641 Travi, Y. (1988). Hydrogéochimie et hydrologie isotopique des aquifères fluorurés du bassin du Sénégal. Origine et
642 conditions de transport du fluor dans les eaux souterraines. In: *Thesis, Mém. Sci. Géol, Strasbourg*
643
644 Travi, Y., Gac, J. Y., Fontes, J. C., Fritz, B. (1987). Reconnaissance chimique et isotopique des eaux de pluie au
645 Sénégal. *Géodynamique, n°spécial, Paris*. 2 : 43-53.
646
647 Travi, Y., Gac, J. Y., Gibert, E., Leroux, M., Fontes, J. C. (1991). Composition isotopique et genèse des précipitations
648 sur Dakar pendant les saisons des pluies 1982 et 1984. *Proceedings of an International Symposium on Isotope*
649 *Techniques in Water Resources Development*. IAEA, Vienna. STI/PUB/875: 495-497.
650
651 Vengosh, A., Pankratov, I. (1998). Chloride/Bromide and Chloride/Fluoride ratios of domestic sewage effluents and
652 associated contaminated groundwater. *Groundwater*. 28, 815-824.
653

- 654 Vengosh, A., Weinthal, E., Kloppmann, E. (2004). The BOREMED team. Natural boron contamination in
655 Mediterranean groundwater, *Geotimes*, 20–25.
- 656
- 657 WHO/UNICEF, (2006). Meeting the MDG drinking water and sanitation target: the urban and rural challenge of the
658 decade, WA 675.
- 659
- 660 Xu, Y., Usher, B.H., (2006). Issues of groundwater pollution in Africa. In: Y. Xu and B.H. Usher, Editors, *Groundwater*
661 *Pollution in Africa*, Taylor & Francis, 3-13.