

1 **Modification of REE distribution of ordinary chondrites from**
2 **Atacama (Chile) and Lut (Iran) hot deserts: insights into the**
3 **chemical weathering of meteorites**

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23 **Abstract-** The behavior of rare earth elements (REEs) during hot desert weathering of
24 meteorites is investigated. Ordinary chondrites (OCs) from Atacama (Chile) and Lut (Iran)
25 deserts show different variations in REE composition during this process.

26 Inductively coupled plasma-mass spectrometry (ICP-MS) data reveals that hot desert OCs tend
27 to show elevated light REE concentrations, relative to OC falls. Chondrites from Atacama are by
28 far the most enriched in REEs and this enrichment is not necessarily related to their weathering
29 degrees. Positive Ce anomaly of fresh chondrites from Atacama and the successive formation of
30 negative anomaly with the addition of trivalent REEs is similar to the process reported from
31 Antarctic eucrites. In addition to REEs, Sr and Ba also show different concentrations in OCs
32 from different hot deserts.

33 The stability of Atacama surfaces and the associated old terrestrial ages of meteorites from this
34 region give the samples the necessary time to interact with the terrestrial environment and to be
35 chemically modified. Higher REE contents and LREE enriched composition sign a
36 contamination by terrestrial soil. Despite their low weathering degrees, special care must be
37 taken into account while working on the REE composition of Atacama meteorites for
38 cosmochemistry applications.

39 In contrast, chondrites from Lut desert show lower degrees of REE modification, despite
40 significant weathering signed by Sr content. This is explained by the relatively rapid weathering
41 rate of the meteorites occurring in Lut desert which hampers the penetration of terrestrial
42 material by forming voluminous Fe oxide/oxyhydroxides shortly after the meteorite fall.

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1. INTRODUCTION

45 As soon as a meteoroid enters the Earth's atmosphere, it is subjected to terrestrial
46 alteration (e.g., Bland et al., 2006). The meteorites which are used for cosmochemical studies are
47 being collected immediately or thousands to several hundred thousands of years after their fall. It
48 has been documented that the mineralogy and chemical compositions of these samples can be
49 affected over a human lifetime (Bland et al. 1998; Pillinger et al. 2013; Socki et al. 1991). Even
50 in shorter time scales of less than decades in laboratory environment (Velbel 2014) or tens of
51 days in a humid natural environment (Bischoff et al. 2011). Initial investigations on the meteorite
52 weathering processes commenced just after the recognition of high number of weathered hot
53 deserts samples (e.g., Buddhue, 1939; Olsen and Fuchs, 1967). Since that time, discovery of
54 large numbers of meteorites from cold and hot deserts has drawn the attention of researchers to
55 document the effects of weathering on meteorites to avoid any inaccurate interpretation on their
56 geochemical compositions.
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58 The main purposes of meteorite weathering studies are: (i) understanding the weathering
59 process occurring within a meteorite and its relationship to the finding place of the meteorite, and
60 (ii) identifying the chemical and mineralogical modification during this process, to account for
61 eventual bias in cosmochemical studies (Bland et al. 2006). Several studies have focused on the
62 effect of weathering on the chemical and mineralogical compositions of meteorites from
63 Antarctic and rare hot desert achondrites (White et al., 1967; Gooding, 1982; Koeberl and
64 Cassidy, 1989; Torigoye-Kita et al., 1995; Swindle et al., 1998; Llorca et al., 2013). Compared to
65 achondrites and considering their importance and also higher abundance such studies on the
66 ordinary chondrites (OCs) have high importance.

89 With an annual precipitation of ≤ 28 mm/yr (in the marginal regions) and maximum
90 ground temperature of up to 70 °C (inside the desert) (Djamali et al. 2011; Mildrexler et al.
91 2011) along with the increasing number of meteorite finds (Pourkhorsandi and Mirnejad 2013;
92 Pourkhorsandi et al. 2016), Lut in Iran has the characteristics of a suitable place for preserving
93 meteorites.

94 For this study, we selected a set of OCs from different regions of Lut whose
95 characteristics and finding places are presented in Table 1 and Fig. 1-a. The analyzed samples
96 comprise meteorites from LL, L and H groups with petrologic types varying between 3 and 6.
97 Intense oxidation of Fe-Ni metal grains (>95 % of them) is the prevalent characteristic of these
98 meteorites (Fig. 2-a). According to the method of Wlotzka (1993), weathering degrees of these
99 samples are between W3 and W5. Iron oxides/oxyhydroxides occur as individual patches, veins
100 (along and/or perpendicular to cracks) and lamellar structures occupying former Fe-Ni metal,
101 troilite and silicate locations. Reflected light optical microscopy reveals the alteration of troilite
102 to pyrite/marcasite - rather than Fe oxides/ oxyhydroxides – in some of these samples.

103 **2.2. Ordinary chondrites from Atacama Desert**

104 The central depression of the Atacama, the driest and oldest hot desert on Earth (Hartley
105 et al. 2005), is responsible for the preservation of very old surfaces containing unique meteorite
106 accumulation areas (e.g., San Juan, and El Médano). Other meteorite dense collection areas
107 (DCAs) occur in the domain of the coastal range, known as La Yesera and Pampa de Mejillones
108 DCAs, with a different climatic condition, due to the presence of oceanic aerosols and morning
109 mist coming from the Pacific Ocean. Extreme aridity and the surface stability of the Atacama
110 lowers the rate of chemical weathering of meteorites compared to those reported from other hot
111 deserts (Gattacceca et al. 2011) - in this case Lut. Unlike the relatively low number of recovered

112 meteorites from Lut, the Atacama hosts the highest number of meteorites per surface area
113 reported so far (Gattacceca et al. 2011; Hutzler et al. 2016).

114 In this work, we used samples previously classified and described by Valenzuela (2011)
115 from La Yesera and Pampa de Mejillones DCAs which are located near to the coast of Pacific
116 (Fig. 1-b) and the considerable presence of coastal fog, which means higher amount of NaCl in
117 the regional soils. Most of these samples were settled on sand dunes, in the context of
118 Pleistocene sandstones with paleo-coastal lines, with a the presence of carbonates in the soils.

119 To compare the chemical behavior of meteorites from different regions of the same
120 desert, samples from San Juan and El Médano DCAs were chosen. Different from the coastal
121 regions, they are in situated in much drier climatic conditions. Volcanic rocks cover the surface
122 of these areas, with remarkable presence of gypsum and anhydrite as caliche layers.

123 Similar to the Lut meteorites, OCs from LL, L and H groups are included (Table 1). The
124 selected meteorites exhibit weathering degrees between W1-W5 with a mode at W3. In
125 comparison to meteorites from other hot deserts, Atacama samples have long terrestrial ages. For
126 instance, based on the $^{36}\text{Cl}/^{41}\text{Ca}$ terrestrial age determinations, Hutzler (2015) presents ages of
127 1900 ± 80 ka and 2590 ± 100 ka for El Médano 049 and Caleta el Cobre 006 meteorites,
128 respectively. Gattacceca et al. (2011) and Valenzuela (2011) also report relatively old terrestrial
129 ages (see Table 1 and Section 5).

130 In addition to these meteorites, an OC from the Sahara (Aridal 006) was selected for
131 analysis.

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3. METHODOLOGY

134 To prepare the powder used for measuring the abundances of REE in bulk meteorites,
135 samples were examined under a stereomicroscope to check for the presence of any terrestrial
136 material (e.g., cemented or enclosed sand grains inside cracks, fusion crust, desert varnish and
137 other sedimentary crusts). Each of these material show different geochemical behaviors and their
138 contamination can affect the analysis result. For instance, desert varnish tends to show high REE
139 contents (Goldsmith et al. 2014). Thus, care was taken to separate the mentioned products from
140 the exterior and interior of the meteorites. The samples extracted from the interior and the
141 freshest parts, away from the possible fusion crust and mentioned terrestrial products. To have a
142 representative aliquot of whole rock composition, approximately 1.5-2 g of each meteorite were
143 finely hand-ground in an agate mortar and then carefully homogenized.

144 Inductively coupled plasma-mass spectrometry (ICP-MS) instruments (VG PQII Plus
145 STE and Perkin-Elmer NexION® 300x) at the Dipartimento di Scienze della Terra of the
146 Università di Pisa were used to determine the trace elements (herein REE, Sr and Ba) contents of
147 the meteorite samples. The sample preparation for the ICP-MS analyses followed in this study, is
148 basically similar to the procedure described by Folco et al. (2007). About 50-100 mg of each
149 powder were dissolved in a mixture of HF and HNO₃ on a hot plate at ~120 °C inside screw-top
150 perfluoroalkoxy (PFA) vessels. This dissolution procedure could incompletely dissolve celestine
151 (SrSO₄) or barite (BaSO₄) eventually occurring into the samples. However, we always obtained
152 very clear final sample solutions. Then the sample solutions were spiked with Rh, Re and Bi as
153 internal standards (20 ng ml⁻¹ in the final solutions) and diluted to 50 mL in polypropylene flasks.
154 In each step of sample preparation, Mill-Q® purified water (18.2 M cm), ultrapure HF and HNO₃
155 were used. The correction procedure included 1) blank subtraction, 2) instrumental drift
156 correction using internal standardization and repeated (every 5 samples) analysis of a drift

157 monitor, and 3) oxide-hydroxide interference correction. The geochemical reference samples
158 with basaltic composition WS-E, PM-S and BIR-1, and the Allende chondrite reference sample
159 (USNM 3529, split 20, position 22) were dissolved and analyzed along with the unknown
160 samples to check the accuracy of the results. In Table 2 are reported the results for the four
161 analyzed reference sample with literature values, and the detection limits for each analyte
162 calculated as six times the standard deviation of the blank counts. The analytical precision is
163 between 5 and 10% RSD for elements with concentrations $> 0.5 \mu\text{g/g}$ and between 10 and 20%
164 RSD for elements with concentrations $< 0.5 \mu\text{g/g}$.

165 To investigate the effect of surface soil chemical composition and to observe any possible
166 significant anomaly, representative soil samples collected from a hilltop close to Estacion
167 Catalina ($25^{\circ}12.673'S$, $69^{\circ}41.894'W$) are used in this study. This region which occurs in the
168 Catalina DCA, is composed of igneous rocks with the presence of caliche layers and has very
169 similar lithology to San Juan and El Médano DCAs. Samples containing gravel and soils were
170 collected from depths of zero to 0.5 cm and 0.5 to 5 cm. Then gravels and soils ($< 2 \text{ mm}$) sieved
171 and separated for subsequent analyses. Soil concentrate samples pulverized to 85% passing 75
172 μm in tungsten-carbide mill. Total chemistry for major and trace elements was determined using
173 ICP-MS and ICP-AES after a lithium metaborate fusion of each soil sample at the ALS
174 Geochemistry©.

175 REE anomalies represent deviations from neighboring elements based on the expected
176 change in REE abundances as a function of atomic number and ionic radii (Lipin and McKay
177 1989). To calculate Ce and Eu anomalies, the values were normalized to the corresponding OC
178 group (LL, L and H) by using the following equations (Dauphas and Pourmand 2015):

179
$$\text{Ce/Ce}^* = \text{Ce}_N / (\text{La}_N^{0.48} \times \text{Pr}_N^{0.52}) \quad (1)$$

$$180 \quad \text{Eu}/\text{Eu}^* = \text{Eu}_N / (\text{Sm}_N^{0.45} \times \text{Gd}_N^0 \text{Gd}_N^{0.55}) \quad (2)$$

181 where N stands for the normalization of REE to mean OC composition.

182 The contents of Sr and Ba have been shown to be suitable chemical proxies to evaluate
 183 the amount of chemical weathering and differentiate the meteorite collection regions (Al-Kathiri
 184 et al. 2005; Folco et al. 2007; Zurfluh et al. 2012). During the same analysis, the concentrations
 185 of these elements were also measured.

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4. RESULTS

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4.1. Meteorite samples

189 REE, Sr and Ba concentrations of the studied meteorites are reported in Table 3. The
 190 REE data of each group are normalized to the corresponding mean composition reported in
 191 Wasson and Kallemeyn (1988). Masuda-Coryell diagrams of the studied meteorites together with
 192 the data of OCs from Oman (Al-Kathiri et al., 2005), UAE (Hezel et al., 2011), Sahara Desert
 193 (Saunier et al., 2010; Folco et al., 2007) and Europe (Folco et al., 2007) are shown in Fig. 3.

194 Both enrichment and depletion elemental patterns compared to the corresponding OC mean
 195 composition can be seen in the diagrams.

196 Σ REE values in the Atacama and Lut samples have a range of 3.55 - 5.77 (avg. 4.46) ppm
 197 and 2.98 - 3.76 (avg. 3.42) ppm for H, 2.97 - 6.45 (avg. 3.97) ppm and 3.26 - 2.93 (avg. 3.63)
 198 ppm for L and 3.21 - 9.82 (5.96) ppm (just for Atacama) for LL OCs, respectively (Table 3).
 199 These data shows a higher amount of REEs in Atacama meteorites than those of Lut.

200 A closer look at the normalized REE patterns (Fig. 3) reveals an enrichment in La for
 201 some samples. It goes up to $2.64 \times H$ in El Médano 049, with an average amount of $1.79 \times H$ (n

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202 = 8) and $1.22 \times H$ ($n = 5$) for Atacama and Lut meteorites, respectively. Likewise, La is enriched
203 up to $2.98 \times L$ in Caleta el Cobre 006 which is an L OC. Average La_N amounts of Atacama and
204 Lut samples are $1.40 \times L$ ($n = 14$) and $1.15 \times L$ ($n = 3$), respectively. Among the studied
205 meteorites, Paposito exhibits a remarkable La enrichment ($4.05 \times LL$).

206 The general shape of the normalized spider diagrams (Fig. 3) indicates the occurrence of
207 a fractionation between light REEs (LREEs) and heavy REEs (HREEs). Fig. 4 displays the
208 values of La_N/Lu_N (as an indicative of LREE/HREE fractionation) vs. La_N for H and L OCs from
209 different regions. Most of the samples show LREE/HREE ratios of more than one. The Atacama
210 and Sahara samples tend to represent higher degrees of REE fractionation than Lut and UAE
211 samples. The Gd_N/Yb_N (as an indicative of MREE/HREE fractionation) ratios for the studied
212 samples are mostly less than one, but since the values are near the error zone, they will not be
213 discussed further.

214 Cerium and Eu show both negative and positive anomalies (Table 3, Fig. 3). In H OCs,
215 the variation range of Ce/Ce^* and Eu/Eu^* values for Atacama samples ($0.60 - 1.69$ and $0.56 -$
216 1.08 , respectively) is broader than those of Lut ($0.92 - 0.96$ and $0.96 - 1.29$, respectively). In
217 general, no pronounced difference in the amounts of Ce/Ce^* and Eu/Eu^* between different OC
218 groups exists. A remarkable Ce positive anomaly is evident in Cobija (H6) and Lutschaunig's
219 Stone (L6) (Fig. 3-a, b). Note that these meteorites are the least weathered samples (W1) studied
220 in Atacama.

221 The negative Tm anomaly among the analyzed samples (Fig. 3) is similar to the Tm
222 anomaly in the CI normalized patterns described from ordinary chondrites (e.g., Dauphas and
223 Pourmand, 2015). However, it is not the case here for OC normalized samples. It should be noted

224 that the seemingly more negative anomalies in H OCs relative to that of L OCs is not real but
225 occurred as the result of the normalization errors.

226 Logarithmic plot of Sr vs. Ba for the OCs from Atacama, Oman and Sahara is shown in
227 Fig. 5. Different chemical trends is particularly more pronounced among those from the Atacama
228 and Oman.

229 4.2. Atacama soil samples

230 The major and trace element composition of the analyzed soil samples are reported in
231 Table 4. As reflected in their SiO₂ content, the parent rock material is basaltic/andesitic but the
232 soil consists more acidic (granitic) material than the gravels, which is blown from neighboring
233 lithologies. Both CI normalized REE patterns of gravels and <2 mm fractions of the soils are
234 fractionated, showing LREE enrichments and a relatively flat HREE (Fig. 6). LREE enrichment
235 in the <2 mm soil is higher than the parent rock gravels (La_N/Lu_N mean values 9.1 vs. 6.7) which
236 is related to the weathering of the parent rock and the higher mobility of LREE into the soils
237 during rock weathering process (Kabata-Pendias and Pendias 2001). The same reason applies
238 for the Eu negative anomaly of the soils compared to the gravels and more resistance of Eu
239 hosting grains during weathering.

240 Comparison of the analyzed samples with the REE composition of Upper Crust
241 Composition (UCC) (Kemp and Hawkesworth 2004) and mean soil composition references
242 (Govindaraju 1994; Kabata-Pendias and Pendias 2001) shows their similarity, which imply a
243 well-mixed texture from various bed rocks. Dissimilar to the Atacama meteorites that generally
244 show higher REE concentrations compared to the other hot desert meteorites, the soils show

245 REE chemical compositions comparable to the UCC and mean soil composition and without any
246 significant compositional difference.

247

248 5. DISCUSSION

249 Our results show that terrestrial weathering may significantly modify the whole rock REE
250 distribution of OCs from hot deserts. However, this modification varies among different deserts.
251 Atacama OCs have higher concentrations of REE and their compositions has been affected to
252 greater extent than those of Lut.

253 5.1. REE mobilization during meteorite weathering: a concise review

254 One of the first plausible evidence for REE mobilization during terrestrial weathering of
255 meteorites was reported by Masuda et al. (1977) and Masuda and Tanaka (1978) on samples
256 from Antarctica. These authors, however, considered the differences in the abundance of REE
257 between the internal and external portions of meteorite samples to be intrinsic and primary rather
258 than produced by terrestrial effects. Further studies on Antarctic achondrites attributed the
259 modified REE compositions to cold desert weathering (Floss and Crozaz 1991; Kagi and
260 Takahashi 1998; Mittlefehldt and Lindstrom 1991; Shimizu et al. 1983; Shinonaga et al. 1994;
261 Swindle et al. 1998). Most prevalent results of weathering effects are LREE-enriched patterns
262 with notable Ce (and in some cases Eu) anomalies produced by aqueous leaching. As proposed
263 by Mittlefehldt and Lindstrom (1991) and confirmed by Floss and Crozaz (1991), during the
264 exposure of a meteorite to atmosphere and its heating by the sun at the surface in Antarctica,
265 surrounding meltwater can penetrate the meteorite through cracks and fractures. In equilibrium
266 with atmospheric CO₂, this fluid becomes a weak carbonic acid solution which leaches the

267 minerals such as Ca-phosphates and causes removal of trivalent REEs. In addition to being in +3
268 state, Ce and Eu occur as tetravalent and divalent ions, respectively. Since Ce⁺⁴ is less soluble
269 than the +3 REEs and Eu is largely retained in the +2 state in plagioclase (which is relatively
270 unaffected by weathering), they are preferentially retained in the sample and thus show their
271 corresponding anomalies. Another study by Kagi and Takahashi (1998) showed a correlation
272 between Ce anomalies and absorbed water content in Antarctic lunar meteorites that
273 demonstrated its terrestrial origin.

274 Following the studies on the Antarctic meteorites, attempts were made to investigate the
275 possible weathering effects on hot desert achondrites (both in mineral and whole rocks). In order
276 to monitor the weathering of recently fallen meteorites, Barrat et al. (1999) analyzed fragments
277 of Tatahouine diogenite recovered just sixty-three years after its fall. By comparing the REE
278 composition with those of the fragments recovered immediately after the fall, elevated amounts
279 of LREE was found in one of the fragments. This study revealed the disturbance of trace element
280 composition of meteorites in relatively short time spans in a terrestrial environment. Crozaz et al.
281 (2003) evaluated the mineral composition of different achondrite groups by secondary ion mass
282 spectrometry (SIMS) and attributed the elevated levels of LREE in olivine and low-Ca pyroxene
283 to terrestrial contamination. However, Barrat et al. (2003) did not find any significant weathering
284 effects on whole rock REE composition of eucrites from the Sahara desert. A similar conclusion
285 was made during a study of NWA 4872 brachinite by Hyde et al. (2014).

286 **5.2. Effects of weathering on the REE composition of Atacama and Lut OCs**

287 The aliquots analyzed in our work were from a large mass (about 2 grams) of well-mixed
288 powder; so the formation of the observed REE patterns by nugget effect of REE-rich components

289 is unlikely. Deviations of more than $\pm 10\%$ from the mean composition of the corresponding
290 meteorite groups assure that the enrichments/depletions are significant.

291 Unlike the Atacama OCs, those from Lut do not show profound changes in ΣREE (Table
292 3). Chemical modification of a meteorite during weathering, including changes in REE content,
293 is controlled by several factors such as, primary chemical composition, modal composition,
294 degree of recrystallization, shock stage, terrestrial age and locality (Croaz et al. 2003). REEs
295 which are mostly concentrated in Ca-phosphates and silicates (Ebihara and Honda 1984) are
296 released during the weathering of these minerals. This process is facilitated by formation of
297 sulfuric acid inside the meteorite as a result of troilite (FeS) oxidation in contact with meteoric
298 water (e.g. Bland et al., 2006). With the development of primary mineral dissolution, REEs tend
299 to concentrate in veins filled with iron oxides/oxyhydroxides (Thiagarajan and Aeolus Lee
300 2004).

301 Contrary to meteorites from Antarctic which release their REE during weathering on ice,
302 an increase in REE concentration is expected for hot desert meteorites as a consequence of
303 residing on a relatively REE-rich soil surface (Croaz and Wadhwa 2001). Once a meteorite with
304 its relatively high initial porosity falls on a hot desert surface, soil salts dissolved by water
305 infiltrate into it by capillary forces triggered by temperature fluctuations (Zurfluh 2012). In
306 addition, wind activity, burial in soil and desert varnish formation introduces terrestrial minerals
307 containing REE, or develops the occurrence of primary minerals with modified REE patterns
308 (Croaz et al. 2003). We suggest that the notable increase in ΣREE contents of the Atacama OCs
309 is caused by these kind of implementations.

310 Is the modification in REEs concentration controlled by the weathering degree of a
311 meteorite? A look at the data of Lut OCs (Tables 1 and 3) shows that despite having high

312 weathering degrees, their Σ REE does not change as much as the Atacama samples. For example,
313 Kerman 001 (W4), which was half-buried in a valley in the Kalout formation of the Lut with a
314 soil composed of clay, sand, evaporites and carbonates (Farpoor and Krouse 2008) at the time of
315 recovery and exhibits well developed parallel weathering veins (Fig. 2-a), does not show any
316 noticeable deviation from the mean composition. In comparison, the Atacama samples which
317 generally are less weathered show higher Σ REE. For example, El Médano 049 which contains
318 the highest Σ REE among the studied H OCs and was recovered from a igneous rocks deflation
319 surface is also highly weathered (W3, Fig. 2-b), although less than Kerman 001. Comparing to
320 samples from other regions, we see that three highly weathered H OCs from UAE (Hezel et al.
321 2011) and even weathered samples from Europe (Folco et al. 2007) represent Σ REE contents
322 similar to Lut samples. However, Σ REE is elevated in highly weathered Saharan OCs (analyzed
323 by Saunier et al., 2010) and Aridal 006 (W4). Therefore, the meteorite weathering as shown by
324 weathering degree (Wlotzka 1993) and veins filled with secondary products is not the only factor
325 responsible for Σ REE increase.

326 As for REEs, different chemical trends in Sr and Ba is also evident in meteorites from
327 different regions (Fig. 5 & 7). Al-Kathiri et al. (2005) and Folco et al. (2007) showed that the
328 weathering of meteorites in Oman and Sahara deserts (respectively) increases the Sr and Ba
329 contents of meteorites. The carbonate basement and the Sr and Ba-rich soils in Oman deserts is
330 responsible for the further enrichments of these elements in the recovered meteorites from the
331 region. In some cases from Oman, high concentrations of Sr and Ba in the soil, leads to the
332 formation of secondary celestine and barite inside the meteorites (Al-Kathiri et al. 2005).
333 Meteorites from Atacama exhibit a different trend compared to Oman meteorites (Fig. 5). The
334 majority of collecting site of the meteorites from Atacama is mostly made of volcanics and

335 evaporites (Valenzuela 2011), which usually have lower concentrations of Sr and Ba than
336 carbonates. The difference in the soil type can explain different trends in the Sr budget.
337 Compared to the Omani and Atacama OCs, the samples from Lut desert have very low Ba
338 contents (< 10 ppm). On the other hand, Sr concentrations of Lut OCs, are in the range of Omani
339 meteorites. Considering Sr and La contents (Fig. 7), again we see difference in the chemical
340 composition of meteorites from different regions. This differences can be used as proxies which
341 allow distinguishing Atacama meteorites from Oman and Lut. Al-Kathiri et al. (2005) report a
342 positive correlation between weathering degree and Sr, Ba concentration for Omani OCs,
343 however we do not see any correlation between these parameters for the meteorites from
344 Atacama and Lut samples.

345 The Atacama desert has exceptionally high meteorite concentrations compared to other
346 hot deserts. In a study of 22 meteorites, Gattacceca et al. (2011) found half of the meteorites with
347 terrestrial ages older than 20 ka, which is much older than the age ranges reported from other hot
348 deserts. We may hypothesize that the generally high terrestrial age of the Atacama meteorites is
349 responsible for their high Σ REE amounts. The similar effect of terrestrial age on REE content
350 can be seen in higher amounts of La in very old meteorites from Atacama and in samples with
351 terrestrial ages of more than >30 kyr from Oman (Al-Kathiri et al. 2005) (Fig. 3-a).

352 Most of the Atacama and some highly weathered Saharan OCs show differentiated REE
353 patterns (Fig. 3 and Fig. 4). Since the soil and sedimentary materials typically exhibit higher
354 LREE contents (Aide and Aide 2012), as observed in our analyzed samples from the Atacama,
355 the residence of a meteorite on a soil surface increases the LREE content of the meteorite (Al-
356 Kathiri et al. 2005; Crozaz et al. 2003; Dreibus et al. 2001). The studied meteorites and their
357 patterns which tend to converge toward that of the surface material is consistent with the

358 assumption that soil has affected their chemical composition (Fig. 6). Taking into account the
359 different soil types of the meteorite collecting sites (carbonate in Oman versus
360 volcanic/evaporitic in Atacama and clay/sand in Lut), it can be deduced that nearly all of the
361 meteorites represent approximately similar LREE-enriched patterns which forms as a result of
362 both: (i) higher abundance of LREE in soils and (ii) higher solubility of LREE (Aide and Aide,
363 2012 - and references therein).

364 Different clay and iron oxide/oxyhydroxide minerals display specific Ce and Eu
365 anomalies. These minerals which are the main components of soils, occur in different
366 proportions based on the soil type; as a result different Ce and Eu anomalies is exhibited in
367 different soils (Laveuf and Cornu 2009). Although most of the Ce anomalies in our data are
368 small and hardly significant, two slightly weathered (W1) meteorites (Cobija and Lutschaunig's
369 Stone) display notable positive Ce anomalies (Fig. 3a, b) that might reveal the incipient phase of
370 REE modification during weathering. Cerium is the most abundant REE in the average soil
371 composition (Aide and Aide 2012). Once the meteorite falls, the initial physical soil
372 contamination through cracks and fractures gives it a positive Ce anomaly. By developing the
373 chemical weathering, trivalent REEs of the soil with higher solubility rates than Ce⁺⁴ infiltrate
374 the meteorite and most of the Ce oxides to Ce⁺⁴ and preferentially retains in the soil. Progressive
375 alteration of the meteorite and the addition of trivalent REEs removes the initial Ce positive
376 anomaly. At higher residence times and more interaction of terrestrial contaminants with
377 meteorite, the amount of La and Pr exceeds the Ce concentration and creates a negative Ce
378 anomaly that is evident in the REE patterns of some meteorites with high terrestrial ages such as,
379 La Yesera 003, Caleta el Cobre 006, Paposo and El Médano 049 (Fig. 3). Variation in Eu

380 anomalies among studied meteorites can result from contamination by different soil components,
381 although the presence of possible primary plagioclase should not be excluded.

382 Putting these data together, we hypothesize that terrestrial age rather than weathering
383 degree is the main governing factor of REE content and its modification in hot desert OCs, even
384 though weathering degree and terrestrial age at a given recovery site show a positive correlation
385 (e.g. Bland et al., 2006). Indeed, the formation of secondary oxidation products decreases the
386 porosity of the meteorite by filling the fractures and cracks which are the pathways for REE
387 containing terrestrial fluid to circulate and solid materials to penetrate (Bland et al. 2006). The
388 earlier this process initiates, the lower the possibility that a terrestrial material is able to affect the
389 REE composition of the meteorite. But in areas with low weathering rates (like Atacama),
390 greater amounts of soil and fluid penetrate the meteorite and modify the composition readily.

391 We observe that since the meteorites from Atacama are older than those from Lut and
392 other regions (Sahara, Oman), and also owing to low weathering rates in the former, more time
393 has been available for the terrestrial environment to affect the chemistry of Atacama OCs by
394 means of different interactions. The terrestrial ages of Saharan meteorites used in this study are
395 not determined, however by comparing weathering degree with terrestrial age of other meteorites
396 collected from this desert (e.g. Welten et al., 2004), their terrestrial ages is probably high. The
397 prevailing humid environment for UAE and Europe and thus higher weathering rates (Folco et
398 al. 2007; Hezel et al. 2011), might have prevented the formation of a pronounced Σ REE
399 increment in meteorites from these area. We believe that the terrestrial age range of highly
400 weathered Lut meteorites shall be in the range of UAE meteorites. The occurrence of
401 archeological sites in the surroundings of Lut (e.g. Muscarella, 2001), points to wetter climatic
402 conditions, at least periodically over the last 10 kyr compared to nowadays. This is in agreement

403 with the high weathering degree and the lack of a pronounced Σ REE change in the chemical
404 composition of Lut OCs.

405 A large number of meteorites from Atacama is expected to be recovered in the future
406 (Gattacceca et al. 2011; Hutzler et al. 2016). So care must be taken into account while working
407 on the REE composition of even “fresh-looking” Atacama meteorites.

408

409 6. CONCLUSIONS

410 The chemical compositions of OCs from Atacama and Lut along with the published data
411 from other hot desert areas reveal a difference in the effects of terrestrial weathering on REE
412 distribution between these populations. Optimal conditions for meteorite preservation in the
413 Atacama desert (i.e., high stability of the surface and low meteorite weathering rates) provide
414 enough time for the meteorites to interact efficiently with terrestrial materials. In contrast, higher
415 weathering rates of the meteorites from other hot deserts (specifically Lut) prevent such
416 interaction. As a consequence, the highly weathered meteorites from Lut desert have lower
417 Σ REE contents than those of the moderately weathered meteorite from Atacama. In some cases,
418 even the OCs with minor weathering from Atacama show modified REE patterns with notable
419 Ce positive anomalies. These anomalies may have been created by the initial physical
420 contamination (the addition of soil or dust directly into the fractures, rather than more time-
421 consuming chemical weathering) and then by the addition of trivalent REEs. Notwithstanding
422 the lack of terrestrial age data for Lut meteorites, it is envisaged on the basis of almost unaffected
423 REE compositions and high weathering degrees that this area was wetter than nowadays during
424 the last tens of thousands of years and generally had a climate like UAE and Oman deserts.

425 Future investigations on the mobilization of REE isotopes, which are being used
426 frequently in cosmochemistry, are needed to estimate their behavior during weathering.

427

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436

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650 **Tables heading:**

651 **Table 1:** Meteorites studied in this work.

652 **Table 2:** Concentrations ($\mu\text{g/g}$) of Sr, Ba and REE in the four geochemical reference materials analyzed
653 by ICP-MS along with the unknown samples. Literature values for these reference materials are also
654 reported. The last column reports the detection limits ($\mu\text{g/g}$) of the used analytical method.

655 **Table 3:** Strontium, Ba and bulk REE composition ($\mu\text{g/g}$) and the calculated ratios of the studied
656 meteorites.

657 **Table 4:** Major and trace element chemical composition of soil and gravel sample from Estacion
658 Catalina in the Atacama Desert. Trace element concentration unit is ppm.

659

660 **Figure captions:**

661 **Fig. 1:** a) Map of the central Lut Desert and the finding places of the studied meteorites. b) Map
662 of the Atacama Desert and the location of the dense collection areas which their meteorites were
663 used in this study.

664 **Fig. 2:** Mosaic pictures of a) Kerman 001 and b) El Médano 049 prepared in reflected polarized
665 light show the different weathering types of the meteorites. Kerman 001 contains clear parallel
666 veins filled with weathering products and represents much strong weathering than El Médano
667 049 whose Fe-Ni metal and troilite are not completely altered yet.

668 **Fig. 3:** Normalized REE spider diagrams of a) H, b) L and c) LL ordinary chondrites. For
669 comparison, the data of OCs from Oman (the average values for different age classes) (Al-
670 Kathiri et al., 2005), U.A.E. (Hezel et al., 2011), Sahara Desert (Saunier et al., 2010; Folco et
671 al., 2007) and Europe (Folco et al., 2007) are used along the data of this work. PdM is the
672 abbreviated form of Pampa de Mejillones.

673 **Fig. 4:** La_N/Lu_N vs. La_N for H and L OCs from different regions. For comparison, the data of OCs
674 from Oman (Al-Kathiri et al., 2005), U A EE (Hezel et al., 2011), Sahara Desert (Saunier et al.,
675 2010; Folco et al., 2007) and Europe (Folco et al., 2007) are used along the data of this work.

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676 **Fig. 5:** Sr vs. Ba logarithmic plot showing OCs from various areas. The data of OCs from Oman
 677 (Al-Kathiri et al., 2005), Sahara Desert (Saunier et al., 2010; Folco et al., 2007) and Europe
 678 (Folco et al., 2007) are used along the data of this work. Some samples from Lut with Ba <10
 679 ppm are also included.

680 **Fig. 6:** REE chemical composition of Estacion Catalina soil composition compared to the mean
 681 soil compositions (Govindaraju 1994; Kabata-Pendias and Pendias 2001). The Atacama H and L
 682 OCs show LREE enrichment in relation to falls in an attempt to reach equilibrium during the hot
 683 desert terrestrial weathering.

684 **Fig. 7:** Different chemical trends of OCs from different regions can be seen in the Sr vs. La_N
 685 plot. The data of L and H OCs from Oman, in different age classes (Al-Kathiri et al., 2005),
 686 Sahara Desert (Saunier et al., 2010; Saunier et al., 2010; Folco et al., 2007) and Europe (Folco et
 687 al., 2007) are used along the data of this work.

688

689 Table 1 (Pourkhorsandi et al.)

Meteorite	Type	Weathering degree	Terrestrial age (ka) ^a
Ordinary chondrites from Atacama			
La Yesera 001	H6	W3	10.79 ± 1.56
La Yesera 002	LL5	W2	25.44 ± 4.45
La Yesera 003	L4	W4	16.98 ± 2.47
La Yesera 004	L6	W3	34.07 ± 1.92
Pampa (a)	L6	W2	25.08 ± 1.46
Pampa (b)	L4/5	W3	21.29 ± 2.45
Pampa (c)	L4	W4	13.89 ± 2.08
Pampa (d)	L5	W2/3	14.18 ± 1.91
Pampa (g)	L5	W3	14.34 ± 1.62
Pampa de Mejillones 002	H5	W3	3.86 ± 1.36
Pampa de Mejillones 004	L6	W4/5	23.98 ± 4.40

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Pampa de Mejillones 007	L6	W4	>27.9
Pampa de Mejillones 010	L5	W3	18.06 ± 1.90
Pampa de Mejillones 011	L5	W5	4.35 ± 1.34
Pampa de Mejillones 012	H4/5	W5	>34.3
Pampa de Mejillones 014	L/LL4-6	W2	12.54 ± 1.95
San Juan 01	L5	W2	>28.1
San Juan 02	H6	W2	19.44 ± 1.69
El Médano 049	H4	W3	1890 ± 80
Estacion Imilac	H5	W1	-
Cobija	H6	W1	19.7 ± 4.20
Rencoret	H6	W3	25.3 ± 6.4
Lutschaunig's Stone	L6	W1	9.2 ± 1.6
Caleta el Cobre 006	L6	W3	2345 ± 80
Paposo	LL6	W2	>49.2
Ordinary chondrites from Lut			
Lut 001	H5	W3	-
Lut 003	L3	W3	-
Lut 006	LL3	W3	-
Lut 008	H4	W5	-
Lut 009	H4	W4	-
Kerman 001	H5	W4	-
Kerman 002	L6	W3	-
Kerman 003	L5	W2	-
Shahdad	H5	W4	-
Ordinary chondrites from Sahara			
Aridal 006	H6	W4	-

690 ^aTerrestrial age data from Valenzuela (2011) by ¹⁴C, and Hutzler (2015) by ³⁶Cl/⁴¹Ca.

691

692 Table 2 (Pourkhorsandi et al.)

Element	Allende (this work)	Allende (literature) ^a	BIR-1 (this work)	BIR-1 (literature) ^b	WS-E (this work)	WS-E (literature) ^c	PM-S (this work)	PM-S (literature) ^c	detection limit
Sr	14.5	12 ± 3	122	108	450	410 ± 5	307	280 ± 5	0.5
Ba	4.5	4 ± 1	< 10	7	342	338 ± 6	139	148 ± 3.2	0.6 - 10
La	0.49	0.52 ± 0.04	0.62	0.62	27.6	27 ± 1.1	2.63	2.8 ± 0.17	0.02
Ce	1.24	1.33 ± 0.08	1.90	1.95	60	61 ± 1.4	6.5	6.8 ± 1.25	0.06
Pr	0.21	0.21 ± 0.01	0.39	0.38	8.0	7.8 ± 0.4	1.05	1.08 ± 0.16	0.007
Nd	1.01	0.99 ± 0.03	2.45	2.5	33.3	33 ± 0.7	5.4	5.5 ± 0.25	0.03

Sm	0.33	0.34 ± 0.02	1.10	1.1	8.8	8.8 ± 0.3	1.78	1.75 ± 0.05	0.02
Eu	0.09	0.11 ± 0.01	0.53	0.54	2.31	2.25 ± 0.04	1.04	1.07 ± 0.04	0.05
Gd	0.39	0.42 ± 0.02	1.96	1.85	7.4	7.2 ± 0.23	2.05	2 ± 0.1	0.004
Tb	0.07	0.081 ± 0.010	0.38	0.36	1.10	1.1 ± 0.04	0.35	0.36 ± 0.02	0.002
Dy	0.46	0.42 ± 0.03	2.78	2.5	6.5	6 ± 0.16	2.18	2 ± 0.1	0.005
Ho	0.10	0.10 ± 0.01	0.59	0.57	1.18	1.2 ± 0.06	0.42	0.42 ± 0.03	0.006
Er	0.29	0.29 ± 0.01	1.81	1.7	3.17	3 ± 0.11	1.19	1.1 ± 0.06	0.01
Tm	0.05	0.0572	0.26	0.26	0.42	0.43 ± 0.03	0.16	0.17 ± 0.01	0.003
Yb	0.31	0.30 ± 0.02	1.70	1.65	2.56	2.5 ± 0.1	0.96	1 ± 0.05	0.007
Lu	0.04	0.052 ± 0.006	0.26	0.26	0.36	0.37 ± 0.01	0.14	0.15 ± 0.01	0.002

693 ^aJarosewich et al. (1987).

694 ^bGovindaraju (1994).

695 ^cGovindaraju (1995).

696 Table 3 (Pourkhorsandi et al.)

Meteorite	Sr	Ba	La	Ce	Pr	Nd	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	ΣREE	Ce/Ce*	Eu/Eu*	Tm/Tm*	La/Lu*	La/Sm*	Gd/Yb*	Sr/Ba
La Yesera 001	30.6	38	0.76	1.63	0.22	1.01	0.27	<0.05	0.31	0.05	0.38	0.08	0.23	0.03	0.22	0.03	5.3	0.92	0.56	0.86	2.27	1.73	0.98	0.79
La Yesera 002	26.3	50	0.54	1.54	0.18	0.81	0.23	0.10	0.31	0.05	0.39	0.08	0.25	0.04	0.25	0.03	4.8	1.08	1.28	1.13	1.73	1.52	0.90	0.53
La Yesera 003	37	12.6	0.45	1.12	0.17	0.86	0.26	<0.05	0.29	0.05	0.38	0.09	0.27	0.04	0.25	0.05	4.3	0.90	0.58	0.92	1.04	1.07	0.83	2.99
La Yesera 004	12.4	7.5	0.31	0.81	0.13	0.61	0.20	0.07	0.25	0.05	0.31	0.08	0.21	0.03	0.21	0.03	3.3	0.93	1.03	0.87	0.93	0.97	0.84	1.67
Pampa (a)	18.1	4.1	0.30	0.85	0.12	0.61	0.21	0.10	0.24	0.05	0.33	0.08	0.22	0.03	0.23	0.03	3.4	0.99	1.41	0.88	1.16	0.93	0.75	4.41
Pampa (b)	21.1	5.0	0.29	0.75	0.12	0.52	0.23	0.07	0.23	0.04	0.29	0.07	0.19	0.03	0.20	0.03	3.06	0.91	0.97	0.93	1.14	0.81	0.84	4.21
Pampa (c)	31.8	14.6	0.40	0.94	0.15	0.70	0.23	0.09	0.29	0.05	0.35	0.08	0.26	0.04	0.25	0.04	3.9	0.86	1.06	0.89	1.17	1.10	0.81	2.18
Pampa (d)	21.3	4.8	0.26	0.70	0.11	0.59	0.19	<0.05	0.24	0.04	0.31	0.07	0.21	0.03	0.19	0.03	3.0	0.95	0.75	0.92	0.87	0.84	0.87	4.39
Pampa (g)	38.6	42	0.55	1.08	0.17	0.82	0.26	<0.05	0.32	0.05	0.35	0.09	0.26	0.03	0.23	0.04	4.2	0.81	0.55	0.83	1.39	1.31	0.99	0.92
Pampa de Mejillones 002	17.6	5.9	0.43	1.22	0.17	0.73	0.23	0.07	0.25	0.05	0.31	0.08	0.23	0.03	0.23	0.03	4.1	1.05	0.95	0.84	1.45	1.19	0.76	2.96
Pampa de Mejillones 004	23.6	8.3	0.39	0.97	0.14	0.68	0.21	0.08	0.25	0.05	0.31	0.07	0.20	0.03	0.21	0.03	3.6	0.92	1.07	0.97	1.24	1.16	0.85	2.82
Pampa de Mejillones 007	42	66	0.66	1.13	0.20	0.92	0.25	0.10	0.32	0.06	0.37	0.08	0.25	0.03	0.23	0.03	4.6	0.70	1.13	0.88	1.91	1.65	0.98	0.64
Pampa de Mejillones 010	26.1	5.3	0.29	0.88	0.12	0.55	0.21	0.08	0.26	0.05	0.31	0.07	0.19	0.03	0.23	0.03	3.3	1.06	1.10	0.85	0.97	0.90	0.80	4.91
Pampa de Mejillones 011	11.3	5.1	0.32	0.82	0.12	0.62	0.22	0.07	0.27	0.05	0.33	0.08	0.24	0.04	0.22	0.03	3.4	0.94	0.94	1.02	1.19	0.90	0.85	2.19
Pampa de Mejillones 012	39	21.3	0.43	1.07	0.14	0.68	0.20	0.07	0.25	0.05	0.33	0.07	0.23	0.03	0.21	0.02	3.8	1.00	1.08	0.84	1.69	1.36	0.80	1.85
Pampa de Mejillones 014	18.4	12.5	0.27	0.92	0.12	0.59	0.19	<0.05	0.25	0.05	0.30	0.08	0.200	0.03	0.19	0.03	3.2	1.11	0.74	1.08	0.92	0.88	0.94	1.47
San Juan 001	71	52	0.57	1.17	0.18	0.77	0.24	0.11	0.27	0.05	0.34	0.07	0.20	0.03	0.21	0.03	4.3	0.83	1.35	1.00	1.77	1.48	0.89	1.37
San Juan 002	39	9.4	0.38	0.91	0.14	0.69	0.18	<0.05	0.24	0.05	0.32	0.08	0.24	0.03	0.21	0.03	3.5	0.92	0.77	0.75	1.12	1.28	0.80	4.24
El Médano 049	45	270	0.78	1.34	0.25	1.22	0.31	0.10	0.44	0.07	0.49	0.10	0.30	0.04	0.27	0.04	5.8	0.69	0.91	0.82	1.87	1.56	1.13	0.17
Estacion Imilac	11.8	9.8	0.44	1.03	0.16	0.71	0.21	0.07	0.27	0.05	0.34	0.08	0.24	0.03	0.24	0.03	3.9	0.90	0.97	0.82	1.35	1.29	0.78	1.20
Cobija	15.8	6.2	0.33	1.48	0.12	0.62	0.19	0.07	0.25	0.04	0.33	0.07	0.21	0.03	0.19	0.03	4.0	1.69	1.05	0.85	1.03	1.08	0.89	2.57
Rencoret	26.1	12.9	0.68	1.62	0.22	1.01	0.27	0.09	0.32	0.05	0.35	0.08	0.25	0.04	0.24	0.03	5.2	0.97	0.97	0.81	2.04	1.58	0.89	2.01
Lutschmannig's Stone	10.4	4.2	0.35	1.83	0.14	0.69	0.22	0.07	0.30	0.05	0.38	0.09	0.25	0.03	0.23	0.03	4.7	1.84	0.92	0.89	1.06	0.98	0.92	2.48
Caleta el Cobre 006	72	326	0.92	1.89	0.28	1.22	0.32	0.10	0.40	0.07	0.48	0.10	0.31	0.04	0.28	0.04	6.4	0.85	0.87	0.90	2.46	1.80	1.01	0.22

Paposo	41	70	1.48	3.49	0.50	1.88	0.46	0.10	0.48	0.08	0.52	0.11	0.31	0.05	0.30	0.05	9.8	0.88	0.68	0.99	2.92	2.06	1.15	0.58
Lut 001	62	<10	0.30	0.72	0.11	0.50	0.18	0.08	0.23	0.04	0.30	0.06	0.20	0.03	0.20	0.03	2.98	0.92	1.28	0.82	1.02	1.06	0.80	6.2
Lut 003	91	<10	0.34	0.81	0.12	0.58	0.18	0.07	0.26	0.04	0.33	0.07	0.20	0.03	0.19	0.03	3.26	0.90	1.01	0.94	1.20	1.20	0.94	9.1
Lut 006	126	20.8	0.37	0.92	0.14	0.68	0.23	0.08	0.29	0.06	0.37	0.08	0.24	0.04	0.26	0.03	3.8	0.86	1.01	0.96	1.10	1.04	0.82	6.1
Lut 008	72	<10	0.31	0.79	0.12	0.62	0.20	0.07	0.26	0.05	0.32	0.07	0.20	0.03	0.20	0.03	3.28	0.96	0.96	0.86	1.01	0.99	0.87	7.1
Lut 009	497	14.3	0.33	0.84	0.12	0.62	0.19	0.07	0.25	0.05	0.31	0.07	0.20	0.03	0.20	0.03	3.3	0.95	1.08	0.86	1.03	1.08	0.85	34
Kerman 001	165	<10	0.38	0.93	0.14	0.70	0.21	0.08	0.30	0.05	0.36	0.08	0.22	0.03	0.22	0.03	3.7	0.93	0.99	0.87	1.16	1.12	0.96	16.5
Kerman 002	33	<10	0.35	0.90	0.14	0.66	0.22	0.08	0.29	0.05	0.38	0.08	0.24	0.03	0.23	0.04	3.7	0.91	1.00	0.89	1.07	1.02	0.88	3.3
Kerman 003	41	<10	0.43	1.00	0.14	0.70	0.22	0.08	0.30	0.05	0.37	0.08	0.24	0.03	0.24	0.04	3.9	0.92	1.06	0.87	1.27	1.23	0.89	4.1
Shahdad	141	<10	0.47	1.04	0.14	0.66	0.20	0.07	0.25	0.05	0.33	0.07	0.20	0.03	0.21	0.03	3.8	0.93	1.07	0.85	1.63	1.46	0.79	14.1
Aridal 006	17.9	23.7	0.29	0.70	0.10	0.49	0.16	0.08	0.21	0.04	0.28	0.06	0.18	0.03	0.18	0.03	2.84	0.94	1.49	0.82	1.01	1.14	0.80	0.76

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705 Table 4 (Pourkhorsandi et al.)

	SV2 < 2mm soil				SV1 gravels			
	CS-Surface-01	CS-Surface-02	CS-1-01	CS-1-02	CS-Surface	CS-1-01	CS-1-02	CS-5
Depth (cm)	0-0.5	0-0.5	0.5-5	0.5-5	0-0.5	0-0.5	0.5-5	>34
SiO₂	59.8	60.4	58	58	53	52.5	57.8	52.5
Al₂O₃	14.65	14.65	14.6	14.65	17.35	17	16.6	17.15
Fe₂O₃	8.09	7.88	7.56	7.62	8.13	8.29	6.99	8.21
CaO	4.42	4.11	3.9	3.97	8.88	8.72	7.46	9.44
MgO	2.11	1.98	2.19	2.23	5.52	5.51	3.82	5.55
Na₂O	2.96	2.84	2.48	2.53	2.94	2.8	2.75	3.04
K₂O	2.42	2.37	2.31	2.31	1.17	1.12	1.3	1.11
Cr₂O₃	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
TiO₂	1.19	1.16	1.08	1.1	1.05	1.06	0.98	1.07
MnO	0.14	0.16	0.12	0.13	0.13	0.14	0.15	0.13
P₂O₅	0.19	0.2	0.21	0.23	0.27	0.23	0.23	0.23
Total %	96.14	95.92	92.6	92.92	98.6	97.53	98.27	98.58
Rb	84.1	85.4	95	91.6	30.5	25.4	35.6	19.2
Sr	525	483	423	421	931	865	878	921
Ba	931	954	789	782	412	384	779	360
La	25.8	26.1	26.1	25.8	19.6	18.8	20.1	17.9
Ce	52	53	52.6	52.4	42.2	40.4	45.6	38.7
Pr	6.3	6.4	6.4	6.4	5.3	5.1	5.3	4.9
Nd	24.3	24.3	25	24.4	21.8	21.1	21.5	20.5
Sm	4.9	4.6	4.9	4.8	4.3	4.3	4.3	4.1
Eu	1.2	1.3	1.2	1.2	1.4	1.4	1.4	1.4
Gd	5	4.7	5	4.7	4.8	4.4	4.8	4.4
Tb	0.7	0.7	0.7	0.6	0.7	0.6	0.7	0.6
Dy	3.7	3.7	3.8	3.8	3.8	3.8	3.9	3.6
Ho	0.7	0.7	0.7	0.7	0.7	0.7	0.7	0.7
Er	2.3	2.2	2.3	2.2	2.2	2.2	2.3	2.1
Tm	0.3	0.3	0.3	0.3	0.3	0.2	0.3	0.3
Yb	2.1	2.1	2.1	2	2	1.9	2.1	1.9
Lu	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3
La_N/Lu_N	9.1	9.2	9.2	9.1	6.9	6.6	7.1	6.3

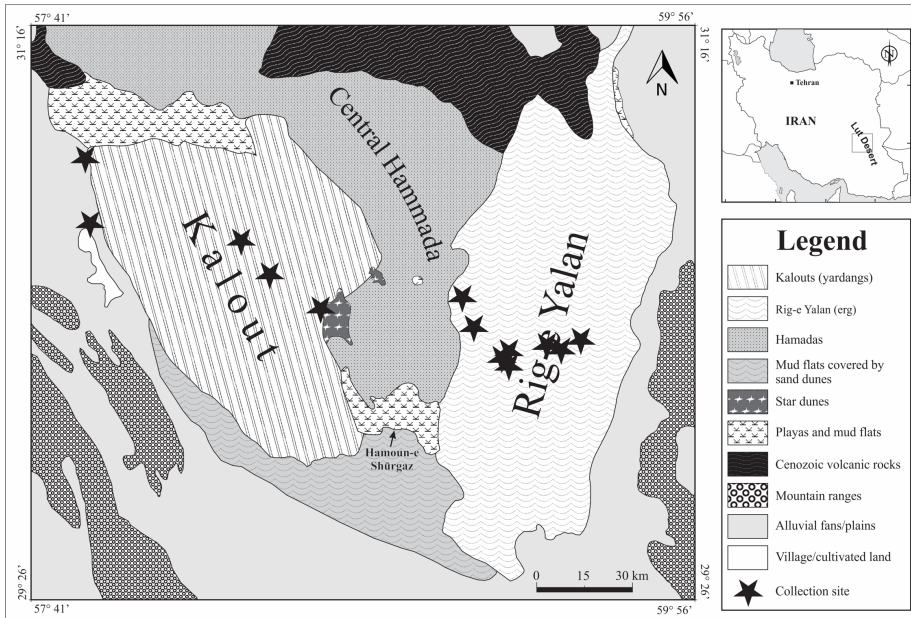
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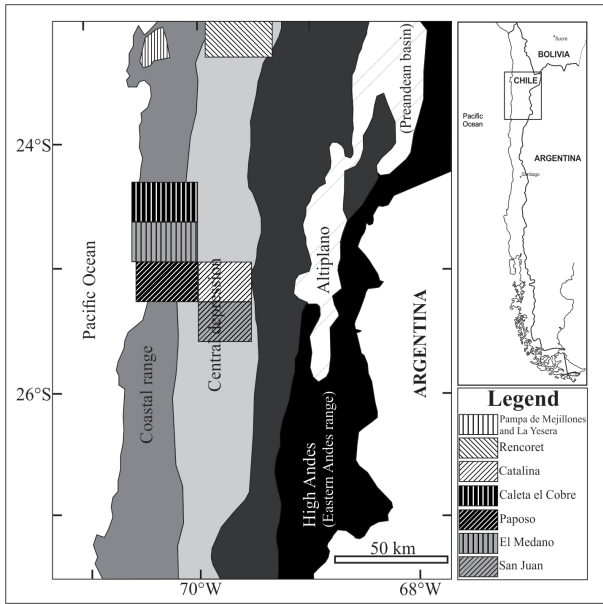
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710 Fig. 1-a. (Pourkhorsandi et al.)



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725 Fig. 1-b. (Pourkhorsandi et al.)



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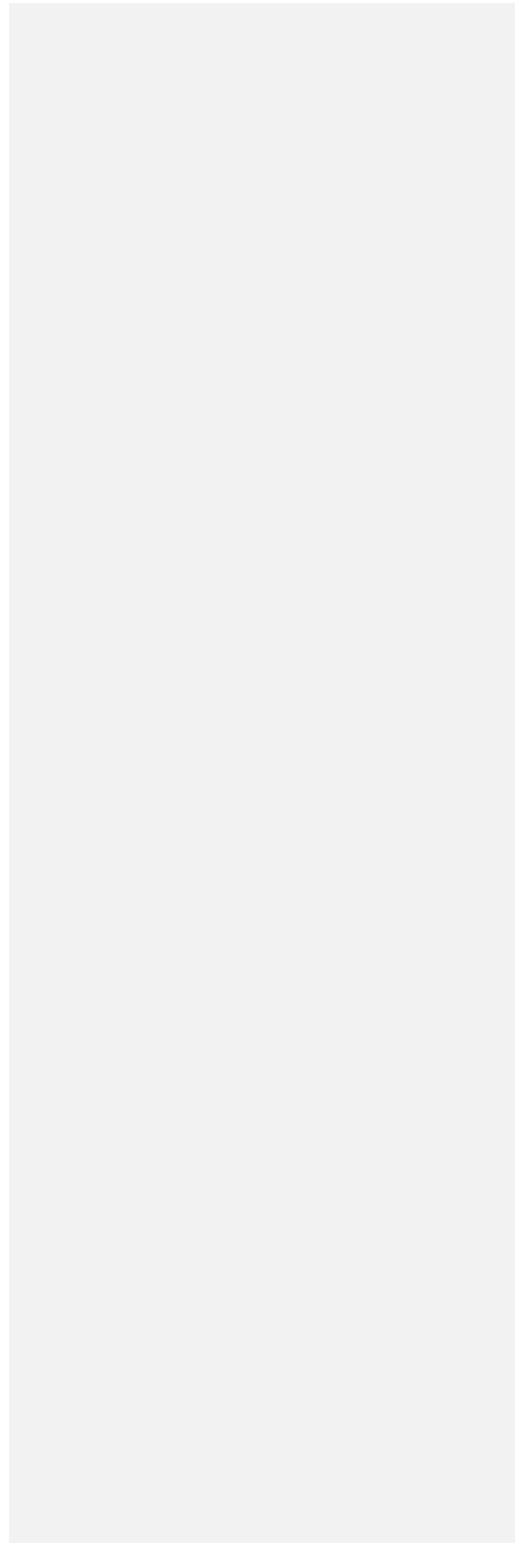
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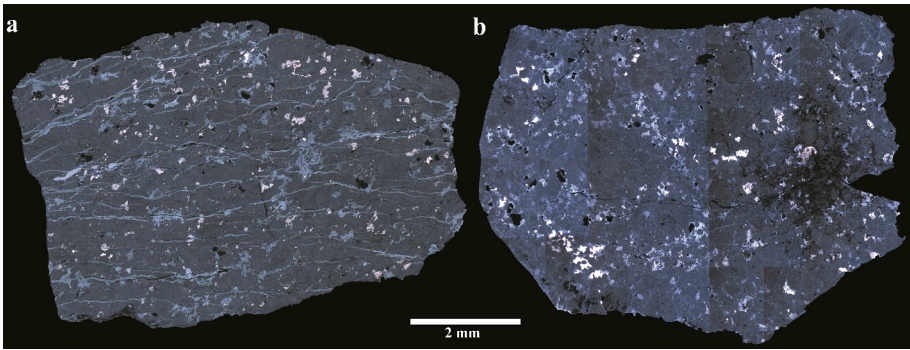
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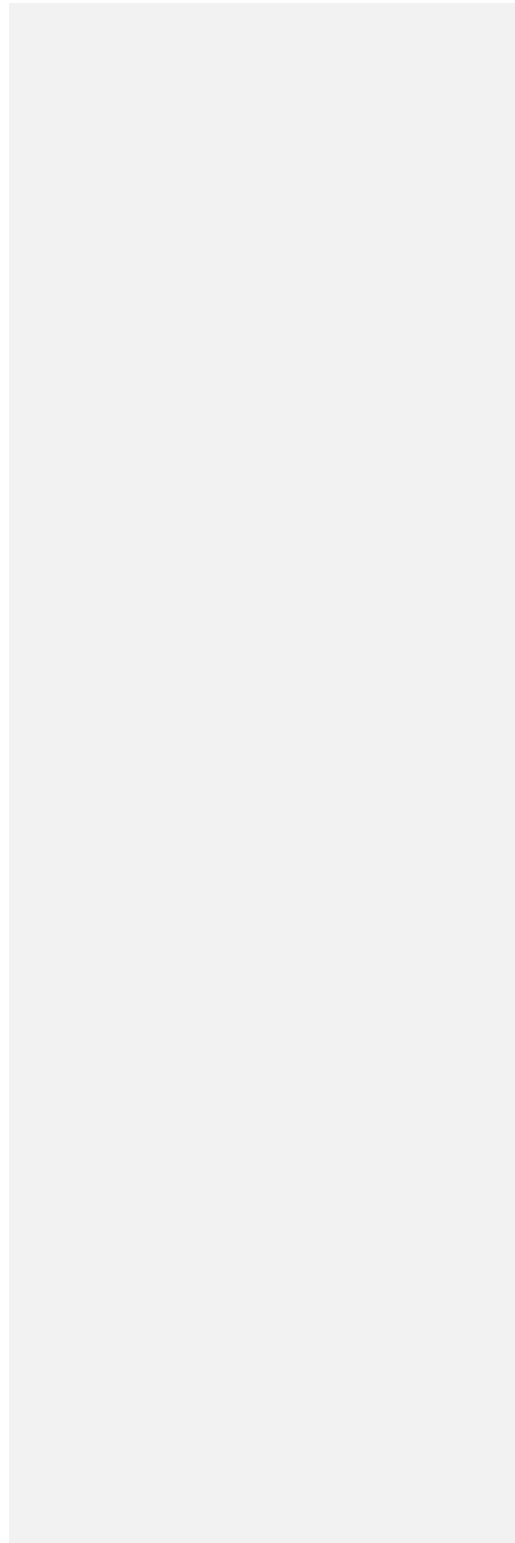
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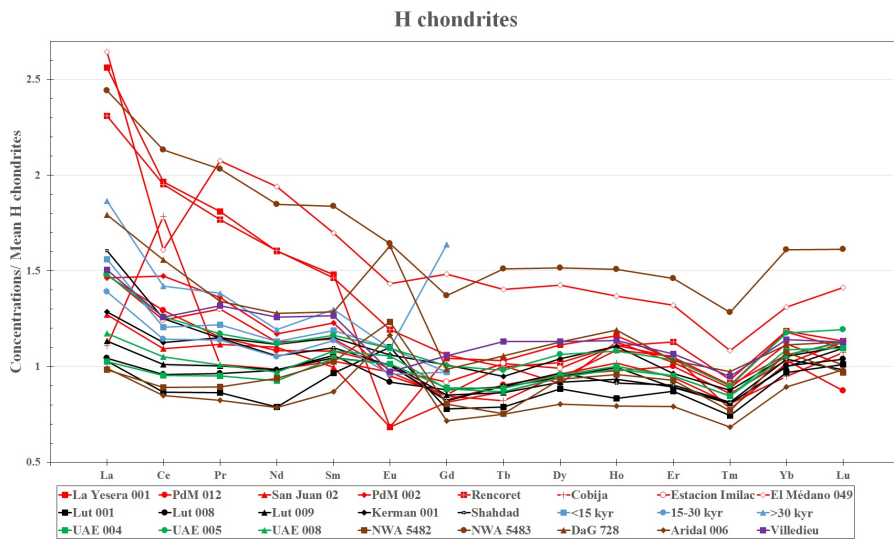
735 Fig. 2. (Pourkhorsandi et al.)



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756 Fig. 3-a. (Pourkhorsandi et al.)



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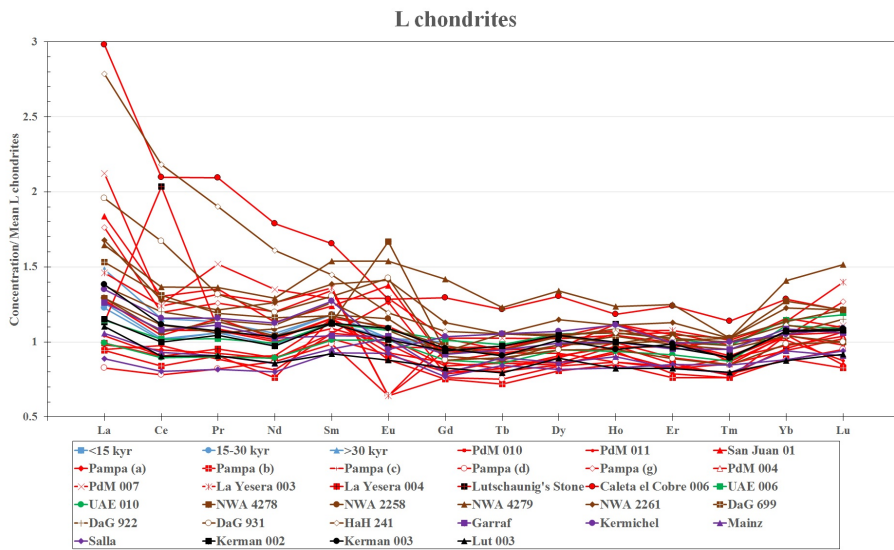
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772 Fig. 3-b. (Pourkhorsandi et al.)



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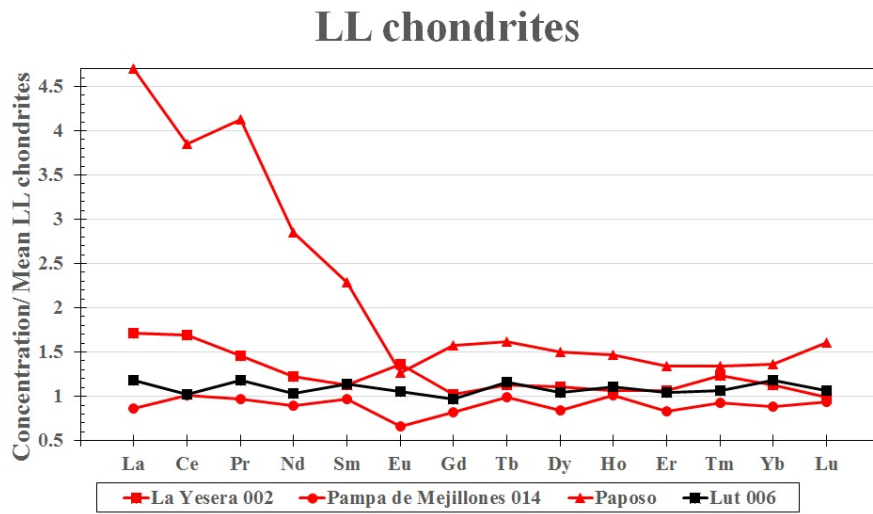
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784 Fig. 3-c. (Pourkhorsandi et al.)



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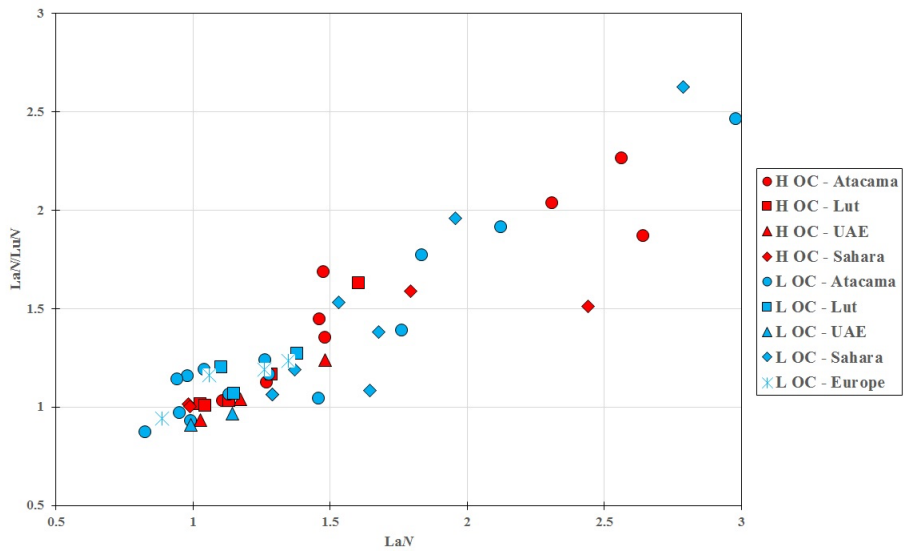
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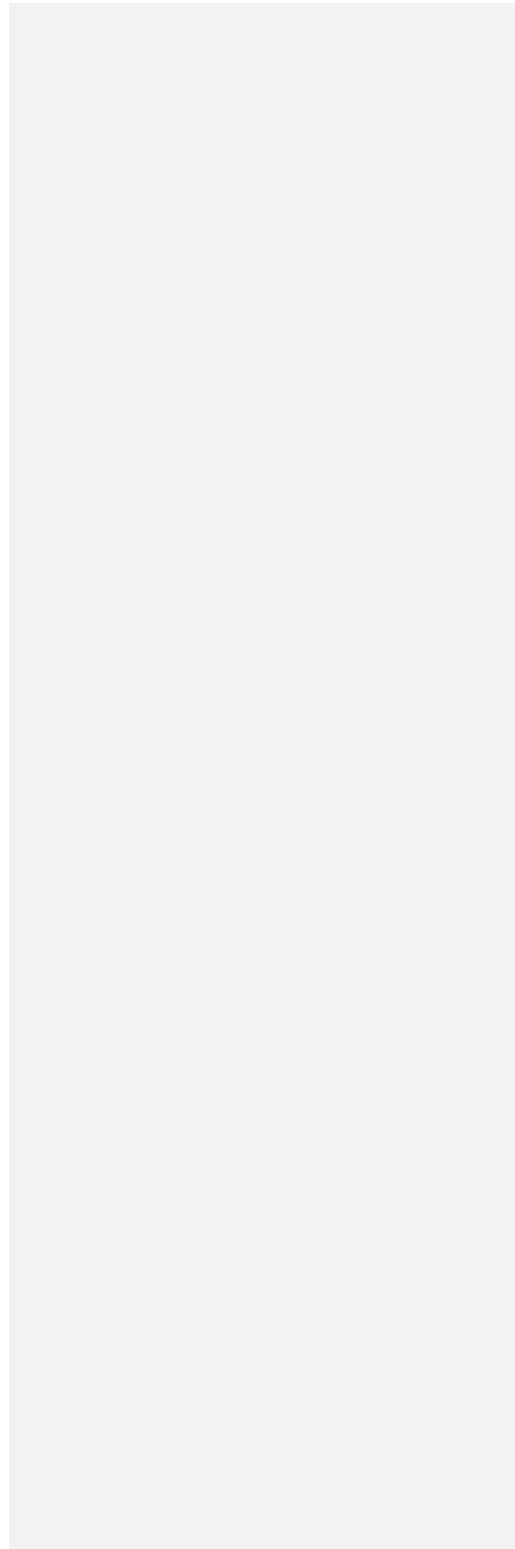
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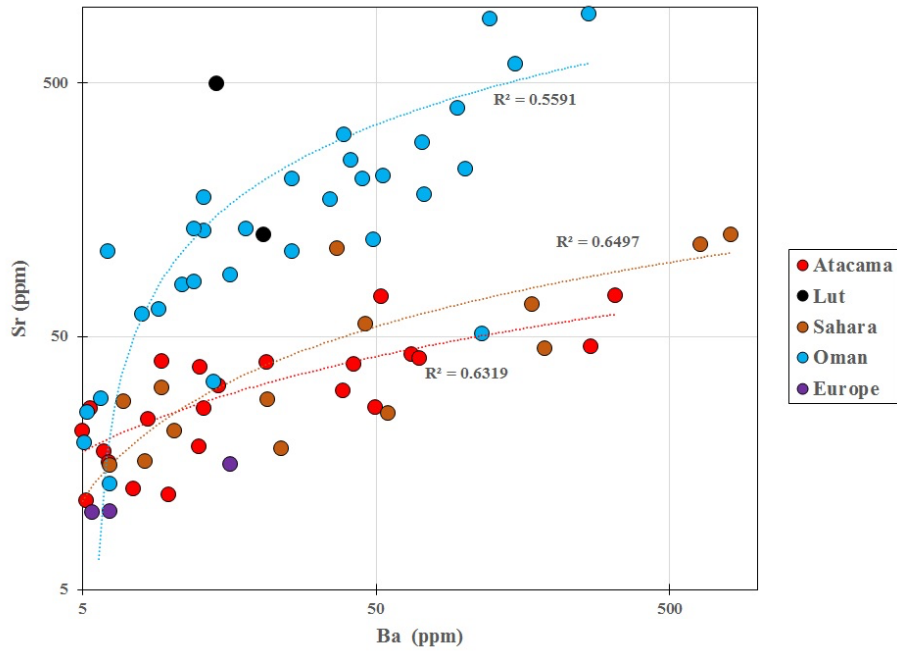
795 Fig. 4. (Pourkhorsandi et al.)



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806 Fig. 5. (Pourkhorsandi et al.)



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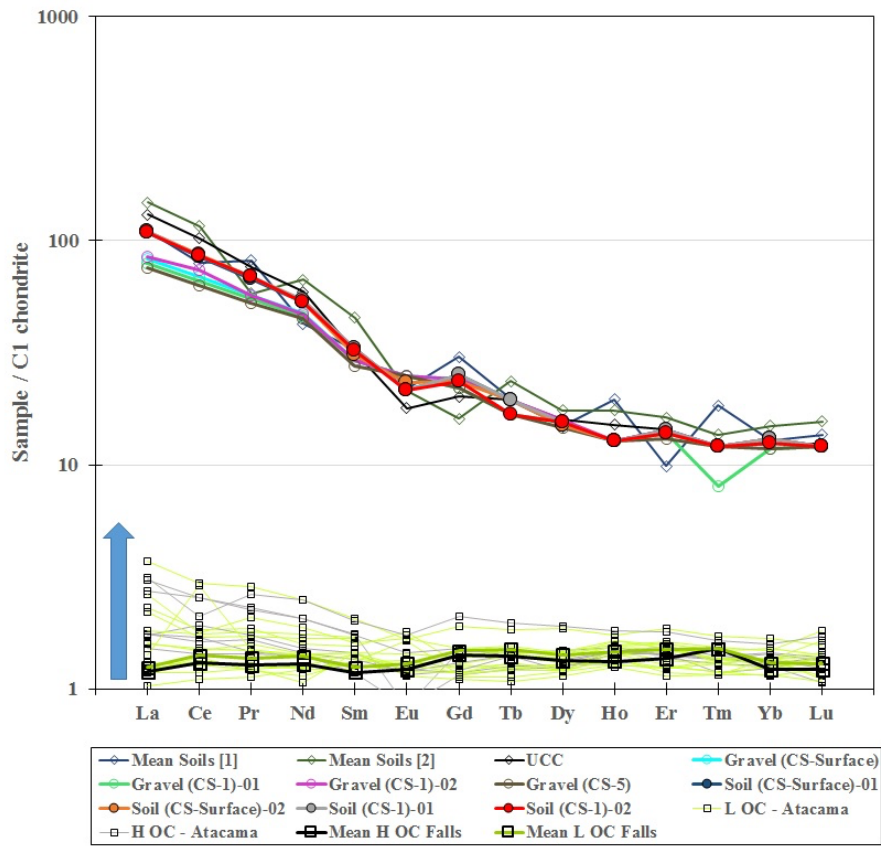
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815 Fig. 6. (Pourkhorsandi et al.)



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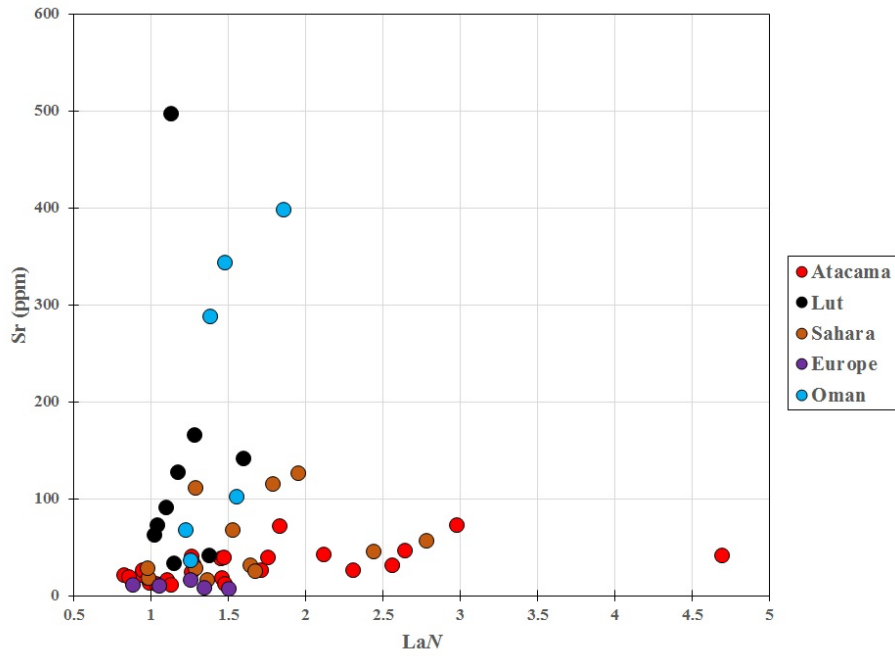
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821 Fig. 7. (Pourkhorsandi et al.)



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