Carba-D,L-allal- and -D,L-galactal-derived vinyl *N*-nosyl aziridines as useful tools for the synthesis of 4-deoxy-4-(*N*-nosylamino)-2,3-unsaturated-5*a*-carbasugars

Valeria Di Bussolo,^{***} Ileana Frau,¹ Lucilla Favero,¹ Vittorio Bordoni,¹

Stefano Crotti,^{b2}Gloria Uccello Barretta,^a Federica Balzano,^aPaolo Crotti^{b,*}

Dipartimento di Chimica e Chimica Industriale, Università di Pisa,

Via G. Moruzzi 3, 56124 Pisa, Italy

Dipartimento di Farmacia, Università di Pisa, Via Bonanno 33, 56126 Pisa, Italy

Abstract: The novel carba-D,L-allal- and carba-D,L-galactal-derived vinyl *N*-nosyl aziridines were prepared and the regio- and stereoselective behavior in opening reactions with *O*- and *N*-nucleophiles examined. The carbaglycosylating ability of the novel aziridines, as deduced by the amount of *1,4-addition products* (1,4-regioselectivity) obtained in the acid-catalyzed methanolysis taken as a model reaction, is similar or superior to that observed with the corresponding carba-D,L-allal- and -D,L-galactal-derived vinyl epoxides, respectively. In all *1,2-* and *1,4-addition products* obtained, a *N*-nosylamino group is regio- and stereoselectively introduced at the C(4) carbon of a 1,2- or 2,3-unsaturated carbasugar, susceptible to further elaborations toward aminocyclitol derivatives. The stereoselective synthesis of the corresponding, enantiomerically pure carba-D,L-allal- and -D,L-galactal-derived vinyl *N*-acetyl aziridines is also described.

Keywords: Vinyl aziridines, 4-(N-nosylamino)-carbasugars, Carbaglycosylation, Regioselectivity.

1.Introduction

Carbasugars, a particular family of cyclitols, are compounds structurally related to carbohydrates with the only difference being the replacement of the endocyclic oxygen with a methylene group.¹ This structural modification is responsible for an increased chemical stability and makes carbasugars valuable mimics of the corresponding carbohydrates.²

^{*} Corresponding author.

^{**} Corresponding author.

E-mail addresses: valeria.dibussolo@unipi.it (V. Di Bussolo),paolo.x.crotti@gmail.com (P. Crotti).

¹ Present address: QuintilesIMS via Roma 108, 20060 Cassina de Pecchi, Milano, Italy.

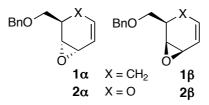
² Present address: Kedrion Biopharma, Gallicano, Lucca, Italy.

Within the carbasugar family, aminocyclitols, also regarded as amino carbasugars, can be found in nature in several families of natural and clinically important antibiotics.³ Moreover, aminocyclitols represent interesting scaffolds in drug discovery. In fact, these structures are present in currently used therapeutic agents,⁴ and turn out to be attractive substrates for targeting many key pathways implicated in disorders such as diabetes, viral infections and cancer.⁵

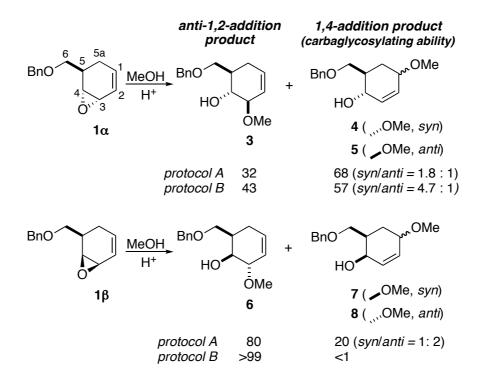
Although the synthesis of amino carbasugars has been investigated and reviewed extensively,^{5,6} new versatile methodologies for their preparation still represent an interesting challenge in synthetic carbohydrate chemistry.

Following our experience with vinyl epoxides⁷ and *N*-activated aziridines,^{7e,8} we herein present a stereoselective synthesis of 2,3-unsaturated-4-(*N*-protected-amino) carbasugars by using carba-D,L-glycal-derived vinyl *N*-nosyl aziridines, as suitable precursors.

Recently, racemic and enantiopure vinyl epoxides 1α and 1β ,⁹ the carba analogs of the previously studied D-allal-and D-galactal-derived vinyl epoxides 2α and 2β ,^{7c-f,h-l} were synthesized and their regio- and stereoselective behavior examined in addition reactions with several *O*-, *C*-, *N*- and *S*-nucleophiles. In particular, by using MeOH as a model *O*-nucleophile, the possibility of considering these epoxides suitable for the stereoselective construction of carba-*O*-glycosides and/or mixed carba-oligo-saccharides was checked. For such an application, a complete or high 1,4-regioselective behavior with associated high stereoselectivity, in the opening reactions, is required. This behavior, defined as carbaglycosylating ability, is determined by the total amount of *1,4-addition products* obtained.⁹

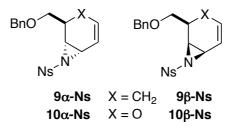


The results obtained after acid methanolysis indicated an interesting carbaglycosylating ability for epoxide 1α , with moderate syn-stereoselectivity, as shown by 68% (1.8:1 *syn/anti* ratio) and 57% 1,4-regioselectivity (4.7:1 *syn/anti* ratio) under *protocol* A (0.2 N H₂SO₄/MeOH) and synthetically more interesting *protocol* B reaction conditions (MeOH, 6 equiv, in 10^{-2} N TsOH/CH₂Cl₂), respectively.^{9a,10} In this framework, diastereoisomeric epoxide 1β turned out to be a very poor carbaglycosyl donor, as shown by the low 1,4-regioselectivity (20%) observed under the corresponding *protocol* A, and by the complete anti-1,2-regioselectivity (>99%) found under the corresponding *protocol* B reaction conditions. These results and the structures of all the addition products obtained in the acid methanolysis of epoxides 1α and 1β (methoxy alcohols 3-8), are shown in Scheme 1.^{9a,10}



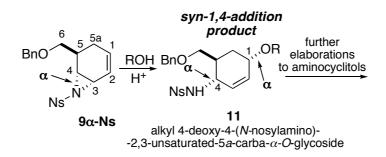
Scheme 1. Regio- and stereoselectivity of vinyl carba epoxides 1α and 1β in the acid methanolysis under *protocol A* and *protocol B* reaction conditions.

As an extension of our interest toward glycal-derived systems and their corresponding carba analogues, and looking for new, effective carbaglycosyl donors, our attention was directed to the diastereoisomeric carba-D,L-allal- and -D,L-galactal-derived vinyl *N*-nosyl aziridines 9α -Ns and 9β -Ns, the carba analogs of the corresponding, previously studied D-allal- and D-galactal-derived *N*-nosyl aziridines 10α -Ns and 10β -Ns.^{7e,8c}



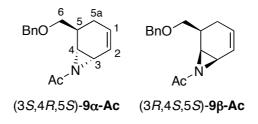
The examination of the regio- and stereoselective behavior in nucleophilic opening reactions could indicate whether aziridines 9α -Ns and 9β -Ns are effective carbaglycosyl donors with

simultaneous, completely regio- and stereoselective introduction of a –(*N*-nosylamino) group at C(4)-carbon of a 2,3-unsaturated carbasugar system as shown in **11**, with high levels of syn-1,4-stereoselectivity, if possible, when alcohols (ROH) are used as carbaglycosyl acceptors (Scheme 2). The presence of the double bond and the easy deprotection of the –(*N*-nosylamino) group,^{8c} could make carbapyranosides **11** suitable for further elaborations toward biologically interesting aminocyclitols,³⁻⁶ through a new synthetic approach (Scheme 2, where only *N*-nosyl aziridine **9** α -**Ns** is shown, for simplicity).



Scheme 2. The conceivable behavior of aziridine 9α -Ns as a carbaglycosyl donor in the acid alcoholysis, with complete 1,4-syn-stereoselectivity and simultaneous complete regio- and stereoselective introduction of a –(*N*-nosylamino) group at C(4) carbon of a 2,3-unsaturated carbasugar.

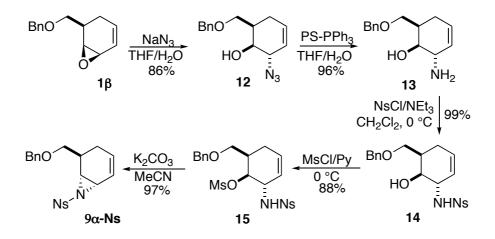
In this preliminary examination of their regio- and stereoselective behavior in nucleophilic addition reactions, *N*-nosyl aziridines 9α -Ns and 9β -Ns were prepared in a racemic form. However, in view of a possible use of these systems for the construction of compounds of biological interest, an enantioselective route to both aziridine systems, in the form of *N*-acetyl analogs, as (3S,4R,5S)- 9α -Ac and (3R,4S,5S)- 9β -Ac, has been disclosed, also.



2. Results and discussion

The stereoselective synthesis of vinyl aziridines 9α -Ns and 9β -Ns was carried out by means of the same protocol starting from azidolysis of the corresponding epoxide with the opposite configuration 1β and 1α , respectively.

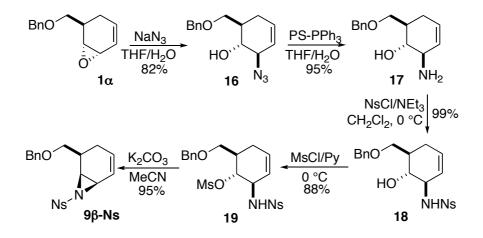
For the synthesis of aziridine 9α -Ns, the azidolysis of epoxide 1β by NaN₃ in a 1:1 THF/H₂O mixture¹¹ turned out to be completely 1,2-regio- and anti-stereoselective affording *trans* 3,4-azido alcohol **12** as the only reaction product (Scheme 3).¹²



Scheme 3. Synthesis of vinyl *N*-nosyl aziridine 9α -Ns.

trans 3,4-Azido alcohol **12** was reduced by polimer-supported PPh₃ (PS-PPh₃, Aldrich)^{8b} in a heterogeneous phase (20:1 THF/H₂O) to the corresponding *trans* 3,4-amino alcohol **13**, which was obtained in good yield (96%) and was sufficiently pure to be directly used in the next step. Regioselective *N*-nosylation of the amino group of **13** by a NsCl/NEt₃/CH₂Cl₂ protocol afforded *N*-nosyl derivative **14**. Subsequent *O*-mesylation by MsCl/Py of the free -OH functionality gave *trans N*-nosyl-*O*-mesyl derivative **15**. The treatment of mesylate **15** under basic conditions (K₂CO₃/MeCN) gave the cyclization to the desired aziridine **9α-Ns** which was obtained with a good overall yield (69%), through a 5 step sequence, starting from epoxide **16** (Scheme 3).

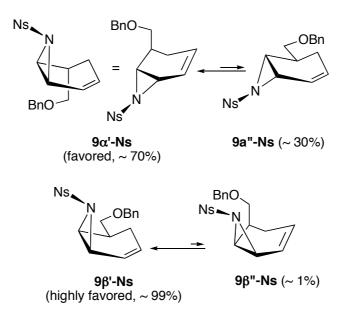
The treatment of epoxide 1α with NaN₃ in 1:1 THF/H₂O¹¹ mixture afforded *trans* 3,4-azido alcohol **16** as the only reaction product.¹² Reduction of azido alcohol **16** to *trans* 3,4-amino alcohol **17** by PS-PPh₃ in THF/H₂O,^{8b} followed by regioselective *N*-nosylation to *N*-nosyl derivative **18** and subsequent *O*-mesylation by MsCl/Py protocol led to *trans N*-nosyl-*O*-mesyl derivative **19**. Base-catalyzed (K₂CO₃/MeCN) cyclization of **20** afforded pure aziridine **9**β-Ns (64% overall yield through 5 steps starting from epoxide **1** α , Scheme 4).



Scheme 4. Synthesis of vinyl *N*-nosyl aziridine 9β-Ns.

Aziridines 9α -Ns and 9β -Ns, even if prudentially stored at -15°C, turned out to be sufficiently stable also at room temperature.

A theoretical conformational study (performed with DF, density functional theory calculation, in vacuum) carried out on simplified models (see Supplemetary data, sections 4.1, 4.2 and 4.3), has indicated that *N*-nosyl aziridine 9α -Ns largely exists as the favored conformer 9α '-Ns (nearly a 70:30 equilibrium between the two conformers 9α '-Ns and 9α "-Ns) whereas aziridine 9β -Ns practically exists as the only, highly favored, conformer 9β '-Ns (about 99 %, Scheme 5). It is worth noting that the favored conformer 9α '-Ns in aziridine 9α -Ns and the highly favored conformer 9β '-Ns in aziridine 9β -Ns have the same ring conformation with the side chain (– CH₂OBn) axial in 9α '-Ns and equatorial in 9β '-Ns (Scheme 5).



Scheme 5. Theoretical conformational analysis of *N*-nosyl aziridines 9α -Ns and 9β -Ns.

2.1. Regio- and stereoselective behavior of N-nosyl aziridines 9α -Ns and 9β -Ns in nucleophilic addition reactions

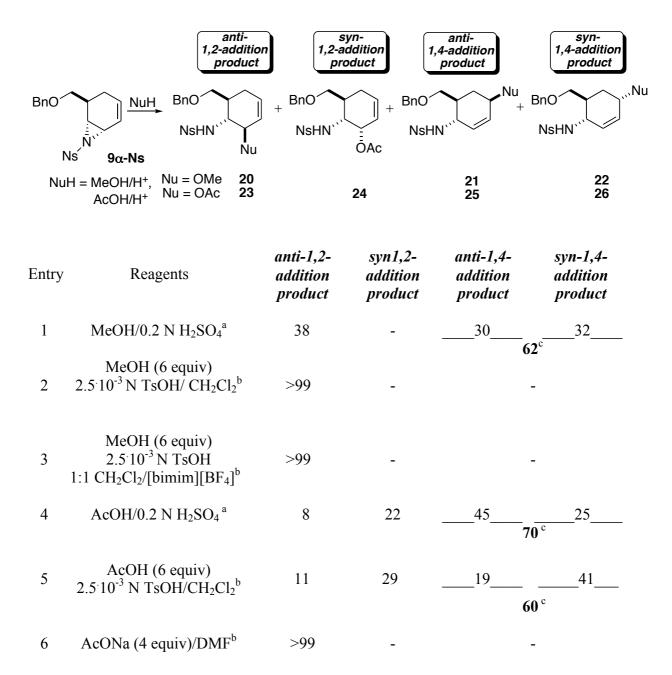
Methanolysis and acetolysis reactions were carried out under acidic conditions (*protocol A* and *B* reaction conditions)¹⁰ and aminolysis reactions were carried out in the presence of a Lewis acid as $Sc(OTf)_3$ (*protocol A*).¹⁰ Conversely, alkaline acetolysis (AcONa/DMF) and azidolysis reactions (NaN₃/THF-H₂O) (*protocol B*)¹⁰ did not necessitate any catalyst. Alkaline methanolysis (MeONa/MeOH) afforded only complex reaction mixtures. The results will be presented in the following sections.

2.1.1. Regio- and stereoselective behavior of N-nosyl aziridines 9α -Ns and 9β -Ns in nucleophilic addition reactions of MeOH (O-nucleophile) under acid conditions

The regio- and stereochemical behavior of aziridine 9α -Ns in the acid methanolysis turned out to be strictly dependent on the reaction conditions and, in particular, on the aziridine:MeOH ratio.

In fact, when the methanolysis was carried out using 0.2 N H₂SO₄ solution in MeOH, that is in the presence of a large amount of the nucleophile (*protocol A* reaction conditions, see Table 1),¹⁰ the reaction did not show regio- and/or stereoselectivity affording a 38:30:32 mixture of *trans*-3methoxy-4-(*N*-nosylamino)-derivative **20** (*anti-1,2-addition product*), *trans*-1-methoxy-4-(*N*nosylamino)-derivative **21** (*anti-1,4-addtion product*) and *cis*-1-methoxy-4-(*N*-nosylamino)derivative **22** (*syn-1,4-addition product*) (¹H NMR spectroscopy) (entry 1, Table 1). When the methanolysis of aziridine **9** α -**Ns** was performed using a 2.5⁻10⁻³ N TsOH solution in CH₂Cl₂ containing a drastically reduced amount of MeOH in such a way that aziridine:TsOH:MeOH ratio = 1:0.1:6 (*protocol B* reaction conditions),¹⁰ a completely 1,2-regio- and *anti*-stereoselective behavior was observed, with the exclusive production of *anti-1,2-addition product* **20** (entry 2, Table 1).

Table 1. Regio- and stereoselectivity of acid methanolysis and acetolysis of aziridine 9α -Ns, under *protocol A* and *B* reaction conditions.^{a,b}



^a *Protocol A* reaction conditions: the nucleophile (MeOH or AcOH) is the reaction solvent in the presence of the acid (H_2SO_4).

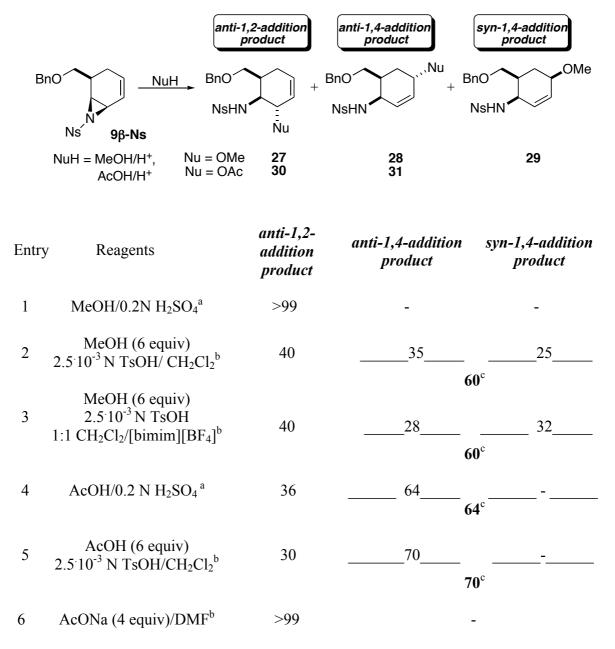
^b *Protocol B* reaction conditions: aziridine:nucleophile (MeOH or AcOH) ratio = 1:6 in CH₂Cl₂ or 1:1 CH₂Cl₂/[bimim][BF₄], as the reaction solvent, in the presence of the acid (TsOH, 0.1 eq); aziridine:AcONa ratio = 1: 4 in the alkaline acetolysis in DMF.

^c Total amount of *1,4-addition products* (carbaglycosylating ability).

Aziridine **9** β -Ns showed a similar regio- and stereochemical behavior but, surprisingly, obtained under diametrically opposed reaction conditions. Actually, under *protocol B* reaction conditions,¹⁰ a 40:35:25 mixture of *trans*-3-methoxy-4-(*N*-nosylamino)-derivative **27** (*anti*-1,2-*addition product*), *trans*-1-methoxy-4-(*N*-nosylamino)-derivative **28** (*anti*-1,4-*addtion product*) and

cis-1-methoxy-4-(*N*-nosylamino)-derivative **29** (*syn*-1,4-addition product) (¹H NMR spectroscopy) was obtained indicating a non-regioselective behavior (entry 2, Table 2). On the contrary, when the methanolysis reaction of aziridine **9** β -Ns was repeated under *protocol A* reaction conditions,¹⁰ a complete 1,2-regio- and *anti*-stereoselective addition was observed with the isolation of *anti*-1,2-*addition product* **27**, as the only reaction product (entry 1, Table 2).

Table 2. Regio- and stereoselectivity of acid methanolysis and acetolysis of aziridine 9β -Ns under *protocol A* and *B* reaction conditions.^{a,b}



^a *Protocol A* reaction conditions: the nucleophile (MeOH or AcOH) is the reaction solvent in the presence of the acid (H_2SO_4).

^b *Protocol B* reaction conditions: aziridine:nucleophile (MeOH or AcOH) ratio = 1:6 in CH₂Cl₂ or 1:1 CH₂Cl₂/[bimim][BF₄], as the reaction solvent, in the presence of the acid (TsOH, 0.1 eq); aziridine:AcONa ratio = 1: 4 in the alkaline acetolysis in DMF.

^c Total amount of *1,4-addition products* (carbaglycosylating ability).

The reason for the different regiochemical behavior of aziridines 9α -Ns and 9β -Ns under the same reaction conditions (see entries 1 and 2, Tables 1 and 2) can be rationalized by considering the influence that the different acid reaction conditions (*protocol A* and *B*) could have on the conformational equilibrium in the two aziridines.

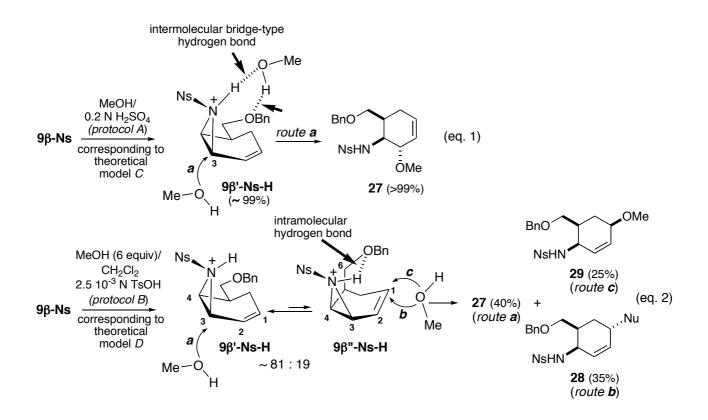
A DFT theoretical conformational study on protonated aziridine 9β -Ns-H, the species likely present under acid conditions, carried out with and without the solvent, simulated in an explicit way by one molecule of MeOH (theoretical model *C* and *D*, respectively), has indicated, first of all, that the protonation occurs inside the ring with the bulky –Ns group outside (Schemes 6 and 7, and Supplementary data, sections 4.3 and 4.4).

In the presence of one molecule of MeOH (theoretical model *C*), the most favored minimum energy structure shows the presence of two bridge-type intermolecular hydrogen bonds in which MeOH is contemporarily hydrogen-bonded to the protonated aziridine nitrogen and the oxygen of the 6-OBn functionality of the cis-disposed equatorial side chain. The occurrence of these hydrogen bonds determines the exclusive presence of the highly favored conformer **9** β **'-Ns-H** with the side – CH₂OBn chain equatorial (about 99%, Scheme 6 and Supplementary data, section 4.4).¹³

On the contrary, in the absence of MeOH (theoretical model *D*), the occurrence of an intramolecular hydrogen bond between the protonated aziridine and the oxygen of the 6-OBn group of the cis-disposed axial side chain, possible only through conformer 9β "-Ns-H, makes this conformer consistently present at the equilibrium (9β '-Ns-H : 9β "-Ns-H conformers ratio nearly 81: 19, Scheme 6 and Supplementary data, section 4.3).

By admitting that theoretical models *C* (presence of one molecule of MeOH) and *D* (absence of MeOH) correspond to experimental *protocol A* (presence of a large amount of MeOH, solvent of the reaction)¹⁰ and *protocol B* reaction conditions (presence of only a small amount of MeOH, 6 equiv, in CH₂Cl₂, the reaction solvent),¹⁰ respectively, nucleophilic attack by MeOH under *protocol A* (corresponding to model *C*) necessarily occurs at aziridine C(3)-carbon of the only present conformer **9β**'-**Ns-H** through a sterically unhindered, highly favored trans diaxial opening of the aziridine ring (*route a*, eq. 1, Scheme 6).¹⁴ In this way, the corresponding *anti-1,2-addition product* **27**, is reasonably the only reaction product, as experimentally observed (>99%, entry 1, Table 2).

On the other hand, under *protocol B* (corresponding to theoretical model *D*) nucleophilic attack can reasonably occur on both conformers 9β '-Ns-H and 9β "-Ns-H, consistently present at the equilibrium (eq. 2, Scheme 6). In this framework, even if nucleophilic attack at the aziridine C(3)-carbon of conformer 9β '-Ns-H remains highly favored (a trans-diaxial opening process, *route a*),¹⁴ nucleophilic attack on conformer 9β "-Ns-H, necessarily occurring at the more reactive vinyl C(1)-carbon in a conjugated fashion through *routes b* and *c*, becomes competitive and a 40:35:25 mixture of *anti-1,2-* 27 (*route a*), *anti-1,4-* 28 (*route b*), and *syn-1,4-addition product* 29 (*route c*) is reasonably obtained, as found experimentally (eq. 2, Scheme 6 and entry 2, Table 2).¹⁵ Nucleophilic attack at C(3)-aziridine carbon of protonated conformer 3β "-Ns-H does not occur because it corresponds to a stereoelectronically disfavored trans-diequatorial ring opening process.^{14,16,17}



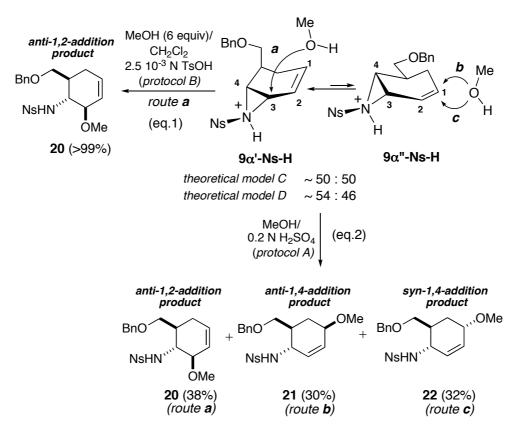
Scheme 6. Theoretical conformational analysis of protonated aziridine 9β -Ns-H in the presence (theoretical model *C*) and absence of one molecule of MeOH (theoretical model *D*), and regio- and stereoselectivity of aziridine 9β -Ns in acid methanolysis under *protocol A* and *B* reaction conditions.

In the case of protonated aziridine 9α -Ns-H, due to the trans relationship between protonated aziridine nitrogen and –CH₂OBn side chain, no intramolecular hydrogen bond of the type observed

with aziridine 9β -Ns-H under acid conditions can occur. As a consequence, almost the same theoretical protonated conformer 9α '-Ns-H : 9α "-Ns-H ratio has been found both in the presence (nearly 50:50, theoretical model *C*) and in the absence of one molecule of MeOH (nearly 54:46, theoretical model *D*) (Scheme 7 and Supplementary data, sections 4.5 and 4.6).¹⁸

On the basis of the conformational analysis, the stereo- and regioselective result observed in the acid methanolysis of aziridine 9α -Ns can be explained by the amount of nucleophile present in the reaction mixture and the different reactivity of the two conformers 9α '-Ns-H and 9α "-Ns-H.

In the presence of only a small amount of MeOH (*protocol B*), nucleophilic attack occurs only at the aziridine C(3)-carbon of the protonated conformer 3α '-Ns-H, due to the possibility of having a highly favored trans-diaxial ring opening process (*route a*, eq. 1, Scheme 7 and entry 2, Table 1).¹⁴ As a consequence, *anti-1,2-addition product* 20 is the only reaction product, as observed. On the contrary, when the reaction is carried out in the presence of a large amount of nucleophile (MeOH as solvent, *protocol A*), nucleophilic attack in a conjugated fashion at the vinyl C(1)-carbon of the protonated aziridine 9α "-Ns-H, equally from the α and β direction, becomes competitive to nucleophilic attack at the C(3)-carbon of conformer 9α '-Ns-H. As a consequence a mixture of all the possible addition product 22 (32%, *route c*) (eq. 2, Scheme 7 and entry 1, Table 1).^{14,17}



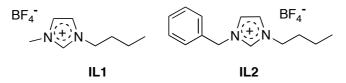
Scheme 7. Theoretical conformational analysis of protonated aziridine 9α -Ns-H in the presence (theoretical model *C*) and absence of one molecule of MeOH (theoretical model *D*) and regio- and stereoselectivity of aziridine 9α -Ns in acid methanolysis under *protocol A* and *B* reaction conditions.

The results obtained with MeOH, as a model *O*-nucleophile, indicate that *N*-nosyl aziridines 9α -Ns and 9β -Ns are not completely 1,4-regioselective, under the experimented reaction conditions, in presence of the nucleophile as a solvent or as a reagent (*protocol A* and *B*). However, the overall amount of *1,4-addition products* (*anti-* and *syn-*adducts) is sufficiently high (62 and 60% from 9α -Ns and 9β -Ns, entries 1 and 2, Tables 1 and 2, respectively) to make aziridines 9α -Ns and 9β -Ns be considered useful in carbaglycosylating processes. Moreover, a comparison with the corresponding results obtained with epoxides 1α and 1β indicates that the carbaglycosylating ability of aziridines 9α -Ns (62%, *protocol A*) and 9β -Ns (60%, *protocol B*) is similar and superior to the maximum value obtained with the corresponding epoxides 1α (68%, *protocol A* and 57%, *protocol B*) and 1β (20%, *protocol A*), respectively.^{9a,19}

2.2. Acid methanolysis of epoxides 1α and 1β and N-nosyl aziridines 9α -Ns and 9β -Ns in the presence of an ionic liquid (IL)

On the basis of the above described results, some different conditions able to influence positively the regioselectivity of the methanolysis reactions toward *1,4-addition products*, and thus the carbaglycosylating ability, not only with aziridines 9α -Ns and 9β -Ns, but also with the related, previously studied, epoxides 1α and 1β ,⁹ were tried. Considering that an increase in the 1,4-addition process could reasonably be obtained by an opening process characterized by a more diffused carbocation-like intermediate, the use of an ionic liquid (IL) was envisaged.²⁰

Two different ILs, butylmethylimidazolium tetrafluoroborate ([bmim][BF_4]) (**IL1**) and benzylbutylimidazolium tetrafluoroborate ([bzbim][BF_4]) (**IL2**), soluble in organic solvents, were checked.



IL1 and **IL2** were used as solvent or cosolvent in different acidic reaction conditions $(10^{-2} H_2SO_4 \text{ or } 2.5 \cdot 10^{-3} \text{ N TsOH})$. The nucleophile (MeOH) was present in a large amount when cosolvent of the reaction with IL (1:1 MeOH/IL ratio, *protocol A*), or only in a controlled amount (6 equiv) with respect to the epoxide, when IL or 1:1 IL/CH₂Cl₂ mixture was the solvent of the reaction (*protocol B*). The result obtained with epoxides 1α and 1β are reported in Tables 3 and 4, respectively, where the results previously obtained under standard conditions are also reported (entries 1 and 2, Tables 3 and 4).^{9a}

Table 3. Regio- and stereoselectivity, with indication of the carbaglycosylating ability, of epoxide 1α in acid methanolysis reactions (*protocol A* and *B*) under standard conditions and in the presence of an IL.

		-addition oduct	a-1,4-addition product	anti-1,4-addition product
	BnO <u>MeOH</u> BnO O' H ⁺ , IL HO''	+ BnO H	O''' Bn	OMe HO'''
	1α	3	4	5
Entry	Reaction conditions	anti-1,2- addition product	anti-1,4-addition product	syn-1,4-addition product
1	$0.2 \text{ N H}_2 \text{SO}_4 / \text{MeOH} (A)^{a,b}$	32	44	24
2	MeOH (6 equiv) in 10^{-2} N TsOH/CH ₂ Cl ₂ (<i>B</i>) ^{a,b}	43	47	8 ^c 10
3	$10^{-2} \text{ N H}_2 \text{SO}_4 \text{ in}$ 1:1 MeOH/[bmim][BF ₄] (A) ^b	31	33	36
4	MeOH (6 equiv) in 2.5 [.] 10 ⁻³ N TsOH / 1:1 CH ₂ Cl ₂ /[bmim][BF ₄] (<i>B</i>) ^b	20	76	4
			80 ^c	
5	MeOH (6 equiv) in 2.5 \cdot 10 ⁻³ TsOH/[bmim][BF ₄] (<i>B</i>) ^b	25	45	30
			75	5 ^c

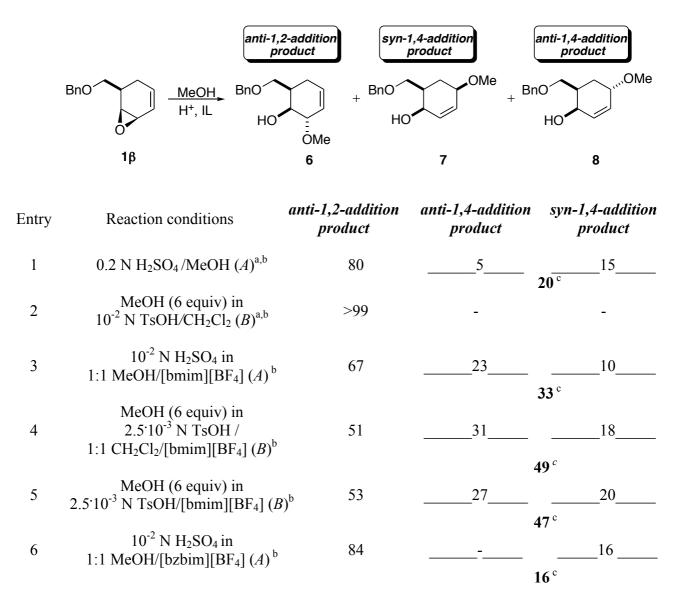
$$6 \frac{10^{-2} \text{ N H}_2 \text{SO}_4 \text{ in}}{1:1 \text{ MeOH}/[\text{bzbim}][\text{BF}_4] (A)^{\text{b}}} 50 \underline{30}_{50} \underline{20}_{50}$$

^a See ref. 9a.

^b(*A*): *protocol A*; (*B*): *protocol B* (see Tables 1 and 2 and ref. 10).

^c Total amount of *1,4-addition products* (carbaglycosylating ability).

Table 4. Regio- and stereoselectivity, with indication of the carbaglycosylating ability, of epoxide 1β in acid methanolysis reactions (*protocol A* and *B*) under standard conditions and in the presence of an IL.



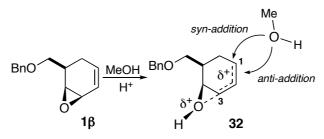
^a See ref. 9a.

^b(*A*): *protocol A*; (*B*): *protocol B* (see Tables 1 and 2 and ref. 10).

^c Total amount of *1,4-addition products* (carbaglycosylating ability).

An examination of the regio- and stereoselectivity obtained in the methanolysis of epoxides 1α and 1β , limited to the more interesting *protocol B* reaction conditions (*B*, Tables 3 and 4), indicates how these new reaction conditions are able to increase the 1,4-regioselectivity in both the epoxides, particularly in epoxide 1β . Actually, a 1.4 (from 57 to 80%, entries 2 and 4, Table 3) and 2.4 times increase (from 20 to 49% entries 1 and 4, Table 4) is observed in epoxide 1α and 1β , respectively.

In accordance with expectations, the presence of IL favors, by delocalization, an increase in partial positive charge at the C(1)-carbon further from the inductive electron- withdrawing effect of the protonated oxirane oxygen (structure **32**, Scheme 8 where only epoxide 1β is shown for simplicity). Subsequent conjugate attack of the nucleophile (MeOH) on that carbon of epoxides 1α and 1β can more easily occur with the corresponding increased formation of *syn*- and *anti-1,4-addition products* (entries 4, Tables 3 and 4).¹⁵



Scheme 8. Effective delocalized intermediate species in the acid methanolysis of epoxide 1β in the presence of IL.

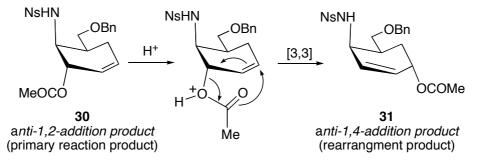
In both epoxides, the best result is obtained when the nucleophile is present only in a limited amount (6 equiv, *protocol B* reaction conditions) in a 1:1 CH₂Cl₂/IL mixture as the solvent in the presence of 2.5^{10⁻³} N TsOH as the acid (entries 4, Tables 3 and 4). Remarkably, the best result obtained with epoxide **1** α (80% 1,4-addition, entry 4, *protocol B*, Table 3) is accompanied, for the first time, with a consistent level of *syn*-stereoselectivity (76%, with *syn-/anti-1,4-addition product* ratio = 19:1).¹⁵ The remaining is constituted by the usual *anti-1,2-addition product* which, in these conditions, reaches its lowest result (20%).

These last results confirm, once again, that epoxide 1α is an efficient carbaglycosyl donor in the reaction with *O*-nucleophiles. At the same time, epoxide 1β shows 1,4-addition values (49%, entry 4, *protocol B*, Table 4) which makes also this system interesting for further synthetic applications.

Differently from what observed with epoxides 1α and 1β , the use of ILs in the methanolysis of aziridines 9α -Ns and 9β -Ns was, unfortunately, unsuccessful and the regioselective results obtained under *protocol B* reaction conditions (entries 3, Tables 1 and 2) are completely similar to those previously obtained under the corresponding standard conditions (entries 2, Tables 1 and 2).

2.3. Acid and basic acetolysis of aziridines 9α -Ns and 9β -Ns

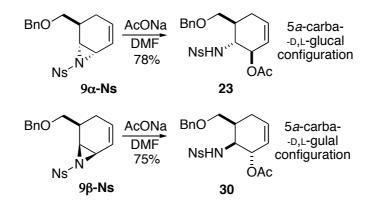
Acid acetolysis reactions, carried out both under *protocol A* and *B* reaction conditions, indicated that the regioselctive behavior of aziridines 9α -Ns and 9β -Ns is almost the same, giving similar mixtures of corresponding *1,2-* and *1,4-addition products* in both cases [about 30/40:70/60 *1,2-/1,4-addition products* ratio from 9α -Ns (entries 4 and 5, Table 1) and 9β -Ns (entries 4 and 5, Table 2)]. In this framework, it is worth noting that, in the case of azridine 9β -Ns, *anti-1,4-adduct* **31** is the only *1,4-addition product* obtained and almost the same mixture of *anti-1,2-* **30** and *anti-1,4-adduct* **31** is obtained both under *protocol A* and *protocol B* reaction conditions. The peculiarity of this result with azridine 9β -Ns, also in a comparison with the results obtained in the corresponding methanolysis reactions, let us think that *anti-1,4-adduct* **31** could not be an addition product, but the result of a partial isomerization of allyl acetate **30** (*anti-1,2-addition product*), the only addition product and primary reaction product, in a type of acid-catalyzed suprafacial, completely stereoselective [3,3]sigmatropic rearrangement, as here shown (Scheme 9).²²



Scheme 9. Hypothesized stereoselecive formation of *anti-1,4-addition product* 31 from corresponding *anti-1,2-addition product* 30 in the acid acetolysis of aziridine 9β -Ns.

Alkaline acetolysis reactions [AcONa (4 equiv)/DMF, *protocol B*] of both aziridines 9α -Ns and 9β -Ns turned out to be completely 1,2-regio- and *anti*-stereoselective with the exclusive isolation of the corresponding *trans*-3-acetoxy-4-(*N*-nosylamino) derivative 23 and 30 from aziridine 9α -Ns and 9β -Ns, respectively (entries 6, Table 1 and 2, and Scheme 10). In this way, a simple, completely regio- and stereoselective procedure is available for the synthesis of 4-deoxy-4-

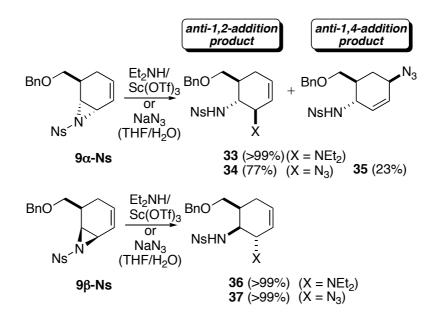
(*N*-nosylamino)-3-*O*-acetyl-derivatives having relative 5*a*-carba-D,L-glucal (**23**) and 5*a*-carba-D,L-gulal configuration (**30**), respectively, and containing easily removable *N*- and *O*-protective groups (Scheme 10).^{8c}



Scheme 10. Alkaline acetolysis of aziridines 9α -Ns and 9β -Ns.

2.4. Regio- and stereoselective behavior of N-nosyl aziridines 9α -Ns and 9β -Ns in nucleophilic addition reactions of N-Nucleophiles

Aziridines 9α -Ns and 9β -Ns react with Et₂NH under *protocol A* reaction conditions (amine as the solvent) only in the presence of a Lewis acid, as Sc(OTf)₃. Under these conditions, both aziridines 9α -Ns and 9β -Ns afforded, in a completely 1,2-regio- and anti-stereoselective fashion, the corresponding *anti-1,2-addition product*, *trans*-3-(*N*,*N*-diethylamino)-4-(*N*-nosylamino) derivatives 33 and 36, respectively (Scheme 11).



Scheme 11. Regio- and stereoselectivity of aminolysis and azidolysis of *N*-nosyl aziridines 9α -Ns and 9β -Ns.

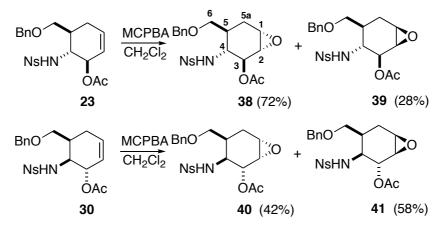
As for the azidolysis reactions, carried out by means of the previously described protocol (NaN₃ in THF/H₂O),¹¹ aziridine 9α -Ns turned out to be mainly 1,2-regio- and anti-stereoselective, affording a 77:23 mixture of *anti-1,2-addition product*, *trans*-3-azido-4-(*N*-nosylamino) derivative **34** and *anti-1,4-addition product*, *trans*-1-azido-4-(*N*-nosylamino) derivative **35** (¹H NMR), whereas aziridine 9β -Ns showed a completely anti-1,2-regioselective behavior with the isolation of *trans*-3-azido-4-(*N*-nosylamino) derivative **37**, as the only reaction product (Scheme 11).

The results obtained indicate that the behaviour of aziridines 9α -Ns and 9β -Ns with an amine and azide ion is dominated by the nucleophilicity of the reagents which directs, completely in 9β -Ns (>99%) and mostly in 9α -Ns (77-99%), the ring opening toward an anti-1,2-addition process.

2.5. Functionalization of the double bond of model anti-1,2-addition products from N-nosyl aziridines 9α -Ns and 9β -Ns

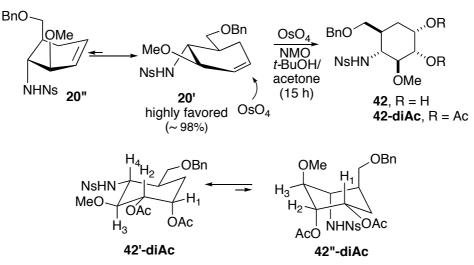
The reactivity in electrophilic reactions of the double bond of model *anti-1,2-addition products*, as *trans*-3-methoxy-4-(*N*-nosylamino) derivatives **20** and **27** and *trans*-3-acetoxy-4-(*N*-nosylamino) derivatives **23** and **30** obtained from aziridines 9α -Ns and 9β -Ns, respectively (Tables 1 and 2 and Scheme 10), was checked. These compounds were subjected to two typical electrophilic addition reactions such as epoxidation and catalytic dihydroxylation.

Epoxidation of *trans*-3-acetoxy-4-(*N*-nosylamino)-1,2-unsaturated derivatives **23** and **30** by the MCPBA/CH₂Cl₂ protocol is not stereoselective and mixtures of the corresponding diastereoisomeric epoxides **38** and **39** (72: 28 ratio), from **23**, and **40** and **41** (42:58 ratio), from **30**, were obtained, then separated by preparative TLC (Scheme 12) (see Supplementary data, sections 3.3 and 3.4, for NMR identifications of epoxides **38-41** and considerations on the absence of diastereoselectivity found in their formation).²³



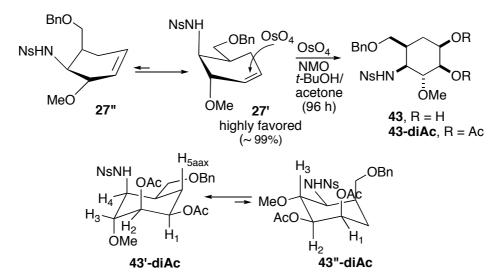
Scheme 12. Epoxidation by MCPBA/CH₂Cl₂ protocol of acetates 23 and 30 (*anti-1,2-addition products*).

On the contrary, catalytic dihydroxylation by OsO_4/NMO protocol of *trans*-3-methoxy-4-(*N*-nosylamino)-1,2-unsaturated derivatives **20** and **27**²⁴ carried out at room temperature in a *t*-BuOH/acetone mixture turned out to be completely stereoselective in both cases and *cis* diol **42**, a 4-deoxy-4-(*N*-nosylamino)-5*a*-carba- α -D,L-glucopyranose derivative, and **43**, a 4-deoxy-4-(*N*-nosylamino)-5*a*-carba- β -D,L-idopyranose derivative, were obtained from **20** and **27**, respectively, as the only reaction products (Schemes 13 and 14).



Scheme 13. Catalytic dihydroxylation of *anti-1,2-addition product* 20 by OsO₄/ NMO protocol.

In order to make the determination (¹H NMR) of the structure and configuration of the addition products easier, *cis* diols **42** and **43** were transformed into the corresponding diacetates **42**-**diAc** and **43-diAc** (Schemes 13 and 14).



Scheme 14. Catalytic dihydroxylation of *anti-1,2-addition product* 27 by OsO₄/ NMO protocol.

An accurate examination of the ¹H NMR spectra of these compounds, together with considerations based on the mechanism of the dihydroxylation reaction (complete *syn*-stereoselectivity) and an appropriate conformational analysis, confirmed the structure and configuration for **42-diAc** and **43-diAc**, and thus for the corresponding *cis* diols **42** and **43**, shown in Schemes 13 and 14.

In particular, the presence in the ¹H NMR spectrum of **42-diAc** of a trans diaxial relationship between the H(3) proton and the vicinal H(2) and H(4) protons ($J_{2,3} = J_{3,4} = 9.7$ Hz) unequivocally indicated the exclusive presence of the corresponding triequatorial conformer **42'-diAc**, the relative configuration at the C(1) and C(2) carbons and, as a consequence, the given structure (Scheme 13).

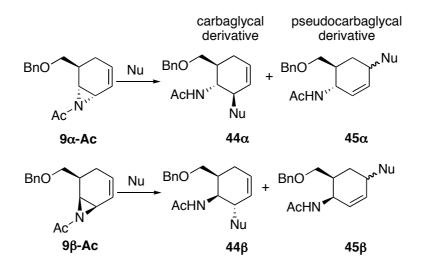
In the case of **43-diAc**, the presence in the corresponding ¹H NMR spectrum of a trans diaxial relationship between the H(1) proton and the axial H(5aax) proton ($J_{1,5aax} = 11.8$ Hz) and of an axial-equatorial relationship with the H(2) proton ($J_{1,2} = 3.1$ Hz) unequivocally indicated the exclusive presence of the corresponding conformer **43'-diAc** with the –CH₂OBn side chain equatorial, the relative configuration at C(1) and C(2) carbons and, as a consequence, the given structure (Scheme 14).

The structures of **42-diAc** and **43-diAc** indicate that dihydroxylation occurred in both diastereoisomeric methoxy derivatives **20** and **27** with a completely *anti* facial selectivity with respect to the direction of the corresponding allyl substituent (-OMe). As for the remaining two substituents, while the $-CH_2OBn$ substituent seems to have no influence on the reaction, the β -directed -NHNs group, axial in conformer **27**^{*},²⁴ could be responsible for the observed longer reaction time (96 h) found for methoxy derivative **27**, when compared with diastereoisomeric

derivative **20** (15 h), by exerting steric hindrance to the β -face-approaching electrophile (Scheme 14).

2.6. Synthesis of N-acetyl aziridines (3S, 4R, 5S)-9 α -Ac and (3R, 4S, 5S)-9 β -Ac

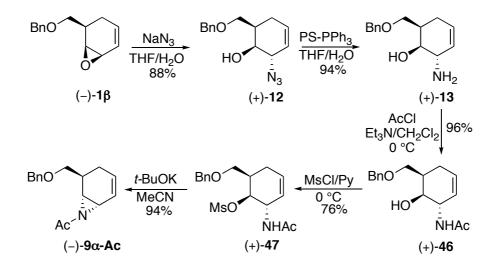
The interesting behavior observed with the racemic *N*-nosyl aziridines 9α -Ns and 9β -Ns in nucleophilic addition reactions, and the increased importance of aminocyclitols in medicinal chemistry³⁻⁶ prompted us to synthesize structurally related enantiomerically pure aziridines. Considering that the –NHAc substituent is biologically interesting, we thought appropriate to direct our efforts toward enantiopure *N*-acetyl aziridines (3S,4R,5S)- 9α -Ac and (3R,4S,5S)- 9β -Ac (Scheme 15). Actually, in a consistent difference with aziridines 9α -Ns and 9β -Ns which transfer at the C(4)-carbon a –NHNs group (Scheme 2 and Tables 1 and 2), when aziridines 9α -Ac and 9β -Ac are subjected to the corresponding ring opening reactions, a -NHAc group is regio- and stereoselectively introduced at the C(4) carbon of an unsaturated carbasugar system [carbaglycal-44 α and 44 β (*1*,*2-addition product*) and/or pseudocarbaglycal-derivative 45 α and 45 β (*1*.4-*addition product*)] with a clear added value to the products (Scheme 15). Both –NHAc and – NHNs^{8c} group can be easily deprotected to corresponding free amino group. However, this step, probably unavoidable in –NHNs substituted derivatives in order to have biologically interesting compounds, could not be necessary in the case of corresponding –NHAc substituted products.



Scheme 15. Hypothesized behavior of aziridines 9α -Ac and 9β -Ac in nucleophilic addition reactions.

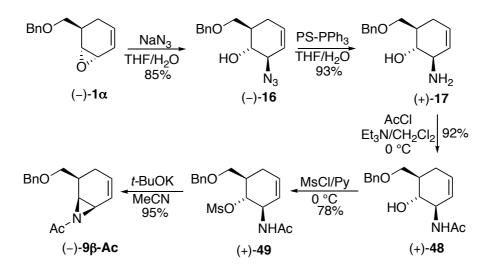
As previously described for the racemic *N*-nosyl aziridines 9α -Ns and 9β -Ns (Schemes 3 and 4), the synthesis of the new, non racemic *N*-acetyl aziridines 9α -Ac and 9β -Ac starts from the corresponding enantiomerically pure vinyl epoxide of opposite configuration (–)-1 β and (–)-1 α , respectively (Schemes 16 and 17).^{9b}

Following the corresponding protocol, azidolysis (NaN₃/THF-H₂O) of epoxide (–)-1 β afforded *trans* 3,4-azido alcohol (+)-12, as the only reaction product, which was then reduced by means of PS-PPh₃ to *trans* 3,4-amino alcohol (+)-13. The treatment of amino alcohol (+)-13 by AcCl /NEt₃ in CH₂Cl₂ at 0°C turned out to be completely regioselective with the exclusive formation of the corresponding *N*-acetyl derivative (+)-46 which was mesylated (MsCl/P) to give the fully protected derivative (+)-47. Base-catalyzed (*t*-BuOK) cyclization of *trans N*-acetyl-*O*-mesyl derivative (+)-47 in anhydrous MeCN afforded the desired, enantiomerically pure *N*-acetyl aziridine (–)-9 α -Ac [57% overall yield, for 5 steps starting from epoxide (–)-1 β] (Scheme 16).



Scheme 16. Synthesis of vinyl *N*-acetyl aziridine $(-)-9\alpha$ -Ac.

The corresponding procedure starting from enatiomerically pure epoxide (–)- $1\alpha^{9b}$ was followed for the synthesis of aziridine (–)- 9β -Ac, as shown in Scheme 17. Azidolysis of epoxide (–)- 1α by the NaN₃/1:1 THF/H₂O protocol afforded, as the only reaction product, *trans* 3,4-azido alcohol (–)-16, which was then reduced (PS-PPh₃ in THF/H₂O) to the corresponding *trans* 3,4-amino alcohol (+)-17. Regioselective *N*-acetylation of *trans* 3,4-amino alcohol (+)-17 by AcCl/Et₃N in CH₂Cl₂ afforded *N*-acetyl derivative (+)-48 which was mesylated (MsCl/Py) to give *trans N*-acetyl-*O*-mesyl derivative (+)-49. The treatment of (+)-49 with *t*-BuOK in anhydrous MeCN afforded enantiomerically pure *N*-acetyl aziridine (–)- 9β -Ac [54% overall yield for 5 steps starting from epoxide (–)- 1α] (Scheme 17).



Scheme 17. Synthesis of vinyl *N*-acetyl aziridine $(-)-9\beta$ -Ac.

As the chiral starting epoxides (–)- 1α and (–)- 1β had been prepared in a stereodivergent fashion starting from commercially available (+)-tri-*O*-acetyl-D-glucal, the common starting material and chiral source,^{9b} the same can be said for enantiomerically pure *N*-acetyl aziridines (–)- 9α -Ac and (–)- 9β -Ac which derive from epoxides (–)- 1β and (–)- 1α , respectively, in a completely stereoselective fashion.

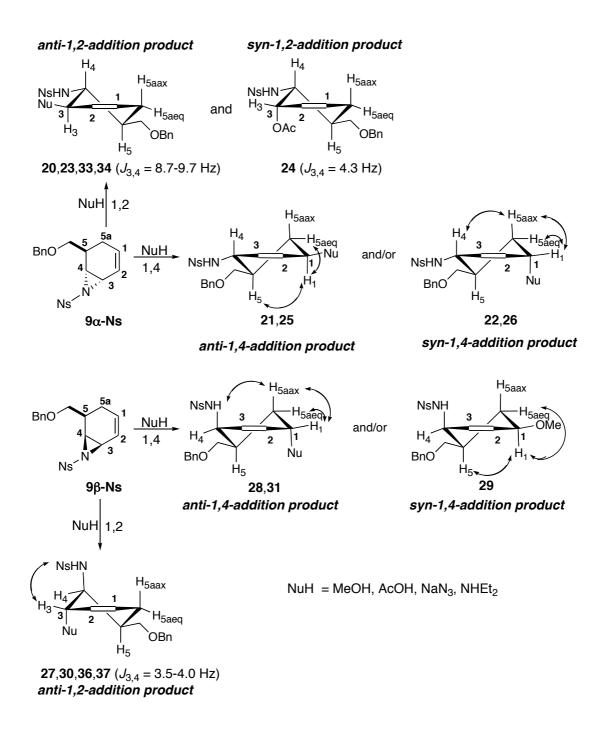
Contrary to expectations based on racemic *N*-nosyl aziridines 9α -Ns and 9β -Ns, the corresponding enantiomerically pure *N*-acetyl aziridines (–)- 9α -Ac and (–)- 9β -Ac turned out to be not particularly stable and had to be stored at –20 °C.

2.7. Structures and configurations

The structure and configuration of regioisomeric 1,2- and 1,4-addition products obtained in the opening reactions of aziridines 9α -Ns and 9β -Ns have been determined by compared analysis of homonuclear scalar (COSY), dipolar (NOESY) and heteronuclear scalar (HSQC) correlations in their 2D NMR maps and by taking into account that the configuration at the C(4)-carbon necessarily corresponds to that of the starting aziridine.

1,2-Addition products obtained in the opening reactions of aziridines 9α -Ns and 9β -Ns have been characterized by the presence of scalar correlations of the corresponding vinyl H(1) proton with the methylene protons at the C(5a)-carbon.

The relative configuration at the C(3)-carbon of *anti-1,2-addition products* **20**, **23**, **33** and **34**, obtained from aziridine **9** α -**Ns**, and **27**, **30**, **36** and **37**, obtained from aziridine **9** β -**Ns**, has been determined by the coupling constant value between the diaxial ($J_{3,4}$ = 8.7-9.7 Hz) or diequatorial H(3)-H(4) proton relationship ($J_{3,4}$ = 3.5-4.0 Hz) in the corresponding compounds from **9** α -**Ns** and **9** β -**Ns**, respectively, and by the presence of NOE between the H(3) proton and the NH in the corresponding compounds from aziridine **9** β -**Ns**. Moreover, a theoretical conformational study had indicated that *anti-1,2-addition products* from both aziridines **9** α -**Ns** and **9** β -**Ns** almost completely exist in the corresponding conformer with the –CH₂OBn side chain equatorial (Scheme 18).²⁴ In this framework, the presence of a diequatorial H(3)-H(4) protons relationship ($J_{3,4}$ = 4.3 Hz) in compound **24**, obtained in the acetolysis of aziridine **9\alpha-Ns**, clearly indicated the corresponding relative configuration at the C(3)-carbon and, as a consequence, the structure of *syn-1,2-addition product* (Scheme 18).¹⁷



Scheme 18. H(3)-H(4) protons coupling constant ($J_{3,4}$) in *anti-1,2-addition products* from aziridines 9α -Ns and 9α -Ns and syn-1,2-addition product from aziridine 9α -Ns, NOEs in *anti-1,2-addition products* from aziridine 9β -Ns and syn- and *anti-1,4-addition products* from both aziridines 9α -Ns and 9β -Ns.

1,4-Addition products obtained in the opening reactions of aziridines 9α -Ns and 9β -Ns have been characterized by means of the corresponding allyl H(1) proton which gives rise to intense NOEs on the methylene ring protons [H(5a_{ax}) and H(5a_{eq})] and on the H(2) vinyl proton (Scheme 18).

The relative configuration at the C(1)-carbon of *anti-1,4-addition products* **21** and **25** from aziridine **9\alpha-Ns** has been established on the basis of the intense H(1)/H(5) dipolar interaction, as expected for a H(1)/H(5) *cis*-1,3-diaxial arrangement. By contrast, in the case of the corresponding *syn-1,4-addition products* **22** and **26**, both H(1) and H(4) protons produce NOE on the methylene proton H(5 a_{ax}), confirming the presence of a cis relationship between the H(1) and H(4) protons.

In accordance with the above described results and with the structures shown in Scheme 18, opposite contacts are observed in the case of *1,4-addition products* obtained by nucleophilic addition to aziridine 9β -Ns: an intense H(1)/H(5) dipolar interaction is found in *syn-1,4-addition product* **29**, whereas in the corresponding *anti-1,4-addition products* **28** and **31**, H(1) and NH protons produce NOE on the methylene proton H(5a_{ax}).

In the case of epoxides 36-41 (Scheme 12), due to the conformational strain of the bicyclo structures, the relative configurations at the C(1) and C(2) carbons have been established on the basis of the comparison of dipolar interactions produced by H(1), H(4) or NH on the methylene ring protons H(5a_{ax})/H(5a_{eq}) (Scheme 12) in a similar way as discussed for the assignment of the configuration at the C(1) carbon of *syn-1,4-addition products* from aziridine 9 α -Ns (epoxides 38 and 40) or *anti-1,4-addition products* from aziridine 9 β -Ns (epoxides 39 and 41) (see Supplementary data, section 3.3).

2.8. Conclusions

The new carba-D,L-allal- and carba-D,L-galactal-derived vinyl *N*-nosyl aziridines 9α -Ns and 9β -Ns have been prepared and their regio- and stereoselective behavior examined in ring opening reactions with *O*- (MeOH and AcOH) and *N*-nucleophiles (azide anion and diethylamine). In particular, by means of acid methanolysis, taken as a model reaction, their tendency to give *1,4-addition products* (the so-called carbaglycosylating ability, as the total amount of *syn*- and/or *anti-1,4-adducts*) was checked, in view of a possible use of these activated carbaglycal system in the regio- and stereoselective construction of alkyl 4-deoxy-4-(*N*-nosylamino)-2,3-unsaturated-5*a*-carba-*O*-glycosides. The results have indicated that the carboglycosylating ability of aziridines 9α -Ns and 9β -Ns is similar (62%, entry 1 Table 1, and 60%, entry 2 Table 2, respectively) to that of the corresponding carba-D,L-allal-derived epoxide 1α (57-68%), and decidedly superior to carba-D,L-galactal epoxide 1β (20%) (Scheme 1).^{9a} The use of an Ionic Liquid (IL), as co-solvent in the acid methanolysis reaction, determines a consistent increase in the carboglycosylating ability of epoxides 1α (80%) and 1β (49%) (entries 4, Tables 3 and 4), but has no effect with aziridines 9α -Ns and 9β -Ns. The catalytic dihydroxylation by OsO₄/NMO of the residual C(1)-C(2) unsaturation

of model *anti-1,2-addition products* from aziridines 9α -Ns and 9β -Ns turned out to be completely stereoselective and could constitute a new protocol for the stereoselective synthesis of 4-*N*-protected-aminocyclitol-derivatives. Alternatively, MCPBA oxidation of further model *anti-1,2-addition products* led to pairs of diastereoisomeric epoxides which were obtained pure by TLC separation. These epoxides could act as effective and stereoselective carbaglycosyl donors for the synthesis of 4-deoxy-4-(*N*-nosylamino)-5*a*-carba-D,L- α - or - β -glycosides,²³ and appropriate studies will be reported in due course on the regio- and stereoselective behavior of these oxirane systems in the corresponding nucleophilic opening reactions in order to evaluate this point.

3. Experimental

3.1.General

All reactions were performed in a flame-dried modified Schlenk (Kjeldahl shape) flask fitted with a glass stopper or rubber septum under a positive pressure of argon. Flash column chromatography was performed employing 230-400 mesh silica gel (Macherey-Nagel). Analytical TLC were performed on Alugram SIL G/UV₂₅₄ silica gel sheets (Macherey-Nagel) with detection by 0.5% phosphomolybdic acid solution in 95% EtOH. NaN₃, PS-PPh₃, NsCl, MsCl, AcOH, HPLC grade MeOH, anhydrous CH₂Cl₂ and MeCN, pyridine and DMF over molecular sieves were purchased from Aldrich and used without purification. Et₂O and THF were distilled from sodium/benzophenone. Racemic epoxides 1α and 1β ,^{9a} and enantiomerically pure epoxides (–)- 1α and (–)- 1β ,^{9b} were prepared as previously described. Racemic *trans* 3,4-azido alcohols 12 and 16 had been previously described.^{9a} Routine ¹H and ¹³C NMR spectra were recorded at 250 and 62.5 MHz, respectively. ¹H NMR COSY, NOESY and HSQC experiments were performed with a spectrophotometer. Elemental analyses were performed at Dipartimento di Farmacia, University of Pisa, by means of Carlo Erba Automated CHN Analyzer model 1106.

3.2. Synthesis of N-nosyl aziridine 9α -Ns

3.2.1. Azidolysis of epoxide $\mathbf{1\beta}$ by NaN₃ in 1:1 THF/H₂O

A solution of epoxide 1β (0.150 g, 0.694 mmol) in distilled THF (1.70 mL) was added to a solution of NaN₃ (0.068 g, 1.041 mmol, 1.5 equiv) in H₂O (1.70 mL) and the reaction mixture was stirred for 20 h at room temperature. After dilution with Et₂O, evaporation of the washed (brine) organic solution gave *trans* 3,4-azido alcohol **12** (0.155 g, 86% yield),^{9a} pure as a yellow liquid, which was used in the next step without any further purification.

3.2.2. 6-O-Benzyl-3-deoxy-3-amino-5a-carba-D,L-gulal (13)

Polymer-supported PPh₃ (PS-PPh₃, 3 mmol/g) (0.339 g, 1.017 mmol, 1.70 equiv) was washed with CH₂Cl₂, acetone and methanol, dried at 60°C for 3 h and allowed to swell in THF/H₂O 20:1 (6.0 mL) for 30 minutes without stirring.^{8b} A solution of *trans* 3,4-azido alcohol **12**^{9a} (0.155 g, 0.598 mmol) in THF (1.0 mL) was added dropwise to the above prepared suspension of PS-PPh₃ in THF/H₂O and the reaction mixture was stirred for 48 h at room temperature. Dilution with Et₂O and evaporation of the filtered (Celite®) organic solution afforded a crude product consisting of *trans* 3,4-amino alcohol **13** (0.134 g, 96% yield), practically pure as a white, crystalline solid, mp 96-98°C; R_f = 0.15 (9:1 AcOEt/MeOH); FTIR v_{max} (Nujol) 3355 (broad), 1453, 1364, 1199, 1176, 1073, 909 cm⁻¹. ¹H NMR (CDCl₃) δ 7.27-7.40 (m, 5H), 5.69 (ddt, 1H, *J* = 9.9, 3.3, 1.3 Hz), 5.54-5.63 (m, 1H), 4.52 (s, 2H), 3.73 (dd, 1H, *J* = 6.7, 2.2 Hz), 3.66 (dd, 1H, *J* = 6.7, 2.9 Hz), 3.59 (dd, 1H, *J* = 9.2, 5.4 Hz), 3.22-3.29 (m, 1H), 2.20-2.35 (m, 1H), 2.05-2.15 (m, 2H), 1.90-2.05 (m, 2H). ¹³C NMR (CD₃OD) δ 137.9, 129.7, 128.6, 128.0, 127.8, 127.2, 76.1, 73.7, 72.6, 52.0, 35.2, 26.2. Anal. Calcd. for C₁₄H₁₉NO₂: C 72.70; H, 8.20; N, 6.00. Found: C, 72.55; H, 8.01; N, 5.74.

3.2.3. 6-O-Benzyl-3-deoxy-3-(N-nosylamino)-5a-carba-D,L-gulal (14)

A solution of *trans* 3,4-amino alcohol **13** (0.134 g, 0.575 mmol) in anhydrous CH₂Cl₂ (4.5 mL) was treated at 0°C with Et₃N (0.096 mL, 0.69 mmol, 1.2 equiv) and NsCl (0.140 g, 0.632 mmol, 1.1 equiv) and the reaction mixture was stirred for 2 h at room temperature. Dilution with CH₂Cl₂ and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude residue consisting of *N*-nosyl derivative **14** (0.238 g, 99% yield), as a yellow viscous liquid, sufficiently pure to be directly used in the next step: R_f = 0.42 (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 3310, 1539, 1416, 1365, 1248, 1070, 788 cm⁻¹. ¹H NMR (CDCl₃) δ 8.09-8.18 (m, 1H), 7.57-7.72 (m, 3H), 7.27-7.37 (m, 5H), 5.78-5.90 (m, 1H), 5.28-5.39 (m, 2H), 4.44 (s, 2H), 4.31-4.39 (m, 1H), 3.52-3.64 (m, 1H), 3.20-3.40 (m, 3H), 1.84-2.36 (m, 2H), 1.46-1.77 (m, 1H). ¹³C NMR (CD₃Cl) δ 147.4, 137.7, 133.8, 133.6, 132.8, 131.3, 130.6, 128.2, 127.5, 127.4, 125.0, 122.6, 73.1, 71.9, 71.1, 54.2, 45.9, 33.9, 24.2. Anal. Calcd. for C₂₀H₂₂N₂O₆S: C 57.41; H, 5.30; N, 6.69. Found: C, 57.26; H, 5.11; N, 6.43.

3.2.4. 6-O-Benzyl-4-O-mesyl-3-deoxy-3-(N-nosylamino)-5a-carba-D,L-gulal (15)

A solution of *N*-nosyl derivative **14** (0.458 g, 1.096 mmol) in anhydrous pyridine (5.0 mL) was treated at 0°C with MsCl (0.169 mL, 2.192 mmol, 2.0 equiv) and the reaction mixture was stirred for 12 h at the same temperature. Dilution with CH₂Cl₂ and evaporation of the washed (10% aqueous HCl, saturated aqueous NaHCO₃ and brine) organic layer afforded a crude residue (0.525 g) consisting of *trans N*-nosyl-*O*-mesyl derivative **15** (H¹ NMR) which was subjected to flash chromatography. Elution with a 1:1 hexane/ AcOEt mixture afforded *trans N*-nosyl-*O*-mesyl derivative **15** (0.478 g, 88% yield), pure as a white solid mp 53-55 °C : $R_f = 0.33$ (1:1 hexane/AcOEt); FTIR v_{max} (Nujol) 3310, 1537, 1414, 1344, 1165, 1072, 918, 844, 731 cm⁻¹. ¹H NMR (CDCl₃) δ 8.22-8.31 (m, 1H), 7.65-7.73 (m, 2H), 7.58-7.64 (m, 1H), 7.30-7.43 (m, 5H), 5.95-6-05 (m, 1H), 5.42-5.52 (m, 1H), 5.35 (d, 1H, *J* = 6.8 Hz), 4.91-4.98 (m, 1H), 4.42 (d, 1H, *J* = 11.6 Hz), 4.37 (d, 1H, *J* = 11.6 Hz), 4.22 (bs, 1H), 3.36-3-47 (m, 2H), 2.97 (s, 3H), 2.27-2.42 (m, 1H),

1.98-2.15 (m, 1H), 1.79-1.97 (m, 1H). ¹³C NMR (CD₃Cl) δ 147.9, 137.9, 134.0, 133.3, 132.6, 132.1, 128.7, 128.2, 128.1, 125.6, 121.4, 77.4, 73.3, 69.4, 52.4, 37.8, 32.8, 23.5. Anal. Calcd. for C₂₁H₂₄N₂O₈S₂: C 50.79; H, 4.87; N, 5.63. Found: C, 50.48; H, 4.61; N, 5.59.

3.2.5 (1S*, 5S*,6R*)-5-(Benzyloxymethyl)-7-nosyl-7-azabicyclo[4.1.0]hept-2-ene (9α -Ns)

A solution of *trans N*-nosyl-*O*-mesyl derivative **15** (0.145 g, 0.292 mmol) in anhydrous MeCN (15.0 mL) was treated with K₂CO₃ (0.121 g, 0.877 mmol, 3.0 equiv) at room temperature and the reaction mixture was stirred for 1 h at the same temperature. After dilution with Et₂O, evaporation of the filtered organic solution afforded *N*-nosyl aziridine **9** α -**Ns** (0.113 g, 97% yield) practically pure, as a pale yellow liquid: R_f = 0.60 (1:1 hexane/AcOEt); FTIR v_{max}(liquid film) 1542, 1364, 1332, 1164, 1123, 1087, 1058, 950 cm⁻¹. ¹H NMR (CDCl₃) δ 8.17-8.27 (m, 1H), 7.67-7-79 (m, 3H), 7.28-7.41 (m, 5H), 5.88-5.99 (m, 1H), 5.78-5.87 (m, 1H), 4.52 (s, 2H), 3.56-3-64 (m, 1H), 3.28-3.48 (m, 3H), 2.10-2.26 (m, 2H), 1.94-2.09 (m, 1H). ¹³C NMR (CD₃Cl) δ 147.8, 134.3, 132.4, 131.2, 128.6, 127.8, 124.7, 124.9, 124.0, 121.1, 73.3, 70.5, 44.9, 38.9, 29.9, 23.7. Anal. Calcd. for C₂₀H₂₀N₂O₅S: C 59.99; H, 5.03; N, 6.99. Found: C, 59.87; H, 4.84; N, 6.71.

3.3. Synthesis of N-nosyl aziridine **9β-Ns**

3.3.1. Azidolysis of epoxide $\mathbf{1}\alpha$ by NaN₃ in 1:1 THF/H₂O

A solution of epoxide 1α (0.138 g, 0.639 mmol) in distilled THF (1.6 mL) was added to a solution of NaN₃ (0.062 g, 0.958 mmol, 1.5 equiv) in H₂O (1.6 mL) and the reaction was stirred for 20 h at room temperature. After dilution with Et₂O, evaporation of the washed (brine) organic solution afforded *trans* 3,4-azido alcohol **16** (0.136 g, 82% yield),^{9a} pure as a yellow liquid, which was used in the next step without any further purification.

3.3.2. 6-O-Benzyl-3-deoxy-3-amino-5a-carba-D,L-glucal (17)

PS-PPh₃ (3 mmol/1 g) (0.617 g, 1.85 mmol, 1.70 equiv) was washed with CH₂Cl₂, acetone and methanol, dried at 60°C for 3 h and allowed to swell in THF/H₂O 20:1 (10.5 mL) for 30 minutes without stirring.^{8b} A solution of *trans* 3,4-azido alcohol **16**^{9b} (0.282 g, 1.088 mmol) in THF (2.0 mL) was added dropwise to the above suspension of PS-PPh₃ in THF/H₂O and the reaction mixture was stirred for 48 h at room temperature. Dilution with Et₂O and evaporation of the filtered (Celite®) organic solution afforded a crude product consisting of *trans* 3,4-amino alcohol **17** (0.241 g, 95% yield), practically pure as a white solid, mp 92-95 °C: $R_f = 0.10$ (9:1 AcOEt/MeOH); FTIR v_{max} (Nujol) 3350 (broad), 1455, 1364, 1192 1176, 1084, 915 cm⁻¹. ¹H NMR [(CD₃)₂CO] δ 7.20-7.39 (m, 5H), 5.66 (ddd, 1H, *J* = 9.5, 4.1, 1.7 Hz), 5.13 (ddd, 1H, *J* = 9.5, 4.5, 2.6 Hz), 4.51 (s, 2H), 3.87-3.95 (m, 1H), 3.76 (dd, 1H, J = 9.2, 3.7 Hz), 3.65-3.72 (m, 1H), 3.60 (dd, 1H, J = 9.2, 6.2 Hz), 2.65-3.00 (m, 3H), 2.26-2.37 (m, 1H), 2.07-2.17 (m, 1H), 1.81-1.93 (m, 1H). ¹³C NMR (CD₃OD) δ 139.9, 129.5, 129.3, 128.7, 128.6, 128.4, 76.4, 74.3, 72.5, 56.7, 41.4, 30.4. Anal. Calcd. for C₁₄H₁₉NO₂: C 72.70; H, 8.20; N, 6.00. Found: C, 72.55; H, 7.91; N, 5.66.

3.3.3. 6-O-Benzyl-3-deoxy-3-(N-nosylamino)-5a-carba-D,L-glucal (18)

A solution of *trans* 3,4-amino alcohol **17** (0.230 g, 0.987 mmol) in anhydrous CH₂Cl₂ (8.0 mL) was treated at 0°C, with Et₃N (0.165 mL, 1.184 mmol, 1.2 equiv) and NsCl (0.240 g, 1.084 mmol, 1.1 equiv) and the reaction mixture was stirred for 4 h at room temperature. Dilution with CH₂Cl₂ and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude residue consisting of *N*-nosyl derivative **18** (0.408 g, 99% yield), as a yellow viscous liquid, sufficiently pure to be directly used in the next step: R_f = 0.24 (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 3357 (broad), 1537, 1439, 1164, 1118, 1070, 1023, 736. cm⁻¹. ¹H NMR (CDCl₃) δ 8.13-8.20 (m, 1H), 7.87-7.93 (m, 1H), 7.68-7.74 (m, 2H), 7.27-7.37 (m, 5H), 5.68 (ddd, 1H, *J* = 10.0, 4.6, 2.3 Hz), 5.58 (d, 1H, *J* = 7.8 Hz), 5.38-5.46 (m, 1H), 4.50 (d, 1H, *J* = 12.0 Hz), 4.45 (d, 1H, *J* = 12.0 Hz), 3.86-3.98 (m, 1H), 3.59 (dd, 1H, *J* = 9.3, 4.2 Hz), 3.52 (dd, 1H, *J* = 9.3, 6.1 Hz), 3.47-3.59 (m, 1H), 3.31-3.35 (m, 1H), 1.82-2.14 (m, 3H). ¹³C NMR (CD₃Cl) δ 148.0, 137.8, 133.5, 133.1, 131.5, 129.1, 128.7, 128.1, 127.9, 126.3, 125.6, 74.5, 73.6, 72.7, 59.6, 39.6, 28.0. Anal. Calcd. for C₂₀H₂₂N₂O₆S: C 57.41; H, 5.30; N, 6.69. Found: C, 57.18; H, 5.06; N, 6.35.

3.3.4. 6-O-Benzyl-4-O-mesyl-3-deoxy-3-(N-nosylamino)-5a-carba-D,L-glucal (19)

A solution of *N*-nosyl derivative **18** (0.458 g, 1.096 mmol) anhydrous pyridine (5.0 mL) was treated at 0°C with MsCl (0.169 mL, 2.192 mmol, 2.0 equiv) and the reaction mixture was stirred for 12 h at the same temperature. Dilution with CH₂Cl₂ and evaporation of the washed (10% aqueous HCl, saturated aqueous NaHCO₃ and brine) organic layer afforded a crude residue (0.525 g) consisting of *trans N*-nosylamino–*O*-mesyl derivative **19** (H¹ NMR) which was subjected to flash chromatography. Elution with a 1:1 hexane/ AcOEt mixture afforded *trans N*-nosyl–*O*-mesyl derivative **19** (0.478 g, 88% yield), pure as a white solid, mp 53-55 °C: $R_f = 0.22$ (1:1 hexane/AcOEt); FTIR v_{max} (Nujol) 3325, 1539, 1441, 1423, 1344, 1167, 1119, 1094, 952, 776 cm⁻¹. ¹H NMR (CDCl₃) δ 8.11-8.19 (m, 1H), 7.87-7.93 (m, 1H), 7.73-7.81 (m, 2H), 7.28-7.37 (m, 5H), 5.79 (d, 1H, *J* = 9.6 Hz), 5.65-5.75 (m, 1H), 4.99 (ddd, 1H, *J* = 10.1, 4.0, 2.3 Hz), 4.82 (dd, 1H, *J* = 9.4, 8.4 Hz), 4.61 (d, 1H, *J* = 11.5 Hz), 4.45 (d, 1H, *J* = 11.5 Hz), 4.26-4.39 (m, 1H), 3.66 (dd, 1H, *J* = 9.4, 4.5 Hz), 3.58 (dd, 1H, *J* = 9.4, 2.7 Hz), 3.22 (s, 3H), 2.15-2.41 (m, 3H). ¹³C NMR (CD₃Cl) δ 148.3, 138.2, 134.1, 133.4, 130.9, 130.8, 130.7, 128.5, 128.1, 127.8, 125.7, 123.9, 81.1, 73.4, 69.5,

56.7, 52.7, 39.7, 29.9. Anal. Calcd. for C₂₁H₂₄N₂O₈S₂: C 50.79; H, 4.87; N, 5.63. Found: C, 50.71; H, 4.92; N, 5.54.

3.3.5. (1R*, 5S*,6S*)-5-(Benzyloxymethyl-)7-nosyl-7-azabicyclo[4.1.0]hept-2-ene (9β-Ns)

A solution of *trans N*-nosyl-*O*-mesyl derivative **19** (0.227 g, 0.458 mmol) in anhydrous MeCN (22.0 mL) was treated with K₂CO₃ (0.190 g, 1.374 mmol, 3.0 equiv) at room temperature, and the reaction mixture was stirred for 1 h at the same temperature. After dilution with Et₂O, evaporation of the filtered organic solution afforded *N*-nosyl aziridine **9**β-Ns (0.175 g, 95% yield) practically pure, as a pale yellow liquid: $R_f = 0.58$ (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 1541, 1364, 1331, 1161, 1123, 1089, 1059, 948 cm⁻¹. ¹H NMR (CDCl₃) δ 8.10-8.19 (m, 1H), 7.59-7-78 (m, 3H), 7.28-7.38 (m, 5H), 5.85-6.04 (m, 2H), 4.49 (d, 1H, *J* = 11.8 Hz), 4.41 (d, 1H, *J* = 11.8 Hz), 3.59 (dt, 1H, *J* = 7.1, 1.8 Hz), 3.42-3.56 (m, 3H), 2.02-2.19 (m, 2H), 1.94-2.09 (m, 1H); ¹³C NMR (CDCl₃) δ 148.3, 138.4, 134.3, 133.6, 132.1, 131.1, 128.6, 127.8, 124.7, 120.4, 73.5, 73.1, 44.5, 40.1, 31.5, 24.3. Anal. Calcd. for C₂₀H₂₀N₂O₅S: C 59.99; H, 5.03; N, 6.99. Found: C, 59.75; H, 4.92; N, 6.68. CONTROLLARE SPETTRO

3.4. Reactions of N-nosyl aziridines 9α -Ns and 9β -Ns with MeOH (O-nucleophile) under acid conditions

3.4.1. Methanolysis of N-nosyl aziridine 9α -Ns with 0.2 N H₂SO₄/MeOH (protocol A)

N-Nosyl aziridine 9α -Ns (0.060 g, 0.15 mmol) was added to 0.2 N H₂SO₄/MeOH (6.0 mL) and the resulting reaction mixture was stirred 2 h at room temperature. After dilution with CH₂Cl₂, solid NaHCO₃ was added. Evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution, afforded a crude product (0.062 g, 96% yield) consisting of a 38:30:32 mixture of (*N*-nosylamino)-methoxy derivatives **20**, **21** and **22** (¹H NMR) (entry 1, Table 1) which was subjected to preparative TLC, using a 9:1 CH₂Cl₂/(*i*-Pr)₂O mixture as the eluant. Extraction of the three most intense bands afforded *trans*-3-methoxy-4-(*N*-nosylamino)- **20** (0.013 g, 20% yield), *trans*- **21** (0.012 g, 19% yield) and *cis*-1-methoxy-4-(*N*-nosylamino) derivative **22** (0.012 g, 18% yield).

6-O-Benzyl-3-O-methyl-4-deoxy-4-(N-nosylamino)-5a-carba-D,L-glucal (**20**): a liquid, $R_f = 0.31$ (6:4 hexane/AcOEt); FTIR v_{max} (liquid film) 3346, 1536, 1441, 1341, 1162, 1086, 732 cm⁻¹. ¹H NMR (CDCl₃) δ 8.10-8.18 (m, 1H), 7.83-7.91 (m, 1H), 7.62-7.75 (m, 2H), 7.28-7.40 (m, 5H), 5.72-5.84

(m, 1H), 5.58 (d, 1H, J = 10.0 Hz), 5.47 (d, 1H, J = 8.2 Hz), 4.56 (d, 1H, J = 11.7 Hz), 4.50 (d, 1H, J = 11.7 Hz), 3.80 (dd, 1H, J = 9.1, 4.4 Hz), 3.51-3.60 (m, 3H), 2.72 (s, 3H), 2.26-2.46 (m, 1H), 1.94-2.19 (m, 2H). ¹³C NMR (CDCl₃) δ 146.6, 137.5, 135.2, 131.9, 131.8, 130.3, 128.0, 127.5, 127.0, 126.8, 124.0, 123.6, 80.4, 72.5, 70.7, 56.8, 54.6, 37.9, 28.5. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.44; H, 5.31; N, 6.14.

Methyl 6-*O*-benzyl-4-(*N*-nosylamino)-2,3,4-trideoxy-5a-carba-β-D,L-erithro-hex-2-enopyranoside (**21**): a liquid, R_f = 0.39 [9:1 CH₂Cl₂/(*i*-Pr)₂O]; FTIR v_{max} (liquid film) 3390, 1537, 1452, 1416, 1344, 1161, 1077, 1061, 1010 cm⁻¹.¹H NMR (CDCl₃) δ 8.10 (dd, 1H, *J* = 7.4, 2.0 Hz), 7.84 (dd, 1H, *J* = 7.4, 1.8 Hz), 7.60-7.74 (m, 2H), 7.26-7.41 (m, 5H), 5.78 (ddd, 1H, *J* = 10.1, 3.7, 1.8 Hz), 5.28-5.38 (m, 2H), 4.39 (s, 2H), 4.00-4.15 (m, 1H), 3.81-3.91 (m, 1H), 3.46-3.60 (m, 2H), 3.33 (s, 3H), 2.21-2.33 (m, 1H), 1.74-1.90 (m, 1H), 1.41-1.58 (m, 1H). ¹³C NMR (CDCl₃) δ 148.0, 138.5, 134.8, 133.7, 133.1, 132.4, 131.1, 129.1, 128.5, 127.7, 127.6, 125.5, 75.2, 73.3, 70.8, 56.0, 53.0, 31.8, 29.9. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.24; H, 5.37; N, 6.08.

Methyl 6-*O*-benzyl-4-(*N*-nosylamino)-2,3,4-trideoxy-5a-carba- α -D,L-erithro-hex-2-enopyranoside (**22**): a liquid, R_f= 0.28 [9:1 CH₂Cl₂/(*i*-Pr)₂O]; FTIR ν_{max} (liquid film) 3357, 1539, 1453, 1410, 1346, 1164, 1074, 935 cm⁻¹. ¹H NMR (CDCl₃) δ 8.09 (dd, 1H, *J* = 7.6, 1.6 Hz), 7.82 (dd, 1H, *J* = 7.9, 1.5 Hz), 7.68 (dt, 1H, *J* = 7.6, 1.6 Hz), 7.60 (dt, 1H, *J* = 7.6, 1.5 Hz), 7.27-7.39 (m, 5H), 5.82-5.91 (m, 1H) 5.52 (dd, 1H, *J* = 9.9, 2.1 Hz), 5.42 (d, 1H, *J* = 8.7 Hz), 4.34 (s, 2H), 4.01-4.12 (m, 1H), 3.65 (dd, 1H, *J* = 7.7, 3.8 Hz), 3.53 (dd, 1H, *J* = 9.4, 4.3 Hz), 3.43 (dd, 1H, *J* = 9.4, 3.7 Hz), 3.33 (s, 3H), 1.93-2.03 (m, 1H), 1.54-1.80 (m, 2H). ¹³C NMR (CDCl₃) δ 147.9, 138.6, 134.9, 133.6, 133.0, 131.7, 130.9, 129.3, 127.7, 127.6, 127.5, 125.4, 73.1, 71.6, 70.5, 56.8, 52.7, 36.4, 29.5. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.46; H, 5.27; N, 6.11.

3.4.2. Methanolysis of N-nosyl aziridine 9α -Ns with 2.5 10^{-3} N TsOH/CH₂Cl₂ containing MeOH (6 equiv) (protocol B)

N-nosyl aziridine 9α -Ns (0.020 g, 0.050 mmol) was added to a CH₂Cl₂ solution (2.0 mL) containing MeOH (0.012 mL, 0.30 mmoli, 6.0 equiv) and TsOHH₂O (0.001 g, 0.005 mmol, 0.1 equiv) (aziridine: TsOH: MeOH = 1: 0.1: 6) and the resulting mixture was stirred 18 h at room temperature. Dilution with CH₂Cl₂ and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product consisting of *trans*-3-methoxy-4-(*N*-nosylamino) derivative **20** (0.019 g, 90% yield) (entry 2, Table 1).

3.4.3. Methanolysis of N-nosyl aziridine 9β -Ns with 0.2 N H₂SO₄/MeOH (protocol A)

N-nosyl aziridine **9β-Ns** (0.020 g, 0.050 mmol) was added to a 0.2 N H₂SO₄/MeOH (2.0 mL) and the resulting reaction mixture was stirred 2 h at room temperature. After dilution with CH₂Cl₂, solid NaHCO₃ was added. Evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution, afforded a crude product (0.020 g, 92% yield) consisting of *6-O-benzyl-3-O-methyl-4-deoxy-4-(N-nosylamino)-5a-carba-D,L-gulal* (**27**) (entry 1, Table 2), pure as a yellow liquid: R_f = 0.85 (9:1 CH₂Cl₂/(*i*-Pr)₂O, 2 runs); FTIR v_{max}(liquid film) 3346, 1538, 1441, 1422, 1361, 1260, 1165, 1079, 1015 cm⁻¹. ¹H NMR (CDCl₃) δ 8.13-8.18 (m, 1H), 7.81-7.87 (m, 1H), 7.59-7.72 (m, 2H), 7.28-7.40 (m, 5H), 5.91 (dt, 1H, *J* = 10.0, 3.4 Hz), 5.64-5.76 (m, 2H), 4.35 (s, 2H), 3.89-3.98 (m, 1H), 3.45 (t, 1H, *J* = 3.7 Hz), 3.29-3.39 (m, 2H), 3.13 (s, 3H), 2.45 (dt, 1H, *J* = 5.8, 2.8 Hz), 2.14-2.32 (m, 1H), 1.81 (dd, 1H, *J* = 19.0, 8.8 Hz). ¹³C NMR (CDCl₃) δ 147.9, 138.2, 135.2, 133.3, 132.9, 131.1, 130.9, 128.6, 127.8, 127.7, 125.3, 124.1, 76.5, 73.2, 70.6, 56.7, 52.9, 33.6, 25.6. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.19; H, 5.33; N, 6.54.

3.4.4. Methanolysis of N-nosyl aziridine 9β -Ns with 2.5 10^{-3} N TsOH/CH₂Cl₂ containing MeOH (6 equiv) (protocol B)

N-nosyl aziridine **9** β -Ns (0.060 g, 0.15 mmol) was added to a CH₂Cl₂ solution (6.0 mL) containing MeOH (0.036 mL, 0.90 mmoli, 6.0 equiv) and TsOHH₂O (0.003 g, 0.015 mmoli, 0.1 equiv) (aziridine: TsOH: MeOH = 1: 0.1: 6) and the resulting mixture was stirred 18 h at room temperature. Dilution with CH₂Cl₂ and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.061 g, 94% yield) consisting of a 40:35:25 mixture of *N*-(nosylamino)-methoxy derivatives **27**, **28** and **29** (¹H NMR) (entry 2, Table 2) which was subjected to preparative TLC using a 9:1 CH₂Cl₂/(*i*-Pr)₂O mixture as the eluant. Extraction of the three most intense bands afforded *trans*-3-methoxy-4-(*N*-nosylamino)- **27** (0.018 g, 28% yield), *trans*- **28** (0.014 g, 22% yield) and *cis*-1-methoxy-4-(*N*-nosylamino) derivative **29** (0.009 g, 14% yield).

Methyl 6-*O*-benzyl-4-(*N*-nosylamino)-2,3,4-trideoxy-5a-carba- α -D,L-threo-hex-2-enopyranoside (**28**): a liquid, R_f = 0.64 (9:1 CH₂Cl₂/(*i*-Pr)₂O, 2 runs); FTIR v_{max}(liquid film) 3392, 1537, 1453, 1416, 1359, 1259, 1164, 1094, 1077, 1012 cm⁻¹. ¹H NMR (CDCl₃) δ 8.10-8.18 (m, 1H), 7.80-7.88 (m, 1H), 7.62-7.74 (m, 2H), 7.27-7.40 (m, 5H), 5.88 (dd, 1H, *J* = 10.1, 4.2 Hz), 5.76 (d, 1H, *J* = 7.8 Hz), 5.64 (dd, 1H, *J* = 10.1, 4.2 Hz), 4.45 (d, 1H, *J* = 12.0 Hz), 4.39 (d, 1H, *J* = 12.0 Hz), 4.07-4.22 (m, 1H), 3.69 (dd, 1H, J = 8.0, 4.0 Hz), 3.57 (dd, 1H, J = 9.4, 7.8 Hz), 3.41 (dd, 1H, J = 9.1, 5.4 Hz), 3.32 (s, 3H), 2.29-2.46 (m, 1H), 1.83 (dt, 1H, J = 14.1, 3.3 Hz), 1.61-1.71 (m, 1H). ¹³C NMR (CDCl₃) δ 147.9, 138.1, 133.5, 135.1, 133.1, 130.9, 129.9, 129.8, 128.6, 127.9, 125.5, 73.3, 71.8, 70.8, 56.6, 50.8, 34.5, 27.3. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.02; H, 5.24; N, 6.12.

Methyl 6-*O*-benzyl-4-(*N*-nosylamino)-2,3,4-trideoxy-5a-carba-β-D,L-threo-hex-2-enopyranoside (**29**): a liquid, R_f = 0.60 (9:1 CH₂Cl₂/(*i*-Pr)₂O, 2 runs); FTIR v_{max} (liquid film) 3341, 1538, 1453, 1420, 1347, 1260, 1166, 1087 cm⁻¹. ¹H NMR (CDCl₃) δ 8.11-8.16 (m, 1H), 7.81-7.88 (m, 1H), 7.62-7.75 (m, 2H), 7.27-7.40 (m, 5H), 5.78-5.85 (m, 1H), 5.55 (d, 1H, *J* = 8.6 Hz), 5.43 (ddd, 1H, *J* = 10.0, 4.8, 2.0 Hz), 4.47 (d, 1H, *J* = 12.0 Hz), 4.40 (d, 1H, *J* = 12.0 Hz), 4.07-4.19 (m, 1H), 3.71-3.82 (m, 1H), 3.59-3.71 (m, 1H), 3.36-3.45 (m, 1H), 3.34 (s, 3H), 2.22-2.48 (m, 1H), 1.92-2.16 (m, 2H). ¹³C NMR (CDCl₃) δ 147.9, 138.3, 135.3, 133.6, 132.9, 130.9, 130.7, 129.8, 128.6, 127.9, 127.8, 125.5, 75.0, 73.5, 71.6, 56.2, 50.6, 36.9, 27.3. Anal. Calcd. for C₂₁H₂₄N₂O₆S: C 58.32; H, 5.59; N, 6.47. Found: C, 58.13; H, 5.74; N, 6.29.

3.5. Reaction of aziridines 9α -Ns and 9β -Ns with AcOH (O-nucleophile) under basic and acid conditions

3.5.1. Acetolysis of aziridine 9α -Ns with AcONa in 20% aqueous DMF (protocol B)

AcONa (0.049 g, 0.60 mmol, 4.0 equiv) was added to a solution of *N*-nosyl aziridine **9α-Ns** (0.060 g, 0.15 mmol) in 20% aqueous DMF (3.5 mL) and the resulting reaction mixture was stirred 2 h at room temperature. Dilution with Et₂O and evaporation of the washed (brine) organic solution afforded a crude reaction product (0.067 g) mainly consisting of *trans* 3-acetoxy-4-(*N*-nosylamino)-derivative **23** (entry 6, Table 1), which was subjected to flash chromatography. Elution with a 1:1 hexane/AcOEt mixture afforded *3-O-acetyl-6-O-benzyl-4-deoxy-4-(N-nosylamino)-5a-carba-D,L-glucal* (**23**) (0.054 g, 78% yield), pure as a pale yellow liquid: R_f = 0.40 (1:1 hexane/AcOEt); FTIR v_{max}(liquid film) 3348, 1727, 1536, 1442, 1342, 1259, 1162, 1074, 1026 cm⁻¹. ¹H NMR (CDCl₃) δ 8.09-8.15 (m, 1H), 7.83-7.88 (m, 1H), 7.59-7.72 (m, 2H), 7.27-7.39 (m, 5H), 5.79-5.88 (m, 1H), 5.42-5.52 (m, 2H), 5.30 (d, 1H, *J* = 8.7 Hz), 4.38 (d, 1H, *J* = 11.7 Hz), 4.30 (d, 1H, *J* = 11.7 Hz), 3.85 (dd, 1H, *J* = 14.0, 8.7 Hz), 3.60 (dd, 1H, *J* = 9.4, 4.1 Hz), 3.49 (dd, 1H, *J* = 9.4, 5.3 Hz), 2.29-2.45 (m, 1H), 2.10-2.26 (m, 2H), 1.61 (s, 3H). ¹³C NMR (CDCl₃) δ 170.6, 147.7, 138.3, 135.9, 133.2, 133.1, 130.6, 130.0, 128.5, 127.7, 127.6, 125.3, 125.0, 73.4, 73.1, 70.9, 56.6, 39.0, 29.0, 20.7. Anal. Calcd. for C₂₂H₂₄N₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.06; H, 5.34; N, 5.87.

3.5.2. Acetolysis of N-nosyl aziridine 9α -Ns with 0.2 N H₂SO₄/AcOH (protocol A)

N-nosyl aziridine **9** α -**Ns** (0.060 g, 0.15 mmol) was added to a 0.2 N H₂SO₄/AcOH (3.0 mL) and the reaction mixture was stirred 1 h at room temperature. Dilution with Et₂O and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.067 g, 97% yield) consisting of an 8:22:45:25 mixture of *trans-* **23** and *cis-*3-acetoxy-4-(*N*-nosylamino)- **24**, *trans-* **25** and *cis-*1-acetoxy-4-(*N*-nosylamino) derivative **26** (¹H NMR) (entry 4, Table 1), which was subjected to preparative TLC using a 9:1 CH₂Cl₂/(*i*-Pr)₂O mixture as the eluant. Extraction of the three most intense bands afforded *cis-*3acetoxy-4-(*N*-nosylamino)- **24** (0.012 g, 17% yield), *trans-* **25** (0.021 g, 30% yield) and *cis-*1-acetoxy-4-(*N*-nosylamino) derivative **26** (0.010 g, 14% yield).

3-O-acetyl-6-O-benzyl-4-deoxy-4-(N-nosylamino)-5a-carba-D,L-altral (**24**), a colourless oil: FTIR v_{max} (liquid film) 3494, 1731, 1540, 1413, 1259, 1093, 1018 cm⁻¹; ¹H NMR δ 8.10-8.15 (m, 1H), 7.83-7.88 (m, 1H), 7.67-7.76 (m, 2H), 7.29-7.40 (m, 5H), 5.90-5.98 (m, 1H), 5.83, (d, 1H, *J* = 9.1 Hz), 5.57-5.66 (m, 1H), 4.72 (t, 1H, *J* = 4.3 Hz), 4.52 (s, 2H), 3.79-3.90 (m, 1H), 3.58-3.71 (m, 2H), 2.32-2.46 (m, 1H), 2.15-2.28 (m, 2H), 1.93 (s, 3H). ¹³C NMR δ 170.2, 147.9, 138.5, 133.9, 133.4, 130.6, 128.6, 128.0, 127.8, 125.7, 123.5, 73.6, 70.8, 67.5, 54.3, 34.8, 31.8, 22.9. Anal. Calcd. for C₂₂H₂₄N₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.21; H, 5.09; N, 6.00.

6-*O*-*Benzyl-4-(N-nosylamino)-2,3,4-trideoxy-5a-carba-β-D,L-erithro-hex-2-enopyranosyl acetate* (**25**), a colourless oil: FTIR v_{max} (liquid film) 3335, 1726, 1539, 1441, 1417, 1360, 1259, 1165, 1068, 1020 cm⁻¹. ¹H NMR δ 8.10 (dd, 1H, *J* = 7.6, 1.5 Hz), 7.25 (dd, 1H, *J* = 7.6, 1.5 Hz), 7.56-7.78 (m, 2H), 7.28-7.41 (m, 5H), 5.61-5.70 (m, 1H), 5.44 (dt, 1H, *J* = 10.2, 2.1 Hz), 5.35 (d, 1H, *J* = 8.7 Hz), 5.29-5.40 (m, *W*_{1/2}= 15.4 Hz, 1H), 4.36 (s, 2H), 4.11-4.27 (m, 1H), 3.56 (dd, 1H, *J* = 9.1, 4.3 Hz), 3.42 (dd, 1H, *J* = 9.1, 3.3 Hz), 2.13-2.34 (m, 1H), 2.02 (s, 3H), 1.81-1.97 (m, 1H), 1.60-1.78 (m, 1H). ¹³C NMR δ 170.7, 148.0, 138.4, 134.8, 133.7, 133.1, 131.0, 130.8, 130.7, 128.5, 127.8, 127.6, 125.5, 73.3, 70.2, 69.2, 52.4, 40.9, 31.6, 22.8. Anal. Calcd. for C₂₂H₂₄N₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.10; H, 4.94; N, 5.69.

6-*O*-*Benzyl-4*-(*N*-nosylamino)-2,3,4-trideoxy-5a-carba-α-D,L-erithro-hex-2-enopyranosyl acetate (**26**), a colourless oil: FTIR v_{max} (liquid film) 3494, 1731, 1540, 1413, 1259, 1093, 1018 cm⁻¹. ¹H NMR δ 8.10 (dd, 1H, *J* = 7.6, 1.5 Hz), 7.85 (dd, 1H, *J* = 7.7, 1.5 Hz), 7.58-7.75 (m, 2H), 7.27-7.40 (m, 5H), 5.81 (ddd, 1H, *J* = 9.6, 4.3, 2.0 Hz), 5.59 (dd, 1H, *J* = 9.6, 2.2 Hz), 5.41 (d, 1H, *J* = 8.5 Hz), 5.14-5.20 (m, 1H), 4.34 (s, 2H), 4.05-4.15 (m, 1H), 3.55 (dd, 1H, *J* = 9.4, 4.1 Hz), 3.45 (dd, 1H, J = 9.4, 3.5 Hz), 2.04 (s, 3H), 1.95-2.02 (m, 1H), 1.86-1.95 (m, 2H). ¹³C NMR δ 170.7, 148.0, 138.4, 134.8, 133.7, 133.3, 133.2, 131.0, 128.6, 127.8, 127.7, 125.5, 73.3, 70.3, 65.6, 52.4, 36.8, 30.5, 21.4. Anal. Calcd. for C₂₂H₂₄N₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.22; H, 5.04; N, 5.95.

3.5.3. Acetolysis of N-nosyl aziridine 9α -Ns with 2.5 10^{-3} N TsOH/CH₂Cl₂ containing AcOH (6 equiv) (protocol B)

N-nosyl aziridine 9α -Ns (0.020 g, 0.050 mmol) was treated with 2.5^{-10⁻³} N TsOH/CH₂Cl₂ solution (0.5 mL) containing AcOH (0.018 g, 0.017 mL, 0.30 mmol, 6 equiv) and the reaction mixture was stirred 16 h at room temperature. Usual work-up afforded a crude reaction product consisting of 11:29:19:41 mixture of *trans-* 23 and *cis-*3-acetoxy-4-(*N*-nosylamino)- 24, *trans-* 25 and *cis-*1-acetoxy-4-(*N*-nosylamino) derivative 26 (¹H NMR) (entry 5, Table 1).

3.5.4. Acetolysis of N-nosyl aziridine 9β -Ns with AcONa in 20% aqueous DMF (protocol B)

AcONa (0.050 g, 0.60 mmol, 4.0 equiv) was added to a solution of *N*-nosyl aziridine **9β-Ns** (0.060 g, 0.15 mmol) in 20% aqueous DMF (3.5 mL) and the resulting reaction mixture was stirred 2 h at room temperature. Dilution with Et₂O and evaporation of the washed (brine) organic solution afforded a crude reaction product (0.068 g) mainly consisting of *trans*-3-acetoxy-4-(*N*-nosylamino) derivative **30** (entry 6, Table 2) which was subjected to flash chromatography. Elution with a 1:1 hexane/AcOEt mixture afforded *3-O-acetyl-6-O-benzyl-4-deoxy-4-(N-nosylamino)-5a-carba-D,L-gulal* (**30**) (0.052 g, 75% yield), pure as a pale yellow liquid: $R_f = 0.45$ (1:1 hexane/AcOEt); FTIR v_{max}(liquid film) 3361, 1731, 1539, 1441, 1424, 1362, 1258, 1236, 1166, 1090, 1013 cm⁻¹. ¹H NMR (CDCl₃) δ 8.15-8.24 (m, 1H), 7.80-7.90 (m, 1H), 7.67, 7.77 (m, 2H), 7.29-7.43 (m, 5H), 5.97-6.07 (m, 1H), 5.56-5.74 (m, 2H), 4.90 (t, 1H, *J* = 3.9 Hz), 4.51 (d, 1H, *J* = 12.0 Hz), 4.42 (d, 1H, *J* = 12.0 Hz), 3.95 (dt, 1H, *J* = 9.1, 3.4 Hz), 3.48-3.59 (m, 1H), 3.42 (dd, 1H, *J* = 9.1, 5.7 Hz), 2.10-2.51 (m, 2H), 1.93 (s, 3H), 1.77-1.92 (m, 1H). ¹³C NMR δ 170.2, 147.0, 138.3, 134.2, 133.7, 133.0, 132.4, 131.4, 128.6, 128.0, 125.4, 122.7, 73.5, 70.6, 68.1, 53.0, 33.7, 29.9, 21.2. Anal. Calcd. for C₂₂H₂₄A₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.54; H, 5.18; N, 6.03.

3.5.5. Acetolysis of N-nosyl aziridine 9β -Ns with 0.2 N H₂SO₄/AcOH (protocol A)

N-nosyl aziridine 9β -Ns (0.040 g, 0.10 mmol) was added to a 0.2 N H₂SO₄/AcOH solution (2.0 mL) and the reaction mixture was stirred 1 h at room temperature. Dilution with Et₂O and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.045 g, 98% yield) consisting of a 36:64 mixture of *trans*-3-acetoxy-4-(*N*-

nosylamino)- **30** and *trans*-1-acetoxy-4-(*N*-nosylamino) derivative **31** (¹H NMR) (entry 4, Table 2) which was subjected to preparative TLC using a 9:1 CH₂Cl₂/(*i*-Pr)₂O mixture as the eluant. Extraction of the two most intense bands afforded *trans*-3-acetoxy-4-(*N*-nosylamino) derivative **30** (0.010 g, 22% yield) and *6-O-benzyl-4-(N-nosylamino)-2,3,4-trideoxy-5a-carba-α-D,L-threo-hex-2-enopyranosyl acetate* (**31**) (0.019 g, 41% yield) pure as a pale yellow oil: $R_f = 0.53$ (9:1 CH₂Cl₂/(*i*-Pr)₂O); FTIR v_{max}(liquid film) 3261, 1731, 1539, 1441, 1424, 1362, 1258, 1236, 1166, 1090, 1013 cm⁻¹. ¹H NMR (CDCl₃) δ 8.08-8.16 (m, 1H), 7.78-7.87 (m, 1H), 7.62-7.74 (m, 2H), 7.27-7.40 (m, 5H), 5.77-5.85 (m, 1H), 5.65-5.77 (m, 2H), 5.21 (dd, 1H, *J* = 7.9, 3.9 Hz), 4.51 (d, 1H, *J* = 12.0 Hz), 4.42 (d, 1H, *J* = 12.0 Hz), 4.10-4.26 (m, 1H), 3.60 (dd, 1H, *J* = 9.3, 8.1 Hz), 3.43 (dd, 1H, *J* = 9.3, 5.3 Hz), 2.29-2.47 (m, 1H), 2.00 (s, 3H), 1.71-1.85 (m, 2H). ¹³C NMR δ 170.6, 148.0, 138.3, 133.6, 133.2, 133.1, 131.4, 130.8, 128.7, 128.3, 128.0, 127.9, 125.6, 73.7, 70.9, 65.9, 50.6, 35.0, 28.1, 21.6. Anal. Calcd. for C₂₂H₂₄N₂O₇S: C 57.38; H, 5.25; N, 6.08. Found: C, 57.14; H, 4.93; N, 5.81.

3.5.6. Acetolysis of N-nosyl aziridine 9β -Ns with 2.5 10^{-3} N TsOH/CH₂Cl₂ containing AcOH (6 equiv) (protocol B)

N-nosyl aziridine **9** β -Ns (0.020 g, 0.050 mmol) was treated with 2.5^{-10⁻³} N TsOH/CH₂Cl₂ solution (0.5 mL) containing AcOH (0.018 g, 0.017 mL, 0.30 mmol, 6 equiv) and the reaction mixture was stirred 18 h at room temperature. Dilution with Et₂O and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.021 g, 91% yield) consisting of a 30:70 mixture of *trans*-3-acetoxy-4-(*N*-nosylamino)- **30** and *trans*-1-acetoxy-4-(*N*-nosylamino) derivative **31** (¹H NMR) (entry 5, Table 2).

3.6. Acid methanolysis of epoxides 1α and 1β and N-nosyl aziridines 9α -Ns and 9β -Ns in the presence of an ionic liquid (IL)

3.6.1. Acid methanolysis of epoxide 1α with 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bmim][BF₄] solution containing MeOH (6 equiv) (protocol B)

Typical procedure. A solution of epoxide 1α (0.011 g, 0.051 mmol) in 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bimim][BF₄] mixture (1.0 mL) containing MeOH (12.0 µL, 0.0098 g, 0.306 mmol, 6 equiv) was stirred 18 h at room temperature. Dilution with Et₂O and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.012 g, 97% yield) consisting of a 20:76:4 mixture of corresponding methoxy alcohols derivatives **3** (*anti*-

1,2-addition product), **4** (*syn-1,4-addition product*) and **5** (*anti-1,4-addition product*),^{9a} respectively (¹H NMR) (entry 4, Table 3).

3.6.2. Methanolysis of epoxide 1β with 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bmim][BF₄] solution containing MeOH (6 equiv) (protocol B)

Following the typical procedure, a solution of epoxide 1β (0.011 g, 0.051 mmol) in 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bimim][BF₄] mixture (1.0 mL) containing MeOH (12.0 µL, 0.0096 g, 0.306 mmol, 6 equiv) was stirred 18 h at room temperature. Usual workup afforded a crude reaction product (0.011 g, 89% yield) consisting of a 51:31:18 mixture of corresponding methoxy alcohols derivatives **6** (*anti-1,2-addition product*), **7** (*syn-1,4-addition product*) and **8** (*anti-1,4-addition product*), ^{9a} respectively (¹H NMR) (entry 4, Table 4).

3.6.3. Methanolysis of N-nosyl aziridine 9α -Ns with 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bmim][BF₄] solution containing MeOH (6 equiv) (protocol B)

Typical procedure. A solution of *N*-nosyl aziridine 9α -Ns (0.010 g, 0.025 mmol) in 2.5 10⁻³ N TsOH in CH₂Cl₂/[bimim][BF₄] mixture (1:1) (1.0 mL) containing MeOH (6.0 µL, 0.15 mmol, 6 equiv) was stirred 18 h at room temperature. Dilution with Et₂O and evaporation of the washed (saturated aqueous NaHCO₃ and brine) organic solution afforded a crude reaction product (0.010 g, 99% yield) consisting of *trans*-3-methoxy-4-(*N*-nosylamino) derivative **20** (¹H NMR) (entry 3, Table 1).

3.6.4. Methanolysis of N-nosyl aziridine 9β -Ns with 2.5 10^{-3} N TsOH in 1:1 CH₂Cl₂/[bmim][BF₄] solution containing MeOH (6 equiv) (protocol B)

Following the typical procedure, a solution of *N*-nosyl aziridine 9β -Ns (0.010 g, 0.025 mmol) in 2.5 $\cdot 10^{-3}$ N TsOH in 1:1 CH₂Cl₂/[bimim][BF₄] mixture (1.0 mL) containing MeOH (6.0 µL, 0.15 mmol, 6 equiv) was stirred 18 h at room temperature. Usual workup afforded a crude reaction product (0.010 g, 99% yield) consisting of a 40:28:32 mixture of *trans*-3-methoxy-4-(*N*-nosylamino)- **27**, *trans*- **28** and *cis*-1-methoxy-4-(*N*-nosylamino) derivative **29** (¹H NMR) (entry 3, Table 2).

3.7. Reactions of N-nosyl aziridines 9α -Ns and 9β -Ns with NaN₃ and NHEt₂ (N-Nucleophiles)

3.7.1. Reaction of N-nosyl aziridine 9α -Ns with NaN₃ in 1:1 THF/H₂O (protocol B)

A solution of *N*-nosyl ziridine 9α -Ns (0.033 g, 0.0825 mmol) in distilled THF (0.3 mL) was added to a solution of NaN₃ (0.0059 g, 0.091 mmol, 1.1 equiv) in H₂O (0.3 mL) and the reaction mixture was stirred 4 h at room temperature. After dilution with Et₂O, evaporation of the washed (brine) organic solution afforded a crude reaction product (0.035 g, 96% yield) consisting of 77:23 mixture of *trans*-3-azido-4-(*N*-nosylamino)- **34** and *trans*-1-azido-4-(*N*-nosylamino) derivative **35** which was subjected to preparative TLC, using a 7:3 hexane/AcOEt mixture as the eluant. Extraction of the two most intense bands afforded *trans*-3-azido-4-(*N*-nosylamino)- **34** (0.015 g, 41% yield) and *trans*-1-azido-4-(*N*-nosylamino) derivative **35** (0.005 g, 14% yield).

6-*O*-*Benzyl-3*, 4-dideoxy-3-azido-4-(*N*-nosylamino)-5a-carba-*D*,*L*-glucal (**34**): a colourless liquid, R_f = 0.56 (7:3 hexane/AcOEt); FTIR ν_{max}(liquid film) 3334, 2094, 1536, 1422, 1358, 1257, 1165, 1119, 1084, 1061, 1027, 916 cm⁻¹. ¹H NMR (CDCl₃) δ 8.16-8.21 (m, 1H), 7.83-7.93 (m, 1H), 7.66-7.76 (m, 2H), 7.27-7.41 (m, 5H), 5.85-5.95 (m, 1H), 5.60 (d, 1H, *J* = 9.0 Hz), 5.51 (d, 1H, *J* = 9.0 Hz), 4.50 (d, 1H, *J* = 11.7 Hz), 4.44 (d, 1H, *J* = 11.7 Hz), 3.72-3.79 (m, 1H), 3.68 (dd, 1H, *J* = 9.7, 4.2 Hz), 3.50-3.62 (m, 2H), 2.30-2.44 (m, 1H), 2.16-2.26 (m, 1H), 2.00-2.15 (m, 1H). ¹³C NMR (CD₃CN) δ 140.4, 136.1, 134.8, 133.3, 132.4, 132.0, 130.6, 129.9, 129.1, 126.9, 125.6, 74.2, 71.5, 58.5, 53.4, 41.4, 33.0. Anal. Calcd. for C₂₀H₂₁N₅O₅S: C 54.16; H, 4.77; N, 15.79. Found: C, 54.09; H, 4.54; N, 15.55.

6-*O*-*Benzyl-4-(N-nosylamino)-2,3,4-trideoxy-5a-carba-β-D,L-erithro-hex-2-enopyranosyl* azide (**35**): a colourles liquid, R_f = 0.36 (7:3 hexane/AcOEt); FTIR v_{max} (liquid film) 3354, 2094, 1646, 1540, 1453, 1356, 1257, 1167, 1093, 1062 cm⁻¹. ¹H NMR (CDCl₃) δ 8.11 (dd, 1H, *J* = 7.3, 2.1 Hz), 7.83-7.88 (m, 1H), 7.62-7.76 (m, 2H), 7.27-7.41 (m, 5H), 5.69 (ddd, 1H, *J* = 10.1, 2.3, 1.8 Hz), 5.45 (dt, 1H, *J* = 10.1, 2.2 Hz), 5.36 (d, 1H, *J* = 8.7 Hz), 4.39 (s, 2H), 4.04-4.22 (m, 1H), 3.87-4.03 (m, 1H), 3.54 (d, 2H, *J* = 4.2 Hz), 2.20-2.30 (m, 1H), 1.61-1.89 (m, 2H). ¹³C NMR (CDCl₃) δ 147.9, 138.3, 134.7, 133.8, 133.2, 131.2, 131.0, 130.0, 128.6, 127.9, 127.7, 125.5, 73.4, 70.2, 57.0, 52.4, 41.1, 31.8. Anal. Calcd. for C₂₀H₂₁N₅O₅S: C 54.16; H, 4.77; N, 15.79. Found: C, 54.29; H, 4.49; N, 15.39.

3.7.2. Reaction of N-nosyl aziridine 9β -Ns with NaN₃ in 1:1 THF/H₂O (protocol B)

A solution of *N*-nosyl aziridine 9β -Ns (0.022 g, 0.055 mmol) in distilled THF (0.2 mL) was added to a solution of NaN₃ (0.004 g, 0.061 mmol, 1.1 equiv) in H₂O (0.2 mL) and the reaction mixture was stirred 4 h at room temperature. After dilution with Et₂O, evaporation of the washed (brine) organic solution afforded a reaction product (0.023 g, 99% yield) consisting of *trans*-3azido-4-(*N*-nosylamino) derivative **37** (0.020 g) which was subjected to preparative TLC using a 7:3 hexane/AcOEt mixture as the eluant. Extraction of the most intense band afforded pure *6-O*benzyl-3,4-dideoxy-3-azido-4-(*N*-nosylamino)-5a-carba-D,L-gulal (**37**), (0.021 g, 86% yield), a liquid, R_f = 0.39 (hexane/AcOEt 7:3); FTIR v_{max} (liquid film) 3342, 2096, 1538, 1442, 1422, 1360, 1242, 1166, 1098, 1078, 1012 cm⁻¹. ¹H NMR (CDCl₃) δ 8.10-8.19 (m, 1H), 7.79-7.89 (m, 1H), 7.61-7.74 (m, 2H), 7.27-7.41 (m, 5H), 6.00-6.10 (m, 1H), 6.90 (d, 1H, *J* = 8.5 Hz), 5.60-5.70 (m, 1H), 4.34 (s, 2H), 3.80-3.90 (m, 1H), 3.72 (t, 1H, *J* = 3.7 Hz), 3.31-3.46 (m, 2H), 2.19-2.47 (m, 2H), 1.76-1.94 (m, 1H). ¹³C NMR (CDCl₃) δ 147.9, 137.9, 134.8, 133.6, 133.1, 132.4, 130.9, 128.6, 127.9, 127.6, 125.5, 122.0, 73.3, 70.3, 59.1, 54.5, 33.9, 25.5. Anal. Calcd. for C₂₀H₂₁N₅O₅S: C 54.16; H, 4.77; N, 15.79. Found: C, 54.11; H, 4.91; N, 15.74.

3.7.3. Reaction of N-nosyl aziridine 9α -Ns with Et₂NH (protocol A)

N-nosyl aziridine **9α**-**Ns** (0.015 g, 0.0375 mmol) was dissolved in distilled Et₂NH (1.5 mL) and the reaction mixture was stirred at room temperature for 2 days in the presence of Sc(OTf)₃ (0.0018 g, 0.00375 mmol, 0.1 equiv). Dilution with Et₂O and evaporation of the washed (brine) organic solution afforded a reaction product (0.016 g, 90% yield) consisting of *trans*-3-(*N*,*N*-diethylamino)-4-(*N*-nosylamino) derivative **33** which was subjected to preparative TLC, using a 6:4 hexane/AcOEt mixture as the eluant. Extraction of the most intense band afforded *6-O-benzyl-3,4-dideoxy-3-(N*,*N*-*diethylamino)-4-(N-nosylamino)-5a-carba-D*,*L-glucal* (**33**) (0.012 g, 68% yield), pure as a pale yellow liquid: R_f = 0.38 (1:1 hexane/AcOEt); FTIR ν_{max} (liquid film) 3372, 1542, 1460, 1401, 1347, 1304, 1170, 1120, 1065, 1055 cm⁻¹. ¹H NMR (CDCl₃) δ 8.09-8.20 (m, 1H), 7.84-7.93 (m, 1H), 7.60-7.70 (m, 2H), 7.28-7.39 (m, 5H), 5.71-5.84 (m, 1H), 5.50-5.60 (m, 1H), 4.51 (d, 1H, *J* = 11.9 Hz), 4.44 (d, 1H, *J* = 11.9 Hz), 3.75 (dd, 1H, *J* = 9.0, 3.4 Hz), 3.47-3.64 (m, 2H), 3.07-3.20 (m, 1H, $W_{1/2}$ = 17.8 Hz), 2.15-2.41 (m, 6H), 1.95-2.09 (m, 2H), 0.67 (t, 6H, *J* = 7.1 Hz). ¹³C NMR (CDCl₃) δ 147.7, 138.7, 137.0, 132.9, 132.7, 130.8, 128.7, 128.5, 127.9, 127.7, 126.0, 125.4, 73.4, 72.6, 64.2, 57.0, 43.9, 41.2, 29.7, 13.3. Anal. Calcd. for C₂₄H₃₁N₃O₅S: C 60.86; H, 6.60; N, 8.87. Found: C, 60.48; H, 6.28; N, 8.50.

3.7.4. Reaction of N-nosyl aziridine 9β -Ns with Et₂NH (protocol A)

N-nosyl aziridine **9** β -Ns (0.015 g, 0.0375 mmol) was dissolved in distilled Et₂NH (1.5 mL) and the reaction mixture was stirred at room temperature for 4 days in the presence of Sc(OTf)₃ (0.0018 g, 0.00375 mmol, 0.1 equiv). Dilution with Et₂O and evaporation of the washed (brine) organic solution afforded a reaction product (0.017 g, 96% yield) consisting of *trans*-3-(*N*,*N*-diethylamino)-4-(*N*-nosylamino) derivative **36** (¹H NMR) which was subjected to preparative TLC,

using a 6:4 hexane/AcOEt mixture as the eluant. Extraction of the most intense band afforded *6-Obenzyl-3,4-dideoxy-3-(N,N-diethylamino)-4-(N-nosylamino)-5a-carba-D,L-gulal* (**36**) (0.011 g, 62% yield), pure as a pale yellow liquid, $R_f = 0.21$ (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 3372, 1541, 1454, 1406, 1347, 1304, 1170, 1120, 1065, 1055 cm⁻¹. ¹H NMR (CDCl₃) δ 8.05-8.12 (m, 1H), 7.78-7.85 (m,1H), 7.57-7.85 (m, 2H), 7.27-7.35 (m, 5H), 5.71-5.82 (m, 1H), 5.48-5.58 (m, 1H), 4.43 (d, 1H, J = 12.1 Hz), 4.36 (d, 1H, J = 12.1 Hz), 3.48-3.70 (m, 2H), 3.45 (t, 1H, J = 9.6 Hz), 2.92-3.06 (m, 1H, $W_{1/2} = 10.8$ Hz), 2.70-2.85 (m,1H), 2.33 (q, 4H, J = 6.6 Hz), 1.91-2.27 (m, 3H), 0.72 (t, 6H, J = 6.6 Hz). ¹³C NMR (CDCl₃) δ 138.7, 134.7, 133.4, 132.8, 131.0, 129.2, 128.5, 127.9, 127.7, 127.6, 125.5, 125.1, 77.4, 73.1, 69.7, 59.1, 43.8, 35.0, 26.4, 13.3. Anal. Calcd. for C₂₄H₃₁N₃O₅S: C 60.86; H, 6.60; N, 8.87. Found: C, 60.59; H, 6.39; N, 8.66.

3.8. Catalytic dihydroxylation of diastereoisomeric trans-3-methoxy-4-(N-nosylamino) derivatives 20 and 27 (anti-1,2-addition products) obtained by aziridines 9α -Ns and 9β -Ns, respectively

3.8.1.Catalytic dihydroxylation by OsO₄/NMO protocol of trans-3.methoxy-4-(N-nosylamino) derivative **20**

Typical procedure. A solution of trans-3-methoxy-4-(N-nosylamino) derivative 20 (0.030 g. 0.070 mmol) in an 1:1 t-BuOH/acetone mixture (0.24 mL) was added, at 0°C under stirring and in the dark, to a 50% p/v aqueous solution of N-methyl morpholine-N-oxide (NMO) (0.060 mL) and the resulting reaction mixture was treated with 2.5% p/v OsO₄ solution in *t*-BuOH (0.060 mL) and stirred for 15 h at room temperature. Dilution with Et₂O and evaporation of the filtered (Celite®) organic solution afforded a crude reaction product (0.032 g) consisting of *cis* diol 42 which was subjected to preparative TLC using a 9:1 $CH_2Cl_2/(i-Pr)_2O$ mixture as the eluant. Extraction of the most intense band afforded pure 6-O-benzyl-3-O-methyl-4-deoxy-4-(N-nosylamino)-5a-carba- α -*D,L-glucopyranose* (42), as a pale yellow oil (0.024 g, 73% yield): FTIR v_{max} (liquid film) 3360 (broad), 1595, 1537, 1417, 1259, 1162, 1088, 1027, 798 cm⁻¹. ¹H NMR (CDCl₃) δ 8.09-8.16 (m, 1H), 7.77-7.88 (m, 1H), 7.56-7.74 (m, 2H), 7.28-7.41 (m, 5H), 5.40 (d, 1H, J = 9.4 Hz), 4.41 (d, 1H, J = 11.7 Hz), 4.36 (d, 1H, J = 11.7 Hz), 3.98-4.09 (m, 1H), 3.72 (t, 1H, J = 4.6 Hz), 3.52-3.57 (m, 2H), 3.47 (dd, 1H, J = 8.4, 2.4 Hz), 3.21 (t, 1H, J = 9.4 Hz), 2.97 (s, 3H), 2.37-2.44 (m, 1H), 1.94-2.12 (m, 4H). ¹³C NMR (CDCl₃) δ 147.0, 138.6, 136.3, 133.0, 132.8, 131.0, 128.7, 128.5, 127.8, 125.0, 84.0, 76.1, 73.3, 70.4, 69.0, 60.5, 58.3, 55.6, 37.1, 30.5, 29.9. Anal. Calcd. for C₂₁H₂₆N₂O₈S: C 54.06; H, 5.61; N, 6.60. Found: C, 53.75; H, 5.37; N, 6.55.

A solution of cis diol 42 (0.019 g, 0.040 mmol) in anhydrous pyridine (1.0 mL) was treated with Ac₂O (0.5 mL) at 0°C and the resulting reaction mixture was stirred at room temperature overnight. Co-evaporation of the reaction mixture with toluene afforded a crude product (0.021 g) consisting of diacetate 42-diAc, which was subjected to preparative TLC, using a 1:1 hexane/AcOEt mixture as the eluant. Extraction of the more intense band afforded 1,2-di-O-acetyl-6-O-benzyl-3-O-methyl-4-deoxy-4-(N-nosylamino)-5a-carba- α -D,L-glucopyranose (42-diAc) (0.016 g, 73% yield) pure as a pale yellow liquid: $R_f = 0.30$ (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 3228, 1736, 1718 (shoulder), 1537, 1454, 1366, 1344, 1249, 1238, 1163, 1075, 1047, 1024, 921 cm⁻¹. ¹H NMR (CDCl₃) δ 8.07-8.15 (m, 1H), 7.81-7.88 (m, 1H), 7.63-7.75 (m, 2H), 7.27-7.40 (m, 5H), 5.38 (d, 1H, J = 8.7 Hz), 5.32-5.37 (m, 1H), 4.71 (dd, 1H, J = 9.7, 2.9 Hz), 4.54 (d, 1H, J = 9.711.7 Hz), 4.45 (d, 1H, J = 11.7 Hz), 3.59-3.69 (m, 2H), 3.53 (t, 1H, J = 9.7 Hz), 3.28 (t, 1H, J Hz), 2.72 (s, 3H), 2.09 (s, 3H), 1.97 (s, 3H), 1.66-1.77 (m, 1H), 1.55-1.65 (m, 2H). ¹³C NMR (CDCl₃) & 170.1, 170.0, 147.8, 138.4, 136.2, 133.1, 132.8, 131.4, 128.5, 128.0, 127.8, 124.9, 81.4, 75.7, 73.5, 70.1, 68.9, 60.1, 58.5, 37.8, 29.9, 21.3, 21.1. Anal. Calcd. for C₂₅H₃₀N₂O₁₀S: C 54.54; H, 5.49; N, 5.08. Found: C, 54.30; H, 5.23; N, 4,73.

3.8.2. Catalytic dihydroxylation by OsO₄/NMO protocol of trans-3-acetoxy-4-(N-nosylamino) derivative 27

Following the typical procedure, the treatment of a solution of *trans*-3-acetoxy-4-(*N*-nosylamino) derivative **27** (0.030 g, 0.070 mmol) in an 1:1 *t*-BuOH/acetone mixture (0.24 mL) with 50% p/v aqueous solution of *N*-methyl morpholine-*N*-oxide (NMO) (0.060 mL) and 2.5% p/v OsO₄ solution in *t*-BuOH (0.060 ml) for 96 h at room temperature afforded a crude reaction product consisting of *cis* diol **43** (0.031 g): FTIR v_{max} (liquid film) 3360 (broad), 1594 (broad), 1537, 1417,1259, 1162, 1088, 1027, 798 cm⁻¹. ¹H NMR (CD₃OD) δ 7.98-8.04 (m, 1H), 7.57-7.67 (m, 2H), 7.24-7.37 (m, 6H), 4.50-4.53 (m, 1H), 4.34 (d, 1H, *J* = 11.9 Hz), 4.27 (d, 1H, *J* = 11.9 Hz), 3.72-3.80 (m, 2H), 3.61-3.71 (m, 2H), 3.34-3.47 (m, 2H), 3.25 (s, 3H), 1.95-2.12 (m, 1H), 1.56-1.79 (m, 2H).

Di-*O*-acetylation of crude *cis* diol **43** (0.031 g, 0.066 mmmol) by anhydrous pyridine (1.0 mL)/Ac₂O (0.5 mL) protocol afforded a crude product (0.035 g) consisting of diacetate **43-diAc** which was subjected to preparative TLC using an 1:1 hexane/AcOEt mixture as the eluant. Extraction of the most intense band afforded *1,2-di-O-acetyl-6-O-benzyl-3-O-methyl-4-deoxy-4-(N-nosylamino)-5a-carba-β-D,L-idopyranose* (**43-diAc**) (0.026 g, 72% yield), pure as a pale yellow liquid: $R_f = 0.34$ (1:1 hexane/AcOEt); FTIR v_{max} (liquid film) 3345, 1739, 1539, 1428, 1370, 1259,

1166, 1086, 1018, 797 cm⁻¹. ¹H NMR (CDCl₃) δ 8.08 (dd, 1H, *J* = 7.7, 1.8 Hz), 7.85 (dd, 1H, *J* = 7.6, 1.5 Hz), 7.70 (dt, 1H, *J* = 7.6, 1.5 Hz), 7.62 (dt, 1H, *J* = 7.6, 1.8 Hz), 7.27-7.40 (m, 5H), 6.06 (d, 1H, *J* = 9.5 Hz), 5.39 (t, 1H, *J* = 3.1 Hz), 5.07 (ddd, 1H, *J* = 11.8, 4.7, 3.1 Hz), 4.33 (d, 1H, *J* = 12.0 Hz), 4.27 (d, 1H, *J* = 12.0 Hz), 3.92 (dt, 1H, *J* = 9.6, 3.1 Hz), 3.46 (dd, 1H, *J* = 9.6, 7.6 Hz), 3.26-3.34 (m, 2H), 3.25 (s, 3H), 2.29-2.46 (m, 1H), 2.24 (s, 3H), 1.98 (s, 3H), 1.69-1.85 (m, 1H), 1.57-1.65 (m, 1H). ¹³C NMR (CDCl₃) δ 170.2, 169.9, 147.8, 138.4, 134.7, 133.6, 133.0, 130.9, 128.6, 127.8, 127.6, 125.5, 79.1, 73.1, 70.9, 69.3, 68.6, 58.3, 51.9, 35.1, 29.9, 24.3, 21.1. Anal. Calcd. for C₂₅H₃₀N₂O₁₀S: C 54.54; H, 5.49; N, 5.08. Found: C, 54.27; H, 5.12; N, 4.92.

Acknowledgments

This work was supported by the University of Pisa (University Funds 2014-2016).

Appendix A: Supplementary data

Supplementary data related to this article can be found at.....

References and notes

- a) Suami T. Pure Appl. Chem. 1987; 59: 1509-1520.
 b) Suami T, Ogawa S. Adv. Carbohydr. Chem. Biochem. 1990; 48: 21-90.
 c) Agrofoglio L, Suhas E, Farese A, Condom R, Challand SR, Earl RA, Guedj R. Tetrahedron 1994; 50: 10611-10670.
 d) Balci M, Sütbeyaz Y, Secen H. Tetrahedron 1990; 46; 3715–3742.
 e) Arjona O, Gómez AM, López LC, Plumet J. Chem. Rev. 2007; 107:1919-2036.
 f) Kilbas B, Balci M. Tetrahedron 2011;67: 2355-2389.
- Ogawa S. In Carbohydrate Mimics: Concepts and Methods; Chapleur, Y., Ed.; Wiley-VCH: Weinheim, Germany, 1998.
- 3. Kudo F, Eguchi T. J. Antibiot. 2009; 62: 471-481.
- a) Chen X, Zheng Y, Shen Y. *Curr. Med. Chem.* 2006; 13: 109-116.
 b) Lin H, Sugimoto Y, Ohsaki Y, Ninomiya H, Oka A, Taniguchi M, Ida H, Eto Y, Ogawa S, Matsuzaki Y, Sawa M, Inoue T, Higaki K, Nanba E, Ohno K, Suzuki Y. *Biochim. Biophys. Acta, Mol. Basis Dis.* 2004;1689: 219-228.
 c) Diaz L, Delgado A. *Curr. Med. Chem.* 2010; 17: 2393-2418.
- a) Delgado A. *Eur. J. Org. Chem.* 2008; 3893-3906.
 b) Kelebekli L, Kara Y, Celik M. *Beilstein J. Org. Chem.* 2010; 6: No. 15.
 c) Harit VK, Ramesh NG. *J. Org. Chem.* 2016; 81: 11574–11586.
 d) Berecibar A, Grandjean C, Siriwardena A. *Chem. Rev.* 1999; 99: 779-844.
- a) Kapferer P, Birault V, Poisson J-F, Vasella A. *Helv. Chim. Acta* 2003; 86: 2210-2227.
 b) Ogawa S, Funayama S, Okazaki K, Ishizuka F, Sakata Y, Doi F. *Bioorg. Med. Chem. Lett.* 2004; 14: 5183-5188.
 c) Curti C, Zanardi F, Battistini L, Sartori A, Rassu G, Auzzas L, Roggio A, Pinna L, Casiraghi G. *J. Org. Chem.* 2006, 71; 225-230.
 d) Gravier-Pelletier C, Maton W, Dintinger T, Tellier C, Le Merrer Y. *Tetrahedron* 2003; 59: 8705-8720.
 - e) Rassu G, Auzzas L, Pinna L, Zambrano V, Zanardi F, Battistini L, Marzocchi L,

Acquotti D, Casiraghi G. J. Org. Chem. 2002; 67: 5338-5342.

7.

f) Vinader V, Haji-Abdullahi MH, Patterson LH, Afarinkia K. PLOS ONE 2013; 8: e82111.

g) Rassu G, Auzzas L, Zambrano V, Burreddu P, Pinna L, Battistini L, Zanardi F, Casiraghi G. *J. Org. Chem.* 2004; 69: 1625-1628.

h) Bwire RN, Majinda RR, Masesane IB, Steel PG. Pure. Appl. Chem. 2009; 81: 105-112.

i) Masesane, IB, Batsanov AS, Howard JAK, Mondal R, Steel PG. *Beilstein. J. Org. Chem.* 2006; 2: No. 9.

a) Di Bussolo V, Frau I, Favero, L, Uccello-Barretta G, Balzano F, Crotti P. *Tetrahedron* 2015; 71: 6276-6284.

b) Di Bussolo V, Fiasella A, Favero, L, Frau I, Crotti P. *Tetrahedron* 2013; 69: 2468-2478.

c) Di Bussolo V, Checchia L, Romano MR, Favero, L, Pineschi M, Crotti P. *Tetrahedron* 2010; 66: 689-697.

d) Di Bussolo V, Fiasella A, Frau I, Favero L, Crotti P. *Tetrahedron Lett* 2010; 51: 4937-4941.

e) Di Bussolo V, Checchia L, Romano MR, Pineschi M, Crotti P. Org. Lett 2008; 10: 2493-2496.

f) Di Bussolo V, Favero L, Romano MR, Pineschi M, Crotti P. *Tetrahedron* 2008; 64: 8188-8201.

g) Di Bussolo V, Fiasella A, Romano MR, Favero, L, Pineschi M, Crotti P. Org. Lett. 2007; 9: 4479-4482.

h) Di Bussolo V, Caselli M, Romano MR, Pineschi M, Crotti P. *J. Org. Chem* 2004; 69: 8702-8708.

i) Di Bussolo V, Caselli M, Romano MR, Pineschi M, Crotti P. *J. Org. Chem* 2004; 69: 7383-7386.

j) Di Bussolo V, Caselli M, Pineschi M, Crotti P. Org. Lett. 2003; 5: 2173-2176.

k) Di Bussolo V, Caselli M, Pineschi M, Crotti P. Org. Lett. 2002; 4: 3695-3698.

See also: 1) Crotti P, Di Bussolo V, Favero L, Macchia F, Pineschi M. *Tetrahedron* 2002; 58: 6069-6091.

8. a) Di Bussolo V, Frau I, Pineschi M, Crotti P. *Synthesis* 2012; 44: 2863-2871.

b) Di Bussolo V, Fiasella A, Favero L, Bertolini F, Crotti P. Org. Lett. 2009; 11: 2675-2678.

c) Di Bussolo V, Romano MR, Pineschi M, Crotti P. *Tetrahedron* 2007; 63: 2482-2489.
d) Di Bussolo V, Romano MR, Favero L, Pineschi M, Crotti P. *J. Org. Chem.* 2006; 71: 1696-1699.

e) Di Bussolo V, Romano MR, Pineschi M, Crotti P. Org. Lett. 2005; 7: 1299-1302.

- 9. a) Di Bussolo V, Frau I, Checchia L, Favero L, Pineschi M, Uccello-Barretta G, Balzano F, Roselli G, Renzi G, Crotti P. *Tetrahedron* 2011; 67:4696-4709 and references therein (racemic 1α and 1β).
 b) Frau I, Di Bussolo V, Favero L, Pineschi M, Crotti P. *Chirality* 2011; 23:820-826 (enantiopure (3*S*,4*R*,5*R*)-1α and (3*R*,4*S*,5*R*)-1β).
- 10. Protocol A reaction conditions: epoxide 1α (or 1β) or aziridine 9α -Ns (or 9β -Ns) is dissolved in the solvent/nucleophile (MeOH, AcOH, Et₂NH, as an example) in the presence of a catalyst, if necessary (H₂SO₄, TsOH or a Lewis acid). Under these conditions, the nucleophilic addition reaction occurs in the presence of a large excess of nucleophile, solvent of the reaction. *Protocol B* reaction conditions: epoxide 1α (or 1β) or aziridine 9α -Ns (or 9β -Ns) is dissolved in a non-nucleophilic solvent (CH₂Cl₂, MeCN, as an example) containing the nucleophile (MeOH, AcOH, azide anion, 3-6 equiv). Under these conditions, the nucleophilic addition reaction occurs in the presence of only a small excess of nucleophile.
- 11. Tenaglia A, Waegell B. *Tetrahedron Lett.* 1988; 29:4851-4854.
- 12. Azidolysis of epoxides 1β and 1α by the commonly used NaN₃/NH₄Cl in MeOH/H₂O protocol turned out to be not regioselective and the desired *anti-1,2-addition product* 12, from 1β (Scheme 3) and 16 from 1α (Scheme 4) were obtained in a mixture with the corresponding *anti-1,4-addition product* (80:20 and 55:45 *anti-1,2-/anti-1,4-addition product* ratio from 1β and 1α , respectively) (see ref. 9a).
- 13. In the presence of MeOH, 9β '-Ns-H: 9β "-Ns-H protonated aziridine conformers ratio is identical (about 99:1) to that present under neutral conditions between corresponding

aziridine conformers 9β '-Ns and 9β "-Ns (Schemes 5 and 6 and Supplementary data, sections 4.1 and 4.4).

a) Crotti P, Di Bussolo V, Favero L, Macchia F, Renzi G, Roselli G. *Tetrahedron* 2002; 58:7119-7133.
b) Crotti P, Favero L, Gardelli C, Macchia F, Pineschi M. *J. Org. Chem.* 1995; 60:2514-2525.
c) Eliel, E. L.; Allinger, N. L.; Angyal, S. J.; Morrison, G. A. In *Conformational Analysis*; Interscience: New York, 1965: pp 102, 296.
d) Fürst, A.; Plattner, P. A. *Abstract of Papers* 12th International Congress of Pure and Applied Chemistry, New York, 1951: p 409.

- 15. The stereoelectronic requirements of the conjugate addition (S_N2 '-type reaction) are satisfied by attack of the nucleophile at C(1)-carbon of C(1)-C(2) π system of vinyl aziridines **9\alpha-Ns** and **9\beta-Ns**, in either of the two possible orientation (*syn* or *anti*) with respect to allyl C(3)-N aziridine bond, acting as the leaving group. This reaction property justifies the reason why an absence of stereoselectivity is often observed in the formation of *1,4-addition products*. See: Atkinson R. S. In *Stereoselective Synthesis*; Wiley: Chichester, England, 1995, pp 110-111.
- 16. Battistini C, Crotti P, Damiani D, Macchia F. *J Org. Chem.* 1978; 44:1643-1647 and references therein.
- 17. Regioisomeric nucleophilic attack at C(4)-carbon in these three-membered vinyl heterocycles, as aziridines 9α -Ns and 9β -Ns and corresponding epoxides 1α and 1β , has never been observed.
- 18. It is worth noting that, compared with the corresponding aziridine conformers 9α'-Ns:9α"-Ns ratio under neutral conditions, an increase of corresponding conformer 9α"-Ns-H, with -CH₂OBn side chain equatorial, is observed under acid conditions (from about 29% to about 46-50%) (Schemes 5 and 7 and Supplementary data, sections 4.2, 4.5 and 4.6).
- 19. The carbagly cosylating ability of aziridine 9β -Ns (60%, entry 2, Table 2) and epoxide

 1α (57%, Scheme 1 and entry 2, Table 3), since obtained by *protocol B* reaction conditions (presence of only a small excess of nucleophile, 6 equiv), is more synthetically interesting than that obtained by *protocol A* (nucleophile as the solvent) with aziridine 9α -Ns (62%, entry 1, Table 1) and epoxide 1β (20%, Scheme 1 and entry 1, Table 4).

20. For the effect of ILs in favoring S_N1-type reaction pathways, see:
a) Creary X, Willis ED, Gagnon M. J. Am. Chem. Soc. 2005; 127:18114-18120 and references therein.
b) Sasaki K, Matsumura S, Toshima K. Tetrahedron Lett. 2004: 45:7043-7047.

21. The results obtained in the acid acetolysis of aziridine 9α -Ns and in particular the presence of *syn-1,2-addition product* 24 (entries 4 and 5, Table 1) can be rationalized as previously admitted in order to justify the results and the presence of a corresponding *syn-1,2-addition product* in the acetolysis of epoxide 1α (see ref. 9a).

- a) Grieco PA, Takigawa T, Bongers SL, Tanaka H. *J. Am. Chem. Soc.* 1980; 102: 7587-7588.
 b) Shanmugam P, Rajasingh P. *Tetrahedron* 2004; 60: 9283-9295 and references therein.
- Epoxides 38-41 appeared as possibly effective carbaglycosyl donors for the preparation of 4-deoxy-4-(*N*-nosylamino)-5*a*-carba-D,L-glycosides, as previously found with other related oxirane systems. See: Bordoni V, Porkolab V, Sattin S, Thépaut M, Frau I, Favero L, Crotti P, Bernardi A, Fieschi F, Di Bussolo V. *RSC Adv.* 2016; 6 : 89578-89584.
- 24. An appropriate conformational study carried out on methoxy derivatives 20 and 27, *anti-1,2-addition products* from aziridines 9α -Ns and 9β -Ns, respectively, has indicated that the corresponding conformers 20' and 27' with the side chain –CH₂OBn equatorial are highly favored and practically the only conformers present at the equilibrium (nearly 98-99%, Schemes 13 and 14).