

**EXPOSURE TO DI-2-ETHYLHEXYL PHTHALATE, DI-N-BUTYL PHTHALATE AND
BISPHENOL A THROUGH INFANT FORMULAE**

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Running title: DEHP, DnBP and BPA in Infant Formulae

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Abstract

Background: Phthalates and Bisphenol A (BPA) are ubiquitous contaminants identified as endocrine disruptors. Phthalates are worldwide used as plasticizers, in particular to improve the mechanical properties of polymers such as polyvinyl chloride. Since they are not chemically bound to the polymer, they tend to leach out with time and use. Di-2-ethylhexyl phthalate (DEHP) and di-n-butyl phthalate (DnBP) are two most common phthalates. BPA is an estrogenic compound used to manufacture polycarbonate containers for food and drink, including baby bottles. It can migrate from container into foods, especially at elevated temperatures. Diet is a predominant source of exposure for phthalates and BPA, especially for infants.

Objective: to test the presence of DEHP, DnBP and BPA in infant formulae.

Methods: DEHP, DnBP and BPA concentrations were measured in 22 liquid and 28 powder milks by gas chromatography with flame ionization detection and high performance liquid chromatography with fluorimetric detection respectively.

Results: DEHP concentrations in our samples were between 0.005 and 5.088 $\mu\text{g/g}$ (median 0.906 $\mu\text{g/g}$), DnBP concentrations were between 0.008 and 1.297 $\mu\text{g/g}$ (median 0.053 $\mu\text{g/g}$), and BPA concentrations were between 0.003 and 0.375 $\mu\text{g/g}$ (median 0.015 $\mu\text{g/g}$). Concentrations of the investigated contaminants in liquid and powder milks were not significantly different, even though samples were packed in different types of containers.

Conclusions: These data point out potential hazards for infants fed with baby formulae. Contamination seems more related to the production of formulae than to a release from containers.

Key Words

Di-2-ethylhexyl phthalate, di-n-butyl phthalate, bisphenol A, infant formula.

1 **Introduction**

2 Endocrine Disruptors (EDs) are chemicals known to mimic steroid hormones' action and to
3 interfere with the synthesis, secretion, transport, activity or elimination of natural hormones
4 [Cobellis et al. 2003, Jenkins et al. 2009, Habert et al. 2009, Latini et al. 2006, Latini et al. 2010]. In
5 particular, they modify the programming of the normal endocrine-signaling pathways during pre-
6 and early post-natal life, thus determining adverse health effects such as neurological and immune
7 effects, reproductive disorders, cancers, lowered fertility and increased incidence of endometriosis
8 [Cobellis et al. 2003, Jenkins et al. 2009, Habert et al. 2009, Latini et al. 2006, Latini et al. 2010].
9 Recent papers show that EDs pose the greatest risk during prenatal and early postnatal
10 development, when organ and neural systems are forming [Jenkins et al. 2009, Habert et al. 2009].
11 The possible relationships between combined exposures to environmental contaminants and
12 diseases are now attracting attention, especially if they occur early in life [Sathyanarayana et al.
13 2013, Wang et al. 2014]. Recently, some studies correlated the combined exposure to phthalates and
14 BPA with human health risks [Sathyanarayana et al. 2013, Wang et al. 2014].
15 Phthalates are widely used in many products to impart softness, flexibility, transparency and
16 longevity to an otherwise rigid polyvinyl chloride (PVC). Since there is not a chemical bond with
17 the polymer, they leach out with time and use, thus becoming ubiquitous environmental
18 contaminants [Latini 2005]. Di-2-ethylhexyl phthalate (DEHP) and di-n-butyl phthalate (DnBP) are
19 two of the most common phthalates [Latini 2005]. Human exposure occurs through ingestion,
20 inhalation, and dermal contact during the whole lifetime, including intrauterine life, but exposure in
21 children exceeds that in adults. Phthalates determine toxic effects in laboratory animals, especially
22 on the developmental and reproductive systems [Sun et al. 2006]. Human studies correlated
23 phthalate exposure with adverse health effects such as liver, kidney and lung damage as well as
24 sexual developmental abnormalities [Cobellis et al. 2003, Latini et al. 2006a, Latini et al. 2006b,
25 Yavaşoğlu et al. 2012, Lovekamp-Swan and Davis 2003 1, 5, 10-12]. Moreover, phthalates may alter
26 the methylation status of DNA and consequently the DNA sequence itself, thus transmitting these

27 effects to future generations [Singh and Li 2012]. Bisphenol A (BPA), 2,2-bis(4-hydroxyphenyl)
28 propane, is at the same time an estrogenic compound and a main monomer for the synthesis of
29 polycarbonate and epoxy resins. Polycarbonate is used for many products like water and baby
30 bottles, children's toys, sport equipment, medical and dental devices etc., whereas coatings of many
31 food and beverage containers consist of epoxy resins [Jenkins et al. 2009, Latini et al. 2005]. BPA
32 tends to migrate from cans containers into foods, especially at elevated temperatures [Jenkins et al.
33 2009, Oldring et al. 2014]. As a consequence, potential risks of exposure to BPA raised concern
34 over the years due to suspicion to affect reproduction, development, and metabolism. There is a
35 consensus that infants are at the greatest risk of harm, even with a low level exposure to BPA
36 [Jenkins et al. 2009]. Recent studies of National Toxicology Program (NTP) and US Food and Drug
37 Administration (FDA) pointed out the potential BPA effects on brain, behavior, and prostate gland
38 in fetuses, infants, and young children [FDA 2010]. Indeed, BPA can affect the hormone-mediated
39 neurologic and behavioral development in early life [Bashore et al. 2001, Chapin et al. 2008, FDA
40 2010, Hengstler et al. 2011, Vom saal and Hughes 2005]. In addition, high BPA exposure has been
41 associated with heart disease, diabetes, abnormally high levels of liver enzymes, and alterations of
42 the thyroid function [Belcher et al. 2012, Rubin 2011, Sriphrapadang et al. 2011]. For these
43 reasons, BPA containing baby bottles have been banned in Europe since March 2011 [Commission
44 Directive 2011/08/EU].

45 Diet remains the predominant source of exposure for both phthalates and BPA especially for
46 infants, since these compounds have been found in breast milk and in baby formulae [Cirillo et al.
47 2011, Cirillo et al. 2013, Latini et al. 2004, Latini et al. 2009]. The present paper analyzed the
48 presence of DEHP, DnBP and BPA in infant formulae to assess possible neonatal exposure and
49 reduce the gap of knowledge in this field.

50

51 **MATERIALS AND METHODS**

52 **Sampling**

53 Fifty infant formula samples were collected at different neonatal nurseries in Naples hospital during
54 three months (May-July 2013). Liquid ready to use (n=22) and powder (n=28) milk samples were
55 collected. Among them, there were 7 special milk samples, i.e. milks for infant with gastrointestinal
56 problems (n=3), rice milk formulae (n=3) and a premature formula. Liquid samples were packed in
57 polyethylene terephthalate (PET) and Tetrapak™, whereas milk powders were contained in
58 aluminum (Al) containers.

59 The infant formula samples were collected in glass vials and rapidly transferred to the laboratory of
60 the Department of Agriculture, where analytical samples were obtained for the different procedures.

61 All samples were labelled. For DEHP and DnBP analysis, aliquots (15 mL) of liquid milk were
62 lyophilized and stored at -18°C until analyses, whereas powder sample aliquots (1 g) were just
63 stored in the dark. For BPA determination, aliquots (5 mL) of liquid milk were stored at -18°C until
64 analyses, whereas powder samples aliquots (2 g) were reconstituted with HPLC water (15 mL), split
65 into 5 mL aliquots and stored at -18°C until analyses. For each reconstituted vial, an additional 5
66 mL vial with HPLC water was stored at -18°C as negative control to avoid possible bias due to a
67 contamination of HPLC water.

68

69 *DnBP and DEHP*

70 **Chemical reagents**

71 Acetonitrile, n-hexane, acetone for organic trace analysis and anhydrous Na₂SO₄ were supplied by
72 Merck (Darmstadt, Germany). Florisil (60/100 mesh) was furnished by Supelco (Bellefonte, PA,
73 USA), and Bondesil (PSA 40UM) by Varian (Palo Alto, CA, USA). Standard solutions of DnBP
74 and DEHP were purchased from Sigma Aldrich (St. Louis, MO, USA).

75

76 **Instrumental parameters**

77 The analyses of phthalates (PAEs, Phthalic Acid Esters) were carried out by a Shimadzu GC-17
78 (Shimadzu, Kyoto, Japan) capillary gas chromatography with a Flame Ionization Detector (GC-
79 FID) and an HP-5 (Crosslinked 5% PHME Siloxane, 30 m length, 0.32 i.d., 0.25 μm film thickness)
80 glass-capillary column. Helium was used as carrier and a hydrogen/air mixture was used to sustain
81 the flame. The volume of injection was 1 μl in splitless mode, the injector and detector temperatures
82 were 260°C and 310°C respectively. The temperature program was 100°C for 1 min, increase of
83 15°C/min up to 280°C, retention of this temperature for 10 min.

84

85

86 **DnBP and DEHP measurement**

87 Because of PAE ubiquity, any contact with plastic was avoided. All the glassware was thoroughly
88 washed, rinsed twice with acetone and n-hexane, heated at 250°C for 2 h and finally stored away
89 from any environmental contamination.

90 In accordance with the method by Cirillo et al. 2013, the lyophilized samples were: 1) extracted
91 three times with 15 mL of acetonitrile in an ultrasound bath for 15 min, 2) centrifuged at 2000 rpm
92 for 10 min and the acetonitrile layer transferred to a separatory funnel, 3) added with 10 mL of n-
93 hexane saturated with acetonitrile and the funnel was vigorously shaken for 5 min. The acetonitrile
94 phase was transferred into a flask and dried under vacuum at 55°C. The extracts were reconstituted
95 by 5 mL of n-hexane and purified through a column containing 2 g of Florisil activated for 2 hours
96 at 200°C, 0.5 g of Bondesil and 1 g of anhydrous Na_2SO_4 . The column was eluted three times with
97 10 mL of n-hexane/acetone mixture (100:5 v/v). The eluates were collected in a flask, evaporated
98 under vacuum at 40°C and reconstituted with 1 mL of n-hexane for GC analysis.

99 The calibration curves were obtained using standard solutions at 0.625, 1.250 and 2.500, 5.00 and
100 10.00 $\mu\text{g}/\text{mL}$ for DEHP, and at 0.312, 0.625, 1.250, 2.500 and 5.00 $\mu\text{g}/\text{mL}$ for DnBP. The
101 regression coefficients (R) were >0.99 for both contaminants. The PAE concentrations in the

102 samples were obtained by comparing the relevant peak areas with calibration curve.

103 Limits of Detection (LODs) and Quantification (LOQs) were evaluated as the mean blank value
104 plus three blank standard deviations and three times the LOD. LODs and LOQs were 5.0 ng/g and
105 15.0 ng/g for DEHP, and 7.5 ng/g and 22.5 ng/g for DnBP respectively.

106 A run without sample was carried out every six determinations to reduce the instrumental
107 background due to contamination. Moreover, solvents used to wash the syringe were frequently
108 replaced.

109 The intra- and inter-day repeatability of the method were evaluated by injecting standard solutions
110 at three different concentration levels (2.50, 5.00 and 10.00 µg/mL for DEHP and 1.25, 2.50 and
111 5.00 µg/mL for DnBP) five times during a day (intra-day) and during five consecutive days (inter-
112 day). The intra-day repeatability ranged from 7.0 to 9.5% for DEHP and from 5.5 to 8.5% for
113 DnBP, whereas inter-day repeatability varied from 6.0 to 8.5% for DEHP and from 4.5 to 6.5% for
114 DnBP.

115 Samples with DEHP and DnBP concentrations lower than LOD were used for recovery tests. Three
116 liquid and three powder milk samples (each in triplicate) were spiked with standard solutions at
117 concentration 2.0, 4.0 and 8.0 µg/mL for DEHP and 1.0, 2.0 and 4.0 µg/mL for DnBP, and then
118 processed as milk samples. Recoveries were for 98 ± 10 % for DEHP and 98 ± 9 % for DnBP.

119 Because of the ubiquity of these compounds, a blank sample (only solvents) for each batch was
120 analysed and the average concentration value was subtracted from PAE detected values.

121

122 *Bisphenol A*

123 **Chemical reagents**

124 Acetonitrile, methanol and water (HPLC grade) were supplied by Merck (Darmstadt, Germany).

125 Solid phase extraction cartridges (Bond Elut C18 SPE, 1g/6mL) were purchased from Agilent

126 Technologies (Palo Alto, CA, USA). A BPA standard (purity $\geq 99\%$) was purchased from Sigma

127 Aldrich (St. Louis, MO, USA).

128

129 **Instrumental parameters**

130 BPA detection was performed through an HPLC (LC-10AT VP Shimadzu, Kyoto, Japan) equipped
131 with a fluorescence detector (Shimadzu RF-10A XL) and a reversed-phase column (Ascentis C18.
132 L × I.D.: 15 cm × 4.6 mm; particle size: 5 μm, Supelco, Bellefonte, PA). The column was kept at a
133 constant temperature of 40°C. The mobile phase consisted of 60% of acidified water (1% of acetic
134 acid), 35% of acetonitrile and 5% of methanol. The flow rate of mobile phase was set at 0.950
135 mL/min (isocratic run). The fluorimetric detection was carried out at an excitation wavelength of
136 275 nm and an emission wavelength of 305 nm.

137

138 **BPA measurement**

139 BPA measurement was performed by adapting the procedure by Sun et al. 2006. An aliquot of each
140 sample (5 mL) was inserted into a 250 mL glass round-bottom flask and added with acetonitrile (20
141 mL). Flasks were placed onto a Heidolph Promax 2020 shaker for 25 min. The content of each flask
142 was then filtered through a filter paper and transferred into a separatory funnel. The flask was rinsed
143 with 5 mL of acetonitrile, which were added to the funnel. Afterwards, 35mL of n-hexane were also
144 added to the separatory funnel, and the resulting mixture was shaken for 25 min. The acetonitrile
145 layer was removed from the funnel and stored in a round-bottom flask, whereas the hexane layer
146 was washed twice with acetonitrile (firstly with 15 mL, then with 10 mL) which was collected and
147 added in the same round-bottom flask. The solvent was removed from the extract through a
148 rotavapor, then the flask was washed with 3 mL of a methanol: water (5:95 v/v) solution to be
149 processed by solid phase extraction. SPE cartridges were firstly conditioned with 5 mL of methanol
150 and then with 5 mL of water. Later the sample was loaded, and the elution was carried out at a flow
151 rate of 3-4 mL/min using a Supelco Visiprep SPE vacuum manifold. The cartridges were then

152 washed with 2 mL of a methanol:water solution (30:70 v/v) and dried under vacuum pump for 1
153 min. Finally, the BPA retained in the cartridge was eluted with 3 mL of a methanol:water (80:20
154 v/v) solution. The eluate was dried by a rotavapor, dissolved with 1 mL of methanol and finally
155 collected in an amber vial before the HPLC analysis.

156 A calibration curve with a correlation coefficient of 0.998 was obtained by injecting standard
157 solutions of BPA at concentrations 10.0, 20.0, 30.0, 40.0 and 50.0 $\mu\text{g/L}$. An instrumental LOD
158 equal to 0,003 $\mu\text{g/g}$ dry weight (dw) was calculated using the standard deviation of the response (σ)
159 and the slope of the calibration curve (S) according to the formula: $3.3 \sigma/S$. Similarly, a LOQ equal
160 to 0,009 $\mu\text{g/g}$ dw was calculated as: $10 \sigma/S$.

161 Recovery percentages at three concentration levels were assessed on six samples (3 liquid and 3
162 powder milk samples with BPA level below the LOD) by spiking each sample with BPA solutions
163 in methanol at concentrations 50.0, 100.0 and 1000.0 $\mu\text{g/L}$. The recoveries were $87 \pm 3\%$. BPA
164 quantification was performed comparing the peak areas obtained in the samples with the BPA
165 standard calibration curve.

166 For each batch of samples, a blank sample was processed according to the procedures mentioned
167 before. A total of 16 blanks were analyzed and all of them showed BPA concentrations well below
168 the LOD value.

169

170 **BPA confirmation by LC MS/MS**

171 Since BPA measurements could be affected by matrix related interferences, a confirmation by LC
172 MS/MS was carried out according to the Shao et al. 2005 method.

173 **Instrumental parameters**

174 Identification was carried out using an alliance 2695 (Waters, USA) liquid chromatography
175 equipped with a Quattro Ultima Pt (Micromass, UK) tandem mass spectrometer and a symmetry C-
176 18 column (150mm \times 2.1mm i.d., 3.5m). The temperature of the column oven was set at 40 $^{\circ}\text{C}$, the

177 flow rate was 0.2 mL/min and the injection volume was 10 μ L. Mobile phases consisted of
178 methanol and water with 0.1% ammonia. The methanol was linearly increased from 10 to 55% in
179 10 min, then increased to 85% in 10 min and held for 7.5 min, finally brought back to 10% and held
180 for 15 min before the following injection. The mass spectrometer was operated in negative mode
181 electrospray ionization in multiple-reaction monitoring (MRM) mode. The capillary voltage was 3.5
182 kV, the cone voltage was 70V and the multiplier voltage was 650V. Nitrogen was used as
183 nebulizing, desolvation and cone gas. In particular, the nebulizing gas was adjusted to the
184 maximum, whereas the flow of the desolvation gas and cone gas were set to 550 L/h and 80 L/h
185 respectively. The source temperature and the desolvation gas temperature were held at 100 and 300
186 $^{\circ}$ C respectively. The RF lens 1 and 2 were set at 50 and 0.5, the ion energy 1 and ion energy 2 were
187 both 0.5, the entrance and exit were zero, the collision gradient was 3.2 eV. UHP argon was used as
188 the collision gas for the tandem mass spectrometric analysis, and the pressure in the collision
189 chamber was kept at 2.8×10^{-3} mbar.

190 A calibration curve in the concentration range 1 to 100 ng/g was obtained by linear regression of the
191 normalized (to the internal standard area) standard solution areas against BPA concentrations. The
192 correlation coefficient was ≥ 0.999 .

193 The intra- and inter-day repeatability of the method were evaluated by injecting standard solutions
194 at three different concentration levels (10, 50 and 100 ng/g) five times during a day (intra-day) and
195 during five consecutive days (inter-day). The intra-day reproducibility ranged from 4.0 to 6.5%,
196 while inter-day reproducibility varied from 4.5 to 6.2%.

197 **Statistical analysis**

198 A power calculation was undertaken to determine an appropriate sample size for this study. Based
199 on literature data [MAFF 1998], considering DEHP as the most abundant phthalate in infant
200 formula, a two-sided test power calculation was performed. Two double of the range value was
201 used as the sigma (0,780 μ g/g dry weight). This power calculation indicated that 11 samples in each

202 group would be necessary to detect a 15% difference in the DEHP concentration with a power of
203 80% at a 5% level of significance.

204 Data distribution was assessed with the Shapiro Wilk's test. One-way ANOVA was performed with
205 SPSS 20.0 software (IBM) to assess the differences between DEHP, DnBP and BPA concentrations
206 in liquid and powder milks. Significance was set at $p < 0.05$. The concentrations below LOD were
207 assumed to be equal to LOD.

208

209 **Dietary intake assessment for Italian infants (age 0–4 months)**

210 Daily intake was estimated as:

$$211 \text{ Intake} = (C \text{ concentration} \times V \text{ volume of milk per day}) / BW \text{ body weight} \quad (1)$$

212 to evaluate DEHP, DnBP and BPA exposure of young children through artificial milk.

213 Dietary exposure was calculated using the blueprint to the budget method (BM) model [WHO
214 2009] in accordance with FAO/WHO, and with the help of weight growth charts by WHO 2006 and
215 pediatric nutrition suggestions for our range of age. We considered two possible scenarios: 1)
216 median concentrations of contaminants, infants with average weight to development at the 50th
217 percentile or at the 95-97th percentile (according to the growth curve by WHO (2006)) who
218 introduce daily a medium quantity of milk (medium case); 2) maximum concentrations of
219 contaminants, children who have grown at the 50th percentile or at the 95-97th percentile and
220 introduce daily a higher quantity of milk (worst case).

221

222 **RESULTS**

223 Most milk samples showed detectable levels of DEHP (92%, 86% of liquid and 96% of powder
224 milks), DnBP (90%, 82% of liquid milks and 96% of powder milks) and BPA (58%, 52% of liquids
225 milks and 67% of powder milks) (**Table 1**).

226 The average concentration of DEHP in all milk samples was $1.327 \pm 0.724 \mu\text{g/g dw}$, and in
227 particular it was $1.112 \pm 0.716 \mu\text{g/g dw}$ in liquid milks and $1.496 \pm 0.729 \mu\text{g/g dw}$ in powder milks.

228 For DnBP, the average concentration in all milk samples was $0.354 \pm 0.305 \mu\text{g/g dw}$, namely 0.384
229 $\pm 0.385 \mu\text{g/g dw}$ in liquid milks and $0.330 \pm 0.229 \mu\text{g/g dw}$ in powder milks. The average
230 concentration of BPA in all milk samples was $0.021 \pm 0.022 \mu\text{g/g dw}$, it was $0.019 \pm 0.037 \mu\text{g/g dw}$
231 in liquid milks and $0.023 \pm 0.028 \mu\text{g/g dw}$ in powder milks (**Table 1**).

232 DEHP concentrations varied from 0.092 to $3.552 \mu\text{g/g}$ (median= $1.136 \mu\text{g/g}$), DnBP concentrations
233 from 0.008 to $1.624 \mu\text{g/g}$ (median= $0.244 \mu\text{g/g}$) and BPA concentrations from 0.003 to $0.169 \mu\text{g/g}$
234 (median= $0.008 \mu\text{g/g}$) (**Table 1**).

235 Similar concentrations of the three analytes were found in liquid and powder milks, even though
236 containers were of different types (**Figure 1**). DEHP, DnBP and BPA concentrations in the HLPC
237 water samples stored as negative controls for reconstituted powder milk were below the LODs.

238 The concentration of DEHP, DnBP and BPA in liquid and powder milks together with the type of
239 packaging are reported in **Table 2** and **3** respectively.

240 Estimates of dietary exposure to DEHP, DnBP and BPA in the medium and worst case are shown in
241 **Table 4** and **5**. The daily intake of DEHP in the medium case ranged from 19.84 to $24.85 \mu\text{g/kg bw}$
242 day at 50th percentile and from 17.63 to $19.14 \mu\text{g/kg bw}$ day at 97th percentile. In the worst case,
243 DEHP intake varied between 42.57 at $54.68 \mu\text{g/kg bw}$ day at the 50th percentile and between 38.80
244 at $46.52 \mu\text{g/kg bw}$ day at the 97th percentile (**Table 4-5**). Estimates of dietary exposure to DnBP in
245 the medium case ranged from $4.15 \mu\text{g/kg bw}$ day to $5.34 \mu\text{g/kg bw}$ day at the 50th percentile and
246 from $3.79 \mu\text{g/kg bw}$ day to $4.54 \mu\text{g/kg bw}$ day at the 97th percentile. In the worst case, the DnBP
247 intake varied between 13.62 and $17.50 \mu\text{g/kg bw}$ day at the 50th percentile and between 12.41 and
248 $14.89 \mu\text{g/kg bw}$ day at the 97th percentile (**Table 4** and **5**). BPA intake in the medium and worst
249 case are shown in **Table 4** and **5**. In the medium case, values ranged from 0.14 to $0.17 \mu\text{g/kg bw}$
250 day at the 50th percentile and from 0.12 to $0.15 \mu\text{g/kg bw}$ day at the 97th percentile. In the worst
251 case, the BPA intake varied between 0.99 and $1.27 \mu\text{g/kg bw}$ day at the 50th percentile and between
252 0.90 and $1.08 \mu\text{g/kg bw}$ day at the 97th percentile. Both in the medium and worst case the highest
253 intake occurred at the 30th day of life, because the amount of consumed milk starts increasing while

254 baby's weight is still pretty low. As for BPA, for both DnBP and DEHP the higher values of intake
255 occurred in children at 30 days of age (**Tables 4-5**).

256 **DISCUSSION**

257 Our data indicate the presence of DEHP, DnBP and BPA in infant formulae. Data relevant to all
258 contaminants showed a wide variability but we found no significant concentration differences for
259 the investigated contaminants between liquid and powder milks, even though samples were packed
260 in different types of containers. This finding would suggest the DEHP, DnBP and BPA
261 contamination to arise from raw materials or manufacturing processes rather than from packaging.
262 Phthalates, in particular, may contaminate milk during the production or preparation of formulae. A
263 main source of contamination results from migration of phthalates from products in contact with
264 food during processing. A number of studies concerned the migration of DEHP from the PVC
265 tubing of the milking machine used in dairy farms [Castle et al. 1990, Feng et al. 2005, Ruuska
266 1987]. PVC tubing contains up to 40% DEHP by weight. A Norwegian study showed a clear
267 difference in DEHP levels between raw milk collected by hand milking (about 5 µg/kg) and
268 machine milking involving PVC tubing (30 µg/kg in milking chamber and 50 µg/kg in collection
269 tank) [Feng et al. 2005].

270

271 **Dietary intake assessment for Italian infants (age 0–4 months)**

272 In order to assess post-natal exposure to phthalates and BPA, the estimation of daily dietary intake
273 of these contaminants was carried out in 0-4 month old children, as milk is the only food introduced
274 in this age group.

275 Four possible nutrition scenarios were possible, namely nutrition with infant powder, liquid
276 formula, breast milk or a combination of these, but we only considered artificially fed babies
277 assuming liquid or powder formulas (or both). The European Food Safety Authority (EFSA)
278 established a Tolerable Daily Intake (TDI) of 50 µg/kg bw for DEHP and 10 µg/kg bw for DnBP
279 [EFSA, 2005a; 2005b]. As expected, the highest intakes of DEHP and DnBP were estimated among
280 infants with growth at the 50th percentile, who have a lower body weight than those at the 97th
281 percentile.

282 Daily intake of DEHP in the medium case varied between 20-25% and 18-21% of TDI at 50th and
283 97th percentile respectively. In the worst case, intake was also lower than TDI, except for the 50th
284 percentile infants aged 30 and 45 days (**Table 5**).

285 Daily intake of DnBP in the medium case varied between 42-53% and 38-45% of TDI at 50th and
286 97th percentile respectively. In the worst case, instead, intake always exceeded TDI, up to 175%.

287 Muller et al. 2003 estimated for 0-6 month old Danish infants a daily intake via infant formulae of
288 9.8 µg/kg bw/day for DEHP and 16.4 µg/kg bw/day for DnBP [Muller et al. 2003]. Our values for
289 DEHP intake were higher than Muller's both in the medium and worst case, whereas DnBP intake
290 levels were lower in the medium case and similar in the worst case. Our estimates of DnBP and
291 DEHP daily intake were higher than those reported by MAFF (1998) for infants 0-3 months old, i.e.
292 13 µg/kg bw/day for DEHP and 2.4 µg/kg bw/day for DnBP.

293 Our estimated BPA daily intakes were well below the temporary Tolerable Daily Intake (t-TDI)
294 established by EFSA in 2014 (5.0 µg/kg bw) [40]. In the medium case, our intakes ranged 2.8-3.4%
295 and 2.4-3.0% of t-TDI for the 50th and 97th percentile respectively, which increased in the worst
296 case to 20-25% and 18-22% of t-TDI for the 50th and 97th percentile respectively.

297 EFSA t-TDI for BPA refers to the adult population, and there isn't a specific TDI for children or
298 infants. Diet is the main source of exposure to BPA in infants aged 0-4 months [EFSA 2014]. Minor
299 pathways of introduction could be the inhalation or ingestion of dust, the dermal contact and the
300 mouthing of toys. Until a few years ago, babies could introduce BPA from polycarbonate baby
301 bottles, especially when bottles were heated and reused multiple times [Jenkins et al. 2009, Wang et
302 al. 2014]. The EU Regulation No. 321/2011 imposed not to use BPA in the manufacture of baby
303 bottles, thus reducing exposure.

304 In 2008, a report of the U.S. National Toxicology Program (NTP) provided daily exposure
305 estimates for infants, children and adults based on realistic scenarios [NTP 2008]. The highest daily
306 exposure to BPA was estimated to occur in infants and children. Formula-fed infants (0-6 months of
307 age) had estimated daily intakes of 1-11 µg/kg bw.

308 In 2010, the FAO and WHO jointly held an Expert Meeting on BPA, whose final report was
309 published in 2011 [FAO/WHO 2011]. The report identified 0-6 month infants fed with liquid
310 formulae in polycarbonate bottles as the sub-population with the highest dietary exposure to BPA,
311 namely 2.4 µg/kg bw per day (average) and 4.5 µg BPA/kg bw per day (95th percentile).

312 In 2012, a probabilistic exposure assessment using data from recent Canadian surveys suggested
313 that daily exposure to BPA in children ranged from 0.083 µg/kg bw (0-1 month old) to 0.164 µg/kg
314 bw (children 4-7 months of age) [Health Canada 2012].

315 Our data resemble those of Health Canada but are lower than those of NTP and FAO/WHO,
316 probably because the problem of BPA migration from baby bottles in Europe has been solved.

317 The different packages (Tetrapack™, PET and aluminum) represent a possible bias of the present
318 study. However, the studied contaminants can be found not only in Tetrapack™ and PET but also in
319 aluminum packages, as these are often internally coated with plastic derivatives.

320 **CONCLUSIONS**

321 Our data show a widespread contamination of infant formulae from the three investigated
322 contaminants, either of environmental or process origin. Our findings demonstrate that infant
323 formulae may represent a main source for the simultaneous exposure to DEHP, DnBP and BPA in
324 babies. This risk is particularly relevant for DEHP and DnBP because intake from formulated milk
325 could exceed in the worst case the TDI from EFSA. In conclusion, potential hazards exist for
326 infants fed with baby formula, as these endocrine disruptors show the highest toxicity in infant
327 population. EFSA established TDIs for the three investigated contaminants only referring to an
328 adult population. We believe that specific TDIs for children would help the protection of the most
329 vulnerable part of the population from a severe public health hazard.

330

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Table 1. DEHP, DnBP and BPA concentrations in $\mu\text{g/g}$ dry weight (mean \pm sd, median and range).

Sample	DEHP				DnBP				BPA			
	POS (%)	Mean \pm sd	Median	Min-max	POS (%)	Mean \pm sd	Median	Min-max	POS (%)	Mean \pm sd	Median	Min-max
Liquid Milk (n = 22)	86	1.112 \pm 0.716	0.926	0.092 – 2.919	82	0.384 \pm 0.385	0.280	0.008 – 1.624	43	0.019 \pm 0.037	0.003	0.003 – 0.169
Powder Milk (n = 28)	96	1.496 \pm 0.729	1.159	0.702 – 3.552	96	0.330 \pm 0.229	0.212	0.101 – 0.812	67	0.023 \pm 0.028	0.011	0.003 - 0.108
Total (n = 50)	80	1.327 \pm 0.724	1.136	0.092 – 3.552	90	0.354 \pm 0.305	0.244	0.008 – 1.624	60	0.021 \pm 0.022	0.008	0.003 - 0.169

Table 2. Concentrations of Bisphenol A (BPA), di-n-butylphthalate (DnBP) and di(2-ethylhexyl)phthalate (DEHP) in liquid milk samples and type of packaging.

Product	Type	Packaging	DEHP (µg/g dry weight)	DnBP (µg/g dry weight)	BPA (µg/g dry weight)
C1	Infant formula	Tetrapak™	0.696	0.075	0.003
C2	Infant formula	PET	0.092	0.082	0.030
C3	Infant formula	Tetrapak™	1.831	0.067	0.003
C4	Infant formula	Tetrapak™	0.219	0.084	0.020
C6	Infant formula	Tetrapak™	0.633	0.142	0.009
C7	Infant formula	Tetrapak™	2.067	0.0075	0.003
C9	Infant formula	PET	1.456	0.287	0.003
C10	Infant formula	PET	0.301	0.067	0.003
C11	Infant formula	PET	2.099	0.624	0.003
C12	Infant formula	PET	0.784	0.482	0.058
C13	Infant formula	PET	0.606	0.216	0.014
C17	Infant formula	PET	1.877	0.787	0.003
C18	Infant formula	PET	1.202	0.351	0.003
C19	Infant formula	PET	0.923	0.899	0.017
C20	Infant formula	PET	0.256	0.14	0.018
C21	Infant formula	PET	0.929	0.088	0.003
C22	Infant formula	PET	2.919	1.624	0.169
C23	Infant formula	PET	0.852	0.423	0.030
C24	Infant formula	PET	1.428	0.384	0.003
C25	Infant formula	PET	0.796	0.807	0.003
C26	Infant formula	PET	1.137	0.272	0.003
C27	Infant formula	PET	1.371	0.548	0.003

Table 3. Concentrations of Bisphenol A (BPA), di-n-butylphthalate (DnBP) and di(2-ethylhexyl)phthalate (DEHP) in powder milk samples and type of packaging.

Product	Type	Packaging	DEHP (µg/g dry weight)	DnBP (µg/g dry weight)	BPA (µg/g dry weight)
C5	Infant formula	Alluminium	1.408	0.321	0.003
C8	Infant formula	Alluminium	1.134	0.199	0.003
C14	Premature formula	Alluminium	0.997	0.201	0.003
C15	Infant formula	Alluminium	0.702	0.155	0.003
C16	Infant formula	Alluminium	0.871	0.212	0.028
C28	Infant formula	Alluminium	1.274	0.161	0.008
C29	Infant formula	Alluminium	0.883	0.137	0.003
C30	Infant formula	Alluminium	3.552	0.809	0.011
C31	Infant formula	Alluminium	2.909	0.765	0.100
C32	Infant formula	Alluminium	1.023	0.101	0.009
C33	Infant formula	Alluminium	1.142	0.356	0.022
C34	Infant formula	Alluminium	0.981	0.392	0.003
C35	Infant formula	Alluminium	1.024	0.161	0.003
C36	Infant formula	Alluminium	0.922	0.337	0.043
C37	Infant formula	Alluminium	1.052	0.709	0.054
C38	Infant formula	Alluminium	1.018	0.575	0.026
C39	Infant formula	Alluminium	2.341	0.187	0.012
C40	Infant formula	Alluminium	0.982	0.123	0.003
C41	Infant formula	Alluminium	1.723	0.118	0.016
C42	Infant formula	Alluminium	1.899	0.704	0.003
C43	Infant formula	Alluminium	1.175	0.148	0.035
C44	Infant formula for gastrointestinal problems	Alluminium	1.213	0.301	0.003
C45	Infant formula for gastrointestinal problems	Alluminium	1.897	0.812	0.108
C46	Rice milk formula	Alluminium	1.723	0.211	0.003
C47	Rice milk formula	Alluminium	2.871	0.321	0.003
C48	Rice milk formula	Alluminium	0.951	0.184	0.046
C49	Infant formula for gastrointestinal problems	Alluminium	1.821	0.201	0.018
C50	Infant formula	Alluminium	2.409	0.349	0.041

Table 4. Medium case, estimated daily dietary intake of Bisphenol A (BPA), di-n-butylphthalate (DnBP) and di(2-ethylhexyl)phthalate (DEHP) in newborns fed with liquid or powder formulae according to the 50th and 97th of infant weight growth curve by WHO (2006).

Age (days)	Infant's average weight		Milk assumption		DEHP intake		DnBP intake		BPA intake	
	(kg)		(g dry weight / day)		(µg/kg bw day)		(µg/kg bw day)		(µg/kg bw ² day)	
	50 th pctl ¹	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl
15	3.70	4.75	67.61	76.06	20.78	18.21	4.46	3.91	0.15	0.13
30	4.25	5.45	92.96	101.41	24.85	21.14	5.34	4.54	0.17	0.15
45	4.76	6.20	101.41	109.86	24.23	20.15	5.20	4.33	0.17	0.14
60	5.41	6.84	98.59	105.63	20.72	17.54	4.45	3.77	0.15	0.12
75	5.76	7.26	105.63	112.68	20.83	17.64	4.47	3.79	0.15	0.12
90	6.10	7.65	112.68	126.76	20.98	18.82	4.51	4.04	0.15	0.13
120	6.70	8.35	114.08	129.58	19.34	17.63	4.15	3.79	0.14	0.12

¹pctl = percentile

²kg bw = kg body weight

Table 5. Worst case, estimated daily dietary intake of Bisphenol A (BPA), di-n-butylphthalate (DnBP) and di(2-ethylhexyl)phthalate (DEHP) in newborns fed with liquid or powder formulae, according to the 50th and 97th of infant weight growth curve by WHO (2006).

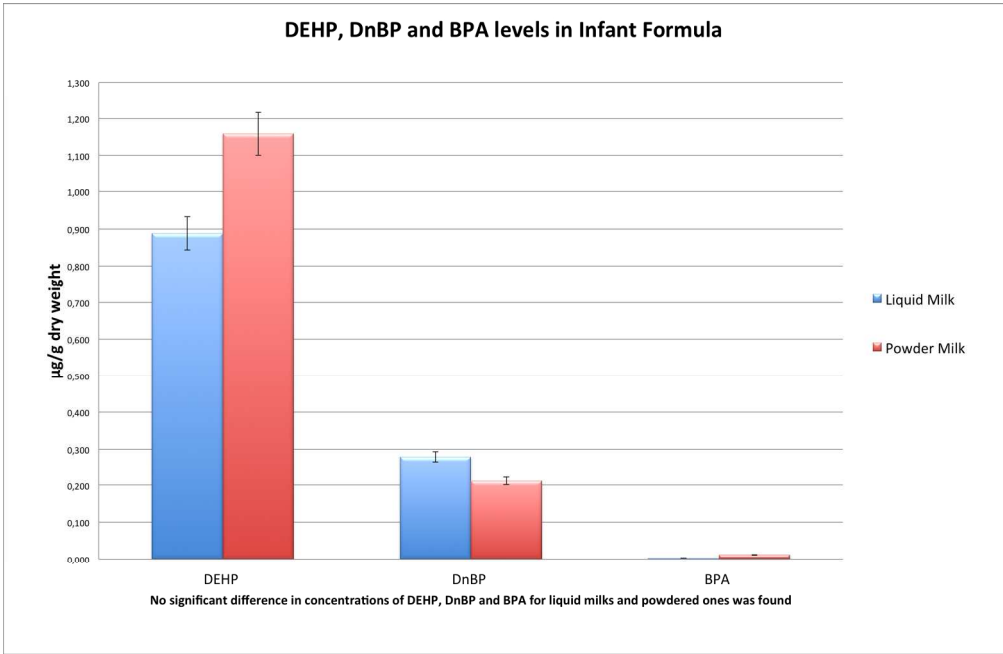
Age (days)	Infant's average weight		Milk assumption		DEHP intake		DnBP intake		BPA intake	
	(kg)		(g dry weight / day)		(µg/kg bw day)		(µg/kg bw day)		(µg/kg bw ² day)	
	50 th pctl ¹	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl	50 th pctl	97 th pctl
15	3.70	4.75	67.61	76.06	45.74	40.07	14.64	12.82	1.06	0.93
30	4.25	5.45	92.96	101.41	54.68	46.52	17.50	14.89	1.27	1.08
45	4.76	6.20	101.41	109.86	53.32	44.33	17.06	14.19	1.24	1.03
60	5.41	6.84	98.59	105.63	45.60	38.61	14.59	12.35	1.06	0.90
75	5.76	7.26	105.63	112.68	45.85	38.83	14.67	12.42	1.06	0.90
90	6.10	7.65	112.68	126.76	46.18	41.43	14.78	13.26	1.07	0.96
120	6.70	8.35	114.08	129.58	42.57	38.80	13.62	12.41	0.99	0.90

¹pctl = percentile

²kg bw = kg body weight

Figure legends

Figure 1. Concentrations of DEHP, DnBP and BPA in liquid and powder milk samples. Data are expressed as median and percentage of SE.



Concentrations of DEHP, DnBP and BPA in liquid and powder milk samples. Data are expressed as median and percentage of SE.