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RESEARCH PAPER

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Carbonic anhydrase activation profile of indole-based derivatives

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ABSTRACT

Carbonic Anhydrase Activators (CAAs) could represent a novel approach for the treatment of Alzheimer's disease, ageing, and other conditions that require remedial achievement of spatial learning and memory therapy. Within a research project aimed at developing novel CAAs selective for certain isoforms, three series of indole-based derivatives were investigated. Enzyme activation assay on human CA I, II, VA, and VII isoforms revealed several effective micromolar activators, with promising selectivity profiles towards the brain-associated cytosolic isoform hCA VII. Molecular modelling studies suggested a theoretical model of the complex between hCA VII and the new activators and provide a possible explanation for their modulating as well as selectivity properties. Preliminary biological evaluations demonstrated that one of the most potent CAA **7** is not cytotoxic and is able to increase the release of the brain-derived neurotrophic factor (BDNF) from human microglial cells, highlighting its possible application in the treatment of CNS-related disorders.

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Carbonic anhydrase activators; indole; microglia; brain associated human CA VII isoform

Introduction

Carbonic anhydrases (CAs) are metalloenzymes that play a crucial role in many physio-pathological conditions, mainly involving the maintenance/alteration of pH homeostasis¹. In the central nervous system (CNS), CAs are involved in neuronal signalling, as the reaction they catalyse, that is the reversible hydration/dehydration of CO_2/HCO_3^- ($CO_2 + H_2O = HCO_3^- + H^+$), contributes to the availability of ions that are essential for GABAergic and glutamatergic neuronal function, such as bicarbonate and protons, able to requlate the pH transitions in both the intra- and extra-cellular compartments²⁻⁵. Specifically, CAs catalytic activity takes place in a reversible two-step process: (i) the conversion of CO_2 to HCO_3^{-1} , via the nucleophilic attack of a Zn-bound hydroxide on CO₂, and the subsequent displacement of the newly formed HCO3⁻ by a water molecule¹; (ii) a proton transfer process mediated by wellordered water molecules and a histidine residue of the active site that regenerates the active enzyme featuring the Zn-bound hydroxide¹.

$$\begin{array}{l} \mathsf{CA}-\mathsf{Zn}(\mathsf{II})-\mathsf{OH}^-+\mathsf{CO}_2\leftrightarrow\mathsf{CA}-\mathsf{Zn}(\mathsf{II})-\mathsf{HCO}_3^-+\mathsf{H}_2\mathsf{O}\\ &\qquad \qquad \leftrightarrow\mathsf{CA}-\mathsf{Zn}(\mathsf{II})-\mathsf{OH}_2+\mathsf{HCO}_3^- \end{array} (i)\\ \\ \mathsf{CA}-\mathsf{Zn}(\mathsf{II})-\mathsf{OH}_2\leftrightarrow\mathsf{CA}-\mathsf{Zn}(\mathsf{II})-\mathsf{OH}^-+\mathsf{H}^+ \end{array} (ii) \end{array}$$

CAs have been considered as promising targets for the development of novel small molecules with therapeutic potential for the treatment of several human diseases. While inhibitors have been widely investigated as antitumor, antiglaucoma, diuretic, anticonvulsant, and antiobesity agents, the development of activators remains a poorly exploited research field until recently^{6,7}.

Recent reports highlighted that CA activation in the brain has implications for Alzheimer's disease and dementia⁸. Specifically, a role of CAs in brain processing of information and memory storage has been proposed, in terms of regulation of CI^-/HCO_3^- potentially associated with hippocampal GABAergic postsynaptic depolarisation, that in turn correlates with the synchronised neuronal activity, called theta rhythm⁹. In addition, CA inhibition was shown to abolish theta activity, as well as antagonism at GABA_A receptor, and to impair learning and memory processes in rats¹⁰. The beneficial effects of CA activation towards enhancing memory and learning observed in ageing rat models, together with the observation that brain CA levels are diminished in age-correlated impairment of CNS functionality¹¹, emphasises the importance of this family of enzymes in both disease and normal neuronal functions, demanding further development of CA activating drugs.

CA inhibitors classically feature a primary sulfonamide group that interacts with the zinc and displaces the catalytic Zn-bound-hydroxide/water necessary for the catalytic conversion of CO_2 into HCO_3^{-12} . On the other hand, most classes of activators identified so far, belong to the amine, amino acid, and oligopeptide chemotypes⁷. Their mechanism, thoroughly clarified by X-ray crystallographic and kinetic studies^{13–17}, basically resides in the participation in the proton transfer reaction from the water bound to the Zn²⁺ ion to the reaction medium, with the generation of the active zinc hydroxide species. This is the rate-determining step

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Figure 1. Structures of known and novel indole-based CAAs.

of the catalytic cycle, physiologically mediated by a His residue (His64 in many isoforms); CA activators (CAAs) function as additive proton shuttle molecules, shifting the equilibrium to the active form of the enzyme^{13–17}. As few examples, phenylalanine (PDB:2FMG)¹³, histamine (PDB:1AVN)¹⁸, histidine (PDB:2ABE)¹⁴ (Figure 1), and other imidazole-based compounds^{19–23} were identified as CA activators providing an alternative or additive site for proton transfer speeding up the reaction. X-ray crystallographic studies indicated the following mechanism^{13–15,18,24–26}:

$$\begin{array}{l} \mathsf{CA}{-}\mathsf{Zn}(\mathsf{II}){-}\mathsf{OH}_2 + \mathsf{CAA} \leftrightarrow [\mathsf{CA}{-}\mathsf{Zn}(\mathsf{II}){-}\mathsf{OH}_2 - \mathsf{CAA}] \\ \leftrightarrow [\mathsf{CA}{-}\mathsf{Zn}(\mathsf{II}){-}\mathsf{OH}^{-}{-}\mathsf{CAA}{-}\mathsf{H}^{+}] \\ \leftrightarrow \mathsf{CA}{-}\mathsf{Zn}(\mathsf{II}){-}\mathsf{OH}^{-}{+}\mathsf{CAA}{-}\mathsf{H}^{+} \end{array}$$

Activators bind to variable sites different from the inhibitors' binding site and far away from the metal ion; the interested region does not necessarily overlap with that occupied by His64 situated in the middle part and extending towards the edge of the active site, but in the opposite part with respect to the active site or in an outer region^{13–15,18,24–26}.

Typically, CAAs' design was based on histidine (Figure 1), the natural proton shuttle, or histamine (Figure 1), one of the first CAAs discovered, as a lead compound and proceeded through modification of the heterocyclic scaffold (including the combinations with additional nuclei), and variable functionalization of the primary amino group also with the insertion of aromatic moieties⁷. In addition, several drugs in clinical use for various therapeutic applications, i.e. sertraline, citalopram, fluoxetine, sildenafil (Figure 1), showed CA-activating effects, which might contribute to their effects or account for some side effects^{27,28}. These agents feature a molecular structure including various aromatic (poly)heterocyclic portions, more complex and not directly correlated with the leads histamine or histidine. Comprehensive SAR studies suggest a

proton-shuttling group bound to a hydrophobic aromatic heterocycle by a short and flexible linker as basic structural requirements to obtain efficient CAAs. To expand the chemical space exploitable for the development of novel CAAs, in the present work, three small series of indole-based derivatives (**A**, **B**, **C**, Figure 1) were synthesised and biologically evaluated for their CA activating properties. Specifically, all the novel compounds featured the indole scaffold, a "privileged scaffold" for drug discovery²⁹, decorated with a benzyl group at N1 atom of the indole ring and a protonatable moiety at different positions as a mimic of the proton shuttling residue (i.e. histidine).

In particular, the A series arises from our longstanding experience in the synthesis of indolglyoxylamides endowed with promising in vivo or in vitro biological activity (nonsedative/hypnotic, nonsedative/anxiolytic, antiinflammatory, neuroprotective)^{30–35}, combined with the widely recognised role of the alpha-keto amide moiety as a privileged motif in medicinal chemistry³⁶. In derivatives 1-4 (series A, Figure 1), two different protonatable (dimethylaminopropyl, diethylaminoethyl) and polar hydroxyldecorated (hydroxypropyl, hydroxyethyl) chains were variously attached to the 5-position and the glyoxamido-nitrogen of the 1benzylindole core. A first round of biological evaluation of compounds 1-4 on human hCA I, II, VA, and VII (vide infra) revealed a similar activation and selectivity profile for all the compounds. In line with these results, and in consideration of the synthetic feasibility, we decided to expand the SARs on these indole-based CAAs with the synthesis of compounds of series **B** and **C** (Figure 1), where the protonatable chain was maintained at 5-position of the indole, while varying the functionalization at 3-position. Specifically, derivatives 5-7 feature a 3-carboxamide moiety decorated with a hydroxyethyl (5 and 6) or lipophilic benzyl (7) group, while a simple ethyl ester moiety characterised derivatives 8 and 9 (Figure 1).

Materials and methods

Chemistry

The uncorrected melting points were determined using a Reichert Kofler hot-stage apparatus. ¹H NMR and ¹³C NMR spectra were recorded on a Bruker AVANCE 400 (¹H, 400 MHz, ¹³C, 100 MHz) in DMSO-d₆, MeOD or CDCl₃. Chemical shifts are expressed in δ (ppm) and coupling constants (J) in Hertz. Magnesium sulphate was used as the drying agent. Evaporations were made in vacuo (rotating evaporator). Analytical TLCs have been carried out on Merck 0.2 mm precoated silica gel aluminium sheets (60 F-254). Silica gel 60 (230-400 mesh) was used for column chromatography. Purity of the target compounds 1-9 was determined using a Shimadzu LC-20AD SP liquid chromatograph equipped with a DDA Detector ($\lambda = 254$ nm) using a column C18 (250 mm \times 4.6 mm, 5 µm, Shim-pack); the mobile phase, delivered at isocratic flow, consisted of 70% of acetonitrile and 30% H₂O (0.1% NH₃); flow rate 1 ml/min. All the compounds showed percent purity \geq 95%. Reagents, starting materials, and solvents were purchased from commercial suppliers and used as received. The intermediate N-benzyl-5-methoxy-1H-indole 10 was prepared according to the reported procedure³⁷.

General procedure for the synthesis of 2-(5-(2-hydroxyethoxy)-1benzyl-1H-indol-3-yl)-N-(dialkylaminoalkyl)glyoxylamides 1 and 2

A solution of the appropriate derivative **16** or **17** (0.31 mmol) was dissolved in 1,4-dioxane (7 ml) and cooled at 0 °C. Then a solution of NaOH 1 M (1.25 ml) was added, and the mixture was stirred at room temperature for 5 h. The reaction mixture was concentrated in vacuum, the residue dissolved in ethyl acetate and washed with saturated aqueous NaHCO₃ solution, water and brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. The crude products were then purified by flash chromatography (DCM/MeOH in ratio 9:1 as eluting system).

2-(5-(2-Hydroxyethoxy)-1-benzyl-1H-indol-3-yl)-N-(3-(dimethylami-

no)propyl)glyoxylamide (1). Yield: 69%; oil; ¹H NMR (400 MHz, CDCl₃) δ 9.05 (s, 1H, Ar-H), 8.24 (t, J = 6.0 Hz, 1H, NH), 7.98 (d, J = 2.4 Hz, 1H, Ar-H), 7.36–7.31 (m, 3H, Ar-H), 7.20–7.18 (m, 3H, Ar-H), 6.93 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.35 (s, 2H, CH₂Ph), 4.21 (t, J = 4.4 Hz, 2H, CH₂O), 4.00 (t, J = 4.4 Hz, 2H, CH₂OH), 3.50–3.45 (m, 2H, CH₂NH), 2.50 (t, J = 6.8 Hz, 2H, CH₂N), 2.34 (s, 6H, 2CH₃), 1.85–1.82 (m, 2H, CH₂CH₂CH₂). ¹³C NMR (100 MHz, CDCl₃) δ 180.23 (C=O), 162.81 (C=O), 156.04 (C-Ar) 141.59 (C-Ar), 135.39 (C-Ar), 131.53 (C-Ar), 129.04 (2 C-Ar), 128.91 (C-Ar), 128.28 (C-Ar) 126.96 (2 C-Ar), 114.19 (C-Ar), 112.16 (C-Ar), 111.51 (C-Ar), 105.63 (C-Ar), 69.90 (CH₂O), 61.55 (CH₂OH), 57.58 (CH₂N), 51.43 (CH₂Ph), 45.12 (2CH₃), 38.11 (CH₂NH), 26.44 (CH₂CH₂CH₂).

2-(5-(2-Hydroxyethoxy)-1-benzyl-1H-indol-3-yl)-N-(2-(diethylamino) ethyl)glyoxylamide (2). Yield: 81%; oil; ¹H NMR (400 MHz, DMSOd₆) δ 8.95 (s, 1H, Ar-H), 8.54 (t, J = 6.0 Hz, 1H, NH), 7.77 (d, J = 2.4 Hz, 1H, Ar-H), 7.48 (d, J = 8.8 Hz, 1H, Ar-H), 7.37–7.28 (m, 5H, Ar-H), 6.90 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.55 (s, 2H, CH₂Ph), 4.00 (t, J = 4.6 Hz, 2H, CH₂O), 3.74 (t, J = 4.4 Hz, 2H, CH₂OH), 3.30–3.23 (m, 2H, CH₂NH), 2.57–2.51 (m, 6H, CH₂N, 2CH₂CH₃), 0.97 (t, J = 7.2 Hz, 6H, 2CH₂CH₃). ¹³ C NMR (100 MHz, DMSO-d₆) δ 181.90 (C=O), 163.60 (C=O), 156.20 (C-Ar), 141.71 (C-Ar), 137.18 (C-Ar), 131.56 (C-Ar), 129.20 (2 C-Ar), 128.44 (C-Ar), 128.28 (C-Ar), 127.80 (2 C-Ar), 113.77 (C-Ar), 112.84 (C-Ar), 111.68 (C-Ar), 105.21 (C-Ar), 70.44 (CH₂O), 60.12 (CH₂OH), 51.48 (CH₂N), 50.43 (CH₂Ph), 47.10 (2CH₂CH₃), 37.19 (CH₂NH), 12.37 (2CH₂CH₃).

General procedure for the synthesis of 2–(5-(dialkylaminoalkoxy)-1-benzyl-1H-indol-3-yl)-N-(hydroxyalkyl)glyoxamides 3 and 4, and ethyl 5-(dialkylaminoalkoxy)-1-benzyl-1H-indole-3-carboxylates 8, 9 and 22

In a flask, Cs_2CO_3 (62 mg, 0.19 mmol) was added to a solution of the appropriate derivative **20** or **21** or **29** (0.15 mmol) in anhydrous DMF (5 ml). In another flask, Cs_2CO_3 (78 mg, 0.24 mmol) was added to a suspension of the appropriate amine (3-dimethylamino-1-propylchloride hydrochloride for **3** and **8**, 2-chloro-*N*,*N*diethylethylamine hydrochloride for **4** and **22**, 2-chloro-*N*,*N*-dimethylethylamine hydrochloride for **9**, 0.24 mmol) in 3 ml of the same solvent. The mixtures were stirred for about 30 min at room temperature and then put together and the resulting mixture was heated at 60 °C for 5 h. The solvent was removed under reduced pressure, then ice was added to the reaction mixture and the solid precipitate formed was collected by vacuum filtration. The crude products were finally purified by flash chromatography (DCM/ MeOH in ratio 9:1 as eluting system).

2-(5-(3-(Dimethylamino)propoxy)-1-benzyl-1H-indol-3-yl)-N-(3-

hydroxypropyl)glyoxamide (3). Yield: 56%; mp = 111–113 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 8.91 (s, 1H, Ar-H), 8.68 (t, J = 5.6 Hz, 1H, NH), 7.76 (d, J = 2.8 Hz, 1H, Ar-H), 7.47 (d, J = 9.2 Hz, 1H, Ar-H), 7.37–7.27 (m, 5H, Ar-H), 6.89 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.55 (s, 2H, CH₂Ph), 4.53 (t, J = 5.2 Hz, 1H, CH₂OH), 4.02 (t, J = 6.4 Hz, 2H, CH₂O), 3.49–3.45 (m, 2H, CH₂OH), 3.29–3.26 (m, 2H, CH₂NH), 2.47 (t, J = 6.8 Hz, 2H, CH₂N), 2.23 (s, 6H, 2CH₃), 1.92–1.88 (m, 2H, NCH₂CH₂CH₂O), 1.69–1.66 (m, 2H, NHCH₂CH₂CH₂OH). ¹³C NMR (100 MHz, DMSO-d₆) δ 182.20 (C=O), 163.87 (C=O), 156.11 (C-Ar), 141.57 (C-Ar), 137.19 (C-Ar), 131.54 (C-Ar), 129.21 (2 C-Ar), 128.41 (C-Ar), 128.27 (C-Ar), 127.80 (2 C-Ar), 113.76 (C-Ar), 112.85 (C-Ar), 111.69 (C-Ar), 105.04 (C-Ar), 66.58 (CH₂O), 59.19 (CH₂OH), 56.01 (CH₂N), 50.41 (CH₂Ph), 45.31 (2CH₃), 36.68 (CH₂NH), 32.42 (NHCH₂CH₂CH₂OH), 27.07 (NCH₂CH₂CH₂O).

2-(5-(2-(Diethylamino)ethoxy)-1-benzyl-1H-indol-3-yl)-N-(2-hydrox-

yethyl)glyoxamide (4). Yield: 57%; mp = 116–118 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.00 (s, 1H, Ar-H), 7.97 (t, J = 6.0 Hz, 1H, NH), 7.94 (d, J = 2.4 Hz, 1H, Ar-H), 7.34–7.31 (m, 3H, Ar-H), 7.19–7.17 (m, 3H, Ar-H), 6.90 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.34 (s, 2H, CH₂Ph), 4.32 (t, J = 5.2 Hz, 2H, CH₂O), 3.85 (t, J = 5.2 Hz, 2H, CH₂OH), 3.60–3.56 (m, 2H, CH₂NH), 3.16 (t, J = 5.2 Hz, 2H, CH₂OH), 3.60–3.56 (m, 2H, CH₂NH), 3.16 (t, J = 5.2 Hz, 2H, CH₂OH), 2.92 (q, J = 6.8 Hz, 2CH₂CH₃), 1.25 (t, J = 6.8 Hz, 6H, 2CH₂CH₃). ¹³C NMR (100 MHz, CDCl₃) δ 179.72 (C=O), 163.48 (C=O), 156.62 (C-Ar), 141.64 (C-Ar), 135.27 (C-Ar), 131.59 (C-Ar), 129.07 (2 C-Ar) 128.88 (C-Ar), 128.33 (C-Ar), 126.97 (2 C-Ar), 113.98 (C-Ar), 112.02 (C-Ar), 111.59 (C-Ar), 105.65 (C-Ar), 61.91 (CH₂O), 51.45 (CH₂OH), 51.10 (CH₂N), 47.60 (CH₂Ph), 42.23 (2CH₂CH₃), 29.69 (CH₂NH), 10.51 (2CH₂CH₃).

Ethyl 5-(3-(dimethylamino)propoxy)-1-benzyl-1H-indole-3-carboxylate (8). Yield: 88%; mp = 58–60 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.81 (s, 1H, Ar-H), 7.69 (d, J=2.4 Hz, 1H, Ar-H), 7.36–7.31 (m, 3H, Ar-H), 7.19–7.14 (m, 3H, Ar-H), 6.88 (dd, J=2.6 Hz, J=9.0 Hz, 1H, Ar-H), 5.31 (s, 2H, CH₂Ph), 4.39 (q, J=7.2 Hz, 2H, OCH₂CH₃), 4.12 (t, J=6.2 Hz, 2H, OCH₂CH₂), 2.69 (t, J=7.6 Hz, 2H, CH₂N), 2.42 (s, 6H, 2CH₃), 2.14–2.07 (m, 2H, CH₂CH₂CH₂), 1.43 (t, J=7.0 Hz, 3H, OCH₂CH₃). ¹³C NMR (100 MHz, CDCl₃) δ 165.11 (C=O), 155.04 (C-Ar), 136.01 (C-Ar), 134.60 (C-Ar), 131.88 (C-Ar), 128.97 (2 C-Ar), 128.10 (C-Ar), 127.76 (C-Ar), 126.95 (2 C-Ar), 113.55 (C-Ar), 111.14 (C-Ar), 107.26 (C-Ar), 104.33 (C-Ar), 66.45 (OCH₂CH₃), 59.62 *Ethyl* 5-(2-(dimethylamino)ethoxy)-1-benzyl-1H-indole-3-carboxylate (9). Yield: 30%; mp = 69–71 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.82 (s, 1H, Ar-H), 7.71 (d, J = 2.4 Hz, 1H, Ar-H), 7.36–7.31 (m, 3H, Ar-H), 7.19–7.14 (m, 3H, Ar-H), 6.93 (dd, J = 2.6 Hz, J = 9.0 Hz, 1H, Ar-H), 5.31 (s, 2H, CH₂Ph), 4.39 (q, J = 7.2 Hz, 2H, OCH₂CH₃), 4.21 (t, J = 5.6 Hz, 2H, CH₂O), 2.85 (t, J = 5.4 Hz, 2H, CH₂N), 2.42 (s, 6H, 2CH₃), 1.42 (t, J = 7.0 Hz, 3H, OCH₂CH₃). ¹³C NMR (100 MHz, CDCl₃) δ 165.15 (C=O), 154.97 (C-Ar), 135.99 (C-Ar), 134.59 (C-Ar), 131.89 (C-Ar), 128.97 (2C-Ar), 128.11 (C-Ar), 127.67 (C-Ar), 126.97 (2C-Ar), 113.85 (C-Ar), 111.16 (C-Ar), 107.27 (C-Ar), 104.14 (C-Ar), 66.10 (OCH₂CH₃), 59.63 (OCH₂CH₂), 58.27 (CH₂N), 50.91 (CH₂Ph), 45.65 (2CH₃), 14.64 (OCH₂CH₃).

Ethyl 5-(2-(*diethylamino*)*ethoxy*)-1-*benzyl*-1*H*-*indole*-3-*carboxylate* (22). Yield: 42%; mp = 48–50 °C; ¹H-NMR (400 MHz, CDCl₃) δ 7.81 (s, 1H, Ar-H), 7.71 (d, J = 2.4 Hz, 1H, Ar-H), 7.35–7.29 (m, 3H, Ar-H), 7.18–7.13 (m, 3H, Ar-H), 6.89 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.29 (s, 2H, CH₂Ph), 4.39 (q, J = 7.2 Hz, 2H, OCH₂CH₃), 4.20 (t, J = 6.2 Hz, 2H, CH₂O), 2.99 (t, J = 6.2 Hz, 2H, CH₂N), 2.74 (q, J = 7.2 Hz, 4H, 2NCH₂CH₃), 1.42 (t, J = 7.0 Hz, 3H, OCH₂CH₃), 1.14 (t, J = 7.2 Hz, 6H, 2NCH₂CH₃).

General procedure for the synthesis of 5-(dialkylaminoalkoxy)-1benzyl-N-(2-hydroxyethyl)-1H-indole-3-carboxamides 5 and 6 and 5-(2-(diethylamino)ethoxy)-N,1-dibenzyl-1H-indole-3-carboxamide 7

To an ice-cooled solution of the appropriate derivative **23** or **24** or **25** (0.25 mmol) in 5 ml of anhydrous DMF, TBTU (120 mg, 0.38 mmol) and DIPEA (87 μ l, 0.50 mmol) were added and the mixture was stirred for 30 min. Then, the appropriate amine (ethanolamine for **5** and **6**, benzylamine for **7**, 0.25 mmol) was added and the resulting mixture was stirred at room temperature overnight. DMF was removed under reduced pressure and ice was added to the flask. The mixture was extracted with DCM, dried over MgSO₄, and concentrated under reduced pressure. The crude products were purified by flash chromatography (DCM/MeOH in ratio 9:1 as eluting mixture).

5-(3-(Dimethylamino)propoxy)-1-benzyl-N-(2-hydroxyethyl)-1H-

indole-3-carboxamide (5). Yield: 23%; oil; ¹H NMR (400 MHz, MeOD) δ 7.88 (s, 1H, Ar-H), 7.69 (d, J = 2.4 Hz, 1H, Ar-H), 7.31–7.24 (m, 4H, Ar-H), 7.17–7.15 (m, 2H, Ar-H), 6.84 (dd, J=2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.31 (s, 2H, CH₂Ph), 4.04 (t, J = 6.0 Hz, 2H, CH₂O), 3.74 (t, J = 5.6 Hz, 2H, CH₂OH), 3.53 (t, J = 5.6 Hz, 2H, CH2NH), 2.55-2.51 (m, 2H, CH2N), 2.28 (s, 6H, 2CH3), 1.99-1.93 (m, 2H, CH₂CH₂CH₂). ¹³C NMR (100 MHz, MeOD) δ 166.98 (C=O), 154.08 (C-Ar), 137.00 (C-Ar), 132.05 (C-Ar), 131.29 (C-Ar), 128.43 (2 C-Ar), 127.50 (C-Ar), 126.78 (2 C-Ar), 112.87 (C-Ar), 111.14 (C-Ar), 109.57 (C-Ar), 103.92 (C-Ar), 60.71 (CH₂O), 55.84 (CH₂OH), 52.26 50.05 (CH₂Ph), 43.01 (2CH₃), 41.52 $(CH_2N),$ $(CH_2NH),$ 23.01 (CH₂CH₂CH₂).

5-(2-(Dimethylamino)ethoxy)-1-benzyl-N-(2-hydroxyethyl)-1H-

indole-3-carboxamide (6). Yield: 27%; oil; ¹H NMR (400 MHz, MeOD) δ 7.90 (s, 1H, Ar-H), 7.72 (d, J = 2.4 Hz, 1H, Ar-H), 7.32–7.26 (m, 4H, Ar-H), 7.19–7.17 (m, 2H, Ar-H), 6.89 (dd, J = 2.6 Hz, J = 9.0 Hz, 1H, Ar-H), 5.35 (s, 2H, CH₂Ph), 4.16 (t, J = 5.2 Hz, 2H, CH₂O), 3.74 (t, J = 5.8 Hz, 2H, CH₂OH), 3.53 (t, J = 5.8 Hz, 2H,

CH₂NH), 2.83 (t, J = 5.2 Hz, 2H, CH₂N), 2.38 (s, 6H, 2CH₃). ¹³C NMR (100 MHz, MeOD) δ 166.86 (C=O), 153.10 (C-Ar), 136.96 (C-Ar), 132.41 (C-Ar), 131.55 (C-Ar), 128.44 (2 C-Ar), 127.52 (C-Ar), 126.78 (2 C-Ar), 112.67 (C-Ar), 111.41 (C-Ar), 109.67 (C-Ar), 104.21 (C-Ar), 62.32 (CH₂O), 60.70 (CH₂OH), 53.37 (CH₂N), 50.10 (CH₂Ph), 42.71 (2CH₃), 41.54 (CH₂NH).

5-(2-(Diethylamino)ethoxy)-N,1-dibenzyl-1H-indole-3-carboxamide

(7). Yield:16%; mp= 57–59 °C; ¹H NMR (400 MHz, MeOD) δ 7.99 (s, 1H, Ar-H), 7.79 (d, J=2.4 Hz, 1H, Ar-H), 7.41–7.30 (m, 5H, Ar-H), 7.29–7.22 (m, 4H, Ar-H), 7.22–7.20 (m, 2H, Ar-H), 6.94 (dd, J=2.6 Hz, J=9.0 Hz, 1H, Ar-H), 5.42 (s, 2H, NCH₂Ph), 4.60 (s, 2H, NHCH₂Ph), 4.35 (t, J=5.0 Hz, 2H, CH₂O), 3.52 (t, J=5.0 Hz, 2H, CH₂N), 3.26 (q, J=7.4 Hz, 4H, 2CH₂CH₃), 1.34 (t, J=7.2 Hz, 6H, 2CH₂CH₃). ¹³C NMR (100 MHz, MeOD) δ 166.48 (C=O), 153.51 (C-Ar), 139.40 (C-Ar), 136.92 (C-Ar), 132.42 (C-Ar), 131.37 (C-Ar), 128.44 (2 C-Ar), 128.09 (2 C-Ar), 127.66 (C-Ar), 127.51 (C-Ar), 127.05 (2 C-Ar), 126.77 (2 C-Ar), 126.63 (C-Ar), 112.73 (C-Ar), 111.26 (C-Ar), 109.60 (C-Ar), 104.26 (C-Ar), 63.02 (CH₂O), 51.14 (CH₂N), 50.14 (2CH₂Ph), 42.48 (2CH₂CH₃), 8.04 (2CH₂CH₃).

General procedure for the synthesis of 2-(1-benzyl-5-methoxy-1Hindol-3-yl)-N-(dialkylaminoalkyl)glyoxylamides 12 and 13

Oxalyl chloride (0.18 ml, 2.0 mmol) was added dropwise at 0 °C to a solution of N-benzyl-5-methoxy-1H-indole **10** (250 mg, 1.0 mmol) in freshly distilled diethyl ether (10 ml). The mixture was maintained at room temperature for 2 h. The generated precipitate was collected by vacuum filtration to give the acyl chloride 11 that was directly used in the subsequent reaction. A solution of the appropriate amine (N,N-dimethyl-1,3-propylenediamine for 12, N,N-diethylethylenediamine for 13, 0.81 mmol) in dry toluene (5 ml) was added dropwise, under a nitrogen atmosphere, to a stirred suspension, cooled at 0°C, of 2-(1-benzyl-5-methoxy-1Hindol-3-yl)glyoxylyl chloride 11 (222 mg, 0.68 mmol) in 5 ml of the same solvent, followed by the addition of NEt₃ (0.12 ml, 0.88 mmol). The reaction mixture was left under stirring at room temperature overnight. The solvent was removed under reduced pressure, and the residue was dissolved with DCM. The organic solution was washed with a 5% solution of NaHCO₃, 10% solution of HCl, H₂O, dried over MgSO₄ and evaporated to dryness, yielding the desired compounds 12, 13 that were used in the subsequent step without further purification.

2-(1-Benzyl-5-methoxy-1H-indol-3-yl)-N-(3-(dimethylamino)propyl)-

glyoxylamide (12). Yield: 48%; mp = $186-188 \,^{\circ}$ C; ¹H NMR (400 MHz, DMSO-d₆) δ 8.93 (s, 1H, Ar-H), 8.89 (t, $J = 6.0 \,$ Hz, 1H, NH), 7.78 (d, $J = 2.8 \,$ Hz, 1H, Ar-H), 7.51 (d, $J = 8.8 \,$ Hz, 1H, Ar-H), 7.38–7.28 (m, 5H, Ar-H), 6.92 (dd, $J = 2.4 \,$ Hz, $J = 8.8 \,$ Hz, 1H, Ar-H), 5.57 (s, 2H, CH₂Ph), 3.80 (s, 3H, OCH₃), 3.31–3.28 (m, 2H, CH₂NH), 3.07–3.03 (m, 2H, CH₂N), 2.73 (s, 6H, 2CH₃), 1.94–1.90 (m, 2H, CH₂CH₂).

2-(1-Benzyl-5-methoxy-1H-indol-3-yl)-N-(2-(diethylamino)ethyl)-

glyoxylamide (13). Yield: 77%; mp = $124-126 \,^{\circ}$ C; ¹H NMR (400 MHz, DMSO-d₆) δ 8.97 (s, 1H, Ar-H), 8.66 (bs, 1H, NH), 7.77 (s, 1H, Ar-H), 7.50 (d, $J = 9.6 \,\text{Hz}$, 1H, Ar-H), 7.34–7.29 (m, 5H, Ar-H), 6.91 (dd, $J = 2.4 \,\text{Hz}$, $J = 8.8 \,\text{Hz}$, 1H, Ar-H), 5.56 (s, 2H, CH₂Ph), 3.79 (s, 3H, OCH₃), 3.37–3.34 (m, 2H, CH₂NH), 2.76–2.63 (m, 6H, CH₂N, 2CH₂CH₃), 1.03 (t, $J = 7.2 \,\text{Hz}$, 6H, 2CH₂CH₃).

General procedure for the synthesis of 2-(1-benzyl-5-hydroxy-1Hindol-3-yl)-N-(dialkylaminoalkyl)glyoxylamides 14 and 15, 2-(1benzyl-5-hydroxy-1H-indol-3-yl)-N-(hydroxyalkyl)glyoxamides 20 and 21 and ethyl 1-benzyl-5-hydroxy-1H-indole-3-carboxylate 29

Boron tribromide (1.0 M in methylene chloride, 0.51 ml, 3.0 mmol) was added dropwise, under a nitrogen atmosphere, to a solution of the appropriate derivative **12**, **13**, **18**, **19** or **28** (0.30 mmol) in anhydrous DCM, cooled at -10 °C. The mixture was left under stirring for 30 min at -10 °C and subsequently at room temperature overnight. The solvent was removed under reduced pressure, and the residue was washed twice with methanol to hydrolyse the excess of BBr₃. The residues obtained were purified by flash chromatography (DCM/MeOH in ratio 9.5:0.5 for **14**, **15**, **20** and **21**, EtOAc/petroleum ether 40–60 °C in ratio 5:5 for **29** as eluting system).

2-(1-Benzyl-5-hydroxy-1H-indol-3-yl)-N-(3-(dimethylamino)propyl)-

glyoxylamide (14). Yield: 42%; oil; ¹H NMR (400 MHz, DMSO-d₆) δ 9.25 (bs, 1H, OH), 8.87–8.85 (m, 2H, NH, Ar-H), 7.67 (d, J = 2.0 Hz, 1H, Ar-H), 7.39–7.27 (m, 6H, Ar-H), 6.75 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.51 (s, 2H, CH₂Ph), 3.30–3.25 (m, 2H, CH₂NH), 3.07–3.03 (m, 2H, CH₂N), 2.74 (s, 6H, 2CH₃), 1.91–1.87 (m, 2H, CH₂CH₂CH₂).

2-(1-Benzyl-5-hydroxy-1H-indol-3-yl)-N-(2-(diethylamino)ethyl)-

glyoxylamide (15). Yield: 70%; mp = $176-178 \,^{\circ}$ C; ¹H NMR (400 MHz, DMSO-d₆) δ 9.25 (s, 1H, OH), 8.91 (s, 1H, Ar-H), 8.71 (bs, 1H, NH), 7.68 (d, $J = 2.4 \,$ Hz, 1H, Ar-H), 7.40–7.28 (m, 6H, Ar-H), 6.75 (dd, $J = 2.4 \,$ Hz, $J = 8.8 \,$ Hz, 1H, Ar-H), 5.52 (s, 2H, CH₂Ph), 3.40–3.37 (m, 2H, CH₂NH), 2.78–2.77 (m, 6H, CH₂N, 2CH₂CH₃), 1.05 (t, $J = 7.2 \,$ Hz, 6H, 2CH₂CH₃).

2-(1-Benzyl-5-hydroxy-1H-indol-3-yl)-N-(3-hydroxypropyl)glyoxa-

mide (20). Yield: 30%; mp = 169–171 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 9.24 (bs, 1H, OH), 8.85 (s, 1H, Ar-H), 8.67 (t, J=6.4 Hz, 1H, NH), 7.66 (d, J=2.4 Hz, 1H, Ar-H), 7.38–7.28 (m, 6H, Ar-H), 6.73 (dd, J=2.4 Hz, J=8.8 Hz, 1H, Ar-H), 5.50 (s, 2H, CH₂Ph), 4.51 (t, J=5.2 Hz, 1H, CH₂OH), 3.49–3.44 (m, 2H, CH₂OH), 3.30–3.26 (m, 2H, CH₂NH), 1.69–1.65 (m, 2H, CH₂CH₂).

2-(1-Benzyl-5-hydroxy-1H-indol-3-yl)-N-(2-hydroxyethyl)glyoxamide (21). Yield: 67%; mp = 217–219 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 9.25 (s, 1H, OH), 8.88 (s, 1H, Ar-H), 8.58 (t, J = 5.8 Hz, 1H, NH), 7.67 (d, J = 2.4 Hz, 1H, Ar-H), 7.38–7.28 (m, 6H, Ar-H), 6.73 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.51 (s, 2H, CH₂Ph), 4.79 (t, J = 5.6 Hz, 1H, CH₂OH), 3.53–3.49 (m, 2H, CH₂OH), 3.31–3.26 (m, 2H, CH₂NH).

Ethyl 1-benzyl-5-hydroxy-1H-indole-3-carboxylate (29). Yield: 54%; mp = 175–177 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 9.08 (s, 1H, OH), 8.16 (s, 1H, Ar-H), 7.39 (d, J=2.0 Hz, 1H, Ar-H), 7.35–7.25 (m, 6H, Ar-H), 6.68 (dd, J=2.4 Hz, J=8.8 Hz, 1H, Ar-H), 5.42 (s, 2H, CH₂Ph), 4.26 (q, J=7.2 Hz, 2H, OCH₂CH₃), 1.33 (t, J=7.2 Hz, 3H, OCH₂CH₃).

General procedure for the synthesis of 2–(1-benzyl-5-acetoxyethoxy-1H-indol-3-yl)-N-(dialkylaminoalkyl)glyoxylamides 16 and 17 To a solution of the appropriate derivative 14 or 15 (0.26 mmol) in acetone (12 ml), Cs_2CO_3 (261 mg, 0.80 mmol) and 2-bromoethyl acetate (0.044 ml, 0.40 mmol) were added. The mixture was stirred at 60 °C overnight, concentrated in vacuum and the residue suspended in ethyl acetate. The suspension was washed

with H_2O , saturated aqueous NaHCO₃ solution and brine. The organic phase was dried over MgSO₄ and concentrated under reduced pressure. The crude products were then purified by flash chromatography (DCM/MeOH in ratio 9:1 as eluting system).

2-(1-Benzyl-5-acetoxyethoxy-1H-indol-3-yl)-N-(3-(dimethylamino)-

propyl)glyoxylamide (16). Yield: 42%; mp = 97–99°C; ¹H NMR (400 MHz, CDCl₃) δ 9.03 (s, 1H, Ar-H), 8.18 (t, J=5.6 Hz, 1H, NH), 7.97 (d, J=2.4 Hz, 1H, Ar-H), 7.37–7.29 (m, 3H, Ar-H), 7.21–7.18 (m, 3H, Ar-H), 6.94 (dd, J=2.4 Hz, J=8.8 Hz, 1H, Ar-H), 5.36 (s, 2H, CH₂Ph), 4.47 (t, J=4.8 Hz, 2H, CH₃COOCH₂), 4.29 (t, J=4.8 Hz, 2H, OCH₂), 3.52–3.48 (m, 2H, CH₂NH), 2.64 (t, J=7.2 Hz, 2H, CH₂N), 2.45 (s, 6H, 2NCH₃), 2.13 (s, 3H, CH₃CO), 1.96–1.91 (m, 2H, CH₂CH₂CH₂).

2-(1-Benzyl-5-acetoxyethoxy-1H-indol-3-yl)-N-(2-(diethylamino)e-

thyl)glyoxylamide (17). Yield: 60%; oil; ¹H NMR (400 MHz, CDCl₃) δ 9.04 (s, 1H, Ar-H), 8.02 (bs, 1H, NH), 7.98 (d, J = 2.4 Hz, 1H, Ar-H), 7.34–7.29 (m, 3H, Ar-H), 7.19–7.16 (m, 3H, Ar-H), 6.92 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.35 (s, 2H, CH₂Ph), 4.45 (t, J = 4.6 Hz, 2H, CH₃COOCH₂), 4.28 (t, J = 4.6 Hz, 2H, OCH₂), 3.48–3.44 (m, 2H, CH₂NH), 2.69 (t, J = 6.2 Hz, 2H, CH₂N), 2.62 (q, J = 3.6 Hz, 4H, 2CH₂CH₃), 2.11 (s, 3H, CH₃CO), 1.09 (t, J = 7.2 Hz, 6H, 2CH₂CH₃).

General procedure for the synthesis of 2-(1-benzyl-5-methoxy-1Hindol-3-yl)-N-(hydroxyalkyl)glyoxamides 18 and 19

To a solution of compound **11** (983 mg, 3.0 mmol) in 1,4-dioxane (3 ml), the appropriate amine (3-amino-1-propanol for **18** or ethanolamine for **19**, 9.0 mmol) was added dropwise at 0 °C and the mixture was stirred for 3 h at room temperature. The dioxane was removed under reduced pressure and water was added to the mixture; the generated precipitate was collected by vacuum filtration. The crude products were finally purified by flash chromatography (EtOAc/petroleum ether 40–60 °C in ratio 7:3 as eluting system).

2-(1-Benzyl-5-methoxy-1H-indol-3-yl)-N-(3-hydroxypropyl)glyoxa-

mide (18). Yield: 71%; mp = 125–127 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.05 (s, 1H, Ar-H), 7.95 (d, J = 2.4 Hz, 1H, Ar-H), 7.85 (t, J = 6.0 Hz, 1H, NH), 7.37–7.32 (m, 3H, Ar-H), 7.20–7.18 (m, 3H, Ar-H), 6.91 (dd, J = 2.8 Hz, J = 9.2 Hz, 1H, Ar-H), 5.37 (s, 2H, CH₂Ph), 3.91 (s, 3H, OCH₃), 3.72 (t, J = 8.0 Hz, 2H, CH₂OH), 3.61–3.55 (m, 2H, CH₂NH), 1.86–1.81 (m, 2H, CH₂CH₂CH₂).

2-(1-Benzyl-5-methoxy-1H-indol-3-yl)-N-(2-hydroxyethyl)glyoxa-

mide (19). Yield: 35%; mp = 154-156 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 8.94 (s, 1H, Ar-H), 8.61 (t, J = 6.0 Hz, 1H, NH), 7.77 (s, 1H, Ar-H), 7.49 (d, J = 9.2 Hz, 1H, Ar-H), 7.37–7.28 (m, 5H, Ar-H), 6.91 (dd, J = 2.4 Hz, J = 8.8 Hz, 1H, Ar-H), 5.56 (s, 2H, CH₂Ph), 4.79 (bs, 1H, OH), 3.79 (s, 3H, OCH₃), 3.52–3.51 (m, 2H, CH₂OH), 3.32–3.29 (m, 2H, CH₂NH).

General procedure for the synthesis of 5-(dialkylaminoalkoxy)-1benzyl-1H-indole-3-carboxylic acids 23-25

LiOH·H₂O (21 mg, 0.49 mmol) was added to an ice-cooled solution of the appropriate derivative **8**, **9**, or **22** (0.81 mmol) in 8 ml of MeOH/H₂O mixture (3:1). The resulting mixture was refluxed

overnight. The suspension was filtered off, and the filtrate was acidified with a 10% aqueous solution of HCl until pH = 1. The mixture was extracted with DCM and, after drying with MgSO₄, the solvent was evaporated to dryness to yield the crude products **23–25**, which did not need any further purification.

5-(3-(Dimethylamino)propoxy)-1-benzyl-1H-indole-3-carboxylic

acid (23). Yield: 97%, oil, ¹H NMR (400 MHz, MeOD) δ 8.00 (s, 1H, Ar-H), 7.64 (d, J = 2.4 Hz, 1H, Ar-H), 7.32–7.29 (m, 4H, Ar-H), 7.27–7.19 (m, 2H, Ar-H), 6.87 (dd, J = 2.4 Hz, J = 9.2 Hz, 1H, Ar-H), 5.41 (s, 2H, CH₂Ph), 4.16 (t, J = 5.8 Hz, 2H, OCH₂), 3.38 (t, J = 7.6 Hz, 2H, CH₂N), 2.94 (s, 6H, 2CH₃), 2.25–2.21 (m, 2H, CH₂CH₂CH₂).

5-(2-(Dimethylamino)ethoxy)-1-benzyl-1H-indole-3-carboxylic acid (24). Yield: 81%; mp = 130-132°C; ¹H NMR (400 MHz, MeOD) δ 8.06 (s, 1H, Ar-H), 7.73 (d, J = 2.0 Hz, 1H, Ar-H), 7.37–7.27 (m, 4H, Ar-H), 7.23–7.21 (m, 2H, Ar-H), 6.97 (dd, J = 8.8 Hz, J = 2.0 Hz, 1H, Ar-H), 5.45 (s, 2H, CH₂Ph), 4.41–4.40 (m, 2H, OCH₂), 3.63–3.61 (m, 2H, CH₂N), 3.01 (s, 6H, 2CH₃).

5-(2-(Diethylamino)ethoxy)-1-benzyl-1H-indole-3-carboxylic acid (25). Yield: 90%; mp = 94–96 °C; ¹H NMR (400 MHz, DMSO-d₆) δ 8.21 (s, 1H, Ar-H), 7.57 (d, J = 2.4 Hz, 1H, Ar-H); 7.47 (d, J = 9.2 Hz, 1H, Ar-H); 7.35–7.29 (m, 2H, Ar-H); 7.27–7.25 (m, 3H, Ar-H); 6.90 (dd, J = 2.4 Hz, J = 9.2 Hz, 1H, Ar-H); 5.48 (s, 2H, CH₂Ph); 4.35 (t, J = 4.6 Hz, 2H, OCH₂); 3.52–3.50 (m, 2H, CH₂N); 3.23–3.21 (m, 4H, 2CH₂CH₃); 1.25 (t, J = 7.2 Hz, 6H, 2CH₂CH₃).

Ethyl 5-methoxy-1H-indole-3-carboxylate (27). To a solution of compound **26** (574 mg, 3.00 mmol) in absolute ethanol (52 ml) a catalytic amount of concentrated sulphuric acid (0.52 ml) was added, and the resulting mixture was allowed to reflux for 20 h. After completion of the reaction, ethanol was removed, and the residue was dissolved with ethyl acetate and washed with saturated NaHCO₃ solution. The organic layer was dried over MgSO₄, filtered, and evaporated to give the corresponding ester. The crude product was then purified by flash chromatography (EtOAc/ petroleum ether 40–60 °C in ratio 5:5 as eluting system). Yield: 31%; mp = 123–125 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.56 (bs, 1H, NH), 7.89 (d, *J* = 3.2 Hz, 1H, Ar-H), 7.70 (d, *J* = 2.8 Hz, 1H, Ar-H), 7.32 (d, *J* = 8.8 Hz, 1H, Ar-H), 6.93 (dd, *J* = 8.8 Hz, *J* = 2.4 Hz, 1H, Ar-H), 4.41 (q, *J* = 7.2 Hz, 2H, OCH₂CH₃), 3.91 (s, 3H, OCH₃), 1.45 (t, *J* = 7.0 Hz, 3H, OCH₂CH₃).

Ethyl 1-benzyl-5-methoxy-1H-indole-3-carboxylate (28). Sodium hydride (40 mg, 1.00 mmol, 60% dispersion in mineral oil) was added portionwise, under nitrogen atmosphere, to an ice-cold solution of compound 27 (200 mg, 0.91 mmol) in 5 ml of DMF and the mixture was stirred for about 30 min at 0 °C. Once hydrogen evolution ceased, benzyl bromide (0.12 ml, 1.00 mmol) was added dropwise, and the reaction was maintained under stirring for 5 h at room temperature. After completion of the reaction, DMF was evaporated under reduced pressure, then ice was added to the flask and the solid precipitate formed was collected by vacuum filtration. The crude product was then purified by flash chromatography (EtOAc - petroleum ether 40-60 °C in ratio 3:7 as eluting system). Yield: 83%; mp = 89–91 °C; ¹H NMR (400 MHz, DMSO- d_6) δ 8.24 (s, 1H, Ar-H), 7.50 (d, J = 2.4 Hz, 1H, Ar-H), 7.44 (d, J = 9.2 Hz, 1H, Ar-H), 7.35–7.25 (m, 5H, Ar-H), 6.84 (dd, J=2.4 Hz, J=8.8 Hz, 1H, Ar-H), 5.47 (s, 2H, CH₂Ph), 4.28 (q, J=7.0 Hz, 2H, OCH₂CH₃), 3.78 (s, 3H, OCH₃), 1.34 (t, J = 7.0 Hz, 3H, OCH₂CH₃).

Carbonic anhydrases activation assays

A stopped-flow method³⁸ has been used for assaying the CA catalysed CO₂ hydration activity with Phenol red as indicator, working at the absorbance maximum of 557 nm, following the initial rates of the CA-catalysed CO₂ hydration reaction for 10–100 s. For each activator, at least six traces of the initial 5–10% of the reaction have been used for determining the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total observed rates. Stock solutions of activator (0.1 mM) were prepared in distilled-deionized water and dilutions up to 0.1 nM were done thereafter with the assay buffer. The activation constant (K_A), defined similarly with the inhibition constant (K_I), was obtained by considering the classical Michaelis–Menten equation (Equation (3), which has been fitted by nonlinear least squares by using PRISM 3:

$$v = v_{max} / \left\{ 1 + K_M / [S] \big(1 + [A]_f / K_A \big) \right\} \eqno(3)$$

where [A]_f is the free concentration of activator.

Working at substrate concentrations considerably lower than K_M ([S] $\ll K_M$), and considering that [A]_f can be represented in the form of the total concentration of the enzyme ([E]_t) and activator ([A]_t), the obtained competitive steady-state equation for determining the activation constant is given by Equation 4:

$$v = v_0 K_A / \{K_A + [A]_t - 0.5 \quad \left\{ ([A]_t + [E]_t + K_A) - ([A]_t + [E]_t + K_A)^2 - 4[A]_t [E]_t^1 / 2 \} \right\}$$
(4)

where v_0 represents the initial velocity of the enzyme-catalysed reaction in the absence of an activator^{39–42}. Enzyme concentrations in the assay system were in the range of 6.5–12.0 nM.

Molecular modeling

To perform the molecular modelling studies on the newly discovered CAAs, the latest version of docking software i.e. AutodockGPU⁴³ along with its Graphical User Interface AutoDockTools (ADT)⁴⁴ were employed. In this study, we attempted to mimic the proposed two-step reversible process. This included the simulation of the ligand-protein complex before and after the proton shuttling has taken place. In this attempt, to mimic the first step, the X-ray crystal structure of hCA VII having the (PDB ID-3MDZ)⁴⁵, solved at 2.32Đ resolution, was downloaded from the RCSB PDB database⁴⁶. The zinc-bound water molecule from hCA II (PDB ID-1AVN)¹⁸ was retained and placed into the hCA VII structure. The preparation of the protein structure was performed using the protein preparation wizard of Maestro Suite^{47,48}. This routine adds the hydrogen atoms and bond orders to produce suitable protonation states; also all the water molecules were deleted apart from the one that coordinates the zinc. Before docking, the co-crystal ligand was also separated from the 3MDZ protein. To mimic the second step (namely after the proton shuttling event), the same X-ray structure of hCA VII-3MDZ was utilised and the H₂O bound to Zn^{2+} converted to OH⁻. The most active compound 7 was built into its protonated and neutral form using the 2D sketcher of Maestro and the optimised geometry of the ligand was obtained using the "minimize ligand" option of the same Maestro suite. Via Maestro, the X-ray crystal structures of hCA I (PDB ID-6EVR)⁴⁹, hCA II (PDB ID-3K34)⁵⁰, were also retrieved and superimposed on the hCA VII coordinates. Both the protein and ligands were translated in the AD4 format (PDBQT) using the python scripts prepare_ligand4.py and prepare_receptor4.py, part of ADT, applying the standard settings. From the literature survey,

it is a well-known fact that the CA activators do not bind the zinc but bind at the entry site far away from the Zn^{2+} ion^{7,51}. For the aforementioned docking, the crystal structure of hCA II bound with the activator of histamine was retrieved and superimposed on the hCA VII and the grid was set at the centre of the histamine molecule. A set of grids of $50 \text{ Å} \times 50 \text{ Å} \times 50 \text{ Å}$ with 0.375 Å spacing was calculated considering the docking area for all the ligand atom types utilising the AutoGrid4 module. 100 independent dockings were accomplished. Each docking calculation included 20 million energy evaluations using the Lamarckian Genetic Algorithm Local Search (GALS) method. This method results in a population of feasible docking solutions and gives the best discrete generation of a binding pose. According to the technique of Solis and Wets, a low-frequency local search method was applied to every docking attempt to make sure that the representation of the final solution is a local minimum. In our docking protocol, 300 iterations of Solis and Wets local search were applied with a probability of 0.06 and a population size of 250. To generate new docking attempts for subsequent generations, a mutation rate of 0.02 and a crossover rate of 0.8 were used, and the best individual from each generation was propagated over the following generation. The docking results from each of the 100 independent docking calculations were clustered based on the root-meansquare deviation (rmsd) between the cartesian coordinates of the atoms (solutions varying by less than 2.0) and scored based on the calculated free energy of binding (ΔG_{AD4}). Using the calculated ΔG_{AD4} and docking cluster population size, the best binding poses of 7, both before and after the proton shuttling step, were chosen among the 100 independent conformations/configuration states resulting from docking experiments.

Biological assay

Human microglial cell line. All materials for cell culturing were obtained by Corning, New York, USA, and all used reagents were purchased by Sigma-Aldrich (Saint Louis, MO, USA). The human microglial cell line C20, originally generated by David Alvarez-Carbonell et al.⁵², was grown in DMEM-F12 medium supplemented with 10% of FBS, 100 U/mL penicillin, 0.1 mg/mL streptomycin, and neomycin ($600 \mu g/mL$), as a selector of the immortalised telomerase-expressing cells. Cells were kept at $37 \,^{\circ}$ C under a humidified atmosphere with 95% O₂ and 5% CO₂. Viable cells (identified by counting trypan-blue-excluding cellular elements) were plated at a density of 100,000 cells each well in a 24-well plate. To evaluate the ability of target compounds to modulate the cellular viability and brain-derived neurotrophic factor (BDNF) release, C20 cells were challenged with CAAs.

Cell viability assay. C20 cells were seeded in 96-well microplates at a density of 5000 cells/well and maintained in complete culture media. The day after the seed, cells were treated with CAAs **7**, **8**, and **9** for 24 h (1 and 10 μ M). The cell viability was determined using the [3–(4,5-dimethylthiazol-2-yl)-5–(3-carboxymethoxyphenol)-2–(4-sulfophenyl)-2H-tetrazolium, inner salt] (MTS) assay according to the manufacturer's instructions (Promega, Milano, Italy). This tetrazolium dye can be reduced by the metabolic reducing agents NADH and NADPH to a water-soluble formazan salt. The amount of formazan produced is considered to be a marker of cell viability. The MTS reagent was added to treated C20 cells and the colorimetric conversion of formazane was quantified after 1 h by measuring the absorbance at 490 nm (EnSightTM multimode plate reader, equipped with Kaleido Data Acquisition and Analysis Software).

BDNF enzyme-linked immunosorbent assay. C20 cells were seeded in 24-well microplates at a density of 100,000 cells/well and maintained in complete culture media. The day after the seed, cells were challenged with compound 7 for 2 h in serumfree medium and then maintained in fresh serum-free medium for 22 h. At the end of the 24 h, the levels of BDNF were measured in the C20 conditioned medium using an enzyme-linked immunosorbent assay (ELISA) Kit (SEA011Mi 96 Tests, Cloude-Clone Corp., CCC, USA), accordingly to manufacture instructions. Briefly, the cell culture supernatants were centrifuged for 20 min at $1000 \times g$, and used in the ELISA assay. The microplate provided in the kit has been pre-coated with an antibody specific for BDNF. Standards or samples were added to the microplate wells and incubated for 1 h at 37 °C. Next, a biotin-conjugated antibody specific to BDNF was added to the well for 1 h. After washing, avidin conjugated to Horseradish Peroxidase was added to each microplate well and following incubation and washing, TMB substrate solution was added. The enzyme-substrate reaction was terminated by the addition of sulphuric acid solution and the colour change was measured spectrophotometrically at a wavelength of $450 \text{ nm} \pm 10 \text{ nm}$. The concentration of BDNF in the samples was then determined by comparing the O.D. of the samples to the standard curve. Levels of BDNF were normalised on the number of cells counted in each well by crystal violet staining, and BDNF levels were reported as pg/mL normalised per total cells.

Data analysis. Values are presented as mean \pm standard error of the mean (SEM) of 3 experiments. Either a Student's *t*-test for independent means or one way ANOVA following by Bonferroni's post-test were used to define statistical differences between absolute values, which were considered significant at p < 0.05.

Results and discussion

Chemistry

Target compounds 1 and 2 were prepared according to the experimental procedure outlined in Scheme 1. Acylation of N-benzyl-5-methoxyindole 10 with oxalyl chloride, in anhydrous diethyl ether, at room temperature, yielded the corresponding indolylglyoxylyl chloride 11, which was allowed to react with the appropriate amine (N,N-dimethyl-1,3-propylenediamine for **12**, N,N-diethylethylenediamine for 13), in the presence of triethylamine, in dry toluene solution, at room temperature, to give compounds 12 and 13. Compounds 14 and 15 were obtained by demethylation of 12 and 13, using boron tribromide in anhydrous DCM at -10°C for 30 min and then at room temperature overnight. Following treatment with 2-bromoethylacetate and Cs₂CO₃ in acetone at reflux overnight, led to compounds 16 and 17, which were finally hydrolysed with 1 M NaOH solution in 1,4-dioxane at room temperature for 5 h, to achieve compounds 1 and 2.

The general synthetic pathway for the target compounds **3** and **4** is outlined in Scheme 2. Condensation of **11** with the appropriate amine (3-amino-1-propanol for **18**, ethanolamine for **19**) in 1,4-dioxane, at room temperature, yielded the amides **18** and **19**, which were then demethylated by treatment with boron tribromide in dry DCM. Subsequent reaction of **20** and **21** with the appropriate chloride (3-dimethylamino-1-propylchloride hydrochloride for **3**, 2-chloro-*N*,*N*-diethylethylamine hydrochloride for **4**) and Cs₂CO₃ in DMF for 5 h at 60 °C furnished the



1, 12, 14, 16: R₁ = -CH₂CH₂CH₂N(CH₃)₂ **2, 13, 15, 17:** R₁ = -CH₂CH₂N(CH₂CH₃)₂

Scheme 1. Reagents and conditions: (i) anhydrous Et₂O, oxalyl chloride, room temperature, 2h; (ii) anhydrous toluene, *N*,*N*-dimethyl-1,3-propylenediamine (for 12) or *N*,*N*-diethylethylenediamine (for 13), NEt₃, room temperature, overnight; iii) a) anhydrous DCM, BBr₃, -10 °C, 30 min.; (b) room temperature, overnight; (iv) acetone, Cs₂CO₃, 2-bromoethylacetate, reflux, overnight; (v) 1,4-dioxane, 1M NaOH, room temperature, 5h.



Scheme 2. Reagents and conditions: (i) 1,4-dioxane, 3-amino-1-propanol (for 18) or ethanolamine (for 19), room temperature, 3 h; (ii) (a) anhydrous DCM, BBr₃, $-10 \,^{\circ}$ C, 30 min.; (b) room temperature, overnight; (iii) anhydrous DMF, Cs₂CO₃, 3-dimethylamino-1-propylchloride hydrochloride (for 3) or 2-chloro-*N*,*N*-diethylethyl-amine hydrochloride (for 4), 60 $^{\circ}$ C, 5h.

target compounds **3** and **4**, finally purified by flash chromatography.

The synthesis of target compounds **5–7** was carried out following the procedure reported in Scheme 3. The ethylester derivatives **8**, **9**, **22**, prepared as outlined in Scheme 4, were hydrolysed with LiOH·H₂O in a solution of MeOH/H₂O (3:1), to obtain **23–25**, which were then treated with 2-(*1H*-benzotriazole-1-yl)-1,1,3,3-

tetramethyluronium tetrafluoroborate (TBTU) as condensing agent, *N*,*N*-diisopropylethylamine (DIPEA) and the appropriate amine (ethanolamine for **5** and **6**, benzylamine for **7**) in anhydrous DMF, at room temperature overnight, to yield compounds **5–7**, finally purified by flash chromatography.

Scheme 4 outlines the synthetic procedure for the obtainment of derivatives **8**, **9**, **22**. The ethyl ester derivative **27** was obtained



5, **8**, **23**: $R_1 = CH_2CH_2OH$; $R_2 = -CH_2CH_2CH_2N(CH_3)_2$ **6**, **9**, **24**: $R_1 = CH_2CH_2OH$; $R_2 = -CH_2CH_2N(CH_3)_2$ **7**, **22**, **25**: $R_1 = CH_2C_6H_5$; $R_2 = -CH_2CH_2N(CH_2CH_3)_2$

Scheme 3. Reagents and conditions: (i) MeOH/H₂O, LiOH H₂O, 80 °C, overnight; (ii) anhydrous DMF, TBTU, DIPEA, ethanolamine (for 5 and 6) or benzylamine (for 7), room temperature, overnight.



Scheme 4. Reagents and conditions: (i) abs. EtOH, conc. H_2SO_4 , 80 °C, 20 h; (ii) (a) NaH, dry DMF, 0 °C, 30 min.; (b) BzBr, room temperature, 5 h; (iii) (a) anhydrous DCM, BBr₃, -10 °C, 30 min.; (b) room temperature, overnight; (iv) anhydrous DMF, Cs₂CO₃, 3-dimethylamino-1-propylchloride hydrochloride (for 8) or 2-chloro-*N*,*N*-dimethylethylamine hydrochloride (for 9) or 2-chloro-*N*,*N*-diethylethylamine hydrochloride (for 22) or 60 °C, 5h.

by refluxing the commercially available 5-methoxyindole-3-carboxylic acid **26** in absolute ethanol for 20 h, in the presence of a catalytic amount of concentrated sulphuric acid. Benzylation of **27** by treatment with benzyl bromide in dry DMF in the presence of NaH, and subsequent demethylation with boron tribromide in anhydrous DCM yielded **29**, which was finally reacted with the appropriate chloride (3-dimethylamino-1-propylchloride hydrochloride for **8**, 2-chloro-*N*,*N*-dimethylethylamine hydrochloride for **9**, 2-chloro-*N*,*N*-diethylethylamine hydrochloride for **22**) in the presence of Cs₂CO₃ in dry DMF at 60 °C for 5 h, yielding **8**, **9**, **22**, finally purified by flash chromatography.

CA activation assays

Compounds **1–9** were assayed for their ability to activate four catalytically active and physiologically relevant hCA isoforms expressed in the human brain, namely the cytosolic hCA I, II, and VII and the mitochondrial hCA VA.

The cytosolic and ubiquitous CA I is expressed in the motor neurons in the human spinal cord⁵³. The physiologically dominant

isoform CA II is located both in the choroid plexus and in oligodendrocytes, myelinated tracts, astrocytes, and myelin sheaths in the vertebrates' brain⁵⁴. Immunocytochemical experiments demonstrated that astrocytes and neurons express the mitochondrial CA VA suggesting that this isozyme has a cell-specific, physiological role in the nervous system⁵⁵. CA VII showed comparable expression in the cortex, hippocampus, and thalamus and might be considered a brain-associated CA being absent in the majority of other tissues⁵⁶.

CA activation data of the indole-based derivatives **1–9** are listed in Table 1. The following structure-activity relationships (SARs) for the activation of these enzymes can be drawn:

- I. the cytosolic isoforms hCA I and II were not significantly activated by these derivatives up to concentrations as high as 100 μ M, with the exception of compounds **2** and **8** which showed K_A values of 88.9 and 69.1 μ M for hCA I, respectively.
- II. The mitochondrial isoform hCA VA was moderately activated by all derivatives with activation constants ranging between

Table 1. CA activation of isoforms hCA I, II, VA, and VII with compounds 1-9, by a stopped-flow CO2 hydrase assay.



 $K_A (\mu M)^a$

Cpd	R1	Ba	hca i Ihca va/hca viii ^d	hCAII [hCA II/hCA VII] ^b	hCA VA ÍhCA VA/hCA VIII ^b	hCA VII
1			> 100 [> 0 2]	> 100 [> 0 2]	25.0.[2.2]	10.9
1	(CH ₂) ₃ N(CH ₃) ₂	(CH ₂) ₂ OH	>100 [>9.5]	>100 [>9.5]	25.0 [2.5]	10.0
2	$(CH_2)_2N(CH_2CH_3)_2$	(CH ₂) ₂ OH	88.9 [8.9]	>100 [>10.0]	28.7 [2.9]	10.0
3	(CH ₂) ₃ OH	$(CH_2)_3N(CH_3)_2$	>100 [>9.4]	>100 [>9.4]	50.9 [4.8]	10.6
4	$(CH_2)_2OH$	$(CH_2)_2N(CH_2CH_3)_2$	>100 [>11.0]	>100 [>11.0]	34.4 [3.8]	9.1
5	$(CH_2)_2OH$	$(CH_2)_3N(CH_3)_2$	>100 [>9.6]	>100 [>9.6]	28.0 [2.7]	10.4
6	$(CH_2)_2OH$	$(CH_2)_2N(CH_3)_2$	>100 [>9.3]	>100 [>9.3]	26.8 [2.5]	10.7
7	$CH_2C_6H_5$	$(CH_2)_2N(CH_2CH_3)_2$	>100 [>13.3]	>100 [>13.3]	24.4 [3.2]	7.5
8	_	$(CH_2)_3N(CH_3)_2$	69.1[9.6]	>100 [>13.9]	59.8 [8.3]	7.2
9	_	$(CH_2)_2N(CH_3)_2$	>100 [>12.2]	>100 [>12.2]	29.1 [3.5]	8.2
Histamine			2.1	125	0.010	37.5

^aMean from 3 different assays, by a stopped flow technique (errors were in the range of $\pm 5-10\%$ of the reported values). ^bK_A ratio for the indicated enzyme isoforms.

24.4 and 59.8 μM ; the best activators were 1, 2, 4–7, 9, whereas 3 and 8 showed a poor activity.

III. The brain associated cytosolic CA isoform (hCA VII) was the most sensitive one to activation by all compounds, with K_A values in the range of 7.2–10.8 μ M, independently from the decoration pattern of the central indole nucleus, suggesting that this good activity might be ascribed to the presence of the protonatable moiety rather than to its position.

Overall, these results pointed out compounds **7**, **8**, and **9** as the most interesting ones in terms of K_A values (7.5, 7.2, and 8.2, μ M, respectively) and selectivity profile towards isoform VII, as they show selectivity ratios of >13.3–9.6 against hCAI, >13.9–>12.2 against hCAII, and 3.2–8.3 against hCA VA.

Molecular docking studies and PK parameters prediction

Molecular modelling studies were undertaken to suggest a theoretical model of the complex between the hCA VII and the newly discovered activators and provide a possible explanation for their modulating as well as selectivity properties. In particular, a sequence of in silico studies was executed on compound 7 that was found to be comparatively the most potent and selective hCA VII activator of the whole set. Docking calculations were performed using the AutoDockGPU software (see molecular modelling methods). Results of the docking studies (Figure 2(A)), attained to model the first step of chemical reaction (before the proton shuttling), revealed that the deprotonated diethylamine is lodged at a site far away from the Zn²⁺ ion and facing the His66 residue; this particular residue is said to be the most important residue in mediating the enzymatic activity of proton shuttle transfer reaction⁵⁷. As per the previous studies reported by De Simone et al.⁵⁸, Pinard et al.⁵⁹, and West et al.⁶⁰, the CA active site has an amphiphilic nature that has both a hydrophobic and a hydrophilic pocket next to the Zn²⁺ ion. The important proton shuttle mediating His66 residue falls into the hydrophilic

region^{59,60}. The ether oxygen at position 5 of the indole ring accepts an H-bond from the Asn64 side chain, while the indole moiety is involved in a π - π interaction with His66 residue. The Nbenzyl ring attached to the indole scaffold is lodged into a cleft that is lined by several non-polar amino acids such as Pro203, Pro204 and is also participating to a parallel displaced π - π interaction with the Tyr22 residue that is present at the entry cleft of the active site. Similarly, the 3-carboxamidobenzyl group is engaging in a π - π interaction with the Trp7 side chain. Since docking calculations were only attained on 7, we postulate that the other compounds reported here to activate hCA VII might establish the same interaction pattern also in light of the similar K_A experimental values (Table 1). Analysis of results attained by modelling the second step of the chemical reaction (i.e. after the proton shuttling) (Figure 2(B)) revealed that the protonated diethylamine and the 3-carboxamidobenzyl group is also lodged in the same protein region predicted to host the unprotonated ligand, forming a similar π - π interaction with Trp7 side chain; the 3-carboxamidobenzyl is also involved in the π - π interaction with His66, with the amide nitrogen donating an H-bond to the backbone CO of Asn64. In this position, the ligand protonated amine is also involved in a cation- π interaction with the His96. All in all, the position of the protonated and deprotonated amine of 7 might support the already postulated theory behind the activation mechanism through the proton shuttling process¹⁴.

To rationalise the selectivity profile of **7**, the binding poses (protonated and neutral ligand) calculated into the hCA VII binding site were rigidly translated into the binding sites of hCA I and hCA II (Figure 3(A,B)). In this particular analysis, both the binding poses of **7** would give rise to multiple steric clashes in hCA I and hCA II isoforms. Most precisely, the neutral ligand's diethylamine and the protonated ligand's *N*-benzyl ring would induce clashes within the hCA I binding site (Figure 3(A)), whereas the steric hindrance of the 3-carboxamidobenzyl group would clash with the hCA II active site. All in all, these data would indicate that the orientation adopted by **7** into the hCA VII active site, favouring



Figure 2. The best binding pose of 7 before (A) and after (B) the proton shuttling step in the hCA VII active site (PDB ID 3MDZ). The ligand is represented as orange (A) or yellow (B) sticks, while the protein structure is depicted in sea-green ribbons and sticks. Interacting residues are labelled. The Zn^{2+} ion is represented as a violet sphere. H-bonds are represented with the red dashed lines.



Figure 3. Binding poses calculated for compound 7 (orange and yellow) rigidly translated into the hCA I structure (PDB ID 6EVR, pink surface, A) and hCA II structure (PDB ID 3K34, green surface, B). All the pictures were rendered using UCSF Chimaera software⁶¹.

the proton shuttling step of the enzyme catalytic process, does not seem to be viable into the hCA I and II isoenzymes. Unfortunately, a similar analysis could not be attained for the hCA V due to the absence of the experimental atomic coordinates for this enzyme.

Given the possible implication of the CA activation for the treatment of CNS diseases, we were further asked if the most interesting compound **7** might also feature specific physicochemical parameters that could allow inferring a CNS activity. These studies were conducted in silico with the Qikprop program (Schrödinger. LLC New York (Table 2). From these studies, **7** should have a high potential of reaching the CNS. Indeed, the QPPMDCK parameter, which estimates the penetration of MDCK cells as a model of the blood-brain barrier (BBB), indicates that **7** should have high brain penetration properties. Also, this compound should be orally bioavailable considering its high calculated QPPCaco, which predicts the apparent Caco-2 permeability as a model of the gut-blood barrier.

Effect of indole-based CCAs on BDNF level secreted by human microglial cells

To explore the neuroprotective potential of the newly synthesised indole-based CAAs, their capacity to promote brain-derived neurotrophic factor (BDNF) release was preliminarily tested on the C20 human microglial cell line. Microglial cells protect physiological brain function by regulating neurogenesis, neuronal survival, synaptic plasticity, and responding to alterations of the extracellular environment⁶². In fact, in the presence of a potentially dangerous signal, microglia rapidly undertake morphological and functional alterations (microglia activation) that may lead to the secretion of specific factors, including BDNF. This latter is one of the key neurotrophic factors responsible for neuronal survival and differentiation, and is also directly involved in the control of synaptic plasticity as a neuromodulator; its levels vary in different regions of the CNS during several neurological disorders, such as viral encephalitis, traumatic injury, ischaemia, multiple sclerosis and Parkinson's disease^{63–66}. Interestingly, a link between BDNF beneficial activities and CA activation has been previously demonstrated in neurons^{61,67}, and correct acidification is a fundamental step during the release of BDNF from secretory granules⁶⁸; this literature suggests a possible and strong relation between BDNF and CAs during neuroprotection. In light of this, microglia, taking an active part in neuroprotection by providing neurons with important survival factors, could represent a model of choice to study CAAs.

In this study, compounds **7**, **8**, and **9**, those with the highest activation towards the brain-related hCA VII isoform, were first tested for their cytotoxicity on microglial cells. To this end, human C20 cells were treated with the target compounds for 24 h with concentrations in the micromolar range, in line with K_A values, and then cell viability was evaluated by MTS assay.

Compound **7** did not exhibit cytotoxic activity on C20 cells at any of the tested concentrations and even a little increase in cell viability was evidenced at $10 \,\mu$ M. On the contrary, compounds **8** and **9** caused a slight, although not statistically significant, reduction in cell viability at the highest tested concentration. These results prompted us to select compound **7** for further determination of its ability to promote the release of BDNF from human microglial cells. Human microglial cells were pre-treated for 2 h with compound **7**, then the medium was replaced and collected after 24 h for the evaluation of BDNF levels. Interestingly, in the medium derived from cells treated with compound **7** a significant increase in BDNF levels was evidenced, at amounts comparable to those reported in the literature for microglia (Figure 4)⁶⁹.

Table 2. Calculated	l physicochemical	and pharmacokinetic	properties of 7
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7	Range or recommended values ^a
11.000	0.0/15.0
455.599	130.0/725.0
6.972	1.0/12.5
1.000	0.0/6.0
5.250	2.0/20.0
6.455	-2.0/6.5
-6.598	-6.5/0.5
1209.191	<25 Poor, $>$ 500 great
-0.076	-3.0/1.2
672.043	<25 Poor, $>$ 500 great
1.261	-1.5/1.5
100.000	(<25% is Poor)
	7 11.000 455.599 6.972 1.000 5.250 6.455 -6.598 1209.191 -0.076 672.043 1.261 100.000

^aFor 95% of known drugs. ^bNumber of non-trivial (not CX3), non-hindered (not alkene, amide, small ring) rotatable bonds. ^cMolecular weight of the molecule. ^fComputed dipole moment of the molecule. ^dComputed dipole moment of the molecule. ^eEstimated number of hydrogen bonds that would be donated by the solute to water molecules in an sueous solution. ^fEstimated number of hydrogen bonds that would be accepted by the solute from water molecules in an aqueous solution. ^gPredicted octanol/water partition coefficient. ^hPredicted aqueous solubility, log S. S in mol dm⁻³ is the concentration of the solute in a saturated solution that is in equilibrium with the crystalline solid. ⁱPredicted apparent MDCK cell permeability in nm/sec. MDCK cells are considered to be a good mimic for the blood-brain barrier. ^lPrediction of binding to human serum albumin. ^mPredicted qualitative human oral absorption.

These preliminary data would indicate that the neuroprotective potential of **7** could properly rely on the increased availability of extracellular BDNF, which in turn could reinforce the protective role on neurons, exerted by resting microglia towards potentially damaging events.

Conclusions

In the present study three small series of indole-based derivatives were investigated for their activation profile on physiologically relevant human CA I, II, VA, and VII isoforms. Several compounds resulted as effective micromolar activators, with promising selectivity profiles towards the brain-associated cytosolic isoform hCA VII. Docking calculations provided a theoretical model explaining the CA activation profile of compound 7, selected as the most potent and selective hCA VII activator of the whole set. Furthermore, in silico studies indicate that compound 7 possesses the physicochemical parameters suitable to reach the CNS and to be orally available. Finally, a preliminary biological evaluation showed its ability to increase the release of the brain-derived neurotrophic factor (BDNF) in microglial cells, combined with a lack of cytotoxicity. All in all these data highlighted 7 as a promising lead compound for the development of novel agents with potential application in the treatment of CNS-related diseases.



Figure 4. Treatment with compound **7** increases the level of BDNF released from human microglial cells. (A) C20 cells were exposed for 24 h at different concentrations of compounds **7**, **8** and **9**. Then the medium was replaced, and an MTS assay was performed to test cell viability. Results are reported as the percentage of the control group, treated with vehicle (DMSO). Data are represented as means \pm SEMs of three independent experiments. The significance of the differences was determined by one-way ANOVA, which was followed by Bonferroni's post-test: ** p < 0.01 vs. CTRL. B) The supernatant derived from C20 cells treated with 10 μ M of compound **7** was collected and used for BDNF quantification by ELISA assay. Levels of BDNF (pg/ml) were then normalised on crystal violet absorbance. Data are represented as means \pm SEMs of three independent experiments and the significance of the differences was determined by Student's *t*-test analysis.

Author contributions

The manuscript was written through the contributions of all authors. All authors have given approval to the final version of the manuscript.

Disclosure statement

No potential conflict of interest was reported by the author(s).

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References

- 1. Supuran CT. Structure and function of carbonic anhydrases. Biochem J 2016;473:2023–32.
- 2. Chesler M. Regulation and modulation of pH in the brain. Physiol Rev Am Physiol Soc 2003;83:1183–221.
- Ruusuvuori E, Kaila K. Carbonic anhydrases and brain pH in the control of neuronal excitability. Subcell Biochem 2014; 75:271–1290.
- Farrant M, Nusser Z. Variations on an inhibitory theme: Phasic and tonic activation of GABA A receptors. Nat Rev Neurosci 2005;6:215–29.
- Makani S, Chen HY, Esquenazi S, et al. NMDA receptordependent afterdepolarizations are curtailed by carbonic anhydrase 14: Regulation of a short-term postsynaptic potentiation. J Neurosci 2012;32:16754–62.
- Supuran CT. Carbonic anhydrases: novel therapeutic applications for inhibitors and activators. Nat Rev Drug Discov 2008;7:168–81.
- Supuran CT. Carbonic anhydrase activators. Future Med Chem 2018;10:561–73.
- 8. Provensi G, Carta F, Nocentini A, et al. A new kid on the block? Carbonic anhydrases as possible new targets in alz-heimer's disease. Int J Mol Sci 2019;20:4724.

- 9. Svichar N, Waheed A, Sly WS, et al. Carbonic anhydrases CA4 and CA14 both enhance AE3-mediated Cl-HCO3exchange in hippocampal neurons. J Neurosci 2009;29: 3252–8.
- 10. Yang MT, Chien WL, Lu DH, et al. Acetazolamide impairs fear memory consolidation in rodents. Neuropharmacology 2013;67:412–8.
- 11. Sun M-K, Alkon DL. Pharmacological enhancement of synaptic efficacy, spatial learning, and memory through carbonic anhydrase activation in rats. J Pharmacol Exp Ther 2001;297: 961–7.
- Supuran CT. How many carbonic anhydrase inhibition mechanisms exist? J Enzyme Inhib Med Chem 2016;31:345–60.
- Temperini C, Scozzafava A, Vullo D, Supuran CT. Carbonic anhydrase activators. Activation of isoforms I, II, IV, VA, VII, and XIV with L- and D-phenylalanine and crystallographic analysis of their adducts with isozyme II: stereospecific recognition within the active site of an enzyme and its consequences for the drug design . J Med Chem 2006;49: 3019–27.
- Temperini C, Scozzafava A, Vullo D, Supuran CT. Carbonic anhydrase activators. Activation of isozymes I, II, IV, VA, VII, and XIV with L- and D-histidine and crystallographic analysis of their adducts with isoform II: engineering proton-transfer processes within the active site of an enzyme. Chemistry 2006;12:7057–66.
- Temperini C, Innocenti A, Scozzafava A, Supuran CT. Carbonic anhydrase activators: Kinetic and X-ray crystallographic study for the interaction of d- and I-tryptophan with the mammalian isoforms I-XIV. Bioorg Med Chem 2008;16: 8373–8.
- 16. Elder I, Tu C, Ming LJ, et al. Proton transfer from exogenous donors in catalysis by human carbonic anhydrase II. Arch Biochem Biophys 2005;437:106–14.
- 17. Maupin CM, Castillo N, Taraphder S, et al. Chemical rescue of enzymes: proton transfer in mutants of human carbonic anhydrase II. J Am Chem Soc 2011;133:6223–34.
- Briganti F, Mangani S, Orioli P, et al. Carbonic anhydrase activators: X-ray crystallographic and spectroscopic investigations for the interaction of isozymes I and II with histamine. Biochemistry 1997;36:10384–92.
- Supuran CT, Scozzafava A. Carbonic anhydrase activators: synthesis of high affinity isozymes I, II and IV activators, derivatives of 4-(arylsulfonylureido-amino acyl)ethyl-1H-imidazole. J Enzyme Inhib 2000;15:471–86.
- Scozzafava A, lorga B, Supuran CT. Carbonic anhydrase activators: Synthesis of high affinity isozymes I, II and IV activators, derivatives of 4-(4-tosylureido-amino acyl)ethyl-1H-imidazole (histamine derivatives). J Enzyme Inhib 2000;15: 139–61.
- Saada MC, Vullo D, Montero JL, et al. Carbonic anhydrase i and II activation with mono- and dihalogenated histamine derivatives. Bioorg Med Chem Lett 2011;21:4884–7.
- Saada MC, Vullo D, Montero JL, et al. Mono- and dihalogenated histamine, histidine and carnosine derivatives are potent carbonic anhydrase I, II, VII, XII and XIV activators. Bioorg Med Chem 2014;22:4752–8.
- 23. Draghici B, Vullo D, Akocak S, et al. Ethylene bis-imidazoles are highly potent and selective activators for isozymes VA and VII of carbonic anhydrase, with a potential nootropic effect. Chem Commun 2014;50:5980–3.
- 24. Temperini C, Innocenti A, Scozzafava A, et al. Carbonic anhydrase activators: I-Adrenaline plugs the active site entrance

of isozyme II, activating better isoforms I, IV, VA, VII, and XIV. Bioorg Med Chem Lett 2007;17:628–35.

- 25. Temperini C, Scozzafava A, Puccetti L, Supuran CT. Carbonic anhydrase activators: X-ray crystal structure of the adduct of human isozyme II with I-histidine as a platform for the design of stronger activators. Bioorg Med Chem Lett 2005; 15:5136–41.
- 26. Temperini C, Scozzafava A, Supuran CT. Carbonic anhydrase activators: The first X-ray crystallographic study of an adduct of isoform I. Bioorg Med Chem Lett 2006;16:5152–6.
- 27. Casini A, Caccia S, Scozzafava A, Supuran CT. Carbonic anhydrase activators. The selective serotonin reuptake inhibitors fluoxetine, sertraline and citalopram are strong activators of isozymes I and II. Bioorganic Med Chem Lett 2003;13: 2765–8.
- Abdülkadir Coban T, Beydemir Ş, Gücin I, et al. Sildenafil is a strong activator of mammalian carbonic anhydrase isoforms I-XIV. Bioorg Med Chem 2009;17:5791–5.
- 29. Welsch ME, Snyder SA, Stockwell BR. Privileged scaffolds for library design and drug discovery. Curr Opin Chem Biol 2010;14:347–61.
- 30. Costa B, Da Pozzo E, Chelli B, et al. Anxiolytic properties of a 2-phenylindolglyoxylamide TSPO ligand: stimulation of in vitro neurosteroid production affecting GABAA receptor activity. Psychoneuroendocrinology 2011;36:463–72.
- 31. Taliani S, Cosimelli B, Da Settimo F, et al. Identification of anxiolytic/nonsedative agents among indol-3-ylglyoxylamides acting as functionally selective agonists at the gamma-aminobutyric acid-A (GABAA) alpha2 benzodi-azepine receptor. J Med Chem 2009;52:3723–34.
- 32. Da Pozzo E, Giacomelli C, Costa B, et al. TSPO PIGA ligands promote neurosteroidogenesis and human astrocyte wellbeing. Int J Mol Sci 2016;17:1028.
- 33. Bruno A, Barresi E, Simola N, et al. Unbinding of translocator protein 18 kDa (TSPO) ligands: from in vitro residence time to in vivo efficacy via in silico simulations. ACS Chem Neurosci 2019;10:3805–14.
- Santoro A, Mattace Raso G, Taliani S, et al. TSPO-ligands prevent oxidative damage and inflammatory response in C6 glioma cells by neurosteroid synthesis. Eur J Pharm Sci 2016; 88:124–31.
- 35. Taliani S, Da Settimo F, Martini C, et al. Exploiting the indole scaffold to design compounds binding to different pharma-cological targets. Molecules 2020;25:2331.
- 36. Robello M, Barresi E, Baglini E, et al. The alpha keto amide moiety as a privileged motif in medicinal chemistry: current insights and emerging opportunities. J Med Chem 2021;64: 3508–45.
- 37. Xu XH, Liu GK, Azuma A, et al. Synthesis of indole and biindolyl triflones: Trifluoromethanesulfonylation of indoles with Tf2O/TTBP (2,4,6-tri-tert-butylpyridine) system. Org Lett 2011;13:4854–7.
- Khalifah RG. The carbon dioxide hydration activity of carbonic anhydrase. I. Stop-flow kinetic studies on the native human isoenzymes B and C. J Biol Chem 1971;246:2561–73.
- 39. Pastorekova S, Vullo D, Nishimori I, et al. Carbonic anhydrase activators: Activation of the human tumor-associated isozymes IX and XII with amino acids and amines. Bioorg Med Chem 2008;16:3530–6.
- 40. Vullo D, Nishimori I, Innocenti A, et al. Carbonic anhydrase activators: An activation study of the human mitochondrial isoforms VA and VB with amino acids and amines. Bioorg Med Chem Lett 2007;17:1336–40.

- 41. Supuran C, Barboiu M, Luca C, et al. Carbonic anhydrase activators. Part 14. Syntheses of mono and bis pyridinium salt derivatives of 2-amino-5-(2-aminoethyl)-and 2-amino-5-(3-aminopropyl)-1, 3, 4- thiadiazole and their interaction with isozyme II. Eur J Med Chem 1996;31:597–606.
- 42. Provensi G, Nocentini A, Passani MB, et al. Activation of carbonic anhydrase isoforms involved in modulation of emotional memory and cognitive disorders with histamine agonists, antagonists and derivatives. J Enzyme Inhib Med Chem 2021;36:719–26.
- 43. Santos-Martins D, Solis-Vasquez L, Tillack AF, et al. Accelerating AutoDock4 with GPUs and gradient-based local search. J Chem Theory Comput 2021;17:1060–73.
- 44. Morris GM, Ruth H, Lindstrom W, et al. Software news and updates AutoDock4 and AutoDockTools4: automated docking with selective receptor flexibility. J Comput Chem 2009; 30:2785–91.
- 45. RCSB PDB 3MDZ: crystal structure of human carbonic anhydrase VII [isoform 1], CA7.
- 46. Berman HM, Westbrook J, Feng Z, et al. The protein data bank. Nucleic acids res. Oxford : Oxford University Press; 2000. p. 235–242.
- Schrödinger Release 2020-2: Protein Preparation Wizard; Epik, Schrödinger, LLC, New York, NY, 2020; Impact, Schrödinger, LLC, New York, NY; Prime, Schrödinger, LLC, New York, NY, 2020.
- Sastry GM, Adzhigirey M, Day T, et al. Protein and ligand preparation: Parameters, protocols, and influence on virtual screening enrichments. J Comput Aid Mol Des 2013;27:221–34.
- Behnke CA, Le Trong I, Godden JW, et al. Atomic resolution studies of carbonic anhydrase II. Acta Crystallogr D Biol Crystallogr 2010;66:616–27.
- 50. Chiaramonte N, Bua S, Ferraroni M, Nocentini A, et al. 2-Benzylpiperazine: a new scaffold for potent human carbonic anhydrase inhibitors. Synthesis, enzyme inhibition, enantioselectivity, computational and crystallographic studies and *in vivo* activity for a new class of intraocular pressure lowering agents. Eur J Med Chem 2018;151:363–75.
- 51. Angeli A, Chiaramonte N, Manetti D, et al. Investigation of piperazines as human carbonic anhydrase I, II, IV and VII activators. J Enzyme Inhib Med Chem 2018;33:303–8.
- 52. Garcia-Mesa Y, Jay TR, Checkley MA, et al. Immortalization of primary microglia: a new platform to study HIV regulation in the central nervous system. J Neurovirol 2017;23:47–66.
- 53. Liu X, Lu D, Bowser R, Liu J. Expression of carbonic anhydrase I in motor neurons and alterations in ALS. Int J Mol Sci 2016;17:1820.
- 54. Halmi P, Parkkila S, Honkaniemi J. Expression of carbonic anhydrases II, IV, VII, VIII and XII in rat brain after kainic acid induced status epilepticus. Neurochem Int 2006;48:24–30.
- 55. Ghandour MS, Parkkila AK, Parkkila S, et al. Mitochondrial carbonic anhydrase in the nervous system: expression in neuronal and glial cells. J Neurochem 2000;75:2212–20.
- 56. Ruusuvuori E, Huebner AK, Kirilkin I, et al. Neuronal carbonic anhydrase VII provides GABAergic excitatory drive to exacerbate febrile seizures. Embo J 2013;32:2275–86.
- 57. Mollica A, Macedonio G, Stefanucci A, et al. Five- and sixmembered nitrogen-containing compounds as selective carbonic anhydrase activators. Molecules 2017;22:2178.
- 58. De Simone G, Alterio V, Supuran CT. Exploiting the hydrophobic and hydrophilic binding sites for designing carbonic

anhydrase inhibitors. Expert Opin Drug Discov 2013;8: 793–810.

- 59. Pinard MA, Mahon B, McKenna R. Probing the surface of human carbonic anhydrase for clues towards the design of isoform specific inhibitors. Biomed Res Int 2015;2015: 453543.
- 60. West D, Kim CU, Tu C, et al. Structural and kinetic effects on changes in the CO(2) binding pocket of human carbonic anhydrase II. Biochemistry 2012;51:9156–63.
- 61. Pettersen EF, Goddard TD, Huang CC, Couch GS, et al. UCSF Chimera-a visualization system for exploratory research and analysis. J Comput Chem 2004;25:1605–12.
- 62. Parkhurst CN, Yang G, Ninan I, et al. Microglia promote learning-dependent synapse formation through brainderived neurotrophic factor. Cell 2013;155:1596–609.
- 63. Wu SY, Pan BS, Tsai SF, et al. BDNF reverses aging-related microglial activation. J Neuroinflammation 2020;17:210–8.
- 64. Fang W, Liao W, Zheng Y, et al. Neurotropin reduces memory impairment and neuroinflammation via BDNF/NF-KB in a transgenic mouse model of Alzheimer's disease. Am J Transl Res 2019;11:1541–54.

- 65. Onodera J, Nagata H, Nakashima A, et al. Neuronal brainderived neurotrophic factor manipulates microglial dynamics. Glia 2021;69:890–904.
- 66. Ferrini F, De Koninck Y. Microglia control neuronal network excitability via BDNF signalling. Neural Plast 2013;2013: 429815.
- Matthews TA, Abel A, Demme C, et al. Expression of the CHOP-inducible carbonic anhydrase CAVI-b is required for BDNF-mediated protection from hypoxia. Brain Res 2014; 1543:28–37.
- Mizui T, Ishikawa Y, Kumanogoh H, et al. BDNF pro-peptide actions facilitate hippocampal LTD and are altered by the common BDNF polymorphism Val66Met. Proc Natl Acad Sci USA 2015;112:E3067–E3074.
- 69. Gomes C, Ferreira R, George J, et al. Activation of microglial cells triggers a release of brain-derived neurotrophic factor (BDNF) inducing their proliferation in an adenosine A2A receptor-dependent manner: A2A receptor blockade prevents BDNF release and proliferation of microglia. J Neuroinflammation 2013;10:16–3.