Supplementary Information:

Experimental evidence of mosaic structure in strongly supercooled molecular liquids

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Supplementary Figure 1. Dielectric loss spectrum of 1-propanol at T=120 K. Red circles: experimental data; black solid line: full model curve fitted to the spectrum; blue dash-dotted line: Debye peak; yellow dash-dotted line: structural relaxation; green dash-dotted line: Johari-Goldstein relaxation.

The measured permittivity function $\epsilon(\nu)$ was analyzed fitting simultaneously its real $\epsilon'(\nu)$ and imaginary $\epsilon''(\nu)$ part, where ν is the frequency. The function used for the fits is reported in the Methods. An example of a dielectric spectroscopy loss spectrum of 1-propanol at T = 120 K is reported in Supplementary Fig. 1 along with the curve obtained from the fitting procedure (black solid line) and the individual contributions of the Debye (blue dashdotted line), α (yellow dash-dotted line) and $\beta_{\rm JG}$ (green dash-dotted line) relaxations.

SUPPLEMENTERY NOTE 2. COMPARISON BETWEEN NEUTRON SCATTERING AND TDI DATA.

Supplementary Fig. 2 reports the τ values extracted from the TDI beating patterns measured at q = 15 and 25 nm^{-1} as in Fig. 2 along with the τ values measured by coherent quasi-elastic neutron scattering (QENS) at $q = 14 \text{ nm}^{-1}$ [1] but at higher temperatures. The two data sets are clearly consistent.



Supplementary Figure 2. Comparison between QENS and TDI relaxation time data. Blue circles and red squares with ± 1 SD errorbars: TDI data, as in Fig. 2. Blue diamonds with ± 1 SD errorbars: τ measured at $q_{\text{max}} = 14 \text{ nm}^{-1}$ by QENS [1]. Dotted-line: scaled shear viscosity data, as reported in [1].

SUPPLEMENTERY NOTE 3. MEAN SQUARED DISPLACEMENT.

In this section the model used to quantitatively relate $q_{\rm DS}$ to the molecular mean-squared displacement, $\Delta r_{\rm JG}$, is discussed, where $q_{\rm DS}$ is the *q*-value at which the $\beta_{\rm JG}$ relaxation characteristic time appearing in the density correlation function matches the one from DS. The connection between $q_{\rm DS}$ and $\Delta r_{\rm JG}$ follows from an argument similar to the one at the basis of the anomalous diffusion model introduced in Ref. [2, 3] to describe sub-diffusion in polymers. The experimental observations for the $\beta_{\rm JG}$ -process reported in the main text, as discussed in the following, are in fact up to a certain extent similar to those reported in [2], given the sub-diffusivity of the $\beta_{\rm JG}$ -process. However, differently from what reported in [3], this comparison will be discussed here only in terms of the characteristic relaxation time probed by TDI and DS without any implication for the underlying distributions of relaxation times.

As a first point it can be noticed that density fluctuations within the β_{JG} -relaxation, at least in the (reduced) time-window accessible to scattering techniques based on X-rays and neutrons, are well described by the KWW model [4–8]:

$$\phi(t) = f_q^{\rm JG} \exp\left[-\left(\frac{t}{\tau}\right)^{\beta_{\rm KWW}}\right] \,. \tag{1}$$

It is also important to point out that in a previous work on 5M2H [4] the KWW expression was found to be able to describe the β_{JG} relaxation, within experimental accuracy, even in a T and q range where the β_{JG} -process i) dominates density fluctuations and ii) completely relaxes them within the time window accessed by TDI. The super-quadratic q-dependence of the β_{JG} -relaxation characteristic time is a further aspect to consider:

$$\tau = Aq^{-n} = \tilde{D}^{-\frac{n}{2}}q^{-n} \tag{2}$$

In Supplementary Eq. 2, \tilde{D} can be regarded as a generalised diffusion coefficient with dimensions $[\tilde{D}] = [L]^2 [t]^{-\frac{2}{n}}$ [3, 9]. If Supplementary Eq. 2 is combined with Supplementary Eq. 1, the intermediate scattering function can be written as:

$$\phi(q,t) = f_q^{\rm JG} \exp\left[-\left(\tilde{D}^{\frac{n}{2}}q^n t\right)^{\beta_{\rm KWW}}\right] \approx f_q^{\rm JG} \exp\left[-\tilde{D}q^2 t^{\beta_{\rm KWW}}\right] \tag{3}$$

The last approximation in Supplementary Eq. 3 is justified by the fact that, close to $q_{\rm DS}$, $n \cdot \beta_{\rm KWW} \approx 2$ for all the samples considered here within one standard deviation and with

Sample	T [K]	$\beta_{\rm KWW}(q_{\rm DS})$	n
5M2H	165.5	0.51	4.2(7)
5M2H	170.4	0.51	4.5(7)
1-propanol	122.5	0.66	3.9(9)
OTP	240	0.6	3.8(7)
OTP	265	0.6	2.9(5)

Supplementary Table I. Stretching parameters and power-law exponents. Stretching parameter β_{KWW} used in the fitting procedure of the TDI beating patterns for $q > q_{\text{max}}$ (and therefore valid at q_{DS}) for 1-propanol (present work), 5-methyl-2-hexanol (5M2H) [4] and o-terphenyl (OTP) [6, 7]. The power-law exponents (*n*) are also reported as obtained from fitting the *q*-dependencies of the characteristic time τ_{p} reported in the present work, recalculated from the data reported in Ref.[4] and taken from Refs.[6, 7].

deviations smaller than 30% (see Supplementary Fig. 3 and Supplementary Tab. I). It is also useful to recall another important result from previous work on 5M2H [4]: around $q_{\rm DS}$ $(25 \text{ nm}^{-1} < q < 40 \text{ nm}^{-1})$ the strength of the $\beta_{\rm JG}$ -relaxation ($f_q^{\rm JG}$) shows no q-dependence. Unfortunately no similar information is available for 1-propanol and o-terphenyl (OTP), as well as no information is available for the entire q-range where 5M2H was investigated. It is nonetheless reasonable to assume that around $q_{\rm DS}$ the relaxation strength does not show a strong q-dependence. Therefore Supplementary Eq. 3 shows approximately a Gaussian shape in q and it is therefore possible to re-write it as [10]:

$$\phi(q,t) \approx \exp\left[-\frac{\langle r^2(t)\rangle q^2}{6}\right]$$
(4)

and interpret $\langle r^2(t) \rangle$ as the mean-squared displacement of the molecule (MSD):

$$\langle r^2(t) \rangle = \langle |\mathbf{r}(t) - \mathbf{r}(0)|^2 \rangle = 6\tilde{D}t^{2/n} \simeq 6\tilde{D}t^{\beta_{\rm KWW}}.$$
 (5)

It is important to remark that Supplementary Eq. 5 is consistent with the assumption that the incoherent approximation is valid. Namely, in the *q*-range here considered the contribution to the intermediate scattering function describing the $\beta_{\rm JG}$ -relaxation can be approximated with its self-part, and therefore collective molecular motions can be neglected: the $\beta_{\rm JG}$ -relaxation at times of the order of $\tau_{\beta_{\rm JG}}^{\rm DS}$ is essentially a single molecule process. This picture, valid at least in the explored *q*-range and before the onset of the α -relaxation, is



Supplementary Figure 3. Justification for the description of the intermediate scattering function in terms of a Gaussian. Product between the power law exponent (n) describing the q-dependence of the characteristic time of the $\beta_{\rm JG}$ relaxation and the stretching exponent $\beta_{\rm KWW}(q_{DS})$ for the three samples analysed in this work: 5-methyl-2-hexanol (5M2H), 1-propanol and o-terphenyl (OTP). The black dashed line indicates the value $n \cdot \beta_{\rm KWW} = 2$. The values of nand $\beta_{\rm KWW}(q_{DS})$ for 5M2H have been recalculated from the data reported in Ref. [4], and those for OTP are taken from Refs. [6, 7]. The plotted errorbars correspond to ± 1 SD.

consistent with two other observations: 1) the q-dependence of $\tau_{\beta_{\rm JG}}^{\rm DS}$ indicates a strongly restricted character of the process and 2) no de Gennes narrowing, hint of strong molecular correlations, was observed in the q-dependence of $\tau_{\beta_{\rm JG}}^{\rm DS}$. From Supplementary Eq. 5 it is then possible to extract the MSD at $\tau_{\beta_{\rm JG}}^{\rm DS}$:

$$\langle r^2(\tau_{\beta_{\rm JG}}^{\rm DS})\rangle = \frac{6}{q_{\rm DS}^2}.$$
 (6)

We can finally use $\langle r^2(\tau_{\beta_{\rm JG}}^{\rm DS}) \rangle$ to evaluate the average center-of-mass molecular root meansquared displacement at the characteristic time of the $\beta_{\rm JG}$ -relaxation, $\tau_{\beta_{\rm JG}}^{\rm DS}$, as:

$$\Delta r_{\rm JG} = \sqrt{\frac{6}{q_{\rm DS}^2}}\,.\tag{7}$$

This expression gives a more solid basis to the previously used concept of typical length-scale for the $\beta_{\rm JG}$ process [4]. It is however important to stress that $\Delta r_{\rm JG}$ has to be considered as the most probable displacement of the molecules participating to the $\beta_{\rm JG}$ -relaxation at $\tau_{\beta_{\rm JG}}^{\rm DS}$. Since the relaxation strength of the $\beta_{\rm JG}$ -process is not known in the whole q-range, it is not currently possible to estimate the full distribution of displacements associated to this relaxation process.

SUPPLEMENTERY NOTE 4. O-TERPHENYL: COMPARISON OF TDI AND DS DATA.

The OTP relaxation map from the TDI measurements reported in [6–8] is plotted in Supplementary Fig. 4, along with the characteristic times for the β_{JG} -relaxation from DS [11] and adiabatic calorimetry (AC) [12] (see also the inset in Supplementary Fig. 4). As it can be observed in Supplementary Fig. 4, the β_{JG} -relaxation decouples from the α one around 273 K. Differently from the case of 1-propanol and 5M2H [4], the β_{JG} -relaxation was observed only at the intra-molecular length-scale [6–8]. It can be inferred that the activation energy of the β_{JG} -relaxation as probed by the TDI data is quite consistent with the one from the DS/AC measurements extrapolated above T_{G} .

From Supplementary Fig. 4 it also clearly appears that molecular re-orientations within the $\beta_{\rm JG}$ -process are faster than density fluctuations at $q = 18 \text{ nm}^{-1}$ and slightly slower than those at 29 nm⁻¹, in line with what observed in 1-propanol and 5M2H. Concerning the q-dependence of the $\beta_{\rm JG}$ -relaxation, a super-quadratic q-dependence was observed also in OTP [6, 7]: n = 2.9(5) at 265 K and n = 3.8(7) at 240 K. $q_{\rm DS}$ was then extracted from these q-dependent data in an analogous way as for 1-propanol and 5M2H (see main text). These $q_{\rm DS}$ data have then been used to extract the average center-of-mass displacement, $\Delta r_{\rm JG}$, at the characteristic time of the $\beta_{\rm JG}$ -relaxation, see Supplementary Eq.7. The mean of the values obtained at 244 and 265 K (see Fig.4) and normalized to the average intermolecular distance is $\left\langle \frac{\Delta r_{\rm JG}}{r_{\rm p}} \right\rangle = 0.125(17)$.

For the specific case of OTP, mean-squared displacement data are available for the crystalline phase at few temperatures [13]. It is then possible to extrapolate those data at the melting temperature, T_m =329 K: the obtained value is 0.604(9)Å², whose square root corresponds to 0.113(2) r_p^c , where r_p^c is the average intermolecular distance estimated from the density of the crystal close to the melting temperature [14]. In the spirit of the Lindemann criterion, this could be considered as the limiting root mean-squared displacement for structural instability in OTP. It jumps to the eyes that this value coincides, within one SD, with the value reported above for $\left\langle \frac{\Delta r_{\rm JG}}{r_{\rm p}} \right\rangle$. This result, which could be verified here only for the case of OTP and



Supplementary Figure 4. Relaxation map of o-terphenyl. Inverse-temperature dependence of the relaxation time measured by TDI at different q-values: 14 nm⁻¹ [6] (grey circles), 18 nm⁻¹ [8] (green circles), 23 nm⁻¹ [6] (blue diamonDS) and 29 nm⁻¹ [8] (yellow right-pointing triangles). The solid lines of the corresponding color show the best fitting curves using the *T*-dependence for the α and $\beta_{\rm JG}$ -relaxation as reported in [6, 8]. The characteristic time of the $\beta_{\rm JG}$ -relaxation as measured by dielectric spectroscopy (DS) [11] (red squares) and adiabatic calorimetry (AC) [12] (red left-pointing triangles) are also reported for the sake of comparison. The red dash-dotted line is the Arrhenius fit to the DS and AC data. Inset: extended inverse-*T* dependence of the DS and AC data. All errorbars correspond to ±1 SD.

that might be fortuitous at this stage, implies that the average mean-square displacement at $\tau_{\beta_{JG}}^{DS}$ for the molecules participating to the β_{JG} -relaxation matches exactly the limiting value for structural instability (in terms of CM-CM distance), a conclusion stricter than the one reported in the main text. If we look at the β_{JG} -relaxation as a distribution of elementary processes with different timescales, all the processes relaxing with a characteristic time longer than $\tau_{\beta_{JG}}^{DS}$ satisfy then the Lindemann criterion for structural instability. It is however

important to underline that the preceding discussion has to be taken with a grain of salt, as the Lindemann limiting mean-square displacement extrapolated at the melting point might differ from the one for the amorphous structure of the glass. It is in fact known that the Lindemann limiting mean-squared displacement does depend on the local structure even for crystalline materials [15], and might therefore differ as well between the glass and the corresponding crystal.

SUPPLEMENTERY NOTE 5. NEAREST NEIGHBORS CALCULATIONS.



Supplementary Figure 5. Center-of-mass pair distribution function. Calculated molecular center-of-mass radial pair distribution function for o-terphenyl [17] at 300 K (blue solid line). The yellow dashed-line enlightens the range corresponding to the first coordination shell.

The average coordination number z can be easily estimated for 1-propanol and OTP integrating their radial intermolecular pair distribution functions, g(r), over the first coordination shell [16]:

$$z = \int_0^{r_{min}} 4\pi \rho r^2 g(r) dr \,.$$
(8)

Here r_{min} marks the position of the first minimum in g(r), and ρ is the number density. In the case of OTP the radial distribution function relative to the molecular center-ofmass (g_{CM}) (see Supplementary Fig. 5) was used, taken from Ref. [17]. In the case of 1propanol, given its more complex structure, the partial g(r) corresponding to inter-molecular correlations between the central carbon atoms of each alkyl-chain was employed instead (D. Bowron, private communication). Both g(r) datasets correspond to room temperature, T = 300 K. The obtained z values are reported and discussed in the main text. These values are not expected to vary by more than 20% at the lower temperatures where the TDI and DS experiments were carried out, based on the corresponding changes in the density of the two samples (see Refs. [17] and [18]).



Supplementary Figure 6. Tg-scaled temperature dependence of the initial beating contrast of the TDI signal. The initial beating pattern f'_q is plotted as measured at two different exchanged wave-vectors: 15 (blue circles with ±1 SD errorbars) and 25 nm⁻¹ (red squares with ±1 SD errorbars).



A. *T*-dependence

Supplementary Figure 7. Wave-number dependence of the initial beating pattern contrast. The q-dependence of the initial beating pattern f'_q is plotted as measured at two different temperatures: T=131.4 (blue circles with ± 1 SD errorbars) and 122.5 K (red squares with ± 1 SD errorbars). The diffuse scattering pattern measured at T=122.5 K is rescaled and reported on the same axes for sake of comparison.

The initial beating contrast, f'_q , corresponding to the τ values reported in Fig. 2 are shown in Supplementary Fig. 6 along with the initial beating contrast values extracted at lower temperatures, where the dynamics is too slow to be probed by TDI. f'_q is related to the strength of the α -relaxation process above $T_{\alpha\beta} \simeq 131$ K whereas below this temperature it accounts for the relaxation strength of both the α and the $\beta_{\rm JG}$ -relaxation. In fact for $T < T_{\alpha\beta}$ the dynamics is too slow to observe a complete decorrelation in the time window accessed by TDI and the relative contributions of the $\beta_{\rm JG}$ and α -processes cannot be disentangled. It can be noticed that, consistently with what already observed in a previous TDI study of 5-methyl-2-hexanol [4], f'_q does not show any discontinuity at $T_{\alpha\beta}$, meaning that when the $\beta_{\rm JG}$ -process separates from the α -one the total relaxation strength is conserved. The *T*-dependence of f'_q shows instead a clear change in slope at *T*g.

B. *q*-dependence

The q-dependence of the initial beating contrast, f'_q , corresponding to the τ values reported in Fig. 3 is plotted in Supplementary Fig. 7. At both probed temperatures, T = 131.4 (blue circles with ± 1 SD errorbars) and 122.5 K (red squares with ± 1 SD errorbars), f'_q displays an oscillation in phase with the S(q) of 1-propanol. This oscillation can be associated to the well-known de Gennes narrowing effect [19–21] and signals the sensitivity to intermolecular correlations of the process(es) contributing to f'_q . As already discussed in the main text, the TDI data are sensitive to the faster relaxation process and to the total strength.

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